Special Hazards

FLAMMABLE.

Enmediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all ignition sources. Notify manufacturer or supplier. Dike to prevent runoff. Stop or reduce discharge if this can be done without risk. Notify environmental authorities.

Protective Clothing and Equipment

Eye protection - goggles or face shield. Gloves - rubber. Boots - rubber. Clothing - suitable for situation.

Fire and Explosion

Use dry chemical, alcohol foam or carbon dioxide to extinguish. Water may be ineffective on fire but may be used to cool fire-exposed containers.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: wash eyes and skin with water and remove contaminated clothing. Inhalation: If breathing has stopped, give artificial respiration; if laboured, give oxygen. Ingestion: give water to conscious victim to drink and induce vomiting. If medical attention is considered necessary, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- Contact manufacturer or supplier for advice.
 Notify environmental authorities to discuss
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Lend-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

ETHYLBENZENE C6H5CH2CH3

IDENTIFICATION

UN No. 1175

Common Synonyms	Observable Characteristics	Manufacturers
PHENYLETHANE EB ETHYLBENZOL	Clear, colourless liquid with an aromatic odour.	Polysar, Sarnia, Ontario. Dow Chemical Canada Inc., Sarnia, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Open (flame arrester) or pressure vacuum. Pump Type: Gear or centrifugal, explosion-proof, grounded.	Label(s): Red and white label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Viton, polypropylene, Teflon, neoprene.	Grades or Purity: Pure grade, 99.5%; technical grade, 99.0%. Containers and <u>Materials</u> : Drums, tank cars, tank trucks; steel, stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.014 g/100 mL (15°C); 0.021 g/100 mL (25°C). Molecular Weight: 106.2 Vapour Pressure: 7 mm Hg (20°C); 12 mm Hg (30°C). Boiling Point: 136.2°C.	Floatability (Water): Floats. Odour: Aromatic (140 ppm, odour threshold). Flash Point: 15°C (c.c.). Vapour Density: 3.7 Specific Gravity: 0.87 at 20°C.	Colour: Colouriess. Explosive Limits: 1.0 to 6.7%. Melting Point: -95°C.

HAZARD DATA

Human Health		
Symptoms: Inhalation: irritation of mucous membrirritation, defatting and dermatitis; eyes - irritation toxicology: Moderately toxic by inhalation, contact TLV* - (inhalation) 100 ppm; 435 mg/m ³ . Short-term inhalation Limits - 125 ppm; 545 mg/m (15 min).	on and burning. It and ingestion. LCso - No information.	Ingestion: symptoms similar to inhalation. Contact: skin - LD50 - Oral: rat = 3.5 g/kg
Fire		
Fire Extinguishing Agents: Use foam, dry chemica down vapours. Behaviour in Fire: Flashback may occur along vapo ignition Temperature: 432°C.		out may be used to cool fire-exposed containers and knock
Reactivity		
With Water: No reaction. With Common Materials: Can react with exidizing Stability: Stable.	materials.	
Environment	_	
Water: Prevent entry into water intakes and water 42.3 mg/L/48 h/fathead minnow/TLm/hard water; Land-Air: No information. Food Chain Concentration Potential: None.	ways. Fish toxicity: 29 ppm/96 h/bluegill/TLm 14 mg/L/96 h/rainbow trout/LC30/freshwater	/freshwater; 32 to 35.1 ppm/48 h/bluegilf/TLm/freshwater; ; BOD: 2.8% 5 days.

Special Hazards

FLAMMABLE.

Immediate Responses

Keep non-involved people away from spill site. Issue warnings: "FLAMMABLE". Call Fire Department. Eliminate all sources of ignition. Call manufacturer. Shut off leak if safe to do so. Dike to prevent runoff. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated protective clothing.

Fire and Explosion

Use foam, dry chemical or carbon dioxide to extinguish. Water may be ineffective, but may be used to cool fire-exposed containers and knock down vapours. Flashback may occur along vapour trail. Do not extinguish fire until leak stopped.

First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; oxygen if laboured. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with water. Ingestions do not induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notlfy environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- I; Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

ETHYL CHLORIDE C2H5C1

IDE	TIFIC	ATT	ON

UN No. 1037

Common Synonyms	Observable Characteristics	Manufacturers
CHLOROETHANE MONOCHLOROETHANE HYDROCHLORIC ETHER	Colourless liquid or gas. Ethereal odour.	Ethyl Corporation of Canada Ltd., Corunna, Ontario. Dow Chemical Canada Inc., Sarnia, Ontario Originating from: Dow Chemical, Freeport, TX.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Flammable gas. Inert Atmosphere: No requirement. Venting: Safety relief. Pump Type: Centrifugal, positive displacement or gear. Steel, stainless steel.	Label(s): Red label - FLAMMABLE GAS; Class 2.1. Storage Temperature: Ambient. Hose Type: Tellon, Viton A, flexible stainless steel.	Grades or Purity: Technical, 99.5 to 100%; USP, 100%. Containers and Materials: Cylinders, tank cars, tank trucks (pressure vessels). Steel, stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): Reacts to form HCl. 0.33 g/100 mL (0°C); 0.57 g/100 mL (20°C). Molecular Weight: 64.5 Vapour Pressure: 457 mm Hg (0°C); 700 mm Hg (10°C); 1 000 mm Hg (20°C); 1 444 mm Hg (30°C). Boiling Point: 12.3°C.	Floatability (Water): Floats and reacts. Odour: Ethereal. Flash Point: -43°C (o.c.); -50°C (c.c.). Vapour Density: 2.2 Specific Gravity: 0.89 (25°C).	Colour: Colouriess. Explosive Limits: 3.8 to 15.4%. Melting Point: -136 to -139°C.

HAZARD DATA						
Human Health						
Symptoms: Inhalation: dizziness, disorientation, incoordination, narcosis, nausea or vomiting. Ingestion: similar to inhalation. Contact: skin - pain, irritation, defatting, frostbite; may be absorbed; eyes - pain and irritation, frostbite burn. Toxicology: Moderately toxic by inhalation, ingestion and contact. TLV* - (inhalation) 1 000 ppm; 2 600 mg/m³. LC50 - No information. Short-term inhalation Limits - 1 250 ppm; LC ₁₀ - inhalation: guinea pig = 4 000 ppm/ 45 min Delayed Toxicity - No information.						
Fire						
Water spray, fog, or foam may be necessary on k	rge before attempting to extinguish fire. Use carbon dioxide or dry chemical to extinguish small fires. arge fires. Use water to cool fire-exposed containers. astion from heat or fire produces toxic hydrochloric acid fumes and phosgene. Flashback may occur along Burning Rate: 3.8 mm/min.					
Reactivity						
With Water: Reacts with water and steam to produce hydrochloric acid. With Common Materials: Can react vigorously with oxidizing materials. Reacts violently with alkaline metals, magnesium, aluminum and zinc. Stability: Stable.						
Environment						
Water: Prevent entry into water intakes and wat- Land-Air: No information. Food Chain Concentration Potential: None.	Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating: >1 000 ppm/96 h/TLm/freshwater; BOD: No information. Land-Air: No information.					

Special Hazards

FLAMMABLE.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". Call Fire Department. Eliminate all sources of ignition. Call manufacturer for advice. Avoid contact and inhalation. If material is burning, use water spray to knock down vapours and fumes; work from upwind. If water spray is used, dike area to contain toxic runoff. Stop or reduce discharge if this can be done without risk. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Boots - high, rubber (pants worn outside boots). Gloves - rubber or plastic.

Fire and Explosion

Stop or reduce discharge before attempting to extinguish fire. Use carbon dioxide to extinguish. Use water to cool fire-exposed containers. Flashback may occur along vapour trail.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; give oxygen if breathing is laboured. Contact: remove contaminated clothing. Wash eyes and affected skin with plenty of warm water for at least 15 minutes. Do not rub areas which appear to have been frostbitten. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

Control - favourable conditions

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If floating, skim and remove.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Control - not possible

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

Control - favourable conditions

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Adsorb residual liquid on natural or
- synthetic sorbents.

 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Control - not possible

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

ETHYLENE C2H4

	IDENTIFICATION	UN No. 1962
Common Synonyms	Observable Characteristics	Marwiscturers
ETHENE OLEFIANT GAS	Colourless liquid or gas. Sweet odour.	Esso Chemicals Canada Ltd., Sarnia, Ontario. Guif Oil Canada Ltd., Varennes, Quebec. Alberta Gas, Joffre, Alberta. Petrosar Ltd., Corunna, Ontario. Dow chemical Canada Inc., Sarnia, Ont., Fort Saskatchewan, Alta.
Transportation and Storage Information		· · · · · · · · · · · · · · · · · · ·
Shipping State: Liquid (compressed gas). Classification: Flammable gas. Inert Atmosphere: No requirement. Venting: Safety relief.	Label(s): Red label - FLAMMABLE GAS; Class 2.1. Storage Temperature: Ambient (gas); -104°C (liquid). Hose Type: Teflon bore 304 stainless steel (300-1 400 psig).	Grades or Purity: Technical, 98 to 99%; CP 99.5 to 100%. Containers and Materials: Cylinders, tube trailers, tank trucks (under special permit), tank cars; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): 0.013 g/100 mL (20°C). Molecular Weight: 28.1 Vapour Pressure: 30 400 mm Hg (0°C). Bolling Point: -103.9°C.	Floatability (Water): Floats and boils. Odour: Sweet (260 to 700 ppm, odour threshold). Flash Point: -136°C approximately. Vapour Density: 0.98 (0°C) Specific Gravity: 0.57 (-104°C); 0.61 (0°C).	Colour: Colourless. Explosive Limits: 2.7 to 36%. Melting Point: -169.2°C.

HAZARD DATA

	4 by prot spour, \$5.4 \$ 4.5					
Human Health						
Symptoms: Inhalation: headache, drowsiness, dizzi pain, burning. Toxicology: Moderately toxic by inhalation. TLV* - (inhalation) Asphyxiant. Short-term Inhalation Limits - No information.	ness, weakness, narcosis, nausea and vomiting, aspl LC50 - Inhalation: mouse = 950 000 ppm. Delayed Toxicity - None known.	hyxia. <u>Contact:</u> skin - liquid can cause frostbite; eyes - LD50 - No information.				
Fire						
Fire Extinguishing Agents: Stop flow of gas before exposed containers. Behaviour in Fire: Flashback may occur along vapolimition Temperature: 450°C.		dry chemical or water fog. Use water to cool fire-				
Reactivity						
With Water: No reaction. With Common Materials: Can react vigorously with oxidizing materials. Reacts violently with aluminum chloride, bromotrichloromethane, carbon tetrachloride, chlorine, chlorine, chlorine dioxide, nitrogen dioxide and ozone. May react with acids. Stability: Stable.						
Environment						
Water: Prevent entry into water intakes and water TLm/freshwater; BOD: No information. Land-Air: Concentrations of 0.5 to 4 ppm cause le Pood Chain Concentration Potential: None.		hwater; Aquatic toxicity rating = 100 to 1 000 ppm/96 h/r inhibits flower opening.				

Special Hazards

FLAMMABLE.

Immediate Responses

Keep non-involved people away. Issue warning: "FLAMMABLE". Call Fire Department. Eliminate sources of ignition. Avoid contact and inhalation. Work from upwind and use water spray to control vapour. Step or reduce discharge if this can be done without risk. Notify manufacturer or supplier. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus. Gloves - rubber or plastic. Acid suit - jacket and pants; rubber or plastic. Boots - high, rubber foants worn outside boots).

Fire and Explosion

Stop flow of gas before attempting extinguishment. Use carbon dioxide, dry chemical or water fog to extinguish. Use water to cool fire-exposed containers. Flashback may occur along vapour trail.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; give oxygen if breathing is laboured. Contact: remove contaminated clothing and wash eyes and affected skin thoroughly with plenty of warm water. Do not rub affected areas. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water-Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

ETHYLENEDIAMINE NH2CH2CH2NH2

MP	NT	FIC.	A	TIO	ě.

UN No. 1604

Common Synonyms 1,2 DIAMINOETHANE 1,2 ETHANEDIAMINE	Observable Characteristics Colourless liquid. Ammonia-like odour.	Manufacturers No Canadian manufacturer. Canadian supplier: Dow Chemical Canada Inc., Sarnia, Ontario. Originating from: Dow Chemical, Midland, MI.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Corrosive liquid. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Centrifugal or positive displacement; stainless steel.	Label(s): Black and white label - CORROSIVE; Class 8, Group II. Storage Temperature: Ambient. Hose Type: Natural rubber, Viton, butyl, flexible stainless steel. No bronze, brass or copper alloy fittings.	Grades or Purity: Technical, 97+%. Containers and Materials: Drums (tin-lined), tank trucks; stainless steel, aluminum.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Completely soluble. Molecular Weight: 60.1 Vapour Pressure: 9 mm Hg (20°C); 16 mm Hg (30°C). Boiling Point: 115-119°C.	Floatability (Water): Floats and mixes. Odour: Ammonia-like odour (1.0 to 11.2 ppm, odour threshold). Flash Point: 50-66°C (o.c.); 40-43°C (c.c.). Vapour Density: 2.1 Specific Gravity: 0.90 (20°C).	Colour: Colourless. Explosive Limits: 4.2 to 14.4% (2.5 to 12% at 100°C). Melting Point: 11°C.

HAZARD DATA					
Human Health					
Symptoms: Inhalation: intense respiratory tract is digestive tract, nausea and vomiting. Toxicology: Highly toxic by inhalation and ingestic TLV*- (inhalation) 10 ppm; 25 mg/m ³ . Short-term inhalation Limits - No information.		. <u>Contact</u> : burns to skin and eyes. <u>Ingestion</u> : irritation to LD50 - Oral: guinea pig = 0.47 g/kg Oral: rat = 1.16 g/kg			
like					
Fire Extinguishing Agents: Use carbon dioxide, dry containers and knock down vapours. Behaviour in Fire: Irritating NO _X gases may be ge ignition Temperature: 380-400°C. Reactivity		neffective, but may be used to cool fire-exposed			
With Water: No reaction; soluble. With Common Materials: May react with oxidizing	g materials. Reacts violently with acetic acid, accidhorohydrin, hydrochloric acid, nitric acid, oleun	etic anhydride, acrolein, acrylic acid, acrylonitrile, allyl n, silver perchlorate, sulluric acid and vinyl acetate.			
Environment					
Water- Prevent entry into water intakes and water	rways. Harmful to aguatic life in very low concen	trations. Fish toxicity: 30 to 60 ppm/24 h/creek chub/			

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life in very low concentrations. Fish toxicity: 30 to 60 ppm/24 h/creek killed/freshwater; Aquatic toxicity rating = 10 to 100 ppm/96 h/TLm/freshwater; BOD: 75%, 5 days (theoretical).

Land-Air: No information.

Food Chain Concentration Potential: None.

Special Hazards

CORROSIVE, FLAMMABLE.

Immediate Responses

Keep non-involved people away from spill site. Issue warnings: "CORROSIVE; FLAMMABLE". Call Fire Department. Avoid contact and inhalation. If water is used to control fire, dike area to prevent runoff. Notify supplier. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated protection suit. Boots - high, rubber (pants worn outside boots). Gloves - rubber.

Fire and Explosion

Use dry chemical, alcohol foam or carbon dioxide to extinguish. Water may be used to cool fire-exposed containers and knock down vapours.

First Aid

Move victim out of splil area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; give oxygen if breathing is laboured. Contact: remove contaminated clothing. Wash eyes and skin with plenty of warm water for at least 30 minutes. Ingestion: give milk or water to conscious victim only. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- i. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

UN No. 1605 Manufacturers

IDENTIFICATION

ETHYLENE DIBROMIDE BrCH2CH2Br

Common Synonyms 1,2-DIBROMOETHANE ETHYLENE BROMIDE

Observable Characteristics

Colourless liquid. Sweet adour.

No Canadian manufacturers. Canadian suppliers:

Basile Import/Export, St. Laurent, Quebec. Chorney Chemical Co., Toronto, Ontario. Dow Chemical Canada Inc.

Originating from:

Dow Chemical, Magnolia, AR.

Transportation and Storage Information

Shipping State: Liquid.

Classification: Poisonous liquid. Inert Atmosphere: No requirement.

Venting: Pressure-vacuum.

Pump Type: Centrifugal or positive displacement,

steel or stainless steel. Physical and Chemical Characteristics

Physical State (20°C, I atm): Liquid.

Solubility (Water): 0.431 g/100 mL (30°C) Molecular Weight: 187.9

Vapour Pressure: 11 mm Hg (20°C); 17 mm Hg

(30°C)

Boiling Point: 131 to 132°C.

Label(s): White label - POISON: Class 6.1, Group II.

Storage Temperature: Ambient Hose type: stainless steel, Viton.

Odour: Sweet (26 ppm, odour threshold).

Floatability (Water): Sinks.

Flash Point: Not flammable.

Vapour Density: 6.5

Grades or Purity: Commercial.

Containers and Materials: Drums: steel. Tanks trucks, tank cars; steel (not aluminum).

Colour: Colourless.

Explosive Limits: Not flammable.

Melting Point: 9 to 10°C

HAZARD DATA

Human Health

Symptoms: Inhalation: shortness of breath, coughing, irritation of respiratory tract, nausea, vomiting, drowsiness, internal injury. Contact: skin - readily absorbed, inflammation, bilisters and symptoms similar to inhalation. Ingestions symptoms similar to inhalation.

Toxicology: Highly toxic by all routes. TLY* - No information; contact should not

LC50 - No information.

LD50 - Orali rat = 0.117 g/kg

be made with ethylene dibromide. Short-term Inhalation Limits - No information. LCLo - Inhalation: rat = 400 ppm/2 h Delayed Toxicity - Suspected carcinogen.

Fire

Fire Extinguishing Agents: Not combustible. in fires involving ethylene bromide, use water spray, dry chemical, foam or carbon dioxide. Water may also be used to disperse vapours.

Behaviour in Fire: When heated to decomposition, emits highly toxic fumes of bromine and hydrogen bromide.

Ignition Temperature: Not combustible.

Burning Rate: Not combustible.

Reactivity

With Water: No reaction.

With Common Materials: Reacts violently with metal powders of aluminum, zinc and magnesium. May react with strong oxidizers.

Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 18 mg/L/48 h/bluegilt/freshwater; BOD: No information. Land-Air: No information.

Food Chain Concentration Potential: None.

Special Hazards

POISON.

immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Avoid contact and inhalation. Notify manufacturer or supplier. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

Not combustible. In fires involving ethylene dibromide, use water spray, dry chemical, foam or carbon dioxide to extinguish. Water may also be used to knock down vapours. When heated to decomposition, emits highly toxic fumes of bromide.

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water for at least 15 minutes. Ingestion: give water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain spill by damming or water diversion.
- liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or
- other barrier.
- 4. Dredge or vacuum pump to remove contaminants, 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
 - 5. Recover undamaged containers.
 - 6. Adsorb residual liquid on natural or synthetic sorbents.
 - 7. Remove contaminated soil for disposal.
 - 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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ETHYLENE DICHLORIDE CH2C1-CH2C1

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UN No. 1184

Сотиноп Бупопуть	Observable Characteristics	Manufacturers
1,2-DICHLOROETHANE ETHYLENE CHLORIDE DUTCH OIL OR LIQUID	Clear, colourless liquid. Sweet, chloroform- like odour.	Dow Chemical Canada Inc., Sarnia, Ontario, Fort Saskatchewan, Alberta.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Standard types for flammable liquids (centrifugal); steel or stainless steel.	Label(s): Red label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Flexible stainless steel, Teflon.	Grades or Purity: Commercial. Containers and Materials: Drums, tank cars, trucks; steel, stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.92 g/100 mL (0°C); 0.869 g/100 mL (20°C). Molecular Weight: 99.0 Vapour Pressure: 40 mm Hg (10°C); 61 mm Hg (20°C); 105 mm Hg (30°C). Boiling Point: 83.5°C.	Ploatability (Water): Sinks. Odour: Chloroform-like (6 to 40 ppm, odour threshold). Flash Point: 18°C (o.c.); 13°C (c.c.). Vapour Density: 3.4 Specific Gravity: 1.26 (20°C).	Colour: Colouriess. Explosive Limits: 6.2 to 16%. Melting Point: -35.3 to -40°C.
	PLANTA DE PARTO	

HAZARD DATA			
Human Health			
system. Ingestion: irritation of the gastrointestinal eyes - irritation, clouding of cornea.	hroat, mental confusion, dizziness, nausea and vomiting tract, nausea and vomiting. <u>Contact</u> : skin - absorbed r	g, diarrhea, coma; damage to liver, lungs and nervous eadily, burning, with symptoms similar to inhalation;	
Toxicology: Moderately to highly toxic by inhalation	n, ingestion and contact. LC5n - No information.	LD50 - Orai: rat = 0.68 g/kg	
TLV* - (inhalation) 10 ppm; 60 mg/m ³ . Short-term inhalation Limits - 15 ppm; 60 mg/m ³ (15 min).	LC _{Lo} - inhalation; human - 4 000 ppm/l h Delayed Toxicity - Suspected carcinogen and damage to the liver.	LCLo - Oral: human = 0.81 g/kg	
Fire .			
down vanours	or carbon dioxide. Water may be ineffective, but may in fires) emits highly toxic fumes of phosgene, vinyl of Burning Rate: 1.6 mm/min.		
Reactivity			
With Water: No reaction. With Common Materials: Can react vigorously with Stability: Stable.	oxidizing materials. Reacts violently with aluminum,	ammonia and dimethylaminopropylamine.	
Environment			
Water: Prevent entry into water intakes and waters Aquatic toxicity rating - 100-1 000 ppm; BOD: 0.00 Land-Air: No information. Food Chain Concentration Potential: None.	ways; Fish toxicity: 225 mg/L/96 h/rainbow trout/LC5, 2%, 5 days.	g/freshwater; 150 ppm/tns/pin perch/TLm/saltwater;	

Special Hazards

FLAMMABLE.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". Call Fire Department. Eliminate all sources of ignition. Avoid contact or inhalation. Contact manufacturer for guidance. Stop or reduce discharge, if this can be done without risk. Contain spill by diking. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

Fire and Explosion

Use dry chemical, foam or carbon dioxide. Water may be used to cool fire-exposed containers and knock down vapours. Flashback may occur along vapour trail. In fires, emits highly toxic fumes of phosgene, vinyl chloride and hydrochloric acid.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: eyes - irrigate with plenty of water for at least 15 minutes; skin - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: give water or milk to conscious victim to drink. Do not induce vomiting. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, 4. Remove material with vacuum pumps or liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop of reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or
- other barrier.
- vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

ANTIFREEZE GLYCOL

Common Synonyms

Observable Characteristics Colouriess liquid. Slight odour. Manufacturers

Dow Chemical Canada Inc., Fort Saskatchewan,

Alberta, Sarnia, Ontario.

Union Carbide, Montreal, Quebec.

MONOETHYLENE GLYCOL 1.2-ETHANEDIOL ETHYLENE DIHYDRATE

Transportation and Storage Information

Shipping State: Liquid. Classification: None.

Inert Atmosphere: No requirement. Venting: Open (flame arrester). Pump Type: Most types.

Label(s): Not regulated.

Storage Temperature: Ambient.

Hose Type: Most types.

Grades or Purity: Industrial.

Containers and Materials: Drums, tank cars,

tank trucks: steel, stainless steel.

Physical and Chemical Characteristics

Physical State (20°C, I atm): Liquid. Solubility (Water): Soluble in all proportions.

Molecular Weight: 62.1

Vapour Pressure: 0.05 mm Hg (20°C); 0.2 mm Hg (30°C).

Bolling Point: 196 to 198°C.

Floatability (Water): Sinks and mixes.

Odour: Slight (0.08 to 25 ppm, odour threshold). Flash Point: 116°C (o.c.); 111°C (c.c.).

Vapour Density: 2.1

Specific Gravity: 1.11 (20°C).

Colour: Colourless.

Explosive Limits: 3.2 to 15.3%.

Melting Point: -13°C.

HAZARO DATA

Human Health

Symptoms: Inhalation: intoxication, headache; prolonged inhalation may cause throat irritation and nervous system disorder. Ingestion: intoxication, headache. vomiting, cyanosis, unconsciousness with convulsions. Contact: skin - absorbed causing intoxication; eyes - irritation.

Toxicology: Moderately toxic by Ingestion, contact and inhalation.

TLV+ - 10 mg/m3 (particulate); 50 ppm;

125 mg/m³ (vapour).

LC 50 - No information. Delayed Toxicity - Fatal kidney injury may result

LDsn - Oral: rat = 5.84 g/kg

Short-term Inhalation Limits - 20 mg/m³ (15 min) from ingestion.

(particulate).

Fire Extinguishing Agents: Use water log, alcohol foam, carbon dioxide, or dry chemical. Water or foam may cause frothing.

Behaviour in Fire: No information.

Ignition Temperature: 398°C.

Burning Rate: 1.0 mm/min.

Reactivity

With Water: No reaction, soluble.

With Common Materials: Can react violently with chlorosulfonic acid, oleum and sulfuric acid. May react with strong oxidizing materials.

Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: >100 ppm/48 h/shrimp/LC50/saltwater; Aquatic toxicity rating = 100 to 1 000/96 h/

TLm/freshwater; >5 000 mg/L/24 h/goldfish/LDso/freshwater; 41 000 mg/L/96 h/rainbow trout/LCso/freshwater; BOD: 16 to 68%, 5 days.

Land-Air: No information. Food Chain Concentration Potential: None.

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Special Hazards

Immediate Responses

Keep non-involved people away from spill site. Dike to prevent runoff. Notify manufacturer for advice. Notify environmental authorities.

Protective Clothing and Equipment

Protective outer clothing as required.

Fire and Explosion

Use water fog, alcohol foam, carbon dioxide or dry chemical to extinguish. Water or foam may cause frothing.

Pirst Ald

Move victim out of from spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated ciothing and wash affected areas with plenty of water; eyes - irrigate with water. Ingestion: give water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

ENVIRONMENTAL PROTECTION MEASURES

Response

l'ater

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

ETHYLENEIMINE CH2CH2NH

IDENT	IFICA	TION
	* **	

UN No. 1185

LD50 - Oral: rat = 0.015 g/kg

Common Synonyms	Observable Characteristics	Manufacturers		
ETHYLENEIMINE (inhibited) AZIRIDINE DIMETHYLENIMINE AZACYCLOPROPANE DIHYDROAZIRINE	Colourless, oily liquid. Ammoniacal odour.	No Canadian or U.S. Manufacturers: Imported from Nippon Shokubai, Japan		
Transportation and Storage Information				
Shipping State: Liquid. Classification: Flammable liquid, poison. Inert Atmosphere: Inerted. Venting: Safety relief. Pump Type: Centrifugal or positive displacement; stainless steel.	Label(s): Red label - FLAMMABLE LIQUID; Class 3.2, Group 1. White label - POISON; Class 6.1, Group 1. Storage Temperature: Ambient. Hose Type: Flexible, stainless steel.	Grades or Purity: 99.0%. Containers and Materials: Drums (plastic-lined), tank cars (with inhibitor; NaOH).		
Physical and Chemical Characteristics				
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Completely soluble. Molecular Weight: 43.1 Vapour Pressure: 160 mm Hg (20°C); 250 mm Hg (30°C). Boiling Point: 55 to 57°C.	Floatability (Water): Floats and mixes. Odour: Ammoniacal (1.96 to 2 ppm, odour threshold). Flash Point: -4°C (o.c.); -11°C (c.c.). Vapour Density: 1.5 Specific Gravity: 0.83 (20°C).	Colour: Colourless. Explosive Limits: 3.6 to 46%. Melting Point: -71°C.		

HAZARD DATA

Human Health

Symptoms: inhalation: difficulty breathing, headache, dizziness, coughing, burning of respiratory tract and mucous membranes. Contact: eyes - burns; skin - severe vesicant-desicant can produce third degree burns, readily absorbed, producing symptoms similar to inhalation. Ingestion: nausea, vomiting and burning of mucous membranes.

Toxicology: Highly toxic by ingestion, inhalation or contact.

TLY* - (skin) 0.5 ppm; 1.0 mg/m³.

LC50 - No information.

Short-term Inhalation Limits - No information.

LCLo - Inhalation: rat = 25 ppm/8 h

Delayed Toxicity - Suspected carcinogen.

Fire

Fire Extinguishing Agents: Use atcohol foam, dry chemical or carbon dioxide. Water may be ineffective, but may be used to cool fire-exposed containers and knock down vapours.

Behaviour in Fire: Containers may rupture when exposed to heat and flame. At high temperatures (in fires), may polymerize with the evolution of heat. Flashback may occur along vapour trail. Toxic nitric oxide fumes produced by burning.

Ignition Temperature: 320°C.

Burning Rate: No information.

Reactivity

With Water: Mild reaction; soluble.

With Common Materials: In the presence of acidic type materials catalytically active metals or chloride lons, a violent exothermic reaction can occur. Reacts violently with acetic acid, acetic anhydride, acrolein, acrylic acid, allyl chloride, carbon disultide, chlorine, chlorosulfonic acid, epichlorohydrin, hydrochloric acid, hydrofluoric acid, nitric acid, oleum, silver, sodium hypochlorite, sulfuric acid and vinyl acetate.

Stability: Unstable. May polymerize violently.

Environment

Water: Prevent entry Into water intakes and waterways. Fish toxicity: No information; BOD: No information.

Land-Air: No information.

Food Chain Concentration Potential: None.

Special Hazards

FLAMMABLE, POISON, UNSTABLE, CORROSIVE.

Immediate Resourses

Keep non-involved people away from spill site. Issue warnings: "FLAMMABLE; POISON". Call Fire Department. Eliminate all sources of ignition. Call supplier or manufacturer. Avoid contact and inhalation. If material is burning, work from upwind. If water spray is used, dike area to contain toxic runoff. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Boots - rubber (pants worn outside boots). Gloves - rubber.

Fire and Explosion

Use dry chemical, alcohol foam or carbon dioxide to extinguish. Water may be ineffective on fire, but may be used to cool fire-exposed containers. Fight fires from behind barrier. Flashback may occur along vapour trail. Toxic nitric oxide fumes produced by burning.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: eyes - immediate and continuous irrigation with flowing water; skin - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: do not induce vomiting; give large quantities of milk or water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- Caution: highly toxic; use non-sparking tools.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Adsorb residual liquid on natural or synthetic sorbents.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

ETHYLENE OXIDE CH2CH2O

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UN No. 1040

Common Synonyms	Observable Characteristics	Manufacturers
OXIRANE 1,2-EPOXYETHANE OXANE DIMETHYLENE OXIDE	Colourless gas or liquid at room temperature. Ethereal odour. Ethereal odour. Montreal, Quebec (sold in cylinders Matheson). Dow Chemical Canada & Sarnia, Ontario, Fort Saskatchewan,	
Fransportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Poisonous gas and flammable gas. Inert Atmosphere: Inerted, with nitrogen. Venting: Safety relief. Pump Type: Centrifugal (not gear), explosion-proof.	Label(s): Red label - FLAMMABLE GAS; Class 2.1. White label - POISON; Class 2.3. Storage Temperature: Ambient or refrigerated. Hose Type: Carbon steel, flexible stainless steel, no natural rubber.	Grades or Purity: Technical, pure (99.7%). Containers and Materials: Cylinders, tank cars; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): Completely miscible. Molecular Weight: 44.1 Vapour Pressure: 1 095 mm Hg (20°C); 1 596 mm Hg (30°C). Bolling Point: 10.6°C.	Floatability (Water): Floats and mixes. Odour: Ethereal (0.8 to 500 ppm, odour threshold). Flash Point: <-18°C (c.c.); -29°C (o.c.). Yapour Density: 1.5 Specific Gravity: 0.89 (7°C).	Colour: Colourless. Explosive Limits: 3.6 to 100.0% (under certain conditions as low as 29%). Melting Point: -!!!*C.

HAZARD DATA

	INTRIBUTE INTER
Human Health	
Symptoms: Inhalation: nose and throat irritation; blistering, frostbite; eyes - burns. Ingestion: sore: Toxicology: Highly toxic by ingestion and moderat TLV* - (inhalation) 1 ppm; 20 mg/m ² .	lely toxic by inhalation. LC50 - Inhalation: rat = 1 462 ppm/4 h LD50 - Oral: rat = 0.072 g/kg
Short-term Inhalation Limits - No information.	TC _{Lo} - Inhalation: human 12,500 ppm/10 sec. Delayed Toxicity - Suspected carcinogen.
Fire	
ethylene oxide nonflammable. Water may be used Behaviour in Fire: Flashback may occur along vap Ignition Temperature: 429°C.	to cool divide or dry chemical. Water may be ineffective because only dilution with 23 volumes of water renders to cool fire-exposed containers. our trail. Can burn inside enclosures with no oxygen. Burning Rate: 3.5 mm/min.
Reactivity With Water: Reacts exothermically unless large d With Common Materials: May polymerize violentl aluminum and alkali metal hydroxides. Can react Stability: Stable, within the limits of the foregoin	ly with evolution of heat on contact with anhydrous chlorides of iron, tin and aluminum, oxides of iron and with acids, bases, alcohols, ammonia, copper, magnesium perchlorate and potassium.
Environment	
Water: Prevent entry into water intakes and water freshwater; BOD: 6%, 5 days. Land-Air: No information. Food Chain Concentration Potential: None.	erways. Aquatic toxicity rating = 10 to 100 ppm/96 h/TLm/freshwater; 90 mg/L/24 h/goldfish/LD50/

Special Hazards

FLAMMABLE; POISON. May polymerize violently.

immediate Responses

Keep non-involved people away from spill site. Issue warnings: "FLAMMABLE; POISON". Call Fire Department. Evacuate from downwind. Eliminate all sources of ignition. Avoid contact and inhalation. Work from upwind and use water spray to control vapour. Stop or reduce discharge, if this can be done without risk. If material is leaking from container, it will inevitably catch fire. Due to the material's ability to burn within vessel with no oxygen, depressurization should be attempted. The best method is to displace ethylene oxide through the leak with water. Contact manufacturer or supplier for assistance. Dike to prevent runoff from rainwater or water application. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and full protective clothing. Gloves - rubber or plastic. Acid suit - jacket and pants, rubber or plastic. Boots - high, rubber (pants worn outside boots). Leather should not be used.

Fire and Explosion

Use dry chemical, alcohol, foam or carbon dioxide to extinguish. Water may be ineffective, but may be used to cool fire-exposed containers and to protect men effecting shutoff. Flashback may occur along vapour trait and into vessels if pressure reduced-

First Ald

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; give oxygen if breathing is laboured. Contact: remove contaminated clothing and immediately flush eyes and skin with plenty of warm water. Ingestion: give milk or water to conscious victim and induce vomiting. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Contact manufacturer or supplier for advice.
- 2. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Contact manufacturer or supplier for advice.
- 2. Dike to prevent runoff from rainwater or water application.
- 3. If liquid, remove material with nitrogen displacement and place in appropriate containers.
- 4. Recover undamaged containers.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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2-ETHYLHEXANOL CH3(CH2)3CH(C2H5)CH2OH

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UN No. 2282 - hexanols

Согитоп Ѕулонуть	Observable Characteristics	Manufacturers
2-ETHYLHEXYL ALCOHOL OCTYL ALCOHOL 2-ETHYL-1-HEXANOL	Colourless liquid. Unpleasant, musty odour.	BASF Canada Limited, Montreal, Quebec.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Not regulated. Inert Atmosphere: No requirement. Venting: Open (flame arrester). Pump Type: Standard, flammable liquid types.	Labei(s): None required. Storage Temperature: Ambient. Hose Type: Natural rubber, polyethylene, butyl.	Grades or Purity: Technical, 99 to 99.7%. Containers and Materials: Cans, drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.1 g/100 mL (20°C). Molecular Weight: 130.2 Vapour Pressure: 0.36 mm Hg (20°C); 1.0 mm Hg (33°C). Boiling Point: 179 to 185°C.	Floatability (Water): Floats. Odour: Unpleasant, musty (0.08 to 0.14 ppm, odour threshold). Flash Point: 85°C (o.c.); 73°C (c.c.). Vapour Density: 4.5 Specific Gravity: 0.83 (20°C).	Colour: Colourless. Explosive Limits: 0.88 to 9.7%. Melting Point: -76°C.

HAZARD DATA

Human Health		
Symptoms: Inhalation and Ingestion: nausea, vomit Texicology: Moderately toxic by ingestion and skir TLV® No information. Short-term Inhalation Limits - No information.	ting, headache, dizziness; <u>Contact</u> : skin and ey n contact. LC 50 - No information. Delayed Toxicity - No information.	res - mildly irritating. LD50 - Oral: rat = 3.2 g/kg
Fire		
Fire Extinguishing Agents: Use foam, carbon diox Behaviour in Fire: No information. Ignition Temperature: 231°C.	ide or dry chemical. Water may be ineffective Burning Rate: 4 mm/min.	e.
Reactivity		
With Water: No reaction. With Common Materials: Can react with oxidizing Stability: Stable.	; materials.	
Environment		
Water: Prevent entry into water intakes and wate Land-Air: No information. Food Chain Concentration Potential None.	rways. Fish toxicity: 19 ppm/24 h/brine shrim	p/TLm/seltwater; BOD: 88%, 5 days (theoretical).

Special Hazards

COMBUSTIBLE.

Immediate Responses

Keep non-invoived people away from spill site. Call Fire Department. Avoid contact. Stop discharge if safe to do so. Notify manufacturer. Notify environmental authorities.

Pretective Clothing and Equipment

In fires, Respiratory protection - use self-contained breathing apparatus. Goggles - (mono), tight-fitting. Face shield may be worn, but must not replace goggles. Gloves - rubber or plastic. Coveralls or acid suit (jacket and pants).

Fire and Explosion

Use dry chemical, alcohol foam or carbon dioxide to extinguish fire. Water may be ineffective on fire, but may be used to cool fire-exposed containers.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: remove contaminated clothing and wash eyes and affected skin thoroughly with warm water. Ingestion: give milk or water to conscious victim to drink. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- I. Stop or reduce discharge it safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum
- equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. incinerate (approval of environmental authorities required).

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FENITROTHION C9H12NO5PS

IDENTIFICATION

UN No. 2783

Danger Group According to Percentage of Active

Substance: Group III, Solid 45 to 100% Liquid 10 to 100%

Common Synonyms

Observable Characteristics

Manufacturers

O.O-DIMETHYL 0-4-NITRO-M-TOLYL

Yellowish to yellowish-brown liquid or soild. Slight odour.

Sumitomo Shoil Canada Ltd.,

PHOSPHOROTHIOATE

Common Trade Names

Toronto, Ontario.

Chemagro Ltd., Mississauga, Ontario.

FOLITHION, SUMITHION, NOVATHION

(An insecticide used largely in Canada for spruce budworm control).

Transportation and Storage information

Shipping State: Solid or liquid (formulation).

Classification: None.

Inert Atmosphere: No requirement.

Venting: Open.

Pump Type: No information.

Label(s): Not regulated.

Storage Temperature: Ambient. Hose Type: No information.

Grades or Purity: Various as described

below.

Containers and Materials: Cans, drums; aluminum

stainless steel, lined steel.

Physical and Chemical Characteristics

Physical State (20°C, I atm): Liquid (technical). Solubility (Water): Insoluble (technical) EC and

WP are dispersible.

Molecular Weight: 277.2

Vapour Pressure: 0.000006 mm Hg (20°C). Boiling Point: 140-145°C (0.1 mm Hg) decomposes.

Floatability (Water): Sinks. EC and WP dispersible

in water. Odour: Slight.

Flash Point: No information. Vapour Density: No information. Specific Gravity: 1.32-1.34 (technical). Colour: Yellow to yellowish-brown. Explosive Limits: SN may form explosive

mixtures in air.

Melting Point: 0.3°C (technical).

HAZARD DATA

Human Health

Symptoms: Inhaiation, Ingestion or Contact (skin): readily absorbed through skint nausea, salivation, tearing, abdominal cramps, vomiting, sweating, slow pulse, muscular tremors.

Toxicology: Highly toxic by ingestion and skin contact.

Short-term Inhalation Limits - No information.

TLY* - No information.

LC50 - No information.

Delayed Toxicity - No information.

LD50 - Oral: rat = 0.25 g/kg LDso - Oral: cat = 0.14 g/kg

Fire

Pire Extinguishing Agents: Foam, carbon dioxide or dry chemical.

Behaviour in Fire: Releases toxic fumes.

Ignition Temperature: No information.

Burning Rate: No information.

Reactivity

With Water: No reaction.

With Common Materials: No information.

Stability: Stable.

Environment

Water: Prevent entry into water intakes or waterways.

Land-Air: LD50 - Oral: Wild bird = 0.025 g/kg; chicken = 0.28 g/kg; duck = 1.19 g/kg.
Food Chain Concentration Potential: Decomposes in tissue to other compounds (Desmethylsumithion and dimethylphosphorothioic acid).

Special Hazards

POISON.

Immediate Responses

Keep non-involved people away from spill site. Stop or reduce discharge if safe to do so. Notify manufacturer or supplier. Dike to contain material or water runoff. Notify environmental authorities.

Protective Clothing and Equipment

In tires or confined spaces - Respiratory Protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, approved pesticide respirator and impervious outer clothing.

Fire and Explosion

Use carbon dioxide, foam or dry chemical to extinguish. Releases toxic fumes in fires.

Pirst Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim to drink and induce vomiting; in the case of petroleum distillates, do not induce vomitting for fear of aspiration and chemical pneumonia. If medical assistance is not immediately available, transport victim to hospital, doctor, or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.

- 3. If possible contain
- discharge by booming.
- 4. If floating, skim and remove.

Sinks or mixes

- 3. If possible contain discharge by damming or water diversions. 4. Dredge or vacuum pump to remove
- contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. If liquid, remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. If solid, remove material by manual or mechanical means.
- 6. Recover undamaged containers.
- 7. Absorb residual liquid on natural or synthetic sorbents.
- 8. Remove contaminated soil for disposal.
- 9. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

Disposal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

Available Formulations

Technical Grade: Purity: 95%+.

Properties: insoluble liquid, sinks in water.

Formulations:

EC - emulsifiable concentrate

LI - flauid SN - solution

WP - wettable powder

Purity: - 10%

- 95 +96 - typically 96% in organic solvents.

- 40%

Properties:

- dispersible in water - insoluble in water.
- combustible.
- low combustibility.

FERRIC CHLORIDE FeCi 3

IDENTIFICATION

UN Ño. Solid 1173 Solution 2583

Common Synonyms	Observable Characteristics	Manufacturers
IRON CHLORIDE FERRIC TRICHLORIDE FERRIC PERCHLORIDE IRON TRICHLORIDE IRON PERCHLORIDE IRON (III) CHLORIDE	Brown to black solid or dark reddish-brown liquid (solution). Odourless or slight acidic odour.	No Canadian Manufacturers Canadian Supplier: Diversey Canada Limited, Mississauga, Ontario.
Transportation and Storage Information		
Shipping State: Liquid (39% solution); solid (96%). Classification: Corrosive. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Gear, centrifugal; rubber-lined, plastic-lined.	Label(s): Black and white label - CORROSIVE SOLID; Class 8, Group II; SOLUTION; Class 5, Group III. Storage Temperature: Ambient. Hose Types Natural rubber, polyethylene, poly- propylene, PVC.	Grades or Purity: Solution (42° Bé; 39%); solid, anhydrous, 96%. Containers and Materials: Solution - drums, tank cars, tank trucks. Anhydrous (solid) - drums.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 74.4 g/100 mL (0°C); 535.7 g/100 mL (100°C). Molecular Weight: 162.2 Vapour Pressure: No information. Boiling Point: 315°C (decomposes).	Floatability (Water): Sinks and mixes. Odour: Odourless or slight acidic odour. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 2.9 (anhydrous) (20°C); 1.4 (42° Bé solution).	Colour: Anhydrous - brown to black; Solution - dark reddish-brown. Explosive Limits: Not flammable. Melting Point: Approximately 300°C.

HAZARD DATA			
Human Health			
Symptoms: Contact: eyes - irritation; skin - irritation: irritation of the gastrointestinal tract. Inhalation: Toxicology: Moderately to highly toxic by ingestion TLV*- I mg/m³ (as Fe) Short-term inhalation Limits - 2 mg/m³ (as Fe) (15 min).	nose and throat irritation.	gestion: may cause burns of the mucous membranes and severe LD50 - Orali mouse = 1.28 g/kg	
Fike			
Fire Extinguishing Agents: Not combustible. Most Behaviour in Fire: When heated to decomposition in ignition Temperature: Not combustible.	fire extinguishing agents, except wate i fire, it may release fumes of hydroge Burning Rate: Not combustible.	r, may be used in fires involving ferric chloride. n chloride.	
Reactivity			
With Water: Reacts exothermically producing corrowith Common Materials: Ferric chloride is a strong with allyl chloride and sodium and potassium metal Stability: Stable.	oxidizing agent and solutions are very	corrosive to iron, copper and most metals. Violent reaction	
Environment			
Water: Prevent entry into water intakes and water magna (water flea)/TLm; BOD: None. Land-Air: No information. Food Chain Concentration Potential: None.	ways. Fish toxicity: 4 ppm (as Fe)/24 to	o 96 h/striped bass larvae/LC5g/freshwater; 15 ppm/96 h/ <u>Daphnia</u>	

Special Hazards

CORROSIVE.

immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Avoid contact and inhalation. Stop or reduce discharge, if this can be done without risk. Contain spill by diking with earth or other available material. Contact manufacturer for advice. Notify environmental authorities.

Protective Clothing and Equipment

Goggles - (mono), tight-fitting. If face shield is used, it must not replace goggles. Gloves - rubber. Boots - high, rubber (pants worn outside boots). Ferric chloride attacks leather boots and shoes rapidly. Outerwear - coveralis, aprons; rubber or vinyl. Respiratory protection required only in the case of fire, where hydrogen chloride may be present.

Fire and Explosion

Not combustible. Most fire extinguishing agents, except water, may be used. When heated to decomposition in fire, fumes of hydrogen chloride may be released.

First Ald

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - irrigate with plenty of water for at least 15 minutes; skin - remove contaminated clothing and wash affected areas with plenty of water. Ingestion: give conscious victim milk or water to drink. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, 4. Remove material by manual or liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

FLUORINE F2

IDENTIFICATION

UN No. 1045

Common Synonyms	Observable Characteristics	•	Manufacturers	
None.	Pale yellow compressed gas. irritating odour.	Sharp, choking,	No Canadian manufacturers Canadian suppliers: Air Products, Brampton, Ontario.	Originating from: Air Products and Chemicals Incorporated, Specialty Gas Department, Allentown, PA, USA
Transportation and St	orage Information			
Shipping State: Liquid Classification: Poison Inert Atmosphere: No Yenting: Safety relief Pump Type: No inform	, oxidizer. requirement.	Label(s): White label Yellow label - OXIDI Storage Temperature Hose Type: No infor	iZER; Class 2.3, 5.1. er Ambient.	Grades or Purity: >98% F ₂ . Containers and Materials: Cylinders, tank trailers; steel.
Physical and Chemica	i Characteristics			
Physical State (20°C Solubility (Water): Re Molecular Weight: 38 Vapour Pressure: 18 I 31 300 mm Hg (20°C) Boiling Point: -188°C	acts to form HF. D 00 mm Hg (0°C);	Floatability (Water): liquid and gas react! Odour: Sharp, chokin () £35 ppm, odour th Flash Point: Not Han Vapour Density: 1.3	g, irritating; reshold).	Colour: Pale yellow. Explosive Limits: Not flammable. Melting Point: -219.6°C. Specific Gravity: (liquid) 1.5 (-188°C).
		Vapour Density: 1.3	ZADO DATA	

HAZARD DATA

Symptoms: Inhalation: irritation and ulceration of mucous membranes, coughing fits, difficulty breathing, respiratory damage, coma, death. Ingestion: painful burning, ulceration of mucous membranes, intense thirst, difficulty breathing, coughing, nausea and vomiting, convulsions, coma, death. Contact: skin burning, blisters, profound tissue damage; eyes - burning, irreparable corneal damage, loss of vision.

Toxicology: Highly toxic by inhalation, Ingestion and contact.

TLV (inhalation) | ppm; 2 mg/m3.

LCg - Inhalation: rat = 185 ppm/1 h

Short-term inhalation Limits - 2 ppm; 4 mg/m³ (15 min).

LCg: - Inhalation: mouse = 150 ppm/1 h

Delayed Toxicity - No information.

LD 20 - No information.

Fire Extinguishing Agents: Once a fire has started with fluorine as the oxidizer, there is no effective way of stopping it other than shutting off the source of fluorine. Clear the area and allow the fire to burn until all of the fluorine is consumed. Do not attempt to extinguish the fire with water or chemicals, since these would act as additional fuel.

Behaviour in Fire: Fluorine acts exothermically with most materials.

Ignition Temperature: Not combustible.

Burning Rate: Not combustible.

Reactivity

Fire

With Water: Reacts exothermically to produce HF and oxygen.

With Common Materials: Reacts violently with many materials; such as most organic matter, hydrogen-containing molecules, oxides of sulfur, nitrogen, phosphorous halides, alkaline metals and alkaline earths. Reacts violently with metals and their sulfides, phosphides, oxides, hydrides, acetylides and carbides. Reacts violently with other halogens and halogen acids, phosphorus, sulfur, hydrazine, coke, potassium nitrate, carbon tetrachioride, silicates, alkenes, alkylbenzenes, carbon disulfide, aluminum, manufactured gas and many polymers. Stability: Stable only under normal circumstances, reacts with many materials.

Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life in very low concentrations. Fish toxicity: 2.3 ppm/tns/trout/TLm/freshwater: BOD: No information.

Land-Air: No information.

Food Chain Concentration Potential: None.

Special Hazards

POISON, OXIDIZER, CORROSIVE.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON and OXIDIZER". Call Fire Department (note caution re: use of water). Consider evacuation of downwind area. Contact manufacturer or supplier immediately for guidance and assistance. Control of discharge should be attempted only by experienced persons. Work from upwind. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Gloves - leather. Boots - high rubber.

Fire and Explosion

Water should be used only to cool containers exposed to fire. Do NOT attempt to extinguish fire with water or chemicals.

First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalation</u>: If breathing has stopped or is laboured, give artificial respiration (not mouth-to-mouth method). Supplemental oxygen should be given also. <u>Contact</u>: eyes - irrigate with plenty of water; skin - flush affected areas with plenty of water and remove contaminated clothing immediately. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Recover undamaged containers.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

FLUOROCHLOROMETHANES

IDENTIFICATION

UN No. 1028 dichlorodifluoromethane 1018 chlorodifluoromethane 1022 chlorotrifluoromethane

Common Synonyms 11; CCl ₃ F - Trichlorofluoromethane (Freon 11) 12; Cl ₂ F ₂ C - Dichlorodifluoromethane (Freon 12; Halon 12) 13; ClF ₃ C - Chlorotrifluoromethane (Freon 13) 21; Cl ₂ FCH - Dichlorofluoromethane (Freon 21) 22; ClF ₂ CH - Chlorodifluoromethane (Freon 22)	Observable Characteristics Colourless, practically odourless gases.	Manufacturers Du Pont Canada, Maitland, Ont. Allied Chemical, Amherstburg, Ont.
Transportation and Storage Information		
Shipping State: Liquelied (compressed gases). Classification: Non-flammable gases. Inert Atmosphere: No requirement. Venting: Safety-relief.	Label(s): Green and white label - NON- FLAMMABLE GAS; Class 2.2. Storage Temperature: Ambient. Hose Type: No information.	Pump Type: No information. Grades or Purity: Technical and refrigerant. Containers and Materials: Cylinders; steel.
Physical and Chemical Characteristics		
Physical State (201C, 1 atm): Gas. Solubility (Water): (11) 0.11 g/100 mL (25°C); (12) 0.028 g/100 mL (25°C); (13) 0.009 g/100 mL (25°C); (21) 0.95 g/100 mL (20°C); (22) 0.30 g/100 mL (25°C). Molecular Weight: (11) 137.4; (12) 120.9; (13) 104.5; (21) 102.9; (22) 86.5. Vapour Pressure: (11) 687 mm Hg; (12) 4 250 mm Hg; (13) 2 400 mm Hg; (21) 1 216 mm Hg; (22) 6 800 mm Hg (all at 20°C).	Boiling Point: (11) 23.8°C; (12) -29.8°C; (13) -81.4°C; (21) 8.9°C; (22) -40.7°C. Floatability (Water): Sink and boil. Odour: Odourless. Flash Point: Not flammable. Vapour Density: (11) No information; (12) 4.2; (13) 3.6; (21) 3.6; (22) 3.1. Specific Gravity: (11) 1.49 (0°C); (12) 1.33 (20°C) liquid; (13) 1.35 (-40°C) liquid; (21) 1.42 (0°C); (22) 1.4 (-40.7°C) liquid.	Colour: Colouriess. Explosive Limits: Not flammable. Melting Point: (11) -111.1°C; (12) -158.0°C; (13) -181.0°C; (21) -135.0°C; (22) -157.4°C.

	HAZARD DATA	
Human Health		
Symptoms: Inhalation: dizziness, loss of coordina dermatitis. Toxicology: Low toxicity by inhalation and content of the conte	ation, narcosis, asphyxia, nausea, vomiting. <u>Contact</u> : skin- act. LC ₅₀ - (11) inhalation: mouse = 10 000 ppm/30 min LC ₅₀ - (12) inhalation: rat = \$0 000 ppm/30 min Delayed Toxicity - No information.	- defatting; liquid causes frostbite, irritation and LD ₅₀ - (11) intraperitoneal: mouse = 1.74 g/kg Short-term inhalation Limits - (12) 1 250 ppm; 6 200 mg/m ³ ; (22) 1 250 ppm; 4 375 mg/m ³ 15 min).
Fire		
	ost fire extinguishing agents may be used in fires involving on heating to decomposition; highly toxic chlorides and flaurning Rate: Not combustible.	
With Water: No reaction.		
with Water to reaction. With Common Materials: (12) (13) react with all Stability: Stable.	uminum.	
Environment		
Water: Prevent entry into water intakes and wa Land-Air: No information. Food Chain Concentration Potential: None.	terways. Aquatic toxicity rating (12) and (22): >1 000 ppm	n/96 h/TLm/freshwater; BOD: No information.

Special Hazards

Immediate Responses

Keep non-involved people away from spill site. If involved in fire, evacuate downwind. Notify manufacturer or supplier. Notify environmental authorities.

Protective Clothing and Equipment

In fires or confined spaces Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, protective clothing as required.

Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving fluorochloromethanes. Toxic by-products are formed on heating to decomposition, highly toxic chlorides and fluorides.

First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; give oxygen if breathing is laboured. Contact: skin - remove contaminated clothing, flush affected areas with plenty of water and treat as for frostbite; eyes - irrigate with plenty of water. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- i. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Remove material by manual or mechanical
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

FLUOSILICIC ACID H2SiF6*xH2O

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UN No. 1778

Cemmon Synonyms	Observable Characteristics	Manufacturers
HYDROFLUOSILICIC ACID HEXAFLUOROSILICIC ACID SILICOFLUORIC ACID HYDROSILICOFLUORIC ACID SAND ACID	Colourless to white liquid. Sharp, acidic odour.	Canadian Industries Ltd., Courtright, Ontario. Cominco Ltd., Trail, B.C. International Minerals and Chemicals, Corporation (Canada) Ltd., Port Maitland, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Corrosive liquid. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Centrifugal; rubber-lined, without packing, PVC, propylene or hard rubber.	Label(s): Black and white label - CORROSIVE LIQUID; Class 8, Group II. Storage Temperature: Ambient. Hose Type: Natural rubber, Hysunite, Hypalon.	Grades or Purity: 22 to 30% solutions in water. Containers and Materials: Drums, tank cars, tank trucks; steel, rubber or plastic-lined.
Physical and Chemical Characteristics		4 ** · · · · · · · · · · · · · · · · · ·
Physical State (20°C, f atm): Liquid. Solubility (Water): Soluble in all proportions. Molecular Weight: 144 (for solute only). Vapour Pressure: No information. Bolling Point: 100 to 105°C (decomposes).	Ploatability (Water): Sinks and mixes. Odour: Sharp, acidic. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 1.2 to 1.3 (20°C).	Colour: Colouriess. Explosive Limits: Not flammable. Melting Point: -31 to -20°C.

HAZARD DATA

Human Health		
difficulty breathing, headache, nausea, coma. Con	tact: skin - burning, blistering, and profound da outh and throat, ulceration of mucous membrane nughing, nausea and vomiting, convulsions, coma	s, ulceration of mucous membranes (nose and throat), mage to tissues; eyes - watering, burning, irreparable damage es, pain in swallowing, intense thirst, painful stomach cramps L. LD50 - No information. LDL0 - Oral: guinea pig = 0.2 g/kg
(15 min) (as F).	• • • • • • • • • • • • • • • • • • • •	
Fire		
Fire Extinguishing Agents: Not combustible; most Behaviour in Fire: If decomposing (boiling), as in fignition Temperature: Not combustible. Reactivity	fire extinguishing agents may be used on tires i ire, corrosive and irritating HF fumes are libera Burning Rate: Not combustible.	nvolving thussincic acid, except water. ated.
	9 4 M. P	•
With Water: Soluble, with evolution of heat and to With Common Materials: Vigorously attacks most and magnesium. Stability: Decomposes with heat (evolution of HF)	common metals, including steel, with evolution	of hydrogen gas. May react with aluminum, alkalis, cyanides
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Fond Chain Concentration Potential: None.	rways. Harmful to aquatic life.	

Special Hazards

CORROSIVE.

immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Contact supplier or manufacturer for advice. Contain spill by diking to prevent runoff. Avoid contact and inhalation of fumes. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and full protective clothing. Gloves - (gauntlet) rubber or neoprene. Boots - high, rubber (pants worn outside boots). Outerwear - acid-resistant aprons or suits.

Fire and Explosion

Not combustible; most fire extinguishing agents may be used on fires involving fluosilicic acid, except water.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: eyes - irrigate with plenty of water for at least 15 minutes; skin - flush with plenty of water and remove contaminated clothing. Ingestion: give conscious victim milk or water to drink. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Neutralize soil and contaminated waters with sodium bicarbonate.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

FORMALDEHYDE HCHO (aqueous)

IDENTIFICATION

UN No. 1198 2209

Common Synonyms	Observable Characteristics	Manufacturers
FORMALITH, FORMALIN FORMALDEHYDE SOLUTION FORMIC ALDEHYDE METHANAL, METHYLENE OXIDE OXOMETHANE	Clear, colourless liquid. Pungent, irritating odour.	Celanese Canada Ltd., Edmonton, Alberta. Reichhold Chemicals Ltd., North Bay, Ont., Thunder Bay, Ontario. Bakelite Thermosets Ltd., Belleville, Ont.
Transportation and Storage Information		
Shipping State: Liquid (aqueous solution). Classification: Not regulated. Inert Atmosphere: No requirement. Yenting: Pressure-vacuum. Pump Type: Centrifugal; stainless steel.	Label(s): None. Class 9.1, Group III. Storage Temperature: Ambient. Hose Type: Polyethylene, butyl, Hypalon, Viton, natural rubber.	Grades or Purity: Pure formaldehyde is not available commercially, because of its tendency to polymerize. It is sold as aqueous solutions containing 37 to 50% formaldehyde by weight and 0 to 15% methanol. Containers and Materials: Drums, tank cars, tank trucks; stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Soluble in all proportions. Molecular Weight: 30.0 (solute only). Vapour Pressure: 10 mm Hg (-88°C) (pure); 760 mm Hg (-19.5°C). Boiling Point: -20°C (pure); 101°C (37% solution).	Floatability (Water): Sinks and mixes. Odour: Pungent, irritating (0.8 to 1 ppm, odour threshold). Flash Point: 37% (no methanol) 85°C (c.c.). (with 15% methanol) 50°C (c.c.); 50% (1.5% methanol) 68.5°C (c.c.) Vapour Density: 1.1 Specific Gravity: 37% solution 1.1 (25°C); liquefied pure 1.1 (25°C).	Colour: Colourless. Explosive Limits: 7 to 73%. Melting Point: 37% solution, -15°C; -118°C (pure).

HAZARD DATA

Symptoms: Inhalation: vapour very irritating to eyes and respiratory tract, sore throat, coughing, allergenic effects, bronchitis. Contact: eyes - liquid can cause severe burns skin - repeated contact causes a tanning effect and dermatitis. Ingestion: severe irritation of the gastrointestinal tract; nausea, coma, abdominal pain. Toxicology: Highly toxic by all routes. TLV* - (inhalation) I ppm; 1.5 mg/m³. LC50 - No information. LD50 - Oral: rat = 0.8 g/kg LCL0 - Inhalation: rat = 250 ppm/4 h Delayed Toxicity - Suspected carcinogen. Fire Fire Extinguishing Agents: Use carbon dioxide, alcohol foam, dry chemical or water spray. Water may be used to cool fire-exposed containers and disperse vapours.

gnition Temperature: 424°C (37% methanol free). Burns

Behaviour in Fire: Formaldehyde vapours may be evolved. Ignition Temperature: 424°C (37% methanol free). Burning Rate: No information.

Reactivity

With Water: No reaction.

With Common Materials: Reacts violently with nitrogen dioxide, (perchloric acid and aniline) and performic acid. Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life in low concentrations. Fish toxicity: 25 ppm/96 h/channel catfish/TLm/freshwater; 15 mg/m³/24 h/striped bass larvae/LC₅₀/freshwater; 30 to 1 000 ppm/48 h/shrimp/LC₅₀/saltwater; BOD: 33 to 106%; 5 days.

Special Hazards

POISON.

Immediate Responses

Keep non-involved people away from spill site. Call Fire Department. Avoid contact and inhalation. Eliminate all ignition sources. Stay upwind and use water spray to control vapours. Stop or reduce discharge, if this can be done without risk. Contact supplier for guidance. Dike or dam spill to prevent runoff. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus. Acid suit - jacket and pants, rubber or plastic, or coveralls. Boots - high, rubber (pants worn outside boots). Gloves - rubber or plastic.

Fire and Explosion

Use water spray, dry chemical, alcohol foam or carbon dioxide. Cool fire-exposed containers with water spray.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped (not mouth-to-mouth method); give oxygen if breathing is laboured. Contact: remove contaminated clothing. Wash eyes and skin with plenty of warm water for at least 15 minutes. Ingestion: give warm water to conscious victim to drink. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

ENVIRONMENTAL PROTECTION MEASURES

Response

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, 4. Remove material with pumps or vacuum liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- equipment and place in appropriate containers.
- 5. Adsorb residual liquid on natural or
- synthetic sorbents. 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposai.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

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FORMIC ACID HOOOH

IDEN	TIFIC	A	TIM

UN No. 1779

Common Synonyms	Observable Characteristics	Manufacturers
FORMYLIC ACID HYDROGEN CARBOXYLIC ACID METHANOIC ACID	Colouriess fuming liquid. Pungent, penetrating odour.	No Canadian Manufacturers. US Manufacturers: Celanese Chemical Co., New York, NY Du Pont, Wilmington, DC Union Carbide, New York, NY
Transportation and Storage Information	· · · · · · · · · · · · · · · · · · ·	
Shipping State: Liquid. Classification: Corrosive. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: No information.	Label(s): Black and white label - CORROSIVE. Storage Temperature: Ambient. Hose Type: No information.	Grades or Purity: Technical, 85 to 95%. Containers and Materials: Carboys, drums, bulk by tank cars, tank trucks; stainless steel.
Physical and Chemical Characteristics	والمراجع	
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Miscible in all proportions. Molecular Weight: 46.0 Vapour Pressure: 35 mm Hg (20°C); 54 mm Hg (30°C). Boiling Point: 100.8°C.	Ploatability (Water): Sinks and mixes. Odour: Pungent, penetrating (21 ppm, odour threshold). Flash Point: 100%, 69°C; 90%, 50°C (c.c.). Vapour Density: 1.6 Specific Gravity: 1.22 (20°C).	Colour: Colouriess. Explosive Limits: 18 to 57% (90% solution only). Melting Point: 8.4°C.

HAZARD DATA

Human Health		
Symptoms: <u>Inhalation</u> : irritation of respiratory tra skin - irritation and burns; eyes - irritation and bur Toxicology: Moderately toxic by all routes.		gestion: nausea, vomiting, dizziness, unconsciousness. Contact:
TLV* - 5 ppm; 9 mg/m ³ .	LCso - No information.	LD50 - Oral: rat = 1.21 g/kg
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	LD30 - Oral: mouse = 1.10 g/kg
Fire		**************************************
Behaviour in Fire: No information. Ignition Temperature: 100%, 539°C; 90%, 434°C.		d to cool fire-exposed containers and disperse vapours.
Reactivity		
With Water: No reaction, soluble. With Common Materials: Reacts violently with alk Stability: Stable.	alis, furfuryl alcohol and hydrogen peroxide.	Reacts with sulfuric acid to produce CO.
Environment		
Water: Prevent entry into water intakes magna/TLm/freshwater; 2 500 mg/L/tns/Gammart Land-Air: No information. Food Chain Concentration Potential: None.	and waterways. Fish toxicity: 175 is pulex/LD100/freshwater; BOD: 2 to 27%, 5	mg/L/24 h/bluegill/TLm/freshwater; 120 ppm/48 h/ <u>Daphnia</u> days.

Special Hazards

CORROSIVE. FLAMMABLE.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department. Avoid contact and inhalation. Eliminate all sources of ignition. Call manufacturer or supplier for advice. Dike to contain material. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

Fire and Explosion

Use carbon dioxide, dry chemical or alcohol foam to extinguish. Water may be used to cool fire-exposed containers and disperse vapours.

First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with water. Ingestion: give water to conscious victim to drink; do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier:
- 4. Remove material with pumps or vacuum equipment in appropriate containers.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

FURFURAL OC3H3C CHO

IDENTIFICATION

UN No. 1199

Common Synonyms Observable Characteristics Manufacturers FURAL. Colouriess to yellow to reddish-brown, No Canadian manufacturers. **FURFURALDEHYDE** liquid. Almond-like odour. Canadian supplier: Originating from: The Ouaker Oats Company, 2-FURALDEHYDE The Ouaker Oats Co. of Canada Ltd., Peterborough, Ont. 2-FURAN CARBONAL Chicago, III. ANT OIL **BRAN OIL** PYROMUCIC ALDEHYDE Transportation and Storage Information Shipping State: Liquid. Labelish: Red and white label - COMBUSTIBLE Grades or Purity: Technical, refined. LIOUID: Class 3.3, Group II. Containers and Materials: Drums, tank cars, tank Classification: Combustible liquid. inert Atmosphere: No requirement. Storage Temperature: Ambient. trucks: steel, stainless steel. Venting: Pressure-vacuum. Hose Type: Butyl, polyethylene, polypropylene. Pump Type: Gear, centrifugal. Steel, stainless steel. Physical and Chemical Characteristics Physical State (20°C, I atm): Liquid. Floatability (Water): Sinks. Colour: Colourless to yellow (pure) will darken to Odour: Almond-like (0.25 ppm odour threshold). Solubility (Water): 8.3 g/100 mL (20°C); reddish-brown on exposure to light. 19.9 g/100 mL (90°C). Flash Point: 60°C (c.c.); 67°C (o.c.) Explosive Limits: 2.1 to 19.3%. Molecular Weight: 96.1 **Vapour Density: 3.3** Melting Point: -36.5°C. Specific Gravity: 1.16 (20°C). Vapour Pressure: I mm Hg (20°C); 3 mm Hg (30°C): 10 mm Hg (50°C). Boiling Point: 161.7°C.

	· HAZARD DATA	
Human Health .		
Symptoms: Inhalation: irritation of throat, respirate eyes - irritation. Ingestion: sore throat, abdominal [Toxicology: Moderately toxic by inhalation, ingestion of the control of the co	pain, vomiting, diarrhea.	act: skin - absorbed readily, dermatitis and eczema; LD50 - Oral: rat = 0.127 g/kg
Short-term inhalation Limits - 10 ppm; 40 mg/m ³ (15 min).		
Fire		
Fire Extinguishing Agents: Water spray, carbon distrapours. Behaviour in Fire: No information. Ignition Temperature: 316°C.	ride, dry chemical, or alcohol foam. Water m Burning Rate: 2.6 mm/min.	nay be used to cool fire-exposed containers and disperse
Reactivity		
With Water: No reaction. With Common Materials: Reacts explosively with a Stability: Stable.	nineral acids or alkalis.	
Environment		
Water: Prevent entry into water intakes and water TLm/freshwater; 1.2 to 32 mg/L/24 h/bluegill/TLm Land-Air: No information.	ways. Harmful to aquatic life in low concentr /freshwater; 24 to 44 mg/L/24 to 96 h/mosqu	rations. Fish toxicity: 24 ppm/96 h/bluegill/ uito fish/TLm/freshwater; BOD: 28 to 77%; 5 days.

Food Chain Concentration Potential: None.

Special Hazards

COMBUSTIBLE. VIOLENT REACTION WITH ACIDS AND ALKALIS.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "COMBUSTIBLE". Call Fire Department. Avoid contact and inhalation. Contain spill by diking. Stop or reduce discharge, if this can be done without risk. Notify manufacturer or supplier. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Gloves - rubber. Boots - high, rubber (pants worn outside boots).

Fire and Explosion

Moderate fire and explosion hazard when exposed to heat and flame. Use water spray, carbon dioxide, dry chemical or alcohol foam to extinguish. Water may be used to cool fire-exposed containers and disperse vapours.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: remove contaminated clothing immediately; flush eyes and affected skin with plenty of warm water. Ingestion: give water or milk to conscious victim to drink and induce vomiting. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or
- water application.
- 4. Dredge or vacuum pump to remove contaminants, 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
 - 5. Recover undamaged containers.
 - 6. Absorb residual liquid on natural or
 - synthetic sorbents.
 - 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

GASOLINE

IDENTIFICATION

UN No. 1203

Common Synonyms	Observable Characteristics	Manufacturers
PETROL, AUTOMOTIVE FUEL AV-GAS - leaded contains tetraethyl lead - lead-free may contain other compounds.	Colouriess (or dyed red, purple) liquid. Typical gasoline odour.	Universally available.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Open (flame arrester) or pressure- vacuum. Pump Type: Standard.	Label(s): Red and white label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Standard.	Grades or Purity: Various octane ratings or use ratings. Leaded or lead-free. Containers and Materials: Cans, drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, f atm): Liquid. Solubility (Water): 1 to 100 ppm/100 mL water. Molecular Weight: Mixture of materials. Vapour Pressure: 300 to 600 mm Hg (20°C). Boiling Point: 40 to 200°C.	Floatability (Water): Floats. Odour: Gasoline (0.25 ppm, odour threshold). Flash Point: -43°C (c.c.) (up to 60 octane); -38°C (c.c.) up to 100 octane; -46°C (c.c.) aviation. Vapour Density: 3 to 4 Specific Gravity: 0.75 to 0.85 at 20°C.	Colour: Colouriess to (dyed red or purple). Explosive Limits 1.4 to 7.6%. Melting Point: -90 to -75°C.

Human Health		
Symptoms: inhalation: vapours cause rapid briconvulsions, coma. Ingestion: gastrointestinal inwatering, stinging and inflammation. Toxicology: Moderately toxic by inhalation.	eathing, excitability, staggering, headache, fa itation, dizziness, fatigue, loss of consciousnes	tigue, nausea and vomiting, dizziness, drowsiness, narcosis, ss, coma. Contact: skin - dryness, cracking, irritation; eyes -
TLV* - 300 ppm; 900 mg/m ³ .	LC50 - No information.	LD50 - No information.
Short-term inhalation Limits - 500 ppm; 1 500 mg/m ³ (15 min).	LCLo - Inhalation: man = 900 ppm/1 h Delayed Toxicity - No information.	
Fire		
containers. Behaviour in Fire: Flashback may occur along valignition Temperature: 280°C, up to 60 octane; 456°C, up to 100 octane; 440°C, 100 to 130 octal (aviation grade); 471°C, 115 to 145 octane (aviat	Burning Rate: 4 mm/min.	
Reactivity		
With Water: No reaction. With Common Materials: Can react vigorously w Stability: Stable.	th oxidizing materials.	
Environment		
Water: Prevent entry into water intakes and wat 91 mg/L/24 h/juvenile American shad/TLm/saltw Land-Air: No information. Food Chain Concentration Potential: None.	erways. Harmful to aquatic life. Fish toxicity: ! rater; 5 to 40 ppm/96 h/rainbow trout/TLm/fresi	90 ppm/24 h/juvenile American shad/TLm/freshwater; hwater; BOD: 8%, 5 days.

Special Hazards

FLAMMABLE.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". Call Fire Department. Eliminate all sources of ignition. Notify manufacturer. Dike to prevent runoff. Shut off leak, if safe to do so. Notify environmental authorities.

Protective Clothing and Equipment

Protective clothing as required.

Fire and Explosion

Use foam, dry chemical or carbon dioxide to extinguish. Water may be ineffective and cause fire to spread, but may be used to cool fire-exposed containers. Flashback may occur along vapour trail.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalation</u>: if breathing has stopped give artificial respiration; if laboured, give oxygen. <u>Ingestion</u>: give water to conscious victim to drink; do not induce vomiting. <u>Contacts skin</u> - remove contaminated clothing and wash affected areas with plenty of warm water; eyes - irrigate with plenty of water. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Adsorb residual liquid on natural or
- synthetic sorbents.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

GLYCERINE CH2OHCHOHCH2OH

IDENTIFICATION

Observable Characteristics	Manufacturers
Colourless oily liquid. Odourless.	Lever Detergents Ltd., Toronto, Ont. Proctor & Gamble Co. of Canada Ltd., Hamilton, Ont. Colgate-Palmolive Limited, Toronto, Ont. Emery Industries (Canada) Ltd., Toronto, Ont. Dow Chemical Canada - from Dow Chemical, Freeport, TX.
Label(s): Not required. Voluntary label - COMBUSTIBLE LIQUID. Storage Temperature: Ambient. Hose Type: Polyethylene, Viton, butyl, Hypalon, Tellon, natural rubber.	Grades or Purity: CP, 99.5%; USP, 96%. Containers and Materials: Drums, tank cars, tank trucks; steel or stainless steel.
Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: 177°C (o.c.); 160°C (c.c.). Vapour Density: 3.2 Specific Gravity: 1.26 (20°C).	Colour: Colourless. Explosive Limits: Not explosive in air. Melting Point: 17.9°C.
	Label(s): Not required. Voluntary label - COMBUSTIBLE LIQUID. Storage Temperature: Ambient. Hose Type: Polyethylene, Viton, butyl, Hypalon, Telion, natural rubber. Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: 177°C (o.c.); 160°C (c.c.). Vapour Density: 3.2

Human Health		
Symptoms: Inhalation: irritation of respiratory tra- Toxicology: Moderately toxic by inhalation, ingest TLV* - (inhalation) 10 mg/m ³ (mist). Short-term inhalation Limits - No information.	ion and contact. LC50 - No information.	ng, diarrhea, fever. <u>Contact</u> : eyes-irritation. LD50 - Oral: rat = 12.6 g/kg
	Delayed Toxicity - No information.	
Fire		
Fire Extinguishing Agents: Use alcohol foam, dry of Behaviour in Fire: No information.	chemical or carbon dioxide. Water or Ioan	may cause frothing.
Ignition Temperature: 370°C.	Burning Rate: 0.9 mm/min.	
Reactivity		
With Water: No reaction; soluble. With Common Materials: Can react with powerful chromium oxide, potassium permanganate, silver p Stability: Stable.		shydride, calcium hypochlorite, chlorine, chromic anhydride,
Environment		
Water: Prevent entry into water intakes and water TLm/freshwater; BOD: 61 to 87%, 5 days. Land-Air: No information. Food Chain Concentration Potential: None.	ways. Fish toxicity: >5 000 mg/L/24 h/goi	ldfish/LD5g/freshwater; Aquatic toxicity rating = >1 000 ppm/96 h/

Special Hazards

COMBUSTIBLE.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "COMBUSTIBLE". Call Fire Department. Stop or reduce discharge, if safe to do so. Contain spill by diking. Notify manufacturer. Notify environmental authorities.

Protective Clothing and Equipment

Goggles - (mono), tight-fitting, and rubber gloves. Protective outerwear as required.

Fire and Explosion

Use alcohol foam, dry chemical or carbon dioxide. Water or foam may cause frothing.

First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: remove contaminated clothing; wash eyes and skin thoroughly with plenty of water. Ingestion: give water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for adice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- i. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. When liquid, remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. When solid, remove material by manual or mechanical means.
- 6. Recover undamaged containers.
- 7. Adsorb residual liquid on natural or synthetic sorbents.
- 8. Notify environmental authorities to discuss

disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

IDENTIFICATION

UN No. 1208

Common Synonyms	Observable Characteristic	\$	Manufacturers	
HEXANE	Clear, colourless liquid. A like odour.	Aild, gasoline-	ESSO Chemicats, Sarnia, Ontario. Texaco Canada, Mississauga, Ontario Canadian suppliers: Hercutes Canada, Varennes, Québec. Polysar, Sarnia, Ontario. Stanchem, Montreal, Québec.	Originating from: Phillips Petroleum, USA Union Oli, USA.
Transportation and S	torage information			
Shipping State: Liquic Classification: Flamm Inert Atmosphere: No Venting: Open (flame or pressure-vacuum. Pump Type: Centrifi	nable liquid. o requirement. : arrester)	LIQUID; Class Storage Temp	and white label - FLAMMABLE 3.1, Group II. erature: Ambient. einforced antistatic rubber, or	Grades or Purity: 85, 95 and 99%. Containers and Materials: Cans, drums, tank cars, tank trucks; steel.
Physical and Chemic	al Characteristics			
Molecular Weight: 86	99095 g/100 mL (20°C). 5.2 1 mm Hg (20°C); 190 mm Hg	Floatability (V Odour: Mild, g Flash Point: - Vapour Densit Specific Grav	gasoline-like. 22°C (c.c.).	Colour: Colourless. Explosive Limits: 1.1 to 7.5%. Melting Point: -94 to -96°C.

HAZARD DATA				
Human Health				
Symptoms: Inhalation: irritation of respiratory tract, nausea, vomiting, headache, depression. Co Toxicology: Low toxicity by all routes.	ict, dizziness, numbness of extremities, intoxication, ntact: skin - defatting and dermatitis; eyes - irritation	unconsciousness. <u>Ingestion</u> : irritation of gastrointestinal on.		
TLV* - 50 ppm; 180 mg/m ³ .	LC50 - No information.	LD50 - No information.		
Short-term inhalation Limits - No information.	LC_{L0} - Inhalation: mouse = 120 000 mg/m ³ . Delayed Toxicity - No information.	LDLo - Intraperitoneal: rat = 9.1 g/kg		
Fire				
Fire Extinguishing Agents: Use dry chemical and of Behaviour in Fire: Fiashback may occur along vapignition Temperature: 223°C.	carbon dioxide. Water may be ineffective but may be our trail. Burning Rate: 7.3 mm/min.	e used to cool fire-exposed containers.		
Reactivity				
With Water: No reaction. With Common Materials: Can react vigorously with Stability: Stable.	th oxidizing materials.			
Environment				
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: None.	erways. Aquatic toxicity rating =>1 000 ppm/96 h/TL	.m/freshwater; BOD: 221%; 5 days.		

Special Hazards

FLAMMABLE.

immediate Responses

Keep non-involved people away from splil site. Issue warning: "FLAMMABLE". Call Fire Department. Eliminate all sources of ignition. Stop or reduce discharge, if safe to do so. Dike to contain the spill. Notify manufacturer. Notify environmental authorities.

Protective Clothing and Equipment

Protective outerwear as required.

Fire and Explosion

Use carbon dioxide or dry chemical to extinguish. Water may be ineffective, but may be used to cool fire-exposed containers.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Ingestion: give water to conscious victim to drink, do not induce vomiting. Contact: skin - remove contaminated clothing and wash affected areas with plenty of water; eyes - irrigate with plenty of water. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. If possible, contain discharge by booming.
- 5. If floating, skim and remove.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

HYDRAZINE NoH4

IDENTIFICATION

UN No. 2029 anhydrous 2030 24% solution

Manufacturers Observable Characteristics Common Synonyms Anhydrous - colourless furning liquid. DIAMIDE Hydrates - colourless to yellow liquids. DIAMINE Ammoniacal odour. HYDRAZINE, BASE HYDRAZINE. AQUEOUS SOLUTION HYDRAZINE, HYDRATE Transportation and Storage information Label(s): Black and white label - POISON. Grades or Purity: Anhydrous 100%; hydrates Shipping State: Liquid. 35 to 64% water solutions. Storage Temperature: Ambient. Classification: Corrosive, poison. Containers and Materials: Carboys, drums; steel Hose Type: No information. Inert Atmosphere: Padded. or aluminum. Venting: Pressure-vacuum. Pump Type: No information. Physical and Chemical Characteristics Floatability (Water): Floats and mixes (saltwater): Colour: Colourless to yellow. Physical State (20°C, I atm); Liquid. Explosive Limits: 2.9 to 98%. sinks and mixes (freshwater). Solubility (Water): Soluble in all proportions. Odour: Ammoniacal (3 to 4 ppm, odour threshold. Melting Points: 1.5°C (anhydrous); Molecular Weights 32.1 -14°C (24% solution). Flash Point: 52°C (o.c.): 37.8°C (c.c.) Vapour Pressure: 16 mm Hg (20°C); 25 mm Hg (30°C). Vapour Density: 1.04 to 1.1 Specific Gravity: 1.004 (anhydrous) (20°C). Boiling Point: Anhydrous 113.5°C: 24% solution 102°C.

HAZARD DATA

Human Health

Symptoms: Inhalation: sore throat, coughing, laboured breathing, dizziness. Ingestion: headache, nausea, abdominal pain, diarrhea, dizziness; symptoms similar to inhalation. Contact: skin - redness, pain, absorbed; eyes - temporary blindness, redness and pain.

Toxicology: Highly toxic by all routes.

TLV (skin) 0.1 ppm; 0.1 mg/m³.

Short-term Inhalation Limits - No information.

LC50 - Inhalation: rat = 570 ppm/4 h LC50 - Inhalation: mouse = 252 ppm/4 h

Delayed Toxicity - Suspected carcinogen.

LD50 - Intraperitoneal: rat = 0.76 g/kg

LDLo - Orali mouse = 0.41 g/kg

Fire

Fire Extinguishing Agents: Use foam, carbon dioxide or dry chemical. Flooding with water may be necessary to prevent reignition. Water may be used to cool fire-exposed containers and knock down vapours.

Behaviour in Fire: Releases toxic NO_x fumes. Can ignite spontaneously in air on contact with porous materials such as earth, asbestos, wood or cloth. Ignition Temperature: Yary with contact of different materials: iron-rust 23°C, black iron 132°C, stainless steel 156°C; glass 270°C.

Burning Rate: I mm/min (est.)

Reactivity

With Water: No reaction; soluble.

With Common Materials: Reacts violently with (alkali metals and ammonia), chlorine, chromates, cupric oxide and salts, fluorine, hydrogen peroxide, metallic oxides, nickel, nitric acid, nitrous oxide, oxygen, potassium dichromate and sodium dichromate.

Stability: May spontaneously ignite under various situations.

Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 146 ppm/0.5 h/rainbow trout/LD10g/freshwater; 136 ppm/22 h/carp/LD10g/freshwater. Land-Air: No information.

Food Chain Concentration Potential: No information.

Special Hazards

CORROSIVE. POISON. Flammable. May spontaneously ignite under certain conditions.

Immediate Responses

Keep non-involved people away from spill site. Issue warnings: "CORROSIYE, POISON". Call Fire Department. Avoid contact and inhalation. Evacuate from downwind. Notify manufacturer or supplier. Dike to contain material. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

Fire and Explosion

Use foam, carbon dioxide, or dry chemical to extinguish. Flooding with water may be necessary to prevent reignition. Water may be used to cool fire-exposed containers and knock down vapours. Releases toxic NO_X furnes in fires.

First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration, (not mouth-to-mouth method) if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with water. Ingestion: give conscious victim water to drink; do not induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Tater

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. With great caution, remove material with pumps or
- vacuum equipment and place in appropriate containers.
- Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal

and cleanup of contaminated materials.

Disnosal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

HYDROCHLORIC ACID HCI (aqueous)

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UN No. 1789

Common Synonyms	Observable Characteristics	Manufacturers
MURIATIC ACID HYDROGEN CHLORIDE (aqueous) CHLOROHYDRIC ACID	Water-white to pale yellow liquid. Sharp, irritating odour. Fumes in humid air.	Canadian industries Limited, Cornwall, Ont., Becancour, Que. Dow Chemical Canada Inc., Sarnia, Ont., Fort Saskatchewan, Alta. Du Pont Canada Limited, Maitland, Ont. FMC Chemicals, Squamish, B.C. Canadian Occidental, Vancouver, B.C. St. Anne Chemicals, Macawick, N.B.
Transportation and Storage Information		
Shipping State: Liquid (aqueous solution). Classification: Corrosive liquid. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Gear, centrifugal, diaphragm. Rubber or plastic-lined.	Label(s): White and black label - CORROSIVE LIQUID; Class 8, Group II. Storage Temperature: Ambient. Hose Type: Natural rubber, polyethylene, polypropylene, PVC, etc.	Grades or Purity: Commercial strengths; 18°Bé (27.9% HCl); 20°Bé (31.5% HCl); 22°Bé (35.2% HCl); 23°Bé (37.1% HCl). Containers and Materials: Drums, tank cars, tank trucks; steel, rubber-lined.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Soluble in all proportions. Molecular Weight: 36.5 (HCl gas). Vapour Pressure: 25 mm Hg (25°C) (20°Bé); Boiling Point: 83°C (20°Bé); 62°C (22°Bé); 51°C (23°Bé); 98°C (18°Bé).	Floatability (Water): Sinks and mixes. Odour: Sharp, Irritating (1 to 10 ppm, odour threshold). Flash Point: Not flammable. Vapour Density: 1.3 Specific Gravity: 1.14 (18°Bé); 1.16 (20°Bé); 1.18 (22°Bé); 1.19 (23°Bé) (15.5°C).	Colour: Colouriess to pale yellow. Explosive Limits: Not Hammable. Melting Point: -42°C (18°Bé); -53°C (20°Bé); -66°C (22°Bé); -74°C (23°Bé).

HAZARD DATA

Symptoms: Inhalation: Vapours can produce severe irritation of the upper respiratory tract, coughing, burning of throat and choking. Contact: eyes - severe irritation of eyes and lids; burning, visual impairment may result; skin - can cause serious burns. Ingestion: very corrosive, can cause serious internal injury. May be fatal.

Toxicology: Moderately toxic by all routes.

TLVº - (inhalation) 5 ppm: 7 mg/m3.

Short-term Inhalation Limits - No information.

LC50 - Inhalation: rat = 3 i24 ppm/30 min

LDso - Oral: rabbit = 0.9 g/kg

Delayed Toxicity - None known.

Fire

Fire Extinguishing Agents: Not combustible. Most fire extinguishing agents may be used on fires involving hydrochloric acid. Water spray may be used sparingly to knock down vapours.

Behaviour in Fire: When heated, toxic and corrosive hydrogen chieride gas is released.

ignition Temperature: Not combustible.

Burning Rate: Not combustible.

Reactivity

With Water: Soluble, with evolution of heat.

With Common Materials: Reacts violently with acetic anhydride, ammonium hydroxide, calcium phosphide, chlorosulfonic acid, ethylenediamine, oleum, perchloric acid, propylene oxide, sodium hydroxide, sulfuric acid and vinyl acetate. Reacts with metals forming flammable H2 gas. Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: 282 ppm/96 h/mosquito lish/TLm/freshwater; 100 to 300 mg/L/48 h/shrimp/LC50/saltwater; BOD: None.

Land-Air: No information.

Food Chain Concentration Potential: None.

Special Hazards

CORROSIVE.

immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIYE". Contact manufacturer for guidance and assistance. Avoid contact and inhalation. Stop or reduce discharge if this can be done without risk. Contain spill by diking to prevent runoff. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and acid suit (rubber). Gloves - rubber. Boots - high, rubber (pants worn outside boots).

Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving hydrochloric acid. Water may be used sparingly to knock down vapours.

First Ale

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: immediately irrigate eyes and flush skin with plenty of water for at least 30 minutes while removing contaminated clothing. Inhalation: If breathing has stopped, give artificial respiration; if laboured, give oxygen. Ingestion: If victim is conscious, give as much water or milk as possible to dilute acid. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Adsorb residual liquid on natural or synthetic sorbents.
- 6. Contaminated soil can be treated with lime or soda ash (calcium hydroxide) to render
- it neutral.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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HYDROFLUORIC ACID HIF aqueous

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	1 I II I I		

UN No. 1790

Common Synonyms	Observable Characteristics	Manufacturers
HYDROGEN FLUORIDE (aqueous) FLUORHYDRIC ACID	Clear, colourless liquid. Strong, irritating odour.	Allied Chemical Ltd., Valleyfield, Quebec, Amherstburg, Ontario. Alcan Smelters and Chemicals, Jonquière, Que.
Transportation and Storage Information		
Shipping State: Liquid (aqueous solution). Classification: Corrosive, poison. inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Centrifugal; stainless steel; all propylene.	Label(s): White and black label - CORROSIVE. White label - POISON; Class 8, 2.3. Storage Temperature: Ambient. Hose Type: Reinforced Halon TFE or Hypalon.	Grades or Purity: Technical, 48%; CP, 70%. Containers and Materials: Bottles, drums, tank cars, tank trucks; polylined.
Physical and Chemical Characteristics		
Physical state (20°C, I atm): Liquid. Solubility (Water): Soluble in all proportions. Molecular Weight: 20.0 (for HF gas). Vapour Pressure: 25 mm Hg (20°C) 48%; 100 mm Hg (15°C), 125 mm Hg (20°C) 70%. Bolling Point: 67°C (70% solution) 198°C (48% solution); 112°C (38% solution).	Floatability (Water): Sinks and mixes. Odour: Strong, irritating (5 ppm, odour threshold). Flash Point: Not flammable. Vapour Density: 0.7 Specific Gravity: 1.26 (70%); 1.19 (48%) (25°C).	Colour: Colourless. Explosive Limits: Not flammable. Melting Point: -37°C (48%); -70°C (70%).

Human Health				
Symptoms: Contact: severe burns to skin and eyes. tissue burns. Can be fatal by all routes. Toxicology: Highly toxic by all routes. TLV* - (inhalation) 3 ppm; 2.5 mg/m³ (as F). Short-term inhalation Limits - 6 ppm; 5 mg/m³ (15 min).	Inhalation: may severely burn respiratory trace LC50 - Inhalation: rat = 1 276 ppm/l h Delayed Toxicity - No information.	t, rapid lung inflammation and congestion. <u>Ingestion</u> : severe LD ₅₀ - No information. LD _{LO} - Intraperitoneal: rat = 0.025 g/kg		
Fixe				
Fire Extinguishing Agents: Not combustible. Water may be used on fires involving hydrofluoric acid. Behaviour in Fire: Not combustible. Toxic and corrosive HF vapours emitted. Ignition Temperature: Not combustible. Burning Rate: Not combustible.				
Reactivity				
With Water: Soluble with evolution of heat. With Common Materials: Reacts violently with calcium oxide, chlorosulfonic acid with acetic anhydride, ammonium hydroxide, ethylenediamine, fluorine, oleum, propylene oxide, sodium, sodium hydroxide, sulfuric acid and vinyl acetate. Stability: Stable.				
Environment				
Water: Prevent entry into water intakes and waterways. Fish toxicity: 60 ppm/tns/fish/lethal/freshwater; >300 mg/L as NaF/48 h/brine shrimp/LC50/saltwater; BOD: None. Land-Air: No information. Food Chain Concentration Potential: None.				

Special Hazards

CORROSIVE, POISON, TOXIC.

Immediate Responses

Keep non-involved people away from spill area. Issue warning: "CORROSIVE, POISON". Contact supplier for advice. Call fire department. Avoid contact and inhalation. Stop or reduce discharge, if this can be done without risk. Stay upwind and if water spray is used to control vapours; dike runoff. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit or acid suit. Boots - high, neoprene (pants worn outside boots). Gloves - neoprene.

Fire and Explosion

Not combustible. Water may be used on fires involving hydrofluoric acid.

Pirst Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped (not mouth-to-mouth method); give oxygen if breathing is laboured. Contact: remove contaminated clothing; wash eyes and skin with plenty of warm water for at least 15 minutes. Ingestion: give milk or water to conscious victim to drink. DO NOT INDUCE VOMITING. If vomiting occurs spontaneously, give more water to further dilute the chemical. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Absorb residual liquid on natural or synthetic sorbents.
- 6. Contaminated soil can be treated with lime (calcium hydroxide) to render it neutral.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

HYDROGEN Ho

IDENTIFICATION

UN No. 1049 (compressed gas)

Manufacturers Observable Characteristics Common Synonyms Colourless gas. Odourless. Canadian manufacturers: Syncrude Canada, Fort McMurray, Alberta. Canadian Fertilizers, Medicine Hat, Alberta. Alberta Gas Chemicals, Medicine Hat, Alberta. Cominco, Carselands, Alberta. Transportation and Storage Information Shipping State: Gas (compressed) and liquid Label(s): Red label - FLAMMABLE GAS; Class 2.1. Grades or Purity: Technical: pure from 99.8% to uitra oure. Storage Temperature: Ambient. (compressed gas). Containers and Materials: As gas in cylinders, tube Hose Type: Gas - braided-high pressure. Classification: Flammable gas. trailers; steel. As liquid in cargo or portable tanks. inert Atmosphere: No requirement. Venting: Safety relief. Physical and Chemical Characteristics Floatability (Water): Floats and boils. Colour: Colourless. Physical State (20°C, 1 atm): Gas. Explosive Limits: 4.0 to 75%. Odour: Odourless. Solubility (Water): 0.00015 g/100 mL Meiting Point: -259.2°C. Flash Point: <-50°C. (20°C). Vapour Density: 0.07 (gas) (25°C). Molecular Weight: 2.0 Specific Gravity: (liquid) 0.07 (-250°C). Vapour Pressure: 8 590 mm Hg (-241°C). Boiling Point: -252.8°C.

HAZARD DATA

Human Health Symptoms: Inhalation: acts as an asphyxlant by displacing air; resulting in cyanosis, diminished mental alertness, impairment and collapse. Contact: skin frostbite. Toxicology: Low toxicity by all routes. LC50 - No information. LDso - No information. TLV* - Asphyxiant. Delayed Toxicity - No information. Short-term inhalation Limits - No information. Fire Extinguishing Agents: Shut off leak before attempting to extinguish fire. Use carbon dioxide, dry chemical, water or halons to extinguish. Water may be used to cool fire-exposed containers. Behaviour in Fire: Containers may rupture violently. Burning Rate: 9.9 mm/min. Ignition Temperature: 500°C. Reactivity With Water: No reaction. With Common Materials: Can react vigorously with oxidizing materials. Reacts violently with bromine, chlorine, chlorine trifluoride, fluorine, lithium, nitrogen trifluoride, 1-pentol, and finely divided platinum in air. Stability: Stable.

Environment

Water: No information. Land-Air: No information. Food Chain Concentration Potential: No information.

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EMERGENCY MEASURES

Special Hazards

FLAMMABLE.

Immediate Responses

Keep non-involved people away and upwind from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Notify manufacturer. Avoid contact with liquid. Stop or reduce discharge, if safe to do so. Notify environmental authorities.

Protective Clothing and Equipment

Outer protective clothing as required.

Fire and Explosion

Shut off leak before attempting to extinguish fire. Use carbon dioxide, dry chemical, water or halons to extinguish. Water may be used to cool fire-exposed containers. Containers may rupture violently.

First AL

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin and eyes - remove contaminated clothing, flush affected areas with plenty of water and treat as for frostbite. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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HYDROGEN CHLORIDE HC1 (anhydrous)

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UN No. 1050

Common Synonyms	Observable Characteristics	Manufacturers
HYDROCHLORIC ACID (anhydrous)	Clear, colourless gas. Pungent, suffocating odour. Fumes in contact with humid air.	Dow Chemical Canada Inc., Sarnia, Ontario. Canadian Industries Ltd., Becancour, Quebec. Du Pont Canada, Maitland, Ontario.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Poisonous gas. Inert Atmosphere: No requirement. Venting: Safety relief.	Label(s): White Label - POISONOUS GAS; Class 2.3. Storage Temperature: Ambient. Hose Type: Monel, stainless steel. Flexible braided, pressure.	Grades or Purity: Technical, 97.5 to 99%. Containers and Materials: Cylinders, tank cars; steel, stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): Soluble 82 g/100 mL (0°C); 56.1 g/100 mL (60°C). Molecular Weight: 36.5 Vapour Pressure: 1 500 mm Hg (-76.6°C); 30 400 mm Hg (17°C). Bolling Point: -85.0°C	Floatability (Water): Liquid, sinks and mixes. Odour: Highly irritating, pungent and suffocating. (I to 10 ppm, odour threshold). Flash Point: Not flammable. Vapour Density: 1.3 Specific Gravity: (liquid) 1.19 (-85°C).	Colour: Colouriess. Explosive Limits: Not flammable. Melting Point: -114.2°C.

Phiman Health		
Symptoms: Inhalation: vapour produces severe in causes severe burns and damage; skin - burns. In Texicology: Texic by all routes. TLY* - (inhalation) 5 ppm; 7 mg/m³. Short-term inhalation Limits - No information.	ritation of the upper respiratory tract, coughing, burgestion: severe irritation, burning of throat and gaste LC50 - Inhalation: rat = 4 701 ppm/30 min. Delayed Toxicity - None known.	ning of throat, choking and possibly death. <u>Contact</u> : eyes rointestinal tract. May be fatal. LD50 - Oral: rabbit = 0.9 g/kg (solution).
Fire		
Fire Extinguishing Agents: Not combustible. Water knock down vapours. Behaviour in Fire: Not combustible. Ignition Temperature: Not combustible.	may be used on fires involving hydrogen chloride. Burning Rate: Not combustible.	Water may be used to cool fire-exposed containers and
Reactivity		
With Water: Soluble; forms hydrochloric acid. With Common Materials: Reacts violently with ca Scability: Stable. Corrosive to metals when wet.	Icium carbide, cesium carbide, lithium silicide, magne	esium boride, mercuric sulfate and sodium.
Environment		
Water: Prevent entry into water intakes and water freshwater; 100 to 330 ppm/48 h/shrimp/LC50/sal Land-Air: No information. Food Chain Concentration Potential: None.	rways. Harmful to aquatic life in low concentrations. itwater; BOD: None.	Fish toxicity: 282 ppm/96 h/mosquito fish/TLm/

Special Hazards

CORROSIVE.

Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department. Evacuate from downwind. Contact manufacturer. Avoid all contact and inhalation. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Boots - high, rubber (pants worn outside boots). Gloves - rubber.

Fire and Explosion

Not combustible. Water may be used on fires involving hydrogen chloride. Water may also be used to cool fire-exposed containers and knock down vapours.

First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: immediately flush eyes and skin with plenty of warm water for 30 minutes or more and remove contaminated clothing. Inhalation: if not breathing, give artificial respiration (not mouth-to-mouth method); if breathing is laboured, give oxygen. Ingestion: give milk or water to conscious victim. Do not induce vomiting. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

ENVIRONMENTAL PROTECTION MEASURES

Response

Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Recover undamaged containers.
- 5. Neutralize contaminated soil with lime
- (calcium hydroxide).
- 6. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

HYDROGEN FLUORIDE HF (anhydrous)

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UN No. 1052

Common Synonyms	Observable Characteristics	Manufacturers
HYDROFLUORIC ACID (anhydrous)	Colourless liquid or gas. Pungent, suffocating odour. Fumes in humid air.	Allied Chemical Ltd., Amherstburg, Ontario, Valleyfield, Quebec. Alcan Smelters and Chemicals, Jonquière, Quebec.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Corrosive liquid, poison. Inert Atmosphere: No requirement. Venting: Safety relief.	Label(s): White and black label - CORROSIVE. White label - POISON; Class 8, 2.3. Storage Temperature: Ambient. Hose Type: Haion TFE plastic hose, Hypalon.	Grades or Purity: Technical, 99.0%. Containers and Materials: Cylinders, tank cars; steel, nickel-steel alloys.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): Soluble in all proportions. Molecular Weight: 20.0 Vapour Pressure: 775 mm Hg (20°C); 1 200 mm Hg (33°C). Boiling Point: 19.5°C	Floatability (Water): Soluble. Odour: Britating, pungent (0.1 ppm, odour threshold) Flash Point: Not flammable. Vapour Density: 1.86 (25°C). Specific Gravity: 0.96 (liquid) (20°C).	Colour: Colouriess. Explosive Limits: Not flammable. Melting Point: -83.4°C.

#### HAZARD DATA

(15 min) (as F).

Symptoms: Inhalation: irritation of mucous membranes, pain in throat, difficulty breathing, headache, fatigue, shock, coma. Contact: skin - readily absorbed, extreme burns, blisters and damage to tissue; eyes - burning, irreparable damage. Ingestion: irritation and burning of lips, mouth and throat, intense thirst, coughing, difficulty breathing, nausea and vomiting, shock, convulsions, coma. Can be fatal by all routes.

Texicology: Highly toxic by inhalation, ingestion and skin absorption.

TLV* - (inhalation) 3 ppm; 2.5 mg/m³ (as F).

LC30 - inhalation: rat = 1 276 ppm/1 h

Short-term Inhalation Limits - 6 ppm; 5 mg/m³

LDsn - No information.

Delayed Toxicity - No information.

LDLo - Intraperitoneal: rat = 0.025 g/kg

Fire Extinguishing Agents: Not combustible; however, water may be used on fires involving HF. Water may be used to cool fire-exposed containers and knock down vapours.

Behaviour in Fire: Not combustible.

ignition Temperature: Not combustible.

Burning Rate: Not combustible.

## Reactivity

With Water: Soluble: forming hydrofluoric acid.

With Common Materials: Reacts violently with arsenic trioxide and phosphorus pentoxide.

Stability: Stable.

## Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life in very low concentrations. Fish toxicity: 60 ppm/tns/fish/lethal/freshwater; >300 ppm/48 h/shrimp/LC so/saitwater.

Land-Air: No information.

Food Chain Concentration Potential: No information.

## Special Hazards

CORROSIVE. POISON.

## Immediate Responses

Keep non-involved people away from spill area. Issue warning: "CORROSIVE, POISON". Call Fire Department. Contact manufacturer or supplier for advice. Avoid contact and inhalation. Stop or reduce discharge, if this can be done without risk. Stay upwind and if water spray is used to control vapours, dike runoff. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated acid suit are mandatory. Boots - high, neoprene (pants worn outside boots). Cloves - neoprene.

## Fire and Explosion

Not combustible. Water may be used to fight fires involving HF, and to cool fire-exposed containers and knock down vapours.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: remove contaminated clothing; wash eyes and skin with plenty of warm water for at least 15 minutes. Ingestions give milk or water to conscious victim. Do not induce vomiting. If vomiting occurs, give more water to further dilute the chemical. Keep patient warm and quiet. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

#### ENVIRONMENTAL PROTECTION MEASURES

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum
- equipment and place in appropriate containers.
- 5. Absorb residual liquid on natural or synthetic sorbents.
- 6. Contaminated soil can be treated with lime
- (calcium hydroxide) to render it neutral.
- 7. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

- I. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## HYDROGEN PEROXIDE H₂O₂ (aqueous)

#### IDENTIFICATION

UN No. 2015, >60% 2014. 8 to 60%

Common Synonyms	Observable Characterist	cs	Manulacturers		
PEROXIDE HYDROGEN DIOXIDE HYDROPEROXIDE	Clear, colourless liquid.	Slightly sharp odour.	No Canadian manufacturers. Canadian suppliers: B.C. Forest Products, Crofto Canadian Industries Ltd., Co Du Pont Canada, Maitland, C	n, B.C. rnwall, Ont.	Originating from: FMC Corp., USA E.I. Du Pont de Nemours, USA
Transportation and Stor	age Information				
Shipping State: Liquid Classification: Oxidize Intert Atmosphere: No Yenting: Safety relief Pump Type: Aluminum steel.	r and corrosive. requirements.	Group I. Black and v Class 8, Group I. Storage Temperatur	ylene, PVC (Tygon), 304 or	50, 70, 989 Containers	Purity: Many concentrations 27.5, 30, 8 as aqueous solutions. s and Materials: Drums (aluminum, ne); tank cars, tank trucks; aluminum.
Physical and Chemical	Characteristics				
Physical State (20°C, solution). Solubility (Water): Solui Molecular Weights 34.0 Vapour Pressure: 27.5% 23 mm Hg; 50%, 18 mm (30°C). Boiling Point: 27.5%, 1	ble in all proportions. (for solute only). , 25 mm Hg; 35%,	Floatability (Water) Odour: Slightly shar Flash Point: Not fla Vapour Density: 1.2 Specific Gravity: 1. 1.29 (70%) all at (20	p. mmable. 13 (35%); 1.20 (50%);	Melting Po	olouriess. Limits: Not fiammable. pint: (35%) -32.8°C; (50%) -50°C; °C; -0.41°C (pure).

#### HAZARD DATA

# Symptoms: Inhalation: sore throat, coughing, laboured breathing, headache, nausea, weakness. Ingestion: pain in swallowing, intense thirst, nausea and vomiting, difficulty breathing, convuisions, coma. Toxicology: Highly toxic by all routes. TLV* - (inhalation) 1 ppm; 1.5 mg/m³. Short-term Inhalation Limits - 2 ppm; 3 mg/m³ (15 min). LC50 - Inhalation: rat = 2 000 mg/m³/4 h (90%). LD50 - Skin: rat = 4.06 g/kg (90%). Delayed Toxicity - None known.

Fire Extinguishing Agents: Not combustible. In fires involving H₂O₂ use water; H₂O₂ may cause other firefighting agents to be ineffective. Use water to coof fire-exposed containers.

Behaviour in Fire: Can cause ignition of combustible material on contact. May decompose with explosive violence on contact with iron, copper, chromium, brass, branze, jead, sliver, manganese and their saits.

Ignition Temperature: Not combustible. Burning Rate: Not combustible.

#### Reactivity

With Water: No reaction. Soluble.

114°C; 70%, 125°C; 150.2°C (pure).

With Common Materials: Reacts violently with acetic acid, acetic anhydride, acetone, antimony and arsenic trisulfide, brass, bronze, t-butylaicohol, cellulose, charcoal, chlorosulfonic acid, chromium, copper, cupric sulfide, ethyl alcohol, ferrous sulfide, formic acid, organic matter, hydrazine, iron, lead, lead sulfide, magnesium, manganese, manganese dioxide, mercuric oxide, nitric acid, platinum, potassium permanganate, silver, sodium, and sodium lodate.

Stability: Stable, within the limits of the foregoing. Higher concentrations are less stable and more reactive.

## Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 40 ppm/tns/(fingerling trout)/toxic/saltwater; BOD: None. Land-Air: No information.

## Special Hazards

OXIDIZER, CORROSIVE, Reactive.

## Immediate Responses

Keep non-involved people away from spill site. Issue warning: "OXIDIZER, CORROSIVE". Call Fire Department. Contact manufacturer or supplier for advice and possible assistance. Avoid contact and inhalation. Stop or reduce discharge, if this can be done without risk. Contain spill by diking with earth. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus. Boots - neoprene, or rubber (pants worn outside boots). Gloves - rubber, neoprene or PVC. Flammable clothing, such as cotton, rayon, leather or wool, should not be worn when in contact with high-strength hydrogen peroxide.

#### Fire and Explosion

Not combustible. In fires involving H2O2, use water; H2O2 may cause other firefighting agents to be ineffective. Use water to cool fire-exposed containers.

## First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. <u>Contact</u>: eyes - irrigate with plenty of water; skin - remove contaminated clothing and flush affected areas with plenty of water. <u>Ingestion</u>: give water to conscious victim to drink. <u>Inhalation</u>: remove at once from further exposure. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain split by diking with earth.
- 4. If concentrated, dilute with at least an equal volume of water.
- 5. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 6. Recover undamaged containers.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# HYDROGEN SULFIDE H₂S

	ATION

UN No. 1053

Common Synonyms	Observable Characteristics	Manufacturers
SULPHURETTED HYDROGEN HYDROGEN SULPHIDE SULFUR HYDRIDE	Colourless gas. Offensive rotten-egg odour.	Thio-Pet Chemicals Limited, Fort Saskatchewan, Alta. Laurentide Chemicals & Sulphur Ltd., Montreal East, Que.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Fiammable, poisonous gas. Inert Atmosphere: No requirement. Venting: Safety relief.	Label(s): Red label - FLAMMABLE GAS; Class 3. White label - POISON; Class 2.3. Storage Temperature: Ambient.	Grades or Purity: Technical, 98.5%; H ₂ S present in many concentrations from gas wells.  Containers and Materials: Cylinders, tank cars; steel.
Physical and Chemical Characteristics		
Physical state (20°C, 1 atm): Gas. Solubility (Water): 0.5 g/100 mL (20°C). Molecular Weight: 34.1 Vapour Pressure: 7 600 mm Hg (-0.4°C); 15 200 mm Hg (25.5°C). Boiling Point: -60.2°C.	Floatability (Water): Liquid floats, boils and dissolves. Odour: Retten-eggs (0.005 ppm, odour threshold). Flash Point: <-50°C. Vapour Density: 1.2 (0°C). Specific Gravity: 0.95 (-60°C).	Colour: Colourless.  Explosive Limits: 4.0 to 44%.  Melting Point: -83 to -86°C.

# HAZARD DATA

Human Health		
Symptoms: Insidious poison, since sense of smell mathroat and eyes, sneezing, headache, dizziness, nau and death. Contact: skin - kritation, painful inflam Toxicology: Highly toxic by Inhalation and contact. TLV* - (inhalation) 10 ppm; 14 mg/m ³ .	sea and vomiting, pale complexion, cold sweat, imation; eyes - kritation, watering, inflammati	diarrhea, muscular weakness, drowsiness, unconsciousness
Short-term Inhalation Limits - 15 ppm; 21 mg/m ³ (15 min).	LC50 - Inhalation: rat = 444 ppm. Delayed Toxicity - No information.	40 July 1.10 Internation
Fire		
Fire Extinguishing Agents: Stop flow of gas. Use w Behaviour in Fire: Flashback may occur along vapor ignition Temperature: 260°C.  Reactivity	arer to keep Irre-exposed containers cool.  Ir trail. Emits fumes of SO _X when heated to de  Burning Rate: (Liquid) 2.3 mm/min.	composition.
With Water: No reaction: soluble.	oxidizing materials. Reacts violently with chic chloride, nitrogen trifluoride.	rine monoxide, chromic anhydride, copper, fluorine,
Environment	A CONTRACTOR OF THE PROPERTY O	
Water: Prevent entry into water intakes and waters freshwater; 10 mg/L/96 h/goldfish/LC100/freshwat Land-Air: No information.	ways. Harmful to aquatic life in low concentration; 0.86 ppm/24 h/trout/LC100/freshwater; BO	ions. Fish toxicity: 1.38 ppm/48 h/fathead minnow/TLm/ D: Not available.

Food Chain Concentration Potential: None.

## Special Hazards

FLAMMABLE. POISON. Odour sensation lost with high concentrations or long period of exposure.

## Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE, POISON". CALL FIRE DEPARTMENT. Call manufacturer for guidance and assistance. Eliminate all ignition sources. If necessary, evacuate people downwind. Work from upwind. Use water spray to control vapours and to cool containers. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self contained breathing apparatus and totally encapsulated suit. Gloves - rubber. Boots - rubber.

## Fire and Explosion

Stop flow of gas, Use water to cool fire-exposed containers. Flash back may occur along vapour trail. Emits toxic SOx fumes when heated to decomposition.

#### First Aid

Move victim out of spill site to fresh air. Call medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Keep patient warm. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water-Land-Air

Because of the properties of hydrogen sulfide, its shipment in refrigerated, pressurized tank cars with specialized valve and relief devices, it is inadvisable for an inexperienced person to try to control an emergency situation.

1. Contact the manufacturer, who will suggest appropriate action or may be able to arrange for on-the-site assistance. Do not attempt to control a leak or deal with an emergency situation without this guidance.

2. Response possibilities include: allowing dispersal into the atmosphere.

3. Vapours can be controlled with log or water spray.

4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## ISOPROPYL ALCOHOL CH3CHOHCH3

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UN No. 1219

Common Synonyms	Observable Characteristics	Manufacturers
ISOPROPANOL DIMETHYLCARBINOL 2-PROPANOL sec-PROPYL ALCOHOL RUBBING ALCOHOL	Clear, colourless liquid. Alcohol odour.	Gulf Oil Canada Ltd., Montreal, Quebec. Shell Canada Ltd., Corunna, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Open (flame arrester) or pressure-vacuum. Pump Type: Gear, centrifugal, flammable liquid types.	Label(s): Red label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Polyethylene, polypropylene, butyl, Hypalon, natural rubber.	Grades or Purity: Commercial, technical, 91%, 95%, 99%. Containers and Materials: Drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Soluble in all proportions. Molecular Weight: 60.1 Vapour Pressure: 32 mm Hg (20°C); 44 mm Hg (25°C); 57 mm Hg (30°C). Boiling Point: 82.4°C.	Floatability (Water): Floats and mixes. Odour: Alcohol-like (40 ppm, odour threshold). Flash Point: 12°C (c.c.); 14°C (c.c. 88% solution). Vapour Density: 2.1 Specific Gravity: 0.79 (20°C).	Colour: Colouriess. Explosive Limits: 2.3% to 12.7%. Melting Point: -86 to -89°C.

## HAZARD DATA

# Human Health

Symptoms: Contact: eyes - severely irritating; skin - defatting and cracking. Inhalation: causes depression of the central nervous system, nausea, vomiting, depressed respiration, abdominal pain. Ingestion: dizzlness, headache, nausea, vomiting. Toxicology: Low toxicity by contact and moderate toxicity by Ingestion.

TLV*- (inhalation, skin) 400 ppm; 980 mg/m3.

LCsp - No Information.

Delayed Toxicity - No information.

LD 50 - Oral: rat = 5.84 g/kg

Short-term inhalation Limits - 500 ppm, 1 225 mg/m³ (15 min).

## Fire

Fire Extinguishing Agents: For fires use dry chemical, carbon dioxide or alcohol foam. Water may be ineffective but may be used to cool fire-exposed

Behaviour in Fire: Flashback may occur along vapour trail.

Ignition Temperature: 399°C.

Burning Rate: 2.3 mm/min.

With Water: No reaction; soluble.

With Common Materials: Can react vigorously with oxidizing materials. Can react violently with oleum, phosgene and potassium-t-butoxide. Stability: Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: 900 to 1 100 ppm/24 h/chub/critical/freshwater; 1 400 mg/ L/48 h/brine shrimp/LC50/saltwater; Aquatic toxicity rating = 100 to 1 000 ppm/96 h/TLm/freshwater; BOD: 129 to 159%; 5 days.

Land-Air: No information.

Food Chain Concentration Potential: None.

## Special Hezards

FLAMMABLE.

## immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all ignition sources. Notify manufacturer or supplier. Dike to contain spill and prevent runoff. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus (in fire). Gloves - rubber or plastic. Outer clothing - suitable for situation, coveralls, etc. Boots -

## Fire and Explosion

Use dry chemical, carbon dioxide or alcohol foam. Water may be ineffective on fire, may be used to cool fire-exposed containers. Flash back may occur along vapour trail.

## First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: remove contaminated clothing; wash eyes and skin with plenty of water. Inhalation: give artificial respiration if breathing has stopped; oxygen, if breathing is laboured. Ingestion: give water to conscious victim and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge vacuum pump to remove contaminants liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Dike to prevent runoff from rainwater or water application.
- 5. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 6. Recover undamaged containers.
- 7. Absorb residual liquid on natural or
- synthetic sorbents.
- 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

DENTIFICATION

UN No. 1223

Manufacturers

Common Synonyms STOVE (RANGE) OIL JET FUEL: JP-1 ILLUMINATING OIL KEROSINE RANGE OIL FUEL OIL NO. 1

COAL OIL

Observable Characteristics

Colourless or pale yellow liquid. Gasoline-

like odour.

Universally available.

Transportation and Shipping Information

Shipping State: Liquid. Classification: Flammable liquid. inert Atmosphere: No requirement. Yenting: Open (flame arrester). Pump Type: Gear or centrifugal.

Labelish: Red label - FLAMMABLE LIQUID: Class 3.3. Group II. Storage Temperature: Ambient. Hose Type: Neoprene, polyethylene, Viton.

Grades or Purity: Mixture of light hydrocarbon distillates. Containers and Materials: Drums, tank cars, tank trucks: steel.

Physical and Chemical Characteristics

Physical State (20°C, I atm): Liquid. Solubility (Water): 0.0002 to 0.0004 g/100 mL (20°C).

Molecular Weight: Variable. Vanour Pressure: "5 mm Hg (20°C). Bolling Point: 150 to 300°C.

Floatability (Water): Floats.

Odour: Gasoline-like (odour threshold about 1 ppm). Flash Point: 43 to 72°C.

**Yanour Density: 4.5** 

Specific Gravity: 0.8 to 0.85 (20°C).

Colour: Colourless or pale yellow. Explosive Limits: 0.7 to 5%. Melting Point: -43 to -49°C.

LD₅₀ - Orais guinea pig = 20 g/kg

HAZARD DATA

Human Health

Symptoms: Inhalation: nausea, vomiting, coughing, headache. Ingestion: nausea, vomiting, weakness, dizziness, slow and shallow respiration, convulsions, unconsciousness. Contact: skin - irritating and defatting; eyes - irritation.

Toxicology: Very low toxicity by all routes.

TLV - No information. Short-term Inhaiation Limits - No Information. LC₅₀ - No information.

Delayed Toxicity - No information.

Fire

Fire Extinguishing Agents: Use dry chemical, foam, carbon dioxide. Water may be ineffective but may be used to cool fire-exposed containers.

Behaviour in Fire: No information.

legition Temperature: 210°C.

Burning Rate: 4 mm/min.

Reactivity

With Water: No reaction.

With Common Materials: Can react with strong oxidizing agents.

Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 2 990 ppm/24 h/bluegill/freshwater; 100-1 000 ppm/96 h/rainbow trout/LCso/freshwater;

BOD: 53%, 5 days.

Land-Air: No Information.

Food Chain Concentration Potential: None.

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## Special Hazards

#### FLAMMABLE.

## immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE", CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Notify supplier or manufacturer. Stop or reduce discharge, if this can be done without risk. Dike spill area to contain runoff. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - in fires or closed spaces, use self-contained breathing apparatus; otherwise, Goggles or face shield. Coveralls. Rubber gloves. Rubber boots (high) - wear pants outside boots.

## Fire and Explosion

Use dry chemical, foam or carbon dioxide to extinguish. Water may be used to cool fire-exposed containers.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped, give oxygen if breathing is laboured. Contact: eyes - irrigate with plenty of water for at least 15 minutes; skin - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: give conscious victim water to drink, do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or

synthetic sorbents.

7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# LATEX

## IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers	
ARCONITRILE BUTADIENE RUBBER PLASTIC LATEX LATEX, LIQUID SYNTHETIC SYNTHETIC RUBBER LATEX NATURAL RUBBER LATEX	White liquid. Variable and characteristic odour.	Universally available.	
Transportation and Shipping Information			
Shipping State: Liquid. Classification: None. Inert Atmosphere: No requirement. Venting: Open. Pump Type: No information.	Label(s): Not regulated. Storage Temperature: Ambient. Hose Type: No information.	Grades or Purity: Regular and concentrated.  Containers and Materials: Drums, tank cars, tank trucks; steel.	
Physical and Chemical Characteristics			
Physical State (20°C, I atm): Liquid. Solubility (Water): Insoluble; but may mechanically mix. Molecular Weight: Variable. Vapour Pressure: No information. Boiling Point: >200°C.	Floatability (Water): Sinks in freshwater; floats in saltwater (may mechanically mix).  Odour: Variable and characteristic.  Flash Point: Not flammable.  Vapour Density: No information.  Specific Gravity: 1.06 (25°C).	Colour: White. Explosive Limits: Not flammable. Melting Point: >150°C.	

Human Health  Symptoms: Inhalation: irritation of mucous members - irritation.	oranes. Ingestion: dangerous to ingest, as coag	gulation may occur internally. <u>Contact</u> : skin - irritation,
Toxicology: Low toxicity by all routes. TLV® No information. Short-term inhalation Limits - No information.	LC ₅₀ - No information. Delayed Toxicity - No information.	LD56 - No information.
Fixe		
Fire Extinguishing Agents: Not flammable. Most i Behaviour in Fire: Splatters when burning. Ignition Temperature: No information.	lire extinguishing agents may be used on fires  Burning Rate: No information.	involving latex.
Reactivity		
With Water: No reaction. With Common Materials: No information. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No information.	•	

# 305

#### EMERGENCY MEASURES

## Special Hazards

## Immediate Responses

Keep non-involved people away from spill site. Call Fire Department. Stop or reduce discharge if safe to do so. Notify manufacturer. Notify environmental authorities.

## Protective Clothing and Equipment

In lires Respiratory protection - self-contained breathing apparatus; otherwise protective clothing as required.

## Fire and Explosion

Not flammable. Most fire extinguishing agents may be used on fires involving latex. Splatters when burning.

#### First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and wash affected areas with plenty of water; eyes - irrigate with water. Ingestion: give water to conscious victim to drink; get prompt medical assistance. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

## Water Floating

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Sinking

- I. Stop or reduce discharge if safe to do.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).
- 4. May be dumped in a municipal landfill (approval of environmental authorities required).

# LEAD ACETATE Pb (CH3COO)2-3H2O

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UN No. 1616

Common Synonyms	Observable Characteristics	Manufacturers
LEAD ACETATE TRIHYDRATE SUGAR OF LEAD PLUMBOUS ACETATE	White to brown or grey powder or flakes. Odourless.	Anachemia Chemicals Ltd., Montreal, Que., Toronto, Ont.
Transportation and Storage Information		
Shipping State: Solid. Classification: Poison. Inert Atmosphere: No requirement. Venting: Open.	Label(s): White label - POISON; Class 6.1, Group III. Storage Temperature: Ambient.	Grades or Purity: Technical, 97%. Containers and Materials: Multiwall paper bags, drums.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 45.6 g/100 mL (15°C); 200 g/100 mL (100°C). Molecular Weight: 379.3 Vapour Pressure: No information. Boiling Point: Decomposes at 200°C.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 2.55 (20°C).	Colour: White to brown or grey.  Explosive Limits: Not flammable.  Melting Point: Loses water at 75°C.

constriction of throat, stomach pains, nausea, ve	ritation of nose and eyes, headache, stomach cramps and omiting, diarrhea, convulsions, coma. <u>Contact</u> : skin - irri	fatigue. Ingestion: metallic taste in mouth, lation, inflammation; eyes - inflammation.
Toxicology: High toxic by ingestion. TLV ² (inhalation) 0.15 mg/m ³ (as Pb). Short-term inhalation Limits - 0.45 mg/m ³ (15 min) (as Pb).	LC ₅₀ - No information.  Delayed Toxicity - Cumulative polson. Suspected carcinogen.	LD50 - intraperitoneal: guinea pig = 0.22 g/kg TDL0 - Oral: rat = 8.52 g/kg
Fire		
Fire Extinguishing Agents: Most fire extinguishi Behaviour in Fire: In fires, toxic PbO fumes are Ignition Temperature: >280°C.	ng agents may be used. Use water sparingly. rejeased. Burning Rate: No information.	-
Reactivity		
With Water: No reaction; soluble. With Common Materials: Reacts violently with Stability: Stable.	KBrO3 (potassium bromate).	
Branthell : Mante.		
Environment		

## Special Hazards

POISON.

## Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Avoid contact and inhalation. Dike to prevent runoff from rainwater or water application. Lightly wet down dry spillage to prevent wind drift of dust. Notify manufacturer or supplier. Notify environmental authorities.

## Protective Clothing and Equipment

In fire or enclosed spaces, Respiratory protection - self-contained breathing apparatus. Otherwise, dust respirators (with suitable filters) or metal fume respirators for normal situations. Rubber gloves and boots will prevent contact. Frequent changes of clothing and footwear should be provided for contaminated clothing, and footwear should be washed immediately after contact.

## Fire and Explosion

Most fire extinguishing agents may be used. Use water sparingly. In fires, toxic PbO fumes are released.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim to drink, and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contained contaminants and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or
- water application. 4. Remove material by manual or mechanical
- means.

  5. Broken and empty bags or containers should
- be handled carefully to avoid scattering of dust.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 302

## LEAD CHROMATE PECTO

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manulacturers
CHROME YELLOW LEAD CHROMATE (IV) PARIS YELLOW	Yellow crystals. Odourless.	Hercules Canada, St. Jean, Quebec. Reed, Ajax, Ontario.
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Closed.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Technical.  Containers and Materials: Kegs, drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.000 007 to 0.000 017 g/100 ml. (20°C). Molecular Weight: 323.2 Vapour Pressure: No information. Boiling Point: Decomposes at 844°C.	Flotability (Water): Sinks. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 6.1	Colour: Yellow. Explosive Limits: Not flammable. Melting Point: 844°C; decomposes.

Human Health		
vomiting, diarrhea, convulsions, coma; Contact:	s, headache, stomach cramps, fatigue. <u>Ingestion</u> skin – irritation and inflammation; eyes – waterin	metallic taste in mouth, stomach pains, nausea and . g, irritation and inflammation.
Toxicology: Highly toxic by ingestion. TLV ^Q (inhalation) 0.05 mg/m ³ (as Cr). Short-term Inhalation Limits - No information.	LC50 - No information. Delayed Toxicity - Suspected carcinogen.	LD ₅₀ - Intraperitoneal: guinea pig = 0.40 g/kg
Fire		
Fire Extinguishing Agents: Not combustible; mos Behaviour in Fire: Decomposes at 844°C, liberati Ignition Temperature: Not combustible. Reactivity	ing oxygen and toxic lead compounds.  Burning Rate: Not combustible.	HANTANG 1020 CIO OLIMANA
With Water: No reaction. With Common Materials: May react with ferric f Stability: Stable.	errocyanide.	
Environment .		
Water: Prevent entry into water intakes or water Land-Air: No information. Food Chain Concentration Potential: May be con		

## Special Hazards

POISON.

## Immediate Responses

Keep non-involved people away from spill site. Avoid contact and inhalation of tumes. Stop or reduce discharge if this can be done without risk. Notify supplier and environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - in confined spaces or fires - self-contained breathing apparatus and totally encapsulated suit, otherwise protective clothing as required. Gloves: rubber or plastic.

## Fire and Explosion

Not combustible, most fire extinguishing agents may be used on fires involving lead chromate. At high temperatures lead chromate decomposes yielding oxygen and toxic lead compounds.

## First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; if laboured, give oxygen. Contact: skin and eyes - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: give conscious victim plenty of water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

## Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other material.
- \$. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Dienos

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# LEAD NITRATE Ph(NO₃)₂

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UN No. 1469

Common Synonyms	Observable Characteristics	Manufacturers
NITRIC ACID, LEAD II SALT	White crystals. Odourless.	No Canadian manufacturers.
Transportation and Storage Information		
Shipping State: Solid. Classification: Oxidizing material. Poison. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Yellow label - OXIDIZER; Class 5.1, Group II; White label - POISON; Class 6.1, Group II. Storage Temperature: Ambient.	Grades or Purity: Technical, 98+%. Containers and Materials: Multiwall paper bags and drums.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm.): Solid. Solubility (Water): 37.7 g/100 mL (0°C); 127 g/100 mL (100°C). Molecular Weight: 331.2 Vapour Pressure: No information. Boiling Point: Decomposes at 470°C.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 4.53 (30°C).	Colour: White. Explosive Limits: Not flammable. Melsing Point: Decomposes at 470°C.

ENGLISHED INTELL	
ritation of nose and eyes, headache, stomach cramp omiting, diarrhea, convulsions, coma. <u>Contact</u> : skin LC50 - No information. Delayed Toxicity - Cumulative poison.	
st fire extinguishing agents may be used. Use water d to decomposition, can emit toxic nitrogen oxide a Burning Rate: Not flammable.	
nite organic materials. Reacts violently with ammo	onium thiocyanate, carbon and lead hypophosphite.
terways. Fish toxicity: 240 ppm/48 h/mosquito fish errestrial animals are capable of concentrating lead	
	ritation of nose and eyes, headache, stomach crampomiting, diarrhea, convulsions, coma. Contact: skin  LC50 - No information. Delayed Toxicity - Cumulative poison.  st fire extinguishing agents may be used. Use water d to decomposition, can emit toxic nitrogen oxide a Burning Rate: Not flammable.  enite organic materials. Reacts violently with ammoditerways. Fish toxicity: 240 ppm/48 h/mosquito fish

## Special Hazards

OXIDIZER. POISON.

## immediate Responses

Keep non-involved people away from spill site. Issue warning: "OXIDIZER. POISON". Avoid contact and inhalation. Call Fire Department. Dike to prevent runoff from rainwater or water application. Lightly wet down dry spillage to prevent wind drift or dust. Notify manufacturer or supplier. Notify environmental authorities.

## Protective Clothing and Equipment

In fire or enclosed spaces, Respiratory protection - self-contained breathing apparatus. Otherwise, dust respirators (with suitable filters) or metal fume respirators for normal situations. Gloves and boots - rubber. Frequent changes of clothing and footwear should be provided. Contaminated clothing and footwear should be washed immediately after contact.

## Fire and Explosion

Not combustible. Most fire extinguishing agents may be used. Use water sparingly.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim to drink, and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

## ENVIRONMENTAL PROTECTION MEASURES

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Broken and empty bags or containers should be handled carefully to avoid scattering of dust.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# LEAD OXIDES

(black) PboO (red) PbyOb (vellow) PbO

			IDENTIFICATION		
Common Synonyms			Observable Characteristics	Manufacturers	
(Yellow) PbO (Red) Pb3O4 Lead Oxide, Yellow Plumbous Oxide Minium Lead Monoxide Plumboplumic Oxide Lead Oxide, Red Lead Tetroxide		(Black) Pb2O Lead Oxide, Black Litharge, Leaded	Yellow, red or black powder. Odourless.	Canada Metal Co. Ltd., Toronto, Ont., Winnipeg, Man. Carter White Lead Co. of Canada Ltd Montreal, Que. Metalex Products, Vancouver, B.C.	
Transportation and Storage Information Shipping State: Solid. Classification: Not regulated. Inert Atmosphere: No requirement. Venting: Open.		Label(s): Not reg label - POISON. Storage Tempera	ulated. Voluntary white Iture: Ambient.	Grades or Purity: Yeilow (litharge) - assay 98% and battery grades, 50 to 95% PbO. Red (red lead) - 85,95,97,98% PbO. Black - usually a mixture of Pb and PbO.  Containers and Materials: Multiwall paper bags, cans, drums.	
Physical and Chemica	l Characteristics				
Physical State (20°C, Solubility (Water); Ye (20°C); black insolubio		Ficatability (Wat Odour: Odourless Flash Point: Not	5.	Colour: Yellow, red, black. Explosive Limits: Not flammable. Melting Point: Yellow, \$88°C; red decomposes	

(25°C). Molecular Weight: Yellow, 223.2; red, 685.6; black, 430.4

Vapour Pressure: No information.

Boiling Point: Yellow. >886°C; red decomposes 500°C: black, decomposes upon heating.

Flash Point: Not flammable. Vapour Density: No information.

Specific Gravity: Yellow, 9.2 to 9.7; red, 8.8 to 9.2; black, 8.3 (20°C).

Melting Point: Yellow, \$88°C: red decomposes 500 to 530°C; black decomposes with heat.

## **HAZARD DATA**

Symptoms: Inhalation: of dust, mist or fumes; irritation of nose and eyes, headache, stomach cramps and fatigue. Ingestion: metallic taste in mouth. constriction of throat, stomach pains, vomiting, diarchea, convulsions and coma. Contact: skin - irritation, inflammation; eyes - Inflammation.

Toxicology: Highly toxic by ingestion. TLV* - (inhalation) 0.15 mg/m3 (as Pb). Short-term Inhalation Limits - 0.45 mg/m³

LCsa - No Information.

Delayed Toxicity - Cumulative poison.

LD50 - Intraperitoneal: guinea pig = 0.22 g/kg LDLo - Intraperitoneal: rat = 0.43 g/kg

Fire

(15 min).

Fire Extinguishing Agents: Not combustible. Most fire extinguishing may be used on fires involving lead oxides. Behaviour in Fire: Toxic Pb and PbO and fumes may be released in fires or when heated to decomposition.

ignition Temperature: Not combustible.

Burning Rate: Not combustible.

Reactivity

Human Health

With Water: No reaction.

With Common Materials: In general, Pb oxides are oxidizing agents. Reacts violently with aluminum, hydrogen trisulfide, sodium, sulfur trioxide, and titanium. Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: (Yellow) > 56 000 ppm/96 h/mosquito fish/TLm/turbid water.

Land-Air: No information.

Food Chain Concentration Potential: Fish and terrestrial animals are capable of concentrating lead.

# Special Hazards

POISON. OXIDIZER.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Notify manufacturer or supplier. Avoid contact or inhalation. Dike to prevent runoff from rainwater or water application. Lightly wet down dry spillage to prevent wind drift of dust. Notify environmental authorities.

# Protective Clothing and Equipment

In fires or enclosed spaces, Respiratory protection - self-contained breathing apparatus, otherwise dust respirators (with suitable filters) for protection against dust under normal conditions. Rubber gloves and boots will prevent contact. Frequent changes of clothing should be provided. Contaminated clothing and footwear to be washed immediately after contact.

#### Fire and Explosion

Not combustible, most fire extinguishing agents may be used.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected areas with plenty of water; eyes irrigate with plenty of water. Ingestion: give water to conscious victim to drink, and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

# ENVIRONMENTAL PROTECTION MEASURES

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoif from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Broken paper bags should be handled carefully to avoid scattering of dust.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 314

# MAGNESIUM HYDROXIDE Mg(OH)2

# IDENTIFICATION

Common Synonyms	Observable Charac	teristics	Manufacturers	
MAGNESIUM HYDRATE MAGNESIUM MAGNA MILK OF MAGNESIA	White powder. Odd	ourless.	No Canadian manufacturers. Canadian suppliers: Dow Chemical Canada Inc., Sarnia, Ontario. Chorney Chemical, Toronto, Ontario. Frank E. Dempsey, Toronto, Ontario.	Originating from: Dow Chemical, Luddington, Mich. Barcroft, USA. Rehels Chemical, USA.
Transportation and Storage	Information			
Shipping State: Solid. Classification: None. Inert Atmosphere: No requir Venting: Open.	ement.		ot regulated. mperature: Ambient.	Grades or Purity: Technical.  Containers and Materials: Multiwall paper bags, drums.
Physical and Chemical Char-	acteristics			
Physical State (20°C, 1 atm Solubility (Water): 0.18 g/10 0.04 g/100 ml. (100°C). Molecular Weight: 58.3 Vapour Pressure: No informa Boiling Point: Loses water a	6 mL (25°C); ation.	Odour: Odo Flash Point Vapour Der	y (Water): Sinks. puriess. t: Not flammable. nsity: No information. ravity: 2.36 (20°C).	Colour: White. Explosive Limits: Not flammable. Melting Point: Loses water at 350°C.

Human Health			
Symptoms: Contact: skin and eyes - irritation and Taxicology: Low toxicity by all routes. TLV® - No information. Short-term inhalation Limits - No information.	redness. <u>Ingestion</u> : abdominal pain and diarring LC50 - No information.  Delayed Toxicity - No information.	LD ₅₀ - No information.	
Fire			
Fire Extinguishing Agents: Not combustible. Most Behaviour in Fire: Upon heating, decomposes to m ignition Temperature: Not combustible. Reactivity	fire extinguishing agents may be used on fire agnesium oxide (MgO) and water vapour.  Burning Rate: Not combustible.	es involving magnesium hydroxide.	· · · · · · · · · · · · · · · · · · ·
With Water: No reaction. With Common Materials: Reacts violently with ma Stability: Stable.	sleic anhydride and phosporus. Reacts with a	icids.	
Environment			
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No information			

# Special Hazards

#### Immediate Responses

Keep non-involved people away from spill site. Notify manufacturer. Dike to prevent runoff. Notify environmental authorities.

# Protective Clothing and Equipment

In fires or confined spaces, Respiratory protection - use self-contained breathing apparatus. Otherwise, protective clothing and equipment as required.

#### Fire and Explosion

Not combustible. Most fire extinguishing agents may be used.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: skin - remove contaminated clothing and wash affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give plenty of water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or
- other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# MALATHION (CH3O)2P(S)SCH(COOC2H5)CH2COOC2H5

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UN No. 2783

Danger Group According to Percentage of Active Substance

Group III

liquid 30 to 100%

Common Synonyms	Observable Characteristics	Manufacturers
S-1,2-BIS (ETHOXYCARBONYL)ETHYL O,O-DIMETHYL PHOSPHORODITHIOATE Common Trade Names CALMATHION, CELTHION, CYTHION, MALATIOZ (A commonly-used general purpose insecticide.)	Yellow to dark brown liquid or white to brown powder. Skunk-like odour.	Canadian industries Ltd., (C.I.L.), Montreal, Quebec. Chipman Chemicals Ltd., Stoney Creek, Ont. Cyanamid Canada, Willowdale, Ont.
Transportation and Storage Information		
Shipping State: Liquid or solid (wettable powder). Classification: None. Inert Atmosphere: No requirement. Venting: Closed. Pump Type: No information.	Label(s): Not regulated. Storage Temperature: Ambient. Hose Type: No information.	Grades or Purity: Various purities as detailed below.  Containers and Materials: Glass bottles, cans, drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid (technical). Solubility (Water): 0.0145 g/100 mL (technical). Molecular Weight: 330 (technical). Vapour Pressure: 0.00004 mm Hg (30°C). Boiling Point: 156°C (at 0.7 mm Hg pressure).	Floatability (Water): Technical sinks, SN floats. Odour: Skunk-like. Flash Point: 54°C (c.c.) SN; 163°C (c.c.) technical; 22 to 32°C (c.c.) EC; PP is also flammable. Vapour Density: No information. Specific Gravity: L23, 25°C (technical).	Colour: Liquid, yellow to dark brown; powder, white to brown.  Explosive Limits: Technical not flammable; SN, PP and EC are flammable.  Melting Point: 2.9°C (technical).

Human Health		
Symptoms: Inhalation: headache, tightness in ches abdominal cramps, nausea, vomiting, diarrhea.  Texicology: Moderately by ingestion and contact.  TLV - (skin) 10 mg/m ³ Short-term inhalation Limits - No information.	t, wheezing, throat constriction. <u>Contact</u> : eyes - Irri LC ₅₀ - No information. Delayed Toxicity - No information.	tation; skin - irritation; may be absorbed. <u>Ingestion</u> :  LD50 - Oral: rat = 0.885 g/kg  LD50 - Oral: man = 0.857 g/kg
Fire		
Fire Extinguishing Agents: Most fire extinguishing containers.  Behaviour in Fire: In fires or when heated to decoming the imperature: Variable.	g agents may be used on fires involving malathion. sposition releases toxic $SO_X$ , $PO_X$ and other fumes. Burning Rate: No information.	Water spray should be used to cool fire-exposed
Reactivity		
With Water: No reaction; only WC insoluble. With Common Materials: May react with oxidizing Stability: Stable.	agents.	
Environment		
freshwater: 0.12 pom/24 h/bluegili/LCso/freshwate	ways. Fish toxicity: 16 ppm/96 h/fathead minnow/LC er; 0.33 to 1 ppm/48 h/brown shrimp/LC50/saltwater; 0 000 ppm; LD50/exposed for 5 days/Oral: pheasant/2 5 m.	.032/24 h/grass shrimp/LC50/saltwater.

# Special Hazards

POISON.

#### immediate Responses

Keep non-involved people away from spill site. Stop or reduce discharge if safe to do so. Notify manufacturer or supplier. Dike to contain material or water runoff. Notify environmental authorities.

# Protective Clothing and Equipment

In fires or confined spaces - Respiratory Protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, approved pesticide respirator and impervious outer clothing.

#### Fire and Explosion

Use carbon dioxide, foam or dry chemical to extinguish. Releases toxic fumes in fires.

#### First Aid

Move victim out of spill site to Iresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim to drink and induce vomiting; in the case of petroleum distillates, do not induce vomiting for fear of aspiration and chemical pneumonia. If medical assistance is not immediately available, transport victim to hospital, doctor, or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

1. Stop or reduce discharge if safe to do so. 2. Contact manufacturer or supplier for advice.

3. If possible contain discharge by booming.

4. If floating, skim and removė.

# Sinks or mixes

3. If possible contain discharge by damming or water diversions. 4. Dredge or vacuum pump to remove

contaminants, liquids and contaminated bottom sediments.

5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier. 4. If liquid, remove material with pumps or vacuum
- equipment and place in appropriate containers.
- 5. If solid, remove material by manual or mechanical means.
- Recover undamaged containers.
- 7. Absorb residual liquid on natural or synthetic sorbents.
- 8. Remove contaminated soil for disposal.
- 9. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

#### Disposal

- 1. Contact manufacturer or supplier for advice on disposai.
- 2. Contact environmental authorities for advice on disposal.

# **Available Pormulations**

Technical Grade: Purity: typically 95% malathion.

Properties: combustible.

# Formulations:

PP - pressurized product

Purity:

- typically 4% malathion, remainder

inerts (e.g. clay, talc). - typically 3% malathion

- typically 20% malathion - typically 3% malathion, large

percentage of petroleum distillates

Properties:

- dispersible in water, low combustibility

- combustible and explosive
- combustible
- combustible, may be flammable;
- floats on water

Type: DU - dust or wettable powder

EC - emulsifiable concentrate SN - solution

# MALEIC ANHYDRIDE (CHCO)20

IDENTIFICATION

UN No. 2215

Common Synonyms	Observable Characteristics	Manufacturers
TOXILIC ANHYDRIDE cis-BUTENEDIOIC ANHYDRIDE 2,5-FURANDIONE MALEIC ACID, ANHYDRIDE	Colouriess to white, crystals or solid. Choking acrid edour.	Monsanto Canada, Montreal, Quebec. BASF Canada, Cornwall, Ontatio.
Transportation and Storage Information		
Shipping State: Solid. Classification: Corrosive. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Black and white label: CORROSIVE; Class 8, Group III. Storage Temperature: Ambient.	Grades or Purity: Commercial; 99.5%. Containers and Materials: Drums; fibre and steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): Reacts "16 g/100 mL (30°C). Molecular Weight: 98.1 Vapour Pressure: 0.00005 mm Hg (20°C); 0.0002 mm Hg (30°C). Boiling Point: 200 to 202°C.	Floatability (Water): Reacts, sinks or floats depending on material. Odour: Choking, acrid (0.3 to 0.5 ppm). Flash Point: 102°C (c.c.) Vapour Density: 3.4 Specific Gravity: 1.3 (60°C); 0.93 (20°C).	Colour: Colourless to white. Explosive Limits: 1.4 to 7.1%. Melting Point: 53°C.

Human Health		
Symptoms: <u>inhalation</u> : irritation of throat and resp severe burning.  Toxicology: Highly toxic upon contact with skin an TLV 9.25 ppm, I mg/m ³ .  Short-term inhalation Limits - No information.		nal pain, vomiting, nausea. <u>Contact</u> : skin - serious burns; eyes - LD50 - Oralirat = 0.48 g/kg
Fire		<del></del>
Fire Extinguishing Agents: Use alcohol foam or ca Behaviour in Fire: No information. Ignition Temperature: 477°C.	rbon dioxide. Water or foam may cause from Burning Rate: 1.4 mm/min.	hing.
Reactivity		
With Water: Reacts, forming maleic acid. With Common Materials: Can react with oxiding materials. Reacts violently with alkaline metals, amines, calcium hydroxide, lithium, potassium, potassium hydroxide, pyridine, sodium and sodium hydroxide. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water freshwater; BOD: 40 to 60%, 5 days. Land-Air: No information. Food Chain Concentration Potential: No informati		TLm/freshwater; 240 mg/L/24 to 96 h/mosquito fish/TLm/

#### Special Hazards

CORROSIVE. Combustible.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning "CORROSIVE". Call Fire Department. Notify manufacturer or supplier. Notify environmental authorities.

#### Protective Clothing and Equipment

In lires or confined spaces, Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, protective clothing as required.

#### Fire and Explosion

Use alcohol foam or carbon dioxide to extinguish. Water or foam may cause frothing.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped (not mouth-to-mouth method); oxygen if breathing is laboured. Contact: skin - remove contaminated clothing and flush affected areas with pienty of water; eyes - irrigate with pienty of water. Ingestion: give conscious victim water to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# 320

# MCPA CH3CIC6H3OCH2COOH

**IDENTIFICATION** 

UN No. 2765

Danger Group According to Percentage of Active Substance: Group III, Liquid 35 to 100%

Common Synonyms	Observable Characteristics	Manufacturers
(4-CHLORO-O-TOLYLOXY) ACETIC ACID	Light brown solid or colourless liquids.	Ciba-Geigy Canada Ltd., Etobicoke, Ont. Chipman Inc. Stoney Creek, Ontario.
Common Trade Names MCPA AMINE		Interprovincial Co-op., Saskatoon, Sask.
(A herbicide used for the control of broadleaf week	is.)	Uniroyal Chemical, Elmira, Ontario.
Transportation and Storage Information		
Shipping State: Solid or liquid (formulation). Classification: None. Inert Atmosphere: No requirement. Venting: Open. Pump Type: No information.	Label(s): Not regulated. Storage Temperature: Ambient. Hose Type: No Information.	Grades or Purity: Various as shown below.  Containers and Materials: Glass bottles, cans, drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Solid (technical). Solubility (Water): 0.083 g/100 mL (solid) (25°C). SN, EC soluble in water. Molecular Weight: 200.6 Vapour Pressure: No information. Boiling Point: No information.	Floatability (Water): Sinks; SN and EC mix. Odour: No information. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 1.56 (25°C) technical.	Colour: Colouriess (liquids); light brown (solid). Explosive Limits: Not flammable. Melting Point: 99 to 107°C (technical).

Human Health		
Symptoms: Ingestion and Skin Contact: readily temperature, lethargy and muscular weakness.  Toxicology: Highly toxic by skin contact; moderat		pain, flushing of skin, vomiting, muscular tremors, abnormal
TLV No information.	LC50 - No information.	LD50 - Oral: rat = 0.70 g/kg
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	LD50 - Oral: mouse = 0.55 g/kg
Fire		
Fire Extinguishing Agents: Use foam, carbon dioxi	de, or dry chemical.	
ignition Temperature: No information.	Burning Rate: No information.	
Reactivity		
With Water: No reaction. With Common Materials: No information. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water freshwater; 75 mg/L/45 h/tongnose killifish/LC50 Land-Ale: No information. Food Chain Concentration Potential: No informat	/saitwater.	m/96 h/TLm/freshwater. Fish toxicities: bluegill/LC50/

# Special Hazards

POISON.

# Immediate Responses

Keep non-involved people away from spill site. Stop or reduce discharge if safe to do so. Notify manufacturer or supplier. Dike to contain material or water runoff. Notify environmental authorities.

#### Protective Clothing and Equipment

In fires or confined spaces - Respiratory Protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, approved pesticide respirator and impervious outer clothing.

#### Fire and Explosion

Use carbon dioxide, foam or dry chemical to extinguish. Releases toxic fumes in fires.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - wrigate with plenty of water. Ingestions give water to conscious victim to drink and induce vomiting; in the case of petroleum distillates, do not induce vomiting for fear of aspiration and chemical pneumonia. if medical assistance is not immediately available, transport victim to hospital, doctor, or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

1. Stop or reduce discharge if safe to do so.

2. Contact manufacturer or supplier for advice.

#### Floats

# 3. If possible contain discharge by booming.

4. If floating, skim and

# Sinks or mixes

3. If possible contain discharge by damming or water diversions.

4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.

5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. If liquid, remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. If solid, remove material by manual or mechanical means.
- 6. Recover undamaged containers.
- 7. Absorb residual liquid on natural or synthetic sorbents.
- 8. Remove contaminated soil for disposal.
- 9. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

#### Disposal

remove.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# **Available Formulations**

Technical Grade: Purity: 85 to 95%.

Properties: insoluble in water.

#### Formulations:

Purity: Type: - typically 50% SN - solution (diethanolamine salt) SN - solution (sodium salt) - typically 50% EC - emulsifiable concentrate (ester)

- typically 40%

#### Properties:

- soluble in water, combustible
- soluble in water
- soluble in water, combustible

Other Possible Ingredients Found in Formulations: Dicamba, mecoprop, linuron, bromacil.

# 322

# MERCURY Hg

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
QUICKSILVER HYDRARGYRUM	Silvery liquid. Odourless.	Johnson Matthey and Mallory Ltd., Brampton, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: None. inert Atmosphere: No requirement. Venting: Open. Pump Type: No information.	Labells): Not regulated. Storage Temperature: Ambient. Hose Type: No information.	Grades or Purity: Pure. Containers and Materials: Bottles, flasks, drums.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Liquid. Solubility (Water): 0.025 g/100 mL (30°C). Molecular Weight: 200.6 Vapour Pressure: 0.0012 mm Hg (20°C); I mm Hg (126°C). Bolling Point: 356.6°C.	Ficatability (Water): Sinks. Odour: Odourless. Flash Point: Not flammable. Vapour Density: 7.0 Specific Gravity: 13.6 at 20°C.	Colour: Silvery. Explosive Limits: Not flammable. Melting Point: -38.9°C.

Human Health		
Symptoms: Inhalation: metallic taste, rapid and diffinausea and vomiting, diarrhea, trembling extremitie Toxicology: Highly toxic by all routes.  TLV [®] (skin - Hg vapour) 0.05 mg/m ³ .  Short-term Inhalation Limits - 0.15 mg/m ³ (15 min)	s. <u>Contact</u> : skin - irritation, inflammation, blisters; LCso - No information.	intense thirst, pain in swallowing, abdominal pain, eyes - Irritation, watering, edema of eyelids.  LD50 - No information.  TD1.0 - intraperitoneal: rat = 0.4 g/kg
as lig vapour.	Delayed Toxicity - Known long-term effects to central nervous system.	
Fire		
Fire Extinguishing Agents: Not combustible. Most fi	re extinguishing agents may be used on fires involvi	ng mercury.
Behaviour in Fire: Vapourizes readily emitting toxic	Hg lumes. Burning Rate: Not combustible.	
Behaviour in Fire: Vapourizes readily emitting toxic ignition Temperature: Not combustible.  Reactivity	Hg Jumes.	
Behaviour in Fire: Vapourizes readily emitting toxic ignition Temperature: Not combustible.	Hg Iumes. Burning Rate: Not combustible.	
Behaviour in Fire: Vapourizes readily emitting toxic lgaition Temperature: Not combustible.  Reactivity  Wish Water: No reaction.  Wish Common Materials: Reacts violently with acet Stability: Stable.  Environment	Hg Jumes. Burning Rate: Not combustible.  ylene, ammonia, chlorine, chlorine dioxide, methyl	

# Special Hazards

POISON.

#### **Immediate Responses**

Keep non-involved people away from spill site. Notify manufacturer or supplier. Dike to contain spill. Notify environmental authorities.

### Protective Clothing and Equipment

In fires or confined spaces, Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

#### Fire and Exptosion

Not combustible. Most fire extinguishing agents may be used on fires involving mercury. Vapourizes readily emitting toxic Hg Jumes.

#### Pirst Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if necessary. Contact: skin - remove contaminated clothing and wash affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water or milk to conscious victim to drink, and induce vomitting. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- Absorb residual liquid on natural or synthetic sorbents.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

#### METHANE CHA

IDENTIFICATION

UN No. 1971 - gas 1972 - liquid

Common Synonyms Observable Characteristics Manufacturers MARSH GAS Colourless or liquid. Odourless when pure Canadian supplier: SEWAGE GAS (commercial forms may have odour compounds Canadian Liquid Air Co. Ltd., **NATURAL GAS*** Montreal, Oue. added). METHYL HYDRIDE *Natural gas is mostly methane, but contains other gases. Transportation and Storage Information Shipping State: Liquid (compressed gas). Label(s): Red label - FLAMMABLE GAS: Class 2.1. Grades or Purity: Technical, 95%, Btu grade. Classification: Flammable gas. Storage Temperature: Ambient. Containers and Materials: Cylinders, tank cars, mert Atmosphere: No requirement. Hose Type: No information. tank trucks: steel. Venting: Salety relief. Pump Type: No information. Physical and Chemical Characteristics Physical State (20°C, I atm): Gas. Floatability (Water): Floats and boils (liquid). Colour: Colouriess. Solubility (Water): 0.0024 g/100 mL (20°C). Odours Odouriess. Explosive Limits: 5.0 to 15.0%. Molecular Weight: 16.0 Flash Points -188°C. Melting Point: -182 to -184°C. Vapour Pressure: Gas. Vapour Density: 0.42 to 0.56 Boiling Point: -161.5°C. Specific Gravity: (Liquid) 0.42 (-164°C).

#### HAZARD DATA

#### Human Health Symptoms: Inhalation: asphyxiation, dizziness, difficult breathing, nausea and vomiting, exhaustion, loss of consciousness, convulsions. Contact: skin - feeling of intense cold, frostbite; eyes - stinging pain, watering of eyes, inflammation. Toxicology: Asphyxiant. TLV simple asphyxiant. LC50 - Man: tolerance limit = 1 000 ppm LD50 - No Information. Short-term Inhalation Limits - No information. $(18 \text{ mg/m}^3)$ . Delayed Toxicity - None known. Fire Fire Extinguishing Agents: Shut off flow of gas if safe to do so. In case of fire use carbon dioxide or dry chemical to extinguish. Behaviour in Fire: Flash back may occur along vapour trail. Burning Rate: 12.5 mm/min. ignition Temperature: 537°C. Reactivity With Water: No reaction. With Common Materials: Reacts with strong oxidizers. Reacts violently with bromine pentafluoride, chlorine, chlorine dioxide, liquid oxygen, nitrogen trifluoride, oxygen difluoride. Stability: Stable. Environment Water: Prevent entry into water intakes and waterways. Not seriously harmful to aquatic life. BOD: 304% (35 days). Land-Air: No information. Food Chain Concentration Potential: None.

Special Hazards

FLAMMABLE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Contact supplier for advice. Eliminate all sources of ignition. Stop or reduce discharge if this can be done without risk. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - in fires, use self-contained breathing apparatus. Otherwise: Gloves - rubber or plastic. Coveralls or acid suit - (jacket and pants).

# Fire and Explosion

Shut off flow of gas if safe to do so. Use carbon dioxide or dry chemical to extinguish. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped; oxygen if breathing is laboured. Contact: (liquified methane) skin and eyes - treat as for frostbite; remove contaminated clothing and wash eyes and skin thoroughly with plenty of water. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water-Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

#### METHANOL CH3OH

**IDENTIFICATION** 

UN No. 1230

Common Synonyms Observable Characteristics Manufacturers METHYL ALCOHOL Clear colourless, liquid, Alcohol-like odour. Alberta Gas Chemical, Medicine Hat, Alta. WOOD ALCOHOL Celanese Canada, Cornwall, Ontario. WOOD NAPHTHA WOOD SPIRIT CARBINOL METHYL HYDROXIDE Transportation and Storage Information Shipping State: Liquid. Label(s): Red and white label - FLAMMABLE: Grades or Purity: Pure, 99,9%; crude. Class 3.2, Group II. Black and white label -Containers and Materials: Cans. drums, tank Classification: Flammable, poisonous liquid. Inert Atmosphere: No requirement. POISON: Class 6.1, Group II. cars, tank trucks; steel. Venting: Open (flame arrester) or Storage Temperature: Ambient. Hose Type: No information. pressure-vacuum. Pump Type: No information. Physical and Chemical Characteristics Floatability (Water): Floats and mixes. Colour: Colourless. Physical State (20°C, I atm): Liquid. Odour: Aicohol-like. Explosive Limits: 6.0% to 36.0%. Solubility (Water): Soluble in all proportions. Flash Point: 11°C (c.c.). Melting Point: -93 to -98°C. Molecular Weight: 32.0 Vapour Pressure: 92 mm Hg (20°C); 160 mm Hg Vapour Density: 1.1 Specific Gravity: 0.79 at 20°C. (30°C). Boiling Point: 64.5°C.

# HAZARD DATA

#### Human Health

Symptoms: Inhalation: asphyxia, headache, fatigue, nausea and vomiting, exhaustion, loss of consciousness, convulsions. Contact: skin - readily absorbed producing symptoms similar to inhalation; eyes - irritation, damage, blurred vision. Ingestion: headache, fatigue, nausea and vomiting, blindness and death. Toxicology: Low toxicity by contact; moderate by inhalation.

TLY* - (skin) 200 ppm: 260 mg/m3.

LC50 - No information.

LDso - Oral: rat = 13 g/kg

Short-term Inhalation Limits - 250 ppm:

TCLo - Inhalation: human = 86 000 mg/m³

310 mg/m³ (15 min).

Delayed Toxicity - None known.

#### Fire

Fire Extinguishing Agents: Use alcohol foam. Water may be ineffective.

Behaviour in Fire: Flashback may occur along vapour trail.

lenition Temperature: 385°C.

Burning Rate: 1.7 mm/min.

# Reactivity

With Water: No reaction; soluble.

With Common Materials: Can react vigorously with oxidizing agents. Reacts violently with chromic anhydride, lead perchlorite, perchloric acid, phosphorus trioxide, and (hydroxides and chioroform).

Stability: Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 13 680 ppm/96 h/rainbow trout fingerling/LC5g/freshwater; 1 700 ppm/96 h/brown shrimp/LC5g/saltwater; 17 000 ppm/24 h/creek chub/LC10g/freshwater; BOD: 48 to 124%; 5 days.

Land-Air: No information.

Food Chain Concentration Potential: None.

#### Special Hazards

FLAMMABLE, POISON.

#### immediate Responses

Keep non-involved people away from spill site. Issue warnings: "FLAMMABLE; POISON". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Notify manufacturer. Stop leak, if safe to do so. Dike to contain material. Notify environmental authorities.

# Protective Clothing and Equipment

Protective outerwear as required.

#### Fire and Explosion

Use alcohol foam to extinguish; water may be ineffective. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected aras with plenty of water; eyes - irrigate with plenty of water. Ingestion: give plenty of water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# METHYL ACRYLATE CH2:CHCOOCH3

HAPP	<b>HTFIC</b>	ATION

UN No. 1919

Common Synonyms	Observable Characteristics	Marwilacturers
ACRYLIC ACID, METHYL ESTER METHYL 2-PROPENOATE METHYL PROPENATE PROPENOIC ACID, METHYL ESTER	Colouriess liquid. Sweet, sharp odour.	No Canadian manufacturers. Celanese Chem. Co., New York, NY. Rohm and Haas Co., Philadelphia, PA.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: Air must be present. Venting: Open (flame arrester). Pump Type: No information.	Label(s): Red and white label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient (if inhibited, <4°C if no inhibiter). Hose Type: No information.	Grades or Purity: Technical 98.5% (minimum purity); 99.9% (with inhibitor; hydroquinone and its methyl ether, in presence of air).  Containers and Materials: Drums, tank cars; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 5.2 g/100 mL (20°C). Molecular Weight: 86.1 Vapour Pressure: 70 mm Hg (20°C); 110 mm Hg (30°C). Boiling Point: 80 to 81°C.	Ploatability (Water): Floats. Odour: Sweet, sharp. Plash Point: -3°C (o.c.). Vapour Density: 3.0 Specific Gravity: 0.96 (20°C).	Colour: Colouriess. Explosive Limits: 2,8 to 25%. Melting Point: -76.5°C.

	IN SECTION DITTIES	
Human Health		
Symptoms: <u>Inhalation</u> : irritation of mucous membrain in swallowing, stomach and abdominal pain, n Inflammation, bilisters; eyes - irritation, watering Toxicology: Highly toxic by inhalation and Ingestic TLV ² (skin) 10 ppm; 35 mg/m ³ .  Short-term Inhalation Limits - No information.	ausea and vomiting, diarrhea, shock, convulsio of eyes, inflammation, chemical burns and lesi	itigue, diarrhea. <u>Ingestion</u> : irritation of lips, mouth and throat, ns. <u>Contact</u> : skin - <u>readily</u> absorbed, itching and irritation, ions.  LD50 - Oral: rat = 0.3 g/kg
Fire		
disperse vapours.  Behaviour in Fire: At temperatures above 21°C po along vapour trail.  Ignition Temperature: 468°C.	·	should be used to keep fire-exposed containers cool and d container, may cause a violent rupture. Flash back may occur
Reactivity		
With Water: No reaction; slightly soluble. With Common Materials: Can react vigorously wit Stability: Polymerizes at high temperatures or unc		
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: None.	rways. Harmful to aquatic life. Aquatic toxicit	ty rating = 100 to 1 000 ppm/96 h/TLm/freshwater.

# Special Hazards

FLAMMABLE. Polymerizes under certain conditions.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Notify manufacturer or supplier. Dike to prevent runoff of material. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

#### Pire and Explosion

Use foam, dry chemical or carbon dioxide to extinguish. Water may be ineffective, but should be used to cool fire-exposed containers and disperse vapours. At temperatures above 21°C, polymerization may take place and, if in a closed container, violent rupture may occur. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected aras with plenty of water; eyes - irrigate with plenty of water. Ingestion: give plenty of water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

# Dissolved

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.

4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.

5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Undissolved

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain spill by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss discuss discussions and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

#### METHYLAMINES

#### IDENTIFICATION

UN No. 1061 Anhydrous 1235 Aqueous solution

Common Synonyms	Observable Characteristics	Manufacturers
MONOMETHYLAMINE (CH3NH2) Aminomethane, Carbinamine DIMETHYLAMINE (CH3)2NH TRIMETHYLAMINE (CH3)3N Methanamine, N,N-Dimethylamine	Colourless gases or liquids. Ammonia-like odours.	Chinook Chemicals, Sombra, Ontario.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas or aqueous solution). Classification: Flammable liquid or gas. Inert Atmosphere: No requirement. Venting: Safety relief.	Label(s): Red label - FLAMMABLE LIQUID OR GAS. Storage Temperature: Ambient.	Grades or Purity: Anhydrous, 96.5%+; aqueous solutions 25, 30, 35, 40%.  Containers and Materials: Anhydrous - cylinders.  Solutions - drums, tank cars, tank trucks; steel, stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas.  Solublity (Water): Soluble in all proportions.  Molecular Weight: 31.1 (mono): 45.1 (di; 59.1 (tri).  Vapour Pressure: (mono) 2 356 mm Hg; (di) 1 292 mm Hg; (tri) 1 444 mm Hg (all at 20°C).  Boiling Point: (mono) -6.5°C; (di) 7.4°C; (tri) 2.9°C.  Specific Gravity: (mono) 0.77 (-70°C); (di) 0.68 (0°C); (tri) 0.66 (-5°C) (anhydrous forms).  Solutions: 25% (mono) 0.93 (15°C); 40%, 0.89 (15°C). 25% (di) 0.92 (15°C); 40%, 0.83 (15°C). 25% (tri) 0.92 (15°C); 40% 0.89 (15°C).	Floatability (Water): Liquid or solutions float and mix.  Odour: Ammonia-like, fishy (0.92 to 3.3 ppm, odour threshold).  Flash Point: gases <0°C. 30% (mono) solution 0.28°C; 40%, -13°C. 25% (di) solution 5.6°C; 40%, -15.6°C; 25% (tri) solution 5°C (all o.c.).  Vapour Density: (mono) 1.97; (di) 1.55; (tri) 2.04.	Colour: Colourless.  Explosive Limits: 4.9 to 20.7% (mono); 2.8 to 14.4% (di); 2.0 to 11.6% (tri).  Melting Point: (anhydrous forms)(mono) -92.5°C; (di) -92 to -96°C; (tri) -177 to -124°C. 25% mono solution -31°C; 40%, -37°C. 25% di solution -17°C; 40%, -37°C. 25% tri solution +6°C; 40%, +4°C.

#### HAZARD DATA

#### Human Health

Symptoms: Inhalation: irritation of nose and eyes, sneezing, coughing. ingestion: burning sensation, pain in swallowing, nausea and vomiting, stomach cramps, rapid breathing, diarrhea. Contact: skin - painful burning, ulceration and shock; eyes - painful irritation, intense watering, burns and irreparable damage.

Toxicology: Moderately toxic by all routes. TLV² (mono) 10 ppm; 12 mg/m³; (di) 10 ppm; 18 mg/m³. Short-term inhalation Limits - No information.

LC50 - No information.

Delayed Toxicity - No information.

LD50 - Oral: rat = 0.698 g/kg (di)

LDLo - Subcutaneous: rat = 0.20 g/kg (mono)

#### Eira

Fire Extinguishing Agents: Stop flow of gas. Use water spray, carbon dioxide, dry chemical and alcohol foam on water solution. Use water to cool fire-exposed containers.

Behaviour in Fire: Flash back may occur along vapour trail.

ignition Temperature: 430°C (mono); 400°C (di); Burning Rate: No information. 109°C (tri).

#### Reactivity

With Water: No reaction; soluble.

With Common Materials: Can react vigorously with oxidizing materials. Aqueous solutions are corrosive to many metals. Stability: Stabie.

#### Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: (mono) 10 to 30 ppm, (di) 30 to 50 ppm/24 h/creek chub/TLm/freshwater; Aquatic toxicity rating = 10 to 100 ppm/96 h/TLm/freshwater; BOD: 130% (di), 5 days.

Land-Air: No information.

Food Chain Concentration Potential: None.

#### Special Hazards

FLAMMABLE. Aqueous solutions are corrosive.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Notify manufacturer or supplier. Dike to prevent runoif from water application. Notify environmental authorities.

#### Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

#### Fire and Explosion

Stop flow of gas. Use water spray, carbon dioxide, dry chemical or alcohol foam on water solutions. Use water to cool fire-exposed containers. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give conscious victim plenty of water to drink; do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or

synthetic sorbents.

7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- I. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# METHYL CHLORIDE CH3CI

UN No. 1063

Common Sysonyms	Observable Character	istics	Manufacturers		
CHLOROMETHANE MONOCHLOROMETHANE	Colourless gas or liqui ethereal odour.	d. Sweet,	No Canadian manufacturer. Canadian suppliers: Domtar, Longford Mills, Ontario. Dow Chemical Canada, Inc., Sarnia, C Ethyl Canada, Sarnia, Ontario.	Ont.	Originating from: Conoco Chem., USA Dow Chemical, Plaguemine, LA, Freeport, TX, Ethyl, USA
Transportation and Storage	nformation				
Shipping State: Liquefied co. Classification: Poisonous gas inert Atmosphere: No requir Venting: Safety relief. Pump Type: Positive displac steel, stainless steel.	s and flammable liquid. ement. ement; carbon	Red label - F Storage Temp	te label - POISONOUS GAS; Class 2.3. LAMMABLE LIQUID; Class 3.1. perature: Ambient. effon, Viton, bronze, stainless	technic Contai	or Purity: Refrigeration grade; cal grade. 99.5+%. ners and Materials: Cylinders, tank ank trucks; steel, stainless steel.
Physical and Chemical Char- Physical State (20°C, 1 atm Solubility: Slight; 0.72 g/100 0.43 g/100 mL (25°C). Molecular Weight: 50.5 Vapour Pressure: 3800 mm is 5090 mm itg (30°C). Boiling Point: -24.2°C.	n): Gas. ) mL (0°C);	Odour: Faint (21 mg/m ³ , o Flash Point: I Vapour Densi		Explos	: Colourless. ive Limits: 10.7 to 17.4%. g Point: -97.7°C.

# HAZARD DATA

Contact: skin - feeling of cold, pain, frostbite; ever	mental confusion, nausea and vomiting, giddiness, exhaustion, loss of consciousness, convulsions.  s - stinging, watering, inflammation.  sc cause serious symptoms may be followed after a few days by more severe effects.  LC50 - knhalation: rat = 152 000 mg/m³ for 30 min. LD50 - No information.  Delayed Toxicity - Subjethal or lethal effects may  occur several days after exposure. Symptoms may not appear until several hours after exposure.
to knock down vapours or cool fire-exposed contain	out fire until leak has been turned off. In fires, use dry chemical, carbon dioxide or foam. Water may be used ners.  ur trail. Decomposition from heat or fire can produce toxic chloride fumes.  Burning Rate: 2.2 mm/min.

# Reactivity

Human Health

With Water: No reaction.

With Common Materials: Can react violently with powdered aluminum, magnesium, sodium and other alkali metals.

Stability: Stable.

# Environment

Water: Prevent entry into water intakes or waterways. Aquatic toxicity rating = >1 000 ppm/96 h/TLm/freshwater; BOD: None. Land-Air: No information.

Food Chain Concentration Potential: None.

### Special Hazards

FLAMMABLE.

#### **Immediate** Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". Call Fire Department. Eliminate all sources of ignition. Call manufacturer for advice. Avoid contact and inhalation. In fire, work from upwind. If water spray used, dike area to contain toxic runoff. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus. Acid suit - jacket and pants, rubber or plastic. Boots - high, rubber; pants worn outside boots. Gloves - rubber or plastic.

# Fire and Explosion

Do not attempt to put out fire until leak has been shut off. For fires, use dry chemical, carbon dioxide or foam. Use water spray to knock down vapour and cool fire-exposed containers. Flashback may occur along vapour trail. Decomposition from heat or fire can produce toxic chloride fumes.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalation</u>: give artificial respiration if breathing has stopped; give oxygen if breathing is laboured. <u>Contact</u>: remove contaminated clothing; wash eyes and affected skin with plenty of warm water. Do not rub areas which appear to have been frosthitten. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic. Victim should be hospitalized for observation because of slowly developing symptoms.

#### **ENVIRONMENTAL PROTECTION MEASURES**

### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- water diversion.

  4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Recover undamaged containers.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

### Dispesal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 333

# METHYL ETHYL KETONE CH3COC2H5

**IDENTIFICATION** 

UN No. 1193

Common Synonyms	Observable Characteristics	Manufacturers
MEK 2-BUTANONE ETHYL METHYL KETONE	Colourless liquid. Sweet, fragrant.	Shell Canada Limited, Montreal, Que. Gulf Oil Canada Limited, Montreal, Que. Esso Chemicals, Sarnia, Ont.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Open (flame arrester) or pressure-vacuum. Pump Type: Gear, centrifugal; explosion- proof motors.	Label(s): Red label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Polyethylene, butyl, polypropylene.	Grades or Purity: Technical, 99.5+%.  Containers and Materials: Drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 35.3 g/100 mL (10°C); 26.3 (20°C); 19.0 g/100 mL (90°C). Molecular Weight: 72.1 Vapour Pressure: 77.5 mm Hg (20°C). Bolling Point: 79.6°C.	Floatability (Water): Floats and mixes. Odour: Sweet, fragrant (2.0 to 6.0 ppm, odour threshold). Flash Point: -5.6°C (o.c.); -9.0°C (c.c.). Vapour Density: 2.4 Specific Gravity: 0.81 (20°C).	Colour: Colourless. Explosive Limits: 1.7 to 11.4%. Melting Point: -85 to 87°C.

	HAZARD DATA	
Human Health		
vomiting, headache and dizziness. Contact: ski Toxicology: Highly toxic by ingestion, moderat	in - defatting and dermatitis; eyes - irritation left toxic by contact.	
TLY - (inhalation) 200 ppm; 590 mg/m ³ .	LC ₅₀ - No Information.	$LD_{50}$ - Oral: rat = 3.4 g/kg
Short-term Inhalation Limits - 300 ppm;	Delayed Toxicity - No information. TC _{Lo} - Inhalation: human = 100 ppm/5	min
885 mg/m ³ (15 min).	tofo - uniquetrout nominat - 100 blants	5575516
Ignition Temperature: 516°C. Reactivity	Burning Rate: 4.1 mm/min.	
With Water: No reaction, soluble. With Common Materials: Can react with oxidi: Stability: Stable.	zing materials. Reacts violently with chlorosu	Ifonic acid, oleum and potassium-t-butoxide.
Environment		
Water: Prevent entry into water intakes and w 5 600 mg/L/48 h/mosquito fish/TLm/freshwate Land-Air: No information.	aterways. Fish toxicity: 5 640 mg/L/48 h/blue er; Aquatic toxicity rating = >1 000 ppm/96 h/	gili/TLm/freshwater; 500 mg/L/24 h/goldfish/LC50/freshwater; TLm/freshwater; BOD: 151 to 224%, 5 days.

#### Special Hazards

#### FLAMMABLE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all ignition sources. Call manufacturer or supplier for advice. Contain spill by diking. Prevent runoff into sewers or watercourses. Avoid inhalation or contact. Stop or reduce discharge, if safe to do so. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus, and full protective clothing. Gloves - rubber. Boots - high, rubber (pants worn outside boots).

### Fire and Explosion

Use dry chemical, alcohol foam or carbon dioxide to extinguish. Water may be ineffective, but may be used to cool fire-exposed containers. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill area, to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: eyes - irrigate with plenty of water; skin - wash affected areas with plenty of water and remove contaminated clothing. Ingestion: give plenty of water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or
- other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# 33

# METHYLCYCLOPENTADIENYL MANGANESE TRICARBONYL C9H7O3Mn

mc i i i	IDENTIFICATION	С9117ОЗМЯ
Common Synonyms	Observable Characteristics	Manufacturers
METHYLCYCLOPENTADIENYLMANGANESE- TRICARBONYL ETHYL MMT ANTIKNOCK MMT	Orange liquids. Sensitive to light. Faint, pleasant, herbaceous odour.	Ethyl Corporation of Canada Limited, Corunna, Ontario; Toronto, Ontario. U.S. manufacturer: Ethyl Corporation, Houston, TX
Transportation and Storage Information		
Shipping State: Liquid. Classification: Not regulated. Voluntary poison label. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Standard types (e.g. centrifugal). For LP grades, flammable solvent equipment required.	Label(s): Voluntary white label - POISON. Storage Temperature: Ambient. Hose Type: Standard neoprene hydrocarbon hose. Also flexible stainless steel, steel, Tefion.	Grades or Purity: Ethyl MMT (neat), Ethyl MMT/LP 62, Ethyl MMT/LP 46. Containers and Materials: Cans, drums; steel.
Physical and Chemical Characteristics		•
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.007 g/100 mL (25°C). Molecular Weight: 218.1 Vapour Pressure: 0.051 mm Hg (20°C); 9.3 mm Hg (100°C). Boiling Point: 233°C (for neat MMT).	Floatability (Water): Sink. Odour: Faint, pleasant, herbaceous. Flash Point: Ethy! MMT, 96°C (c.c.); Ethy! MMT/LP 62, 90°C (c.c.); Ethy! MMT/LP 46, 65°C (c.c.) Vapour Density: No information. Specific Gravity: Ethy! MMT, 1.38 at 20°C; Ethy! MMT/LP 62-1.11; Ethy! MMT/LP 46-1.02.	Colour: Dark orange. Explosive Limits: Not determined for neat MM solvents in other grades may be explosive. Melting Point: -2.2 to 1.5°C.
	HAZARD DATA	
Human Health		
	che, metallic taste, nausea. <u>Ingestion</u> : symptoms similar	to inhalation. Contact: skin - readily absorbed;
eyes - Irritation.  Toxicology: Highly toxic by inhalation and ingest TLV* - (skin) 0.2 mg/m ³ (as Mn).  Short-term inhalation Limits - 0.6 mg/m ³ (15 min (as Mn).	$LC_{co}$ - (inhalation) rat = 247 mg/m ³ /1 h	LD59 - Oral: rat = 0.058 g/kg
Fire		
Fire Extinguishing Agents: Dry chemical, carbon Behaviour in Fire: Decomposes slowly at 200°C; ignition Temperature: No information.	dioxide, foam, water spray.  [airly rapidly at 300°C yielding toxic gases.  Burning Rate: No information.	
Reactivity		
With Water: No reaction.		

With Common Materials: No information.

Food Chain Concentration Potential: No information.

Water: Prevent entry into water intakes and waterways. BOD: No information. Land-Air: No information.

Stability: Stable. Environment

# Special Hazards

POISON, FLAMMABLE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warnings: "POISON; FLAMMABLE". Call fire department. Eliminate all ignition sources. Call manufacturer for guidance and assistance. Stop or reduce discharge, if this can be done without risk. Avoid contact and inhalation. Contain spill area by diking. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - use self-contained breathing apparatus and totalty encapsulated suit. Gloves - neoprene or vinyl. Boots - rubber (pants worn outside boots).

# Fire and Explosion

Use dry chemical, carbon dioxide, foam or water spray to extinguish.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: eyes irrigate with pienty of water; skin - flush with pienty water and remove contaminated clothing. Ingestion: give conscious victim plenty of water to drink and induce vomiting. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Absorb residual liquid on natural or synthetic sorbents.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# METHYLENE CHLORIDE CH2C12

IDENTIFICATION Manufacturers

Canadian supplier:

No Canadian manufacturers.

Stanchem, Montreal, Quebec.

Dow Chemical Canada Inc., Sarnia, Ontario.

UN No. 1593

Common Synonyms

Dichloromethane

Methylenedichloride

Methane dichloride

Methane dichloride

Transportation and Storage information

Shipping State: Liquid.

Classification: Poison.

Inert Atmosphere: No requirement.

Venting: Pressure-vacuum.

Pump Type: Centrifugal or positive displacement: or cross-line

steel or stainless steel.

Label(s): Black and white label - POISON; Class 6.1, Group III.

Storage Temperature: Ambient.
Hose Type: Stainless stee; Teflon, bronze. Viton
or cross-linked polyethylene may be used with caution.

Physical and Chemical Characteristics

Physical State (20°C, 1 atm): Liquid. Solubility: 2.0 g/100 ml. (20°C); 1.67 g/100 ml. (25°C). Molecular Weight: 84.9

Vapour Pressure: 349 mm Hg (20°C); 500 mm Hg (30°C). Boiling Point: 39 to 41°C. Floatability (Water): Sinks.

Odour: Ethereal (25 to 220 ppm, odour threshold). Flash Point: Not flammable. Vanour Density: 2.9

Specific Gravity: 1.33 (20°C).

Grades or Purity: Technical.

Containers and Materials: Drums, tank cars, tank trucks: steel.

Originating from: PPG Industries, USA

Colour: Colouriess.
Explosive Limits: 12 to 22%.

Melting Point: -96.7°C.

#### HAZARD DATA

Human Health

Symptoms: Inhalation: irritation of respiratory tract, headache, dizziness, nausea, unconsciousness; Ingestion: abdominal pain, readily absorbed producing symptoms similar to inhalation. Contact: skin - readily absorbed with symptoms similar to inhalation; eyes - irritation, slight burns.

Toxicology: Moderately toxic by all routes. TLV* - (inhalation) 100 ppm; 350 mg/m³. Short-term inhalation Limits - 500 ppm; 1 400 mg/m³ (15 min).

LC50 - Inhalation: rat = 88 000 mg/m 3 /0.5 h Delayed Toxicity - No Information. LD50 - Oral: rat = 1.67 g/kg

Fire

Fire Extinguishing Agents: Most fire extinguishing agents may be used on fires involving methylene chloride, including water spray or fog and foam. Water may be used to cool fire-exposed containers.

Behaviour in Fire: When heated to decomposition, emits toxic fumes of hydrogen and chloride.

ignition Temperature: 556°C.

Burning Rate: No information.

Reactivity

With Water: No reaction.

With Common Materials: Reacts violently with lithium, sodium/potassium alloy and potassium-t-butoxide. May react with aluminum, magnesium and their alloys.

Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating = 100 to 1 000 ppm/96 h/TLm/freshwater; BOD: None.

Land-Air: No information.

Food Chain Concentration Potential: None.

#### Special Hazards

POISON.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Call Fire Department. Call manufacturer for assistance. Avoid contact and inhalation. Dike to prevent runoff. Notify environmental authorities.

#### Protective Clothing and Equipment

In fires or enclosed spaces, <u>Respiratory protection</u> - self-contained breathing apparatus and totally encapsulated suit. Otherwise, suitable respirator and protective outerwear as required.

# Fire and Explosion

Most fire extinguishing agents may be used on fires involving methylene chloride. Water may be used to cool fire-exposed containers. When heated to decomposition, emits toxic fumes of hydrogen and chloride.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water for at least 15 minutes. Ingestion: give plenty of water to drink. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge is safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth
- or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or
- synthetic sorbents.
- 7. Remove contaminated soil for disposal.
- 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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# METHYL ISOBUTYL KETONE (CH3)2CHCH2COCH3

IDENTIFICATION

UN No. 1245

Manufacturers Observable Characteristics Common Synonyms Shell Canada, Montreal, Quebec. Colourless liquid. Sharp, pleasant odour. HEXONE Gulf Canada, Montreal, Quebec. 4-METHYL-2-PENTANONE ISOBUTYL METHYL KETONE MIBK 2-METHYL-4-PENTANONE Transportation and Storage Information Label(s): Red and white label - FLAMMABLE LIQUID: Grades or Purity: Technical. 98.5%. Shipping State: Liquid. Containers and Materials: Cans, drums, tank cars: Classification: Flammable liquid. Class 3.2, Group II. steel. Storage Temperature: Ambient. inert Atmosphere: No requirement. Venting: Open (flame arrester) or pressure-vacuum. Hose Type: Polyethylene, butyl, polypropylene. Pump Type: Gear, centrifugal; explosionproof motors. Physical and Chemical Characteristics Floatability (Water): Floats. Colour: Colouriess. Physical State (20°C, 1 atm): Liquid. Explosive Limits: 1.2 to 8.0%. Odour: Sharp, pleasant (0.1 to 7.8 ppm). Solubility (Water): 1.8 g/100 ml. (20°C). Melting Point: -80 to -85°C. Flash Point: 18°C (c.c.). Molecular Weight: 100.2 Vapour Density: 3.5 Vapour Pressure: 6 mm Hg (20°C): Specific Gravity: 0.80 (20°C). 10 mm Hz (30°C). Boiling Point: 116 to 119°C.

#### HAZARD DATA

#### Human Health Symptoms: Inhalation: irritation of respiratory tract, headache, nausea, dizziness, unconsciousness, CNS depressant. Ingestion: irritation of gastrointestinal tract, nausea and vomiting. Contact: skin - defatting and irritation; eyes - irritation and burning sensation. Toxicology: Moderately toxic by all routes. LD50 - Oral: rat = 2.08 g/kg LCsn - No information. TLY (inhalation) 50 ppm; 205 mg/m3. Short-term Inhalation Limits - 75 ppm; Delayed Toxicity - No information. LCLo - Inhalation: rat = 4 000 ppm/15 min. 300 mg/m³ (15 min). Fire Fire Extinguishing Agents: Alcohol foam, dry chemical, carbon dioxide. Water may be ineffective, but may be used to cool fire-exposed containers. Behaviour in Fire: Flash back may occur along vapour trail. Burning Rate: No information. ignition Temperature: 448°C. Reactivity With Water: No reaction. With Common Materials: Can react vigorously with reducing or oxidizing materials. Reacts violently with potassium-t-butoxide. Stability: Stable. Environment Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating = >1 000 ppm/96 h/TLm/freshwater; Fish toxicity: 460 mg/L/24 h/goldfish/ LCso/freshwater; BOD: 12 to 60%, 5 days. Land-Air: No information.

Food Chain Concentration Potential: None.

# Special Hazards

FLAMMABLE.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning "FLAMMABLE". CALL FIRE DEPARTMENT. Call manufacturer for assistance. Dike to prevent runoff. Call environmental authorities.

# Protective Ciothing and Equipment

In fires Respiratory protection - self-contained breathing apparatus. Otherwise, suitable respirator and outer protective clothing as required.

# Fire and Explosion

Use alcohol foam, dry chemical or carbon dioxide to extinguish. Water may be ineffective, but may be used to cool fire-exposed containers. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill site. Call for medical assistance but start first aid at once. <u>Inhalation</u>: if breathing has stopped, give artificial respiration; if laboured, give oxygen. <u>Contact</u>: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. <u>Ingestion</u>: give conscious victim plenty of water to drink. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss discossal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# METHYL MERCAPTAN CH3SH

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UN No. 1064

Common Synonyms	Observable Characteristics	Manufacturers
METHANETHIOL MERCAPTOMETHANE (Methyl mercaptan is frequently added to natural gas to provide odour).	Colourless gas. Strong unpleasant (rotten cabbage) odour.	Canadian supplier: Canadian Liquid Air Co. Ltd., Montreal, Que.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Flammable gas. Inert Atmosphere: No requirement. Venting: Safety relief.	Label(s): Red label - FLAMMABLE GAS; Class 3. White label - POISON; Class 2.3. Storage Temperature: Ambient.	Grades or Purity: 98.0% purity; 99.5+% reagent. Containers and Materials: Cylinders, tank cars; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): 2.4 g/100 mL (15°C). Molecular Weight: 48.1 Vapour Pressure: 760 mm Hg (6.8°C); 1 520 mm Hg (26°C); 3 800 mm Hg (60°C). Boiling Point: 6 to 7.6 °C.	Floatability (Water): Floats and boils.  Odour: Strongly unpleasant (rotten cabbage), (0.001 to 0.002 ppm, odour threshold).  Flash Point: -17.8°C.  Vapour Density: 1.7  Specific Gravity: 0.87 (21°C) (liquid).	Colour: Colouriess. Explosive Limits: 3.9 to 21.8%. Metting Point: -123°C.

Symptoms: Inhalation: irritating to respiratory tra unconsciousness and respiratory paralysis. Contac	act; may cause dizziness, suffocation, headache tt skin and eyes - irritation and frostbite.	e and vomiting; CNS depressant, muscular weakness, tremors,
Texicology: Moderately toxic by inhalation. TLV [®] (inhalation) 0.5 ppm; I mg/m ³ . Short-term Inhalation Limits - No information.	i.C ₅₀ - inhalation: rat = 675 ppm Delayed Toxicity - No information.	LD ₅₀ - Subcutaneous: mouse = 0.0024 g/kg
Fire		
Fire Extinguishing Agents: Alcohol foam, carbon of Behaviour in Fire: Emits highly toxic SO _X fumes in Ignition Temperature: No information.  Reactivity	lioxide, dry chemical. Water may be used to con high temperatures. Flash back may occur alor Burning Rate: 3.8 mm/min.	ool fire-exposed containers and knock down vapours.  ng vapour trail.
With Water: Reacts with water to produce toxic S With Common Materials: Can react vigorously with Stability: Stable.	$\Theta_{\mathbf{x}}$ fumes. th oxidizing materials. Reacts with acids to pro-	oduce toxic and flammable SO _X vapours.
Environment		
Water: Prevent entry into water intakes and wate Salmonides/TLm/freshwater. Land-Air: No information. Food Chain Concentration Potential: No informat		ppm/120 h/ <u>Daphnia</u> sp./TLm/freshwater; 0.55 to 0.9 mg/L/

# Special Hazards

FLAMMABLE; POISON.

# **Immediate** Responses

Keep non-involved people away from spill site. Issue warnings: "FLAMMABLE; POISON". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Contact supplier and request advice or assistance. Avoid contact and inhalation. Stop or reduce discharge, if safe to do so. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus, and totally encapsulated suit.

# Fire and Explosion

Use alcohol foam, dry chemical or carbon dioxide to extinguish fire. Use water to cool fire-exposed containers. Emits highly toxic SO_X fumes in high temperatures. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped (not mouth-to-mouth method); and oxygen if breathing is laboured. Contact: remove contaminated clothing and wash eyes and skin thoroughly with plenty of warm water. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- i. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Recover undamaged containers.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# METHYL METHACRYLATE CH2:C(CH3)COOCH3

# IDENTIFICATION

UN No. 1247 (monomer, inhibited)

Common Synonyms	Observable Characteristics	Manufacturers	•	
METHACRYLIC ACID, METHYL ESTER METHYL-α-METHACRYLATE METHYL 2-METHYL-2-PROPENOATE MME METHACRYLATE MONOMER	Colourless liquid. Sharp, sweet adour.	No Canadian manufacturers.  Canadian suppliers: Chemacryl Plastics, Niagara Falls, Ont. DuPont, Canada, Toronto, Ontario. Rohm & Haas Canada, Toronto, Ontario. Rohm & Haas Canada, Toronto, Ontario.		Cy-Ro Industries, USA El duPont de Nemours, USA
Transportation and Storage Information				
Shipping State: Liquid. Classification: Plammable liquid. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: No information.	Label(s): Red and white label LiQUID; Class 3.2, Group II. Storage Temperature: Ambien Hose Type: No information.		Grades or Purity: Technical (inhibited: hydro- quinone 22 to 65 ppm; hydroquinone methyl ester 22 to 120 ppm; dimethyl-t-butylphenol 45 to 65 ppm). Containers and Materials: Drums, tank cars, trucks; steel.	
Physical and Chemical Characteristics				
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 1.5 g/100 mL (25°C). Molecular Weight: 100.1 Vapour Pressure: 28 mm Hg (20°C); 40 mm (26°C); 49 mm Hg (30°C). Boiling Point: 100 to 101°C.	Ficatability (Water): Ficats. Odour: Sharp, sweet, (0.05 to Flash Point: 10°C (o.c.). Hg Vapour Density: 3.5 Specific Gravity: 0.94 (20°C).	••	Colour: Colouries: Explosive Limits: Melting Point: -42	1.7 to 8.2%.

Human Health	
Symptoms: inhalation: headache, drowsiness, nausea, irritability, narcosis. Contact: skin - irritati Toxicology: Moderately toxic by all routes.  TLV 2 100 ppm; 410 mg/m ³ .  LC 50 - Inhalation: rat = 3 750 ppm Short-term Inhalation Limits - 125 ppm; 510 mg/m ³ Delayed Toxicity - No information.  (15 min).	ion and dermatitis; eyes - irritation.  LD50 - Oral: guinea pig = 6.3 g/kg
Fixe	
Fire Extinguishing Agents: Foam, carbon dioxide, dry chemical. Water may be ineffective but may vapours.  Behaviour in Fire: Flash back may occur along vapour trail. At high temperatures (as in fire) polylenition Temperature: 421°C.  Burning Rate: 2.5 mm/min.	
Reactivity	•
With Water: No reaction. With Common Materials: Can react with oxidizing materials. Reacts violently with benzoyl perox Stability: Stable.	xide.
Environment	
Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: toxicity rating x 100 to 1 000 ppm/96 h/TLm/freshwater; BOD: 14%, 5 days. Land-Air: No information. Food Chain Concentration Potential: No information.	232 to 368 mg/L/24 to 96 h/bluegill/TLm/freshwater; Aquatic

# Special Hazards

#### FLAMMABLE.

#### tramediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Contact manufacturer for assistance. Dike to prevent runoff, Notify environmental authorities.

# Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; and protective outerwear as required.

### Fire and Explosion

Foam, carbon dioxide or dry chemical. Water may be ineffective but may be used to cool fire-exposed containers and knock down vapours. Flash back may occur along vapour trail. At high temperatures (as in fire) polymerization may occur and cause container rupture.

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected area with plenty of water; eyes - irrigate with plenty of water. Ingestion: give conscious victim plenty of water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain split by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

**IDENTIFICATION** 

UN No. 2054

DIETHYLENIMIDE OXIDE TETRAHYDRO-2H-1.4-OXAZINE fishy ammonia odour. TETRAHYDRO-p-ISOXAZINE TETRAHYDRO-1,4-OXAZINE TETRAHYDRO-1,4-ISOXAZINE

Colourless, oily liquid with

Manufacturers Canadian supplier:

Canadian Industries Ltd., Toronto, Ont. McArthur Chemical, Montreal, Ont. Texaco Chemical Canada, Toronto, Ont.

BASF Canada, Montreal, Quebec. Kingsley and Keith, Montreal, Ouebec. Originating from: Texaco, United Kingdom.

BASF. W. Germany. Chemische Werke Huis. W. Germany.

Transportation and Storage Information

Shipping State: Liquid. Classification: Flammable liquid.

mert Atmosphere: No requirement. Venting: Open.

Pump Type: No information.

Labelish Red and white label - FLAMMABLE

LIQUID: Class 3.3, Group II. Storage Temperature: Ambient. Hose Type: No information.

Grades or Purity: Technical, 98%.

Containers and Materials: Drums, tank cars:

steel.

Physical and Chemical Characteristics

Physical State (20°C. 1 atml: Liquid. Solubility (Water): Soluble in all proportions.

Molecular Weight: 87.1

Vapour Pressure: 4.3 mm Hg (10°C); 8.0 mm He

(20°C): 13.4 mm Hg (30°C). Bolling Point: 127 to 129°C. Floatability (Water): Mixes.

Odour: Fishy (0.01 to 0.14 ppm, odour threshold).

Flash Point: 38°C (e.c.). Vanour Density: 3.0

Specific Gravity: 1.00 at 20°C (liquid).

Colour: Colourless.

Explosive Limits: 2.0 to 11.2%. Melting Point: -3 to -5°C.

HAZARD DATA

Human Health

Symptoms: Contact: liquid causes skin and eye burns, readily absorbed by skin. Inhalation: vapours may cause nausea, headache and irritation to respiratory tract, logestion; abdominal pain, diarrhea, vomiting,

Toxicology: Highly toxic by skin absorption and moderately toxic by ingestion.

TLV (skin) 20 ppm; 70 mg/m³.

damage.

Delayed Toxicity - May produce liver and kidney

Short-term inhalation Limits - 30 ppm:  $105 \text{ m g/m}^3 (15 \text{ min}).$ 

LC₅₀ - Inhalation: mouse = 1 320 mg/m³.

LD50 - Oral: rat = 1.05 g/kg

Fire

Fire Extinguishing Agents: Water spray, alcohol foam, dry chemical or carbon dioxide. Use water to cool fire-exposed containers and disperse vapours. Behaviour in Fire: Flash back may occur along vapour trail. In fires, toxic NO₂ fumes are produced.

Burning Rate: 1.9 mm/min. legition Temperature: 310°C.

Reactivity

With Water: No reaction, soluble.

With Common Materials: Can react with oxidizing materials.

Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating = 100 to 1 000 ppm/96 h/TLm/freshwater; 300 ppm/96 h/bluegill/TL50/ freshwater; BOD: 2%, 5 days.

Land-Air: No information.

Food Chain Concentration Potential: None.

#### Special Hazards

#### FLAMMABLE.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Avoid contact and inhalation. Call manufacturer or supplier. Dike to prevent runoff. Call environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

#### Fire and Explosion

Use water spray, alcohol foam, dry chemical or carbon dioxide to extinguish fire. Water may be used to cool fire-exposed containers and knock down vapours. Flash back may occur along vapour trail. In fires, toxic NO_X fumes are produced.

#### First Aid

Move victim out of spill area. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected area with plenty of water; eyes - irrigate with plenty of water. Ingestion: give plenty of water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### ENVIRONMENTAL PROTECTION MEASURES

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

#### **NAPHTHA SOLVENT**

IDENTIFICATION

UN No. 1255 - petroleum

1256 - solvent 1300 - V. M&P Manufacturers

Common Synonyms

PETROLEUM SOLVENT

PETROLEUM ETHER

MINERAL SPIRITS

Class 3.2.

synthetic rubber.

IOSOL LIGHT LIGROIN

PETROLEUM SPIRITS CLEANING SOLVENTS DRY CLEANERS NAPHTHA LIGHT NAPHTHA SAFETY SOLVENT WHITE SPIRITS

BENZIN

Naphtha: A generic term applied to refined, partly refined, or unrefined, petroleum products and liquid products of natural gas which distill between the temperatures of 35 and 205°C.

Observable Characteristics

Colourtess liquid. Gasoline-like odour.

Esso Chemical Canada. Div. of Imperial Oil Ltd., Sarnia, Ont.

Gulf Oil Canada Ltd., Montreal, Oue. Shell Canada Limited. Montreal, Oue.: Sarnia, Ont.:

North Burnaby, B.C.

Transportation and Storage Information

Shipping State: Liquid.

Classification: Flammable liquid.

inert Atmosphere: No requirement. Venting: Open (flame arrester) or pressure-

Pump Type: Gear, centrifugal, etc.

Physical and Chemical Characteristics Physical State (20°C, I atm): Liquid.

Solubility (Water): insoluble.

Molecular Weight: Variable, mixtures of

hydrocarbons.

Vapour Pressure: Variable, 0 to 67 mm Hg (38°C). Boiling Point: 30 to 202°C (variable).

Floatability (Water): Floats.

Storage Temperature: Ambient.

Odour: Gasoline-like ("5 ppm, odour threshold).

Label(s): Red label - FLAMMABLE LIQUID:

Hose Type: Buna-N, polyethylene, Viton and

Flash Point: Variable, -20 to 50°C. Vapour Density: Variable 2.5 to 4.8.

Specific Gravity: 0.75-0.87 (20°C) (variable).

Grades or Purity: By flashpoint, petroleum naphtha f.p. >38°C; naphtha solvent f.p. "43°C; naphtha vm&p -7 to -13°C.

Containers and Materials: Drums, tank cars,

tank trucks: steel.

Colour: Colourless.

Explosive Limits: 0.8 to 7.0%. Melting Point: <-30°C.

HAZARD DATA

Human Health

Symptoms: Inhalation and Ingestion: nausea, vomiting, coughing, irritation of respiratory tract, weakness, dizziness, unconsciousness. Contact: eyes - slightly irritating, skin - dermatitis.

Toxicology: Moderately toxic by inhalation.

TLV (Inhalation) No information.

Short-term Inhalation Limits - No Information.

LCso - No information.

LCLo - Inhalation: rat = 1 600 ppm Delayed Toxicity - No information.

LD50 - Oral: rat = 0.5 to 5.0 g/kg

LDLo - Intraperitoneal: mammal = 2.5 g/kg

Fire Extinguishing Agents: Use foam, carbon dioxide or dry chemical. Water may be ineffective but may be used to cool fire-exposed containers and to protect personnel.

Behaviour in Fire: Flash back may occur along vapour trail.

Burning Rate: 4 mm/min. ignition Temperature: 229 to 293°C.

Reactivity

With Water: No reaction.

With Common Materials: Can react with oxidizing materials. Reacts violently with chromates.

Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Toxic to fish at approximately 10 ppm. BOD: Not available.

Land-Air: No information.

#### Special Hazards

FLAMMABLE.

# Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE", CALL FIRE DEPARTMENT. Eliminate all sources of ignition, Call supplier or manufacturer. Call environmental authorities.

### Protective Clothing and Equipment

Respiratory protection - in tires - self-contained breathing apparatus, otherwise; Coveralls. Gioves - polyethylene or suitable plastic. Boots - high (not natural rubber). Gorgies - or face shield.

### Fire and Explosion

Use foam, dry chemical or carbon dioxide to extinguish. Water may be ineffective, but may be used to cool fire-exposed containers and protect personnel. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: do not induce vomiting in conscious victim. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to doctor. hospital or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming-
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# NAPHTHALENE CIGHS

	IDENTIFICATION	UN No. 1334 refined or crude		
Common Synonyms	Observable Characteristics	Manufacturers		
NAPHTHALIN TAR CAMPHOR MOTH FLAKES MOTH BALLS	Colouriess powder or flakes. Aromatic *mothball* odour.	Record Chemical Co. Ltd., Montreal, Que. Kent Laboratories Ltd., Vancouver, B.C.		
Transportation and Storage Information				
Shipping State: Solid. Classification: Flammable solid. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Red and white label - FLAMMABLE SOLID; Class 4.1, Group III. Storage Temperature: Ambient.	Grades or Purity: Pure: mp = 80°C, crude: mp = 74 to 80°C. Containers and Materials: Cans, drums; steel.		
Physical and Chemical Characteristics				
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.0034 g/100 mL (25°C). Molecular Weight: 128.2 Vapour Pressure: 1 mm Hg (53°C). Boiling Point: 218°C (sublimes).	Floatability (Water): Sinks. Naphthalene in some forms includes sufficient air pockets to allow floatation. Odour: Aromatic mothball odour (0.03 ppm, odour threshold). Flash Point: 88°C (o.c.); 79°C (c.c.). Vapour Density: 4.4 Specific Gravity: 1.15 (20°C).	Colour: Colourless. Explosive Limits: 0.9 to 5.9%. Melting Point: 80.2°C, volatilizes at room temperature.		

	HAZARD DATA	
Human Health		
Symptoms: Inhalation: vapours may cause cou and symptoms similar to inhalation. Contact: Toxicology:	ghing, headache, mental confusion, nausea and vom skin – irritation and dermatitis; eyes – irritation an	niting. <u>Ingestion:</u> nausea, vomiting, diarrhea, general fati <b>gue</b> ad watering.
TLV * (inhalation) 10 ppm; 50 mg/m³. Short-term Inhalation Limits - 15 ppm; 75 mg/m³ (15 min).	LC50 - Not available. Delayed Toxicity - No information.	LD50 - Oral: rat = 1.78 g/kg
Fire		
Fire Extinguishing Agents: Water spray, carbo Behaviour in Fire: No information. Ignition Temperature: 526°C.,	on dioxide, dry chemical or foam. Water spray may Burning Rate: 4.3 mm/min.	cause extensive foaming.
Reactivity		
With Water: No reaction. With Common Materials: May react with exid Stability: Stable.	izing materials. Reacts violently with chromic anh	ydride.
Environment		
Water: Prevent entry into water intakes and TLm/freshwater; 150 to 220 mg/L/48 h/mosq Land-Air: No information. Food Chain Concentration Potential: None.	waterways. Fish toxicity: 150 ppm/96 h/sunfish/TLr uito fish/TLm/freshwater; 1.8 ppm/72 h/fingerling	m/freshwater; Aquatic toxicity rating = 1 to 10 ppm/96 h/saltwater; BOD: 0%, 5 days; 192%, 25 days.

#### Special Hazards

COMBUSTIBLE.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Avoid contact and inhalation. Call manufacturer. Stop or reduce discharge if safe to do so. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - in fires or enclosed spaces, self-contained breathing apparatus; otherwise: Goggles -(mono), tight fitting. If face shield is used, it must not replace goggles. Gloves - rubber. Boots - rubber (pants worn outside boots). Coveralls or acid suit - rubber, if in fire or major spill.

### Fire and Explosion

Use water spray, dry chemical, alcohol foam or carbon dioxide to extinguish. Water spray may cause extensive foaming.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin and eyes - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: give water to conscious victim. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

UN No. 1971 compressed 1972 refrigerated liquid

Manufacturers Observable Characteristics Common Synonyms Universally available. Colourless, odourless gas if pure. Commercial LNG varieties have methyl mercaptan added as an odour. GAS Typical composition 85% methane, 10% ethane, 3% propane, 2% butane, pentanes and others. Unprocessed varieties may contain up to 50% HoS, which is very toxic. Transportation and Storage Information Grades or Purity: Variable. Label(s): Red label - FLAMMABLE GAS. Shipping State: Gas or liquid (compressed Containers and Materials: Cylinders, tank trucks, gas or refrigerated). Storage Temperature: Ambient or at -162°C. tank cars, tank vessels; steel. Classification: Flammable gas-Inert Atmosphere: No requirement. Venting: Safety-relief. Physical and Chemical Characteristics Colour: Colouriess. Floatability (Water): Floats and boils. Physical State (20°C, 1 atm): Gas. Explosive Limits: 3.8 to 17%. Odour: Odourless unless methyl mercaptan has been Solubility (Water): 0.006 g/mL (20°C). Melting Point: -182 to -150°C. Molecular Weight: Variable. added. Vapour Pressure: 2 900 mm Hg (-140°C): Flash Point: <-50°C. Vapour Density: 0.7 to 1.40 16 600 mm Hz (-100°C). Specific Gravity: (liquid) 0.4 to 0.5 (at -162 Boiling Point: -162 to -130°C. to -130°C).

# HAZARD DATA

Symptoms: <u>Inhalation</u> : asphyxiant, headache, labor Toxicology: Low toxicity by all routes. TLV ² (inhalation) No information. Short-term Inhalation Limits - No information.	LC 50 - No information.  Delayed Toxicity - No information.	LD50 - No information.
Fire		
Fire Extinguishing Agents: Stop flow of gas. Most Behaviour in Fire: Containers may rupture violent Ignition Temperature: 482 to 632°C.	fire extinguishing agents may be used. Water by in fires. Burning Rate: No information.	should be used to cool fire-exposed containers.
Reactivity		
With Water: No reaction. With Common Materials: No information. Stability: Stable.		

Environment

Human Health

Water: Prevent entry into water intakes and waterways. No information on aquatic toxicity.

Land-Air: No information.

Food Chain Concentration Potential: None.

#### Special Hazards

#### FLAMMABLE.

# Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Notify manufacturer or supplier. Notify environmental authorities.

# Protective Clothing and Equipment

In fires Respiratory protection - self-contained breathing apparatus; otherwise, protective outer clothing as required.

#### Fire and Explosion

Stop flow of gas. Most fire extinguishing agents may be used. Water should be used to cool fire-exposed containers. Containers may rupture violently in fires.

#### Pleat Ale

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing, flush affected areas with plenty of water and treat as for frostbite; eyes - irrigate with plenty of water. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water-Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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# NHIRIC ACID HNO:

#### IDENTIFICATION

UN No. 2032 furning 2031 >40% acid 1760 <40% acid

Canadian Industries Ltd., Carseland, Alta., Courtwright, Ont., McMasterville, Que. Cyanamid of Canada Ltd., Niagara Falls, Ont. Esso Chemical Canada, Redwater, Alta. Nitrochem, Maltland, Ontario.  Grades or Purity: 36°Bé, 52.3% HNO3; 40°Bé, 67.38% min. HNO3; 48.5°Bé, 95.1% min.
40°Bé, 67.38% min. HNO3; 48.5°Bé, 95.1% min.
40°Bé, 67.38% min. HNO3; 48.5°Bé, 95.1% min.
HNO3. Red furning (strengths >85.7); White furning (strengths >97.5).  Containers and Materials: Tank cars, tank trucks, drums; stainless steel, aluminum.
Colour: Colourless to light yellow. Explosive Limits: Not flammable. Meiting Point: 36°Bé, -19.5°C; 40°Bé, -24.5°C; 42°Bé, -33.0°C; 48.5°Bé, -52.0°C. Specific Gravity: 1.33 (36°Bé); 1.38 (40°Bé); 1.41 (42°Bé); 1.50 (48.5°Bé).

#### HAZARD DATA

#### Human Health

Symptoms: Inhalation: irritation of mucous membranes, difficulty breathing, coughing fits, nausea, and muscular weakness. Ingestion: irritation and burning, intense thirst, abdominal cramps, nausea and vomiting, difficulty breathing, convulsions and coma. Contact: skin - yellow discolouration, burning sensation, inflammation, blisters; eves - burning, watering, ulceration, loss of sight.

Toxicology: Highly toxic to skin eyes and mucous membranes.

TLV - (inhalation) 2 ppm; 5 mg/m³. Short-term Inhalation Limits - 4 ppm; 10 mg/m³ (15 min).

LCsn - Inhalation: rat = 65 ppm/4 h red turning (NO₂); inhalation: rat = 242 ppm/ 30 min white furning (NO₂)

LD50 - Oral: rat = 0.05 to 0.5 g/kg LDLo - Orait human = 0.43 g/kg Delayed Toxicity - No information.

Fire Extinguishing Agents: Not combustible. Use water spray in fires involving nitric acid. Water spray may be used to knock down vapours.

Behaviour in Fire: in fires toxic NO. Jumes are evolved.

lenition Temperature: Not combustible.

Burning Rate: Not combustible.

#### Reactivity

With Water: No reaction; very soluble. Mixes exothermically and may release toxic NO_X fumes.

With Common Materials: Strong exidizer, may react with many organics to cause fires or release toxic NO_x fumes. Reacts violently with acetic acid, acetic anhydride, (acetone & acetic acid), acetylene, acrolein, acrylonitrile, allyl alcohol, allyl chloride, ammonia, ammoniam hydroxide, aniline, arsine, calcium hypophosphite, carbon, chlorosulfonic acid, cresol, cumene, cyanides, cyclic ketones, cyclohexanol, cyclohexanone, epichlorohydrin, ethyl alcohol, ferrous oxide, fluorine, furturyal alcohol, hydrazine, hydrogen iodide, hydrogen peroxide, hydrogen sulfide, lithium, magnesium, manganese, nitrobenzene, oleum, phosphine, phosphorus, phosphorus trichloride, phthalic acid, phthalic anhydride, potassium hypophosphite, propylene oxide, pyridine, sodium, sodium hydroxide, sulfamic acid, (sulfuric acid & glycerides), titanium, vinyl acetate and zinc. Stability: Reactive and oxidizing.

Water: Prevent entry into water intakes and waterways. Fish toxicity: 72 ppm/96 h/mosquito fish/TLm/freshwater; Aquatic toxicity rating = 10 to 100 ppm/ 96 h/TLm; 750 pom/48 h/goldfish/LC190/freshwater; BOD: None. Food Chain Concentration Potential: None. Land-Air: No information.

# Special Hazards

CORROSIVE, OXIDIZER, POISON, Reactive.

#### immediate Responses

Keep non-involved people away from spill site. Issue warnings: "CORROSIVE; OXIDIZER; POISON". Call Fire Department. Avoid contact and inhalation. Contact manufacturer for guidance and assistance. Evacuate from downwind if oxides of nitrogen present. Stop or reduce discharge, if this can be done without risk. Contain spill by diking to prevent runoff. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and protective clothing. Gloves - rubber. Boots - high, rubber (pants worn outside boots).

### Fire and Explosion

Not combustible. Use water spray on fires involving nitric acid. Water spray may be used to knock down vapours. In fires, toxic NO_x fumes are evolved.

# First Aid

Move victim from spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact eyes - immediately irrigate with plenty of water; skin - remove contaminated clothing and flush with plenty of water. Ingestion do not induce vomiting. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or
- other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- Soils contaminated by nitric acid might be neutralized with time (environmental authorities' approval required).
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal of material.
- 2. Contact environmental authorities for advice on disposal.

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# NITRILOTRIACETIC ACID N(CH2COOH)3

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manulacturers
AMINOTRIACETIC ACID TRIGLYCINE NTA TRIGLYCOLLAMIC ACID N,N - bis (CARBOXYMETHYLGLYSINE)	White solid. Odourless.	Clough Chemical Co. Ltd., St. Jean, Quebec.
Transportation and Storage Information		
Shipping State: Solid. Classification: Not regulated. Inert Atmosphere: No requirement. Venting: Open.	LabeKs): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Commercial 99.5 +%. Containers and Materials: Cans, drums; steel.
Physical and Chemical Characteristics	,	
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.13 g/100ml. (22.5°C) Molecular Weight: 191.2 Vapour Pressure: No information. Boiling Point: Decomposes 240°C.	Ficatability (Water): Sinks. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: >1 at 20°C (solid).	Colour: White. Explosive Limits: Not flammable. Melting Point: Decomposes 240°C.

Human Health		
Symptoms: Contact: skin and eyes - irritation - li Texicology: No information. TLV No information. Short-term inhalation Limits - No information.	ttle information available. LC50 - No information. Delayed Toxicity - Suspected carcinogen.	LD ₅₀ - Oral: rat = 1.47 g/kg
Fire		
Fire Extinguishing Agents: Most fire extinguishing Behaviour in Fire: Releases toxic fumes when hea ignition Temperature: No information.	agents may be used on fires involving nitrilotriace ted. Burning Rate: No information.	etic acid.
Reactivity		
With Water: No reaction. With Common Materials: No information. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information.	erways. Fish toxicity: 346 ppm/24 h/guppy/lethal	concentration/freshwater; BOD: 0%, 5 days.

#### Special Hazards

# Immediate Responses

Keep non-involved people away from spill site. Call Fire Department. Avoid contact and inhalation. Call manufacturer. Call environmental authorities.

#### Protective Clothing and Equipment:

In fires or confined spaces Respiratory protection - self-contained breathing apparatus; otherwise, protective clothing as required.

## Fire and Explosion

Most fire extinguishing agents may be used on fires involving nitrilotriacetic acid. Releases toxic fumes when heated.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - krigate with plenty of water. Ingestion: give conscious victim water to drink. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom rediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Remove materials by manual or mechanical means.
- 5. Recover undamaged containers.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# NITROGLYCERINE CH2NO3CHNO3CH2NO3

IDENTIFICATION

UN No. 0143 (desensitized)

1204 (<1% nitroglycerin in alcohol)

Observable Characteristics Manufacturers Common Synonyms Canadian Industries Ltd., Colouriess to yellow liquid. 1.2.3-PROPANETRIOL TRINITRATE (C-I-L) Montreal, Quebec. DuPont of Canada Ltd., Montreal, Ouebec. GLYCERYL TRINITRATE **BLASTING OIL** GLYCEROL, NITRIC ACID TRIESTER NITROGLYCEROL, BLASTING GELATIN TRINITROGLYCEROL, NITROL Transportation and Storage Information Shipping State: Liquid: Solid (liquid absorbed Label(s): Orange and black label -Grades or Purity: Inerted < 5% nitroglycerin: EXPLOSIVE: Class 1.1, Group D. White <1% solution nitroglycerin in alcohol. on inert materials). label - POISON; Class 6.1 (for desensitized). Containers and Materials: Special containers. Classification: Explosive, Poison. mert Atmosphere: No requirement. Red and white label - FLAMMABLE LIQUID: Class 3.2, Group II (for <1% nitroglycerin Ventine: Open. in alcohol). Physical and Chemical Characteristics Floatability (Water): Sinks. Colour: Colourless to yellow. Physical State (20°C, I atm): Liquid. Explosive Limits: Explosive. Odour: No information. Solubility (Water): 0.18 g/100 mL (20°C). Melting Point: 3 to 13°C. Molecular Weight: 227.1 Flash Point: Explosive. Vapour Pressure: 0.00025 mm Hg (20°C); Vapour Density: 7.8 0.0073 mm Hg (50°C); 0.098 mm Hg (80°C). Boiling Point: 260°C (explodes). Specific Gravity: 1.60 (20°C).

	HAZARD DATA	
Human Health		
Symptoms: inhalation: headache, flushing of skin, similar to inhalation. Contact: skin - readily absortoxicology: Highly toxic by all routes.	vomiting, dizziness, collapse, cyanosis, convu bed yielding symptoms similar to inhalation; e	disions, coma and respiratory paralysis. <u>Ingestions</u> symptoms eyes - burning and irritation.
TLV - (skin) 0.05 ppm; 0.5 mg/m ³ .  Short-term inhalation Limits - 0.1 ppm; 1 mg/m ³ (15 min).	LC50 - No information. Delayed Texicity - No information.	LD ₁₀ - No information. LD ₁₀ - Subcutaneous: rabbit = 0.4 g/kg
Fire		
Fire Extinguishing Agents: Material typically would Behaviour in Fire: Explodes. Ignition Temperature: 260°C explodes.	i explode in any situation involving fire.  Burning Rate: Detonation velocity 7 600 m/s	S
Reactivity		
With Water: No reaction. With Common Materials: No information Stability: Unstable, when shocked or exposed to be	at or flame, may explode.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No information		gna/LDLo/freshwater; BOD: No information.

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#### **EMERGENCY MEASURES**

Special Hazards

EXPLOSIVE. POISON. FLAMMABLE. Can be detonated by shock, heat or fire.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "EXPLOSIVE; POISON; FLAMMABLE". Evacuate area. Notify manufacturer and supplier. Notify environmental authorities.

### Protective Ciothing and Equipment:

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

# Fire and Explosion

Material typically would explode in any situation involving fire.

#### First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give conscious victim water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water-Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated

materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# NONYL PHENOL

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
NONYLPHENOL 2,6-DIMETHYL-4-HEPTYLPHENOL 0-NONYL PHENOL p-NONYL PHENOL	Pale yellow liquid. Medicinal odour.	Domtar (CDC Division), Longford Milis, Ontario. Hart Chemical, Guelph, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: None. Inert Atmosphere: No requirement. Venting: Open. Pump Type: No information.	Label(s): Not regulated. Storage Temperature: Ambient. Hose Type: No information.	Grades or Purity: 90% para-isomer plus 4% ortho-isomer and 5% 2,4-dinonylphenoi.  Containers and Materials: Drums, tank cars; steel, stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Insoluble. Molecular Weight: 220.3 Vapour Pressure: No information. Boiling Point: 290 to 304°C.	Floatability (Water): Floats. Odour: Medicinal. Flash Point: 141°C (c.c.). Vapour Density: 7.6 Specific Gravity: 0.95 (20°C).	Colour: Pale yellow. Explosive Limits: No information. Melting Point: -10 to 2°C.

Human Health		
Symptoms: Inhalation: sore throat, coughing, laboring the ingestion: sore throat, abdominal pain, nausea, different toxicology: Moderately toxic by Ingestion and cor	urrhea.	kin - severely irritating, redness, pain, burns; eyes - redness, pain;
TLV - No information.	LC 50 - No information.	LD50 - Oral: rat = 1.62 g/kg
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	==>4 - 1
Fire		
Fire Extinguishing Agents: Use alcohol foam, carl Behaviour in Fire: No information. Ignition Temperature: No information.	Burning Rate: No information.	may cause frothing.
Reactivity		
With Water: No reaction. With Common Materials: Can react with oxidizing Stability: Stable.	ş materials.	
Environment		
Water: Prevent entry into water intakes and wate Land-Air: No information. Food Chain Concentration Potentials None.	rways. Aquatic toxicity rating = 10 to 100 p	pm/96h/TLm/freshwater.

# Special Hazards

COMBUSTIBLE. CORROSIVE.

### mmediate Responses

Keep non-involved people away from spiil site. Call Fire Department. Avoid contact and inhalation. Notify manufacturer. Dike to contain material. Notify environmental authorities.

# Protective Clothing and Equipment

In fires or confined spaces - Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, protective clothing as required.

# Fire and Explosion

Use alcohol foam, carbon dioxide or dry chemical to extinguish. Water or foam may cause frothing.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalations</u> if breathing has stopped, give artificial respirations if laboured, give oxygen. <u>Contacts</u> skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. <u>Ingestions</u> give water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If Hoating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal
- and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# OILS, CRUDE

IDENTIFICATION

UN No. 1267

Common Synonyms	Observable Characteristics	Manufacturers
PETROLEUM CRUDE	Dark brown to black oily liquids.  Acrid, tarry, offensive odour.	Universally available.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Not regulated. Inert Atmosphere: No requirement. Venting: Open (ilame arrester). Pump Type: Centrifugal, gear, etc.	Label(s): None. Storage Temperature: Ambient. Hose Type: Neoprene, butyl, Viton.	Grades or Purity: Wide variety, depending on oil field where produced.  Containers and Materials: Tankers, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.901 mg/100 mL. Molecular Weight: Various (mixture of hydrocarbons). Vapour Pressure: >3 mm Hg (20°C) (typically). Solling Point: 150 to 300°C.	Floatability (Water): Floats. Odour: Acrid, tarry, offensive (If high in H ₂ S). Flash Point: (-10 to 20°C). Vapour Density: Variable (typically 3 to 5). Specific Gravity: 0.8-0.9 (15°C).	Colour: Dark brown to black. Explosive Limits: Variable 1 to 7%. Melting Point: Various (-60 to -20°C).

Human Health	
Symptoms: Inhalation: dizziness, headache. Ingestion: nausea and vomiting. Contact: skin - irritation; eyes - irritation.  Toxicology: Generally low toxicity if little H2S is present.  TLV ^Q (as a mineral oil mist) 5 mg/m ³ LD ₅₀ - No information.  Short-term Inhalation Limits - 10 mg/m ³ (15 min)  Delayed Toxicity - No information.  (as mineral oil mist).	
fire the second	
Fire Extinguishing Agents: Use dry chemical, foam or carbon dioxide. Water may be ineffective, but should be used to cool fire-exposed containers.  Behaviour in Fire: Can give off toxic fumes.  Ignition Temperature: Variable; typically >400°C. Burning Rate: 4 mm/min.	
Reactivity	
With Water: No reaction; insoluble. With Common Materials: May react with oxidizing materials. Stability: Stable.	
Environment	
Water: Prevent entry into water intakes and waterways. Fish toxicities: I to 30 mg/L/96 h/rainbow trout/LC50/treshwater; 330 mg/L/48 h/brown shrimp/LC50/saltwater. Land-Air: Fouling to landscape. Food Chain Concentration Potential: None.	!

# Special Hazards

COMBUSTIBLE.

### Immediate Responses

Keep non-involved people away from spill site. Call Fire Department. Stop or reduce discharge, if this can be done without risk. Contain spill, if possible, to prevent or reduce runoff and pollution. Notify supplier and appropriate environmental authorities.

# Protective Clothing and Equipment

Gogzles or face shield. Coveralls. Gloves - rubber. Boots - rubber.

# Fire and Explosion

Use dry chemical, foam or carbon dioxide to extinguish. Water may be ineffective, but should be used to cool containers exposed to fire or heat.

#### First Ald

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: skin - remove oil-soaked clothing. Wipe off as much oil as possible. Wash skin thoroughly with warm water (and soap, if available) for at least 15 minutes; eyes - irrigate with water; ingestion: give water to conscious victim to drink. Transport victim to hospital, doctor or clinic if necessary.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Collect by skimming, pumping off or sorbing on sorbent.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Prevent product from entering sewers.
- 5. Collect, pump off or use sorbents for residuals.
- 6. Collect recovered product in available containers.
- 7. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

#### Oleocal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).
- 4. Oil and slightly contaminated oil may be recycled in a refinery.

Colours Yellow-brown

#### IDENTIFICATION

**AVIATION GASOLINE** 

Common Synonyms

80, 100, 115, 130 (similar to gasoline,

see gasoline).

JP-4, JP-5, JP-6. Type A, Type A-1, Type B, Type C (turbofuel A-1, A, B, C), (Jet A-1, A, B, C)

Transportation and Storage Information

Shipping State: Liquid.

Classification: Flammable liquid. hert Atmosphere: No requirement.

Venting: Open (flame arrester).

Pump Type: Gear, centrifugal; steel, stainless

Molecular Weight: Variable (mixture of

hydrocarbons).

Boiling Point: (A, A-1) 163 to 259°C. (B) 72 to 235°C: (4. 5. 6) 121 to 260°C. Observable Characteristics

Yellow - brown liquids. Gasoline-like odours.

Labelish Red and black label - FLAMMABLE

LIOUID.

Storage Temperature: Ambient.

Hose Type: Neoprene, Viton, polyethylene.

Grades or Purity: Type A-1, type A, type B, type C; JP-4, JP-5, JP-6.

Containers and Materials: Drums, tank cars, tank

trucks: steel.

UN No. 1863

Manufacturers

Universally available.

### Physical and Chemical Characteristics

Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.003 mg/L (20°C).

Vapour Pressure: typically <93 mm Hg (20°C).

Floatability (Water): Float. Odour: Gasoline-like.

Flash Point: (A-1, A) 43 to 66°C; (B, 4) -23 to -1°; (5) 35 to 63°C; (6) 38°C (o.c.).

Vapour Density: typically 3 to 5

Specific Gravity: (A-1) 0.806; (A) 0.816; (B) 0.764

#### HAZARD DATA

# fiaman Health

Symptoms: Inhalation: dizziness, headache. Ingestion: nausea and vomiting. Contact: skin - irritation; eyes - irritation.

Toxicology: Low texicity.

TLV 300 ppm; 900 mg/m³ (gasoline). Short-term inhalation Limits - 500 ppm; 1 500 mg/m³ (gasoline).

LCsn - No information.

Delayed Toxicity - No information.

LDso - No information.

# Fire

Fire Extinguishing Agents: Foam, carbon dioxide or dry chemical. Water may be ineffective, but should be used to cool fire-exposed containers.

Behaviour in Fire: Flash back may occur along vapour trail. legition Temperature: JP-4: 240°C: JP-5: 246°C: Burning Rate: 4 mm/min.

3P-6: 230°C

#### Reactivity

With Waters No reaction.

With Common Materials: May react with oxidizing agents.

Stabilitys Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 2 ppm/96 h/bluegitl/LCsn/treshwater: 100 ppm/96 h/grass shrimp/LCsn/saltwater. Land-Air: No information.

Food Chain Concentration Potential: No information.

Special Hazards FLAMMABLE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". Call fire department. Eliminate all sources of ignition. Notify supplier. Stop or reduce discharge if safe to do so. Dike to prevent runoff. Motify environmental authorities.

## Protective Clothing and Equipment:

Gloves and Boots: neoprene, butyl rubber. Appropriate goggles or face shield.

# Fire and Explosion

the foam, carbon dioxide or dry chemical to extinguish. Water may be ineffective, but should be used to cool fire-exposed containers. Flashback may occur along vapour trail.

## First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with soap and water; eyes - irrigate with water. Ingestion: give conscious victim water to drink; do not induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain spill by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
  6. Adsorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

- I. Contact supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).
- 4. Oil and slightly contaminated oil may be recycled in a retinery.

# OILS, FUEL (distillates 1, 2 and 2D)

IDEN	1.0			1.3		а
LAKE		r.	л.	4.5	VЛ	٧

UN No. 1233 kerosene (fuel oil no. 1)

Common Synonyms	Observable Characteristics	Manufacturers
KEROSENE (FUEL OIL NO. 1) FUEL OIL NO. 1, 2, 2-D DIESEL OIL LIGHT HOME HEATING OIL Fuel Oil No. 2 and 2-D DIESEL OIL MEDIUM	Oily liquids. Light brown to brown colour. Characteristic diesel fuel-like odour.	Universally available.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid (kerosene). Inert Atmosphere: No requirement. Venting: Open (flame arrester). Pump Type: Gear or centrifugal. Steel or stainless steel.	Label(s): Red label - FLAMMABLE LIQUID (kerosene). Storage Temperature: Ambient. Hose Type: Neoprene, Viton, polyethylene.	Grades or Purity: Kerosene, diesel fuel 1-D, diesel fuel 2; diesel fuel 2-D. Containers and Materials: Drums, tank cars, tank trucks, tankers; steel.
Physical and Chemical Characteristics		
Physical state (20°C, 1 atm): Liquid. Solubility (Water): insoluble (about 30 ppm). Molecular Weight: Variable. Vapour Pressure: <1 mm Hg (20°C). Boiling Point: 150 to 350°C.	Floatability (Water): Float.  Odour: Characteristic diesel-like (about 0.1 ppm, odour threshold).  Flash Point: (Fuel 1) 43 to 72°C; (Fuel 2) 52 to 96°C. Vapour Density: about 3 to 5  Specific Gravity: 0.81 to 0.90 (20°C).	Colour: Light brown to brown.  Explosive Limits: 0.7 to 5% (fuel 1).  Melting Point: -18 to -46°C.

Human Health				
Symptoms: Inhalation: dizziness, headache. Ingestion: nausea and vomiting. Contact: skin - irritation; eyes - irritation.  pneumonitis. Dermatitis may result from prolonged and repeated skin exposure.				
Texicology: Low toxicity by all routes. TLV ² (inhalation) 5 mg/m ³ (mineral oil particulate mist). Short-term inhalation Limits - 10 mg/m ³ for 15 min (oil particulate mineral mist).	LC50 - No information. Delayed Toxicity - No information.	LD50 - Oral: rat = 28 g/kg LD50 - Oral: rabbit = 0.2 g/kg		
fae				
Fire Extinguishing Agents: Foam, carbon dioxide or dry chemical. Water may be ineffective but should be used to cool fire-exposed containers.  Behaviour in Fire: Flash back may occur along vapour trail.  Ignition Temperature: (Fuel 1) 210 °C;  Burning Rate: 4 mm/min.  (Fuel 2) 257 °C.				
Reactivity				
With Water: No reaction. With Common Materials: Reacts with oxidizing age Stability: Stable.	nts.			
Environment				
Water: Prevent entry into water intakes or waterways. Fish toxicity: 10 ppm/96 h/rainbow trout/lethal/LC50/freshwater; 95 to 135 ppm/96 h/bluegill/LC50/freshwater (fuel oil no. 2); 2 ppm/96 h/grass shrimp/LC50/saltwater; BOD: Not available.  Land-Air: No information.  Food Chain Concentration Potential: None.				

## Special Hazards

#### FLAMMABLE.

#### immediate Responses

Keep non-involved people away from spill site. issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Notify supplier. Stop or reduce discharge if this can be done without risk. Dike to contain spill and prevent runoff. Notify environmental authorities.

## Protective Clothing and Equipment

Gloves and Boots - neoprene, butyl rubber. Appropriate goggles or face shield.

# Fire and Explosion

Use foam, carbon dioxide or dry chemical to extinguish. Water may be ineffective, but should be used to cool fire-exposed containers. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalation</u>: give artificial respiration if necessary. Contact: skin - remove contaminated clothing follow by washing affected areas with soap and water; eyes - irrigate eyes with plenty of water. <u>Ingestion</u>: do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss discossl and cleanup of contaminated materials.

#### Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact supplier for advice on disposal-
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).
- 4. Oil and slightly contaminated oil may be recycled in a refinery.

# 300

# OILS, FUEL (Residual Oils or Fuels 4, 5 and 6 Bunker)

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
RESIDUAL FUEL OIL NO. 4 (Bunker A) RESIDUAL FUEL OIL NO. 5 (Bunker B) RESIDUAL FUEL OIL NO. 6 (Bunker C) BUNKER FUEL OIL	Heavy, oily, high-viscosity liquids. Brown to black. Characteristic diesel fuel-like odour.	Universally available.
Transportation and Storage information		
Shipping State: Liquid. Classification: None. Inert Atmosphere: No requirement. Venting: Open (flame arrester). Pump Type: Gear or centrifugal.	Label(s): Not regulated. Storage Temperature: Ambient or may be heated. Hose Type: Neoprene, Yiton, polyethylene.	Grades or Purity: Fuel oil (residual) No. 4, fuel oil (residual) No. 5, fuel oil (residual) No. 6 Containers and Materials: Drums, tank trucks tank cars, tankers; steel.
Physical and Chemical Characteristics	##	
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Insoluble, <10 ppm. Molecular Weight: Variable. Vapour Pressure: No information; <1 mm Hg, 20°C approximately). Boiling Point: 185 to 500°C.	Finatability (Water): Floats.  Odour: Characteristic diesel fuel-like (about 0.1 ppm, odour threshold).  Flash Point: (No. 4) 61 to 116°C; (No. 5 light) 69 to 169°C, (No. 5 heavy) 70 to 121°C; (No. 6) 66 to 132°C.  Vapour Density: Variable; 3 to 5  Specific Gravity: 0.90 to 1.05 (15.5°C).	Colour: Brown to black. Explosive Limits: 1 to 5% (approximate). Melting Point: -29 to 10°C.

Human Health				
Symptoms: Inhalation: dizziness, headache. Inge Texicology: Low toxicity by all routes. TLV ² (inhalation) 5 mg/m ³ (mineral oil mist). Short-term inhalation Limits - 10 mg/m ³ for 15 min (mineral oil particulate mist).	stion: nausea and vomiting. Contact: skin - LC50 - No information. Delayed Toxicity - No information.	irritation; eyes - irritation.  LD5g - No information.		
Fire				
Fire Extinguishing Agents: Dry chemical, foam or carbon dioxide. Water may be ineffective but may be used to cool fire-exposed containers.  Behaviour in Fire: in fixes, toxic fumes may be emitted.  Ignition Temperature: No.4 263°C; No.6 407°C.  Burning Rate: 4 mm/min (approx.).				
Reactivity				
With Water: No reaction. With Common Materials: May react with strong or Stability: Stable.	xidizing agents.			
Environment				
Water: Prevent entry into water intakes and water 26 ppm/96 h/grass shrimp/CL5g/saltwater; 21 ppm Land-Air: No information. Food Chain Concentration Potential: None.	rways. Fish toxicity: 40 to 190 ppm/96 h/rai n/96 h/mosquito fish/LC5g/freshwater; BOD:	nbow trout/LC50/ireshwater. Fuel oil no. 6, Not available.		

# Special Hazards

FLAMMABLE, COMBUSTIBLE.

## mmediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Notify supplier. Stop or reduce discharge if this can be done without risk. Dike spill area to contain runoff. Notify environmental authorities.

# Protective Clothing and Equipment

Gloves and Boots - neoprene, butyl rubber. Appropriate goggles or face shield.

# Fire and Explosion

Use foam, carbon dioxide or dry chemical to extinguish. Water may be ineffective, but should be used to cool fire-exposed containers. In fires, toxic fumes may be emitted.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: remove from exposure and give artificial respiration if necessary. Contact: skin - remove contaminated clothing and wash affected areas with soap and water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and
- cleanup of contaminated materials.

- 1. Contact supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. incinerate (environmental authorities' approval required).
- 4. Oil and slightly contaminated oil may be recycled in a refinery.

# OLEUM H2SO4 + SO3

MENTIFICATION

11A1 NA 1931

	EDENTIFICATION	UN NO. 1831
Common Synonyms	Observable Characteristics	Manufacturers
FUMING SULPHURIC ACID SULPHURIC ACID, FUMING DISULPHURIC ACID (see sulfuric acid for dilute solutions).	Colourless to brown; heavy, oily liquid.  Sharp, choking odour. Fumes on contact with moist air.	Canadian Industries Ltd., Montreal, Quebec. Du Pont Canada Ltd., Montreal, Quebec. Cominco Ltd., Vancouver, B.C. Imperial Oil Ltd., Toronto, Ontario. International Minerals and Chemicals Corp. Ltd., Toronto, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Corrosive, poisonous liquid. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Centrifugal. Alloy 20, Durimet, Worthite.	Label(s): Black and white label - CORROSIVE; Class 8, Group I. White label - POISON; Class 6.1, Group I. Storage Temperature: Ambient. Hose Type: Up to 30%-polypropylene, Teflon- lined. Above 30%, flexible stainless steel or rigid piping with swivel joints.	Grades or Purity: 20 to 65% free SO3 (104.5 to 114.6% equivalent H ₂ SO ₄ ); technical, commercial.  Containers and Materials: Bottles, drums, tank cars, tank trucks; steel, stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Liquid. Solubility (Water): Soluble in water with vigorous reaction. Molecular Weight: Variable. Vapour Pressure: No information. Boiling Point: Decomposes.	Floatability (Water): Reacts vigorously, releasing heat. Odour: Sharp, choking (1 mg/m ³ odour threshold). Flash Point: Not flammable. Vapour Density: Approx. 2.8 (20°C). Specific Gravity: 20% 1.88; 30% 1.92; 40% 1.95; 65% 1.98 (38°C).	Colour: Colourless to dark brown, depending on purity. Explosive Limits: Not fiammable. Melting Point:-9°C, 20% free SO3; 15.5°C, 30% free SO3; 33.0°C, 40% free SO3; and 3.6°C, 65% free SO3.

HAZARD DATA				
Human Health				
Symptoms: Inhalation: sore throat, coughing, laboured breathing. Contact: eyes or skin - produces severe burns. Ingestion: sore throat, abdominal pain, nausea and vomiting.  Toxicology: Highly toxic by all routes.  TLV*- (inhalation) 1 mg/m³ (as sulfuric acid).  Short-term inhalation Limits - No information.  Delayed Toxicity - None known.				
Fire				
Fire Extinguishing Agents: Not combustible. Use Behaviour in Fire: Not combustible. Toxic $SO_X$ full ignition Temperature: Not combustible.	dry chemical on adjacent fires. Avoid use of water. mes are released in fires. Burning Rate: Not combustible.			
Reactivity				
acetonitrile, acrolein, acrylic acid, acrylonitrile, a	producing toxic SO _X fumes.  contact with combustible materials. Can react violently with reducing agents, acetic acid, acetic anhydride, allyl alcohol, allyl chloride, ammonium hydroxide, aniline, n-butyraldehyde, cresol, cumene, diethylene glycol, hydrochloric acid, hydrofluoric acid, isoprene, propylene oxide, pyridine, sodium hydroxide, stryene, vinyl			

Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life in very low concentrations. Aquatic toxicity = 10 to 100 ppm/96 h/TLm/freshwater; 43 ppm/48 h/prawn/LC5g/saltwater; 24 mg/L/24 h/bluegili/lethal/freshwater; BOD: None.

Land-Air: No information.

Food Chain Concentration Potential: None.

Stability: Stable (within the limits of the foregoing).

#### Special Hazards

CORROSIVE. POISON. Reacts with water. Reacts with many materials.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warnings: "CORROSIVE; POISON". Call Fire Department. Evacuate from downwind. Avoid contact and inhalation. Contact manufacturer. Contain spill area by diking. Stop discharge if possible. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit (acid suit, rubber). Gloves - gauntlet type, rubber. Boots - high, rubber spants worn outside boots).

#### Fire and Explosion

Not combustible. Use dry chemical on adjacent fires. Avoid use of water. Toxic SOx fumes are released in fires.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: eyes - irrigate immediately with plenty of water. If pain persists, continue irrigation; skin - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: give conscious victim plenty of water to drink. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### ENVIRONMENTAL PROTECTION MEASURES

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Remove material by manual or mechanical means.
- 6. Remove contaminated soil for disposal or neutralize
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# OXYGEN 02

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UN No. 1072 (compressed gas)

Common Synonyms	Observable Characteristics	Manufacturers
LOX LIQUID OXYGEN	Colouriess gas or liquid. Odouriess.	Canadian Liquid Air Ltd., Hamilton, Ontario. Union Carbide Canada Ltd., Sault SteMarie, Ontario. Inco, Copper Cliff, Ontario. Dofasco, Hamilton, Ontario.
Transportation and Storage Information		<del></del>
Shipping State: Gas (compressed); liquid (compressed or liquefied gas).  Classification: Nonflammable compressed gas; oxidizer.  Inert Atmosphere: No requirement.  Venting: Safety relief.	Label(s): Green label - NON-FLA MMABLE GAS; Class 2.2. Yellow Label - OXIDIZER; Class 5.1. Storage Temperature: Ambient. Hose Type: Braided, high pressure, degreased for oxygen service.	Grades or Purity: 99.5 + %. Containers and Materials: Cylinders, tank cars; steel (gas) cylinders, tank truck; steel (liquid).
Physical and Chemical Characteristics		· · · · · · · · · · · · · · · · · · ·
Physical State (20°C, 1 atm): Gas. Solubility (Water): 0.004 g/100mL (25°C). Molecular Weight: 32.0 Vapour Pressure: 1 200 mm Hg (-178°C); 31 600 mm Hg (-123°C). Boiling Point: -183°C.	Floatability (Water): Floats and boils (liquid). Odour: Odouriess. Flash Point: Not flammable. Vapour Density: 1.1 (25°C). Specific Gravity: Liquid, 1.14 (-183°C).	Colour: Colourless. Explosive Limits: Not flammable. Melting Point: -218.8°C.

Human Health		
Symptoms: Contact: with liquid may produce fromucous membranes due to dryness.  Toxicology: Relatively nontoxic.  TLV*- No information.  Short-term inhalation Limits - No information.	LC50 - No information.  TCL0 - Inhalation: human = 100%/14 h Delayed Toxicity - No information.	e generally no toxic effect, overdoses may cause irritation of LD50 - Not pertinent.
Fire		
Fire Extinguishing Agents: Stop flow of gas before may be used to cool fire-exposed containers.  Behaviour in Fire: Not flammable or combustible in lightion Temperature: Not flammable or combustible.	• •	
Reactivity		•
	prinated hydrocarbons and many fuels; carbon t	with aluminum compounds, ethers, hydrogen, trichloroethane, etrachloride, hydrazine, lithium hydride, magnesium, methane,
Environment		
Water: Little hazard. Land-Air: Little hazard. Food Chain Concentration Potential: None.		•

#### Special Hazards

OXIDIZER.

# immediate Responses

Keep non-involved people away from spill site. Issue warning: "OXIDIZER". Call Fire Department. Eliminate all sources of ignition. Call manufacturer or supplier. Avoid contact with liquid. Stop or reduce discharge if safe to do so. Notify environmental authorities.

## Protective Clothing and Equipment

Coveralls, gloves (asbestos or non-flammable) and goggles (mono, tight fitting). All exposed clothing must be ventilated thoroughly after exposure (30 min) to eliminate oxygen.

## Fire and Explosion

Stop flow of gas before attempting to extinguish fire. Most fire extinguishing agents may be used on fires involving oxygen. Water may be used to cool fire-exposed containers.

#### First Aid

Move victim out of spill area. Call for medical assistance, but start first aid at once. Inhalation: move victim to fresh air; apply artificial respiration if necessary. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; treat as for frostbite (liquid); eyes - irrigate with plenty of water. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water-Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Allow to disperse.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Material can be safely dispersed into the environment.

### PENTACHLOROPHENOL C6C15OH

**IDENTIFICATION** 

UN No. NA 2020

Common Trade Names and Synonyms Observable Characteristics Manufacturers **DOWICIDE 7** White to light-brown crystalline solid or Uniroval Chemicals Ltd.. **SANTOPHEN 20** powder. Weak odour. Strong pungent odour Edmonton, Alberta. PCP when heated. PENCHLOROL PENTA PERMATOX (Also see sodium pentachlorophenate). Transportation and Storage Information Shipping State: Solid. Label(s): White label - POISON. Grades or Purity: 86 to 100%. Classification: Poison. Storage Temperature: Ambient. Containers and Materials: Bags, fibre drums. Inert Atmosphere: No requirement. bulk by rail or truck. Venting: Open. Physical and Chemical Characteristics Physical State (20°C, I atm): Solid. Floatability (Water): Sinks. Colour: White to light brown. Solubility (Water): Slightly soluble. Odour: Weak. Strong pungent odour when Explosive Limits: Not flammable. 0.0005 g/100 mL (0°C); 0.0014 g/100 mL beated. Melting Point: 188 to 191°C. (20°C); 0.0035 g/100 mL (50°C). Molecular Weight: 266.4 Flash Point: Not flammable. Vapour Density: 9.2 Vapour Pressure: 0.00011 mm Hg (20°C): Specific Gravity: 1.98 (22°C). 40 mm Hg (211.2°C). Boiling Point: Decomposes at 310°C.

#### HAZARD DATA

## Human Health Symptoms: Inhalation: coughing, nausea, vomiting, abdominal cramps, sweating, fever, rapid pulse, convulsions, loss of consciousness. Contact: skin - may be absorbed, irritation, inflammation; eyes - irritation and watering. Toxicology: Highly toxic by all routes. TLV*- (skin) 0.5 mg/m³. LCsa - No information. LDso - Oral: rat = 0.050 g/kgShort-term Inhalation/Skin Limits ~ 1.5 me/m³ Delayed Toxicity - May cause liver and (15 min). kidney damage. Fire Fire Extinguishing Agents: Use water spray, dry chemical, foam or carbon dioxide. Use water to cool fire-exposed containers. Behaviour in Fire: When heated to decomposition, toxic fumes of chlorides are evolved. ignition Temperature: Not combustible. Burning Rate: Not combustible. Reactivity With Water: No reaction. With Common Materials: No information. Stability: Stable. Environment Water: Prevent entry into water intakes and waterways. Fish toxicity: 5 ppm/3 h/trout/tethal/freshwater: 0.24 mg/L/24 h/fathead minnow/LCsn/freshwater: 0.267 mg/L/24 h/goldfish/LCsn/ireshwater; EPA criteria: 6.2 mg/L/24 h (conc. should not exceed 14 ug/L fresh H2O at any time); BOD: Not available.

Land-Air: Toxic to plant life; 4 500 ppm/LC5g/mallard. Food Chain Concentration Potential: No information.

# Special Hazards

POISON.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Avoid contact and inhalation. Dike to prevent runoff from rainwater or water application. Notify supplier. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - in fires or enclosed spaces use self-contained breathing apparatus and totally encapsulated suit. Gloves - gauntiet type, rubber, unlined. Boots - rubber, high (pants worn outside boots).

### Fire and Explosion

Use water spray, dry chemical, foam or carbon dioxide to extinguish. Use water to cool fire-exposed containers. When heated to decomposition, toxic fumes of chlorides are evolved.

# First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin and eyes - remove contaminated clothing and wash eyes and affected skin with plenty of water. Ingestion: give water to conscious victim to drink and induce vomiting. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove liquids, contaminants and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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# PENTAERYTHRITOL C(CH2OH)4

# IDENTIFICATION

Common Trade Names and Synonyms	Observable Characteristics	Manufacturers
2,2-bis (HYDROXYMETHYL)-1,3 - PROPANEDIOL TETRAMETHYLOLMETHANE PENTAERYTHRITE TETRAHYDROXYMETHYLMETHANE PE mono PE, PENTEK TETRAKIS (HYDROXYMETHYL) METHANE	White solid. Odourless.	Celanese Canada, Edmonton, Alberta.
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Labei(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Technical: 86 to 90%; involving 10 to 40% di-pentaerythritol. Pure: 98+ %. Containers and Materials: Bags (multiwall paper); drums, cariots; steel.
Physical and Chemical Characteristics	· · · · · · · · · · · · · · · · · · ·	**************************************
Physical State (20°C, 1 atm): Solid. Solubility (Water): 5.56 g/100 mL (15°C). Molecular Weight: 136.2 Vapour Pressure: No information. Boiling Point: Sublimes (276°C).	Floatability (Water): Sinks and slowly mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: 4.7 Specific Gravity: 1.40 (25°C).	Colour: White. Explosive Limits: Not flammable. Melting Point: Pure, 260 to 269°C; test grade, 220-230°C (sublimes from 130°C at reduced pressure).

Human Health		
Symptoms: No information.  Texicology: Low order of toxicity.  TLV*- (nuisance particulate) 10 mg/m ³ .  Short-term inhalation - (nuisance particulate) 20 mg/m ³ .	LC50 - No information.  Delayed Toxicity - No Information.	LD50 - Oral: mouse = 25.5 g/kg LD50 - Oral: guinea pig = 11.3 g/kg
Fire Fire Extinguishing Agents: Most fire extinguishin Behaviour in Fire: No information. Ignition Temperature: 45°C (dust cloud).	g agents may be used on fires involving pentaery  Burning Rate: No information.	thritol.
Reactivity		
With Water: No reaction; slightly soluble. With Common Materials: May react with oxidizin Stability: Stable.	g materials. Reacts violently with acetic anhydr	ride and nitric acid.
Environment		
Water: prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: None.	erways. No information on fish toxicity available	e. BOD: 95 to 96%, 5 days.

# Special Hazards

COMBUSTIBLE.

#### buspediate Responses

Keep non-involved people away from spill site. Call Fire Department. Notify supplier or manufacturer. Dike to prevent runoff. Notify environmental authorities.

# Protective Clothing and Equipment

Protective outerwear as required.

# Fire and Explosion

Most fire extinguishing agents may be used on fires involving pentaerythritol.

# First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: move to fresh air. Contact: skin-remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water ingestion: give conscious victim water to drink and induce vomiting. Transport victim to hospital, doctor or clinic if necessary.

# ENVIRONMENTAL PROTECTION MEASURES

# Response

#### Water

- 1. Stop or reduce discharge if sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, 5. Recover undamaged containers. liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application. 4. Remove material by manual or mechanical means.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

#### Cimoral

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# PERCHLOROETHYLENE C12C:CC12

DEN	nfic	٨	TION
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UN No. 1897

Common Trade Names and Synonyms	Observable Characteristics	Manufacturers
TETRACHLOROETHYLENE CARBON BICHLORIDE ETHYLENE TETRACHLORIDE "per" DRY CLEANING FLUID	Colouriess liquid. Sweet, ethereal odour.	Dow Chemical Canada Inc., Sarnia, Ontario. Canadian Industries Ltd. (C-I-L), Shawinigan, Quebec.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Poison. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Centrifugal or positive displacement; steel or stainless steel.	Label(s): Black and white label - POISON; Class 6.1, Group III. Storage Temperature: Ambient. Hose Type: Seamless stainless steel, Teflon, bronze. Viton and cross-linked polyethylene may be used with caution.	Grades or Purity: Dry cleaning and industrial, 95+%. Containers and Materials: Cans, drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.015 g/100 mL (25°C). Molecular Weight: 165.8 Vapour Pressure: 14 mm Hg (20°C); 24 mm Hg (30°C), 45 mm Hg (40°C). Boiling Point: 121°C.	Ploatability (Water): Sinks. Odour: Sweet etheral (5 to 50 ppm, odour threshold). Flash Point: Not flammable. Vapour Density: 5.8 Specific Gravity: 1.63 (20°C).	Colour: Colouriess.  Explosive Limits: Not flammable.  Melting Point: -22.4°C.

	HAZARD DATA	
Human Health		
irritation, nausea, vomiting, diarrhea, drows inflammation.	es and throat, headache, intoxication, narcosis, iness, narcosis. <u>Contact</u> : skin – absorbed, dry	loss of consciousness and coma; Ingestion: gastrointestinal skin, inflammation, blisters; eyes - irritation, watering,
Toxicology: Moderately toxic by all routes. TLV*- (skin) 50 ppm; 335 mg/m ³ .	LC 50 - 5 040 ppm (8 h) rat.	LDsn - Oral: rat = 2.60 g/kg
Short-term Inhalation Limits - 200 ppm;	TCI o - Inhalation: human = 10 000 mg/m ³ .	
1 340 mg/m ³ .	Delayed Toxicity - May cause liver and	
	kidney damage.	
Behaviour in Firet When heated to decompositi ignition Temperature: Not combustible. Reactivity	Burning Rate: Not combustible.	
With Water: No reaction. With Common Materials: Reacts violently with Stability: Stable.	barium, beryllium and lithium.	
Environment		
Water: Prevent entry into water intakes and w Land-Air: No information. Food Chain Concentration Potential: Known to	aterways. Aquatic toxicity rating = 10 to 100 ppm/9	6 h/TLm/freshwater; BOD: 5 to 6%, 5 days.

#### Special Hazards

POISON.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Avoid contact and inhalation. Notify manufacturer or supplier. Shut off leak if safe to do so. Dike to contain material. Notify environmental authorities.

### Protective Clothing and Emipment

In lives or enclosed spaces, Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, outer protective clothing as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving perchloroethylene. Water may be used to cool fire-exposed containers. When heated to decomposition, emits toxic fumes of chlorides.

#### First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give conscious victim plenty of water to drink; do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminated and place in appropriate containers. bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal
- and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# PHENOL C6H50H

	ATION

UN No. Solid 1671

Common Synonyms	Observable Characteristics	Manufacturers
CARBOLIC ACID OXYBENZENE HYDROXYBENZENE PHENIC ACID PHENYLIC ACID PHENYL HYDROXIDE	White to pink solid. Liquid at temperatures above 41°C. Sharp, medicinal, sweet, tarry odour.	Guif Oil Canada Ltd., Montreal, Quebec. Dow Chemical Canada Inc., Delta, B.C.
Transportation and Storage Information		
Shipping State: Solid or liquid (molten). Classification: Poison. Inert Atmosphere: No requirement. Venting: Pressure-vacuum.	Label(s): White label - POISON; Class 6.1, Group II. Storage Temperature: Ambient, solid; 54°C, molten.	Grades or Purity: Technical: 82 to 92%, containing cresol.  Containers and Materials: Drums, tank cars, tank trucks; stainless steel, lined carbon steel, nickel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): In all proportions at 66°C; 8.2 g/100 ml. (15°C). Molecular Weight: 94.1 Vapour Pressure: 0.2 mm Hg (20°C); 0.35 mm Hg (25°C); 1 mm Hg (40°C). Boiling Point: 182°C.	Floatability (Water): Floats and mixes. Odour: Sharp, medicinal, sweet, tarry. (0.005 to 0.5 ppm odour threshold). Flash Point: 85°C (o.c.), 79°C (c.c.). Vapour Density: 3.2 Specific Gravity: 1.07 (20°C).	Colour: White to pink. Explosive Limits: 1.8 LEL. Melting Point: 41°C.

#### HAZARD DATA

		533

Symptoms: Inhalation: irritation of the nose, mouth, throat, difficulty breathing, headache, nausea, weakness, serious injury. Contact: skin - readily absorbed, severe burning, inflammation, painful blisters, unconsciousness, death; eyes - stinging, watering, severe burning, loss of vision. Ingestion: irritation and burning in throat, pain in swallowing, intense thirst, abdominal cramps, difficulty breathing, convulsions, coma, death.

Toxicology: Highly toxic by all routes. TLV - (skin) 5 ppm; 19 mg/m³.

Short-term Inhalation Limits - (skin)

LC50 - No information.

Delayed Toxicity - Causes liver and kidney damage. LDLo - Oral: human = 0.14 g/kg

LD50 - Oral: rat = 0.414 g/kg

Fire Extinguishing Agents: Use water spray, carbon dioxide, dry chemical or foam.

Behaviour in Fire: Emits toxic (phenoi) vapours when heated.

ignition Temperature: 715°C.

10 ppm; 38 mg/m³ for 15 min.

Burning Rate: 3.5 mm/min.

Reactivity

With Water: No reaction; soluble.

With Common Materials: Can react with oxidizing materials. Reacts violently with (aluminum chloride and nitrobenzene), butadiene and calcium hypochlorite. When hot, reacts with aluminum, magnesium, lead and zinc.

Stability: Stable.

Environment

Water: Harmful to aquatic life in very low concentrations. Prevent entry into water intakes and waterways. Aquatic toxicity rating: 10 to 100 ppm/96 h/TLm/ freshwater; 23.5 ppm/14 h/shrimp/LC30/saltwater; 9 mg/L/1 h/perch/LC30/freshwater; 46 mg/L/24 h/goldfish/LC30/freshwater; BOD: 140 to 180%, 5 days. Land-Air: No information.

Food Chain Concentration Potential: None.

#### Special Hazards

POISON.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Call Fire Department. Notify supplier. Avoid contact and inhalation. Stop or reduce discharge, if this can be done without risk. Dike spill to contain water runoff. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing aparatus and totally encapsulated suit. Boots - rubber. Gloves - rubber.

## Fire and Explosion

Use water spray, carbon dioxide, dry chemical or foam to extinguish. Emits toxic phenol vapours when heated.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin and eyes - flush immediately with plenty of water for at least 30 minutes while removing contaminated clothing. Ingestion: give conscious victim water to drink; Induce vomiting immediately. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water Sinking

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming
- or water diversion.

  4. Bredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

### Floating

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. Skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

### Land-Air

- 1. Stop or reduce discharge if safe to do so and using protective equipment.
- 2. Contact manufacturer or supplier for advice.
- 3. Centain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or synthetic sorbents.
- 7. Remove contaminated soil for disposal.
- 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- i. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# PHOSGENE COCI2

# IDENTIFICATION

UN No. 1076

Common Synonyms	Observable Characteristics	Manufacturers
CARBONYL CHLORIDE CARBON OXYCHLORIDE CHLOROFORMYL CHLORIDE CARBONIC DICHLORIDE	Colourless to light yellow gas. Odour sweet/sharp, like newly mown grass in dilute concentrations. Strong and stifling in heavy concentrations.	No Canadian manufacturer. U.S. manufacturers: Allied Chemical, Morristown, N.J. Dow Chemical, Midiand, MI Uplohn, Kalamahzoo, Mi
Note: PHOSGENE gas is a common combustion product resulting from the burning of many . chlorinated compounds.		Opposity (variations), int
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Poison and corrosive. Inert Atmosphere: No requirement. Venting: Safety relief.	Label(s): White label - POISON; Class 2.3. White and black label - CORROSIVE; Class 8. Storage Temperature: Ambient.	Grades or Purity: Commercial, 99.0%. Containers and Materials: Cylinders; steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Gas. Solubility (Water): (moisture) Decomposes to produce CO2 and HCl. Molecular Weight: 98.9 Vapour Pressure: 568 mm Hg (0°C); I 215 mm Hg (20°C); I 418 mm Hg (25°C); I 672 mm Hg (30°C). Boiling Point: 7.6°C.	Floatability (Water): Liquid sinks and decomposes to produce CO ₂ and HCl.  Odour: Sweet/sharp, like newly mown grass (dilute) (0.1 to 1.0 ppm odour threshold). Stiffling (heavy conc.). Sense of smell may be fatigued.  Flash Point: Not flammable.  Vapour Density: 3.5  Specific Gravity: 1.37 (liquid) (20°C).	Colour: Colouriess to light yellow. Explesive Limits: Not flammable. Melting Point: -128°C.

## HAZARD DATA

	nacaru data	
Human Health	<u></u>	
Symptoms: Sense of smell may be fatigued, and finhalation: Dryness or burning in throat, numbres and possible death. Contact: skin - burning and in Toxicology: Highly toxic by inhalation and irritation TLY*- (inhalation) 0.10 ppm; 0.4 mg/m³. Short-term inhalation Limits - No information.	ss, chest pain, coughing, difficult breathing, cyanosi flammation: eyes - burning.	ly irritating, even when fatai concentrations are inhaled. is, headache, nausea, vomiting, pulmonary edema, shock  LD50 - No information.
Fire		
Behaviour in Fire: Not combustible. Toxic, corrosi Ignition Temperature: Not combustible.	water spray to cool fire-exposed containers and knoc we (CO, CO ₂ , Cl ₂ , HCl, COCl ₂ ) fumes are produced Burning Rate: Not combustible.	k down vapours. when heated to decomposition.
Reactivity	•	
With Water: Decomposes, to form hydrochloric ac With Common Materials: Reacts violently with al Stability: Stable.	id and carbon dioxide. uminum, isopropyl alcohol, potassium and sodium.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information.	rways. No information on fish toxicities. BOD: None	•

Food Chain Concentration Potential: None.

#### Special Hazards

POISON, CORROSIVE.

#### immediate Responses

Keep non-involved people away from spill site. Issue warnings: "POISON; CORROSIVE". Avoid contact. Evacuate area. Call Fire Department. Call manufacturer. Stop discharge if safe to do so. Stay upwind and use water spray to control vapour. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

## Fire and Explosion

Not combustible. Use water to cool fire-exposed containers and knock down vapours. Toxic, corrosive (CO, CO₂, Cl₂, HCl, COCl₂) fumes are produced when heated to decomposition.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration (not mouth-to-mouth method) if breathing has stopped, and oxygen if breathing is laboured. Keep victim warm and maintain absolute rest. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - brigate with plenty of water. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water-Land-Air

- 1. Contact manufacturer or supplier for advice.
- Stop or reduce discharge if sale to do so (should only be attempted by an experienced person).
- 3. Use water spray to control vapours.
- 4. Notify environmental authorities.

- I. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# PHOSPHORIC ACID H3PO&

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UN No. 1805

Common Synonyms	Observable Characteristics	Manufacturers
ORTHOPHOSPHORIC ACID	Colourless liquid. Odourless.	International Minerals and Chem. Corp., Port Maitland, Ont. Cominco Ltd., Trail and Kimberley, B.C. Esso Chemical Canada, Redwater, Alta. Belledune Fertilizer, Belledune, N.B.
Transportation and Storage Information		
Shipping State: Liquid (aqueous solution) or >100 % H3PO4 Classification: Corrosive. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Positive displacement, gear, centrifugal; stainless steet.	Label(s): Black and white label - CORROSIVE; Class 8, Group III. Storage Temperature: Ambient. If >80% may be heated Hose Type: Polyethylene, butyl, Hypaion.	Grades or Purity: Aqueous solutions from 35 to 85%; 75 and 85% most common; 100% crystalline solid. Containers and Materials: Carboys, drums, tank cars, tank trucks; stainless 316; rubber-lined tanks.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid (aqueous solution), solid (100%).  Solubility: (Water): Completely soluble.  548 g/100 mL (solute) (20°C); aqueous solutions soluble in all proportions.  Molecular Weight: 98 (solute).  Vapour Pressure: 6.4 mm Hg (21°C) 75%;  2.2 mm Hg (21°C) 85%; 0.0285 mm Hg (20°C) (100%).  Boiling Point: 135°C (75%); 158°C (85%);  260°C (100%); at 213°C decomposes.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: 3.4 Specific Gravity: 1.58 (75%); 1.88 (80%); 1.69 (85%); 1.83 (100%) (20°C).	Colour: Colouriess liquid or transparent solid.  Explosive Limits: Not flammable.  Melting Point: 2i.1°C (85%); 42°C (100%); -17.5°C (75%), 4.6°C (80%).

Human Health		
Symptoms: Inhalation: irritation of mucous merabdominal pain, nausea; Contact: skin - burning Toxicology: Moderately toxic by ingestion and of TLV*- (inhalation) i mg/m³.  Short-term Inhalation Limits - 3 mg/m³ (15 min).	, inflammation, blisters; eyes - Durning, watering.	LD50 - Oral: rat = 1.53 g/kg
Fire Fire Extinguishing Agents: Not combustible. U Behaviour in Fire: Not combustible. When heat Ignition Temperature: Not combustible.	se water in fires involving phosphoric acid. Use well to decomposition, emits toxic POx fumes.  Burning Rate: Not combustible.	vater to cool fire-exposed containers and knock down vapours.
Reactivity		
With Water: Soluble with mild exothermic reac With Common Materials: Reacts with some me Stability: Stable.	tion. tals to liberate flammable hydrogen gas.	
Environment		
Water: Prevent entry into water intakes and water ppm/96 h/TLm/freshwater; BOD: None. Land-Air: No information. Food Chain Concentration Potential: None.	aterways. Fish toxicity: 138 ppm/24 h/mosquito fi	sh/TLm/freshwater; Aquatic toxicity rating = 100 to 1 000

# Special Hazards

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Avoid contact. Stop or reduce discharge, if this can be done without risk. Contain spill by diking to prevent runoff. Notify supplier and environmental authorities.

## Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus protective clothing. Otherwise, Goggles - (mono), tight fitting. Gloves - rubber or plastic coated. Boots - high, rubber (pants worn outside boots).

### Fire and Explosion

Not combustible. Use water in fires involving phosphoric acid. Use water to cool fire-exposed containers and knock down vapours. When heated to decomposition, toxic POx fumes are emitted.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin and eyes - remove contaminated clothing and flush skin and eyes with plenty of warm water. Ingestion: give conscious victim water or milk to drink. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material with pumps or vacuum equipment
- and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Neutralize soil with lime.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Neutralization of HaPOs with a suitable alkali (lime) will provide a product which could be buried in a landfill site or placed in other appropriate areas.
- 3. Contact environmental authorities for advice on disposal.

# PHOSPHORUS, RED P4

HAR	JTIFK	CATION	ŝ

UN No. 1338

Common Synonyms	Observable Characteristics	Manufacturers
AMORPHOUS PHOSPHORUS	Reddish-brown powder. Odourless.	Erco Industries Limited, Long Harbour, Nfid., Varennes, Que.
Transportation and Storage Information		
Shipping State: Solid. Classification: Flammable solid. Inert Atmosphere: No requirement. Yenting: Open.	Label(s): White and red striped label - FLAMMABLE SOLID; Class 4.1. Storage Temperature: Ambient.	Grades or Purity: Technical (99.9%). Containers and Materials: Cans, drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Solid. Solubility (Water): Insoluble in warm water. Extremely low solubility in cold water. Molecular Weight: 123.9 (P4). Vapour Pressure: I mm Hg (237°C). Boiling Point: 280°C (ignites at 202 to 280°C).	Floatability (Water): Sinks. Odour: Odourless. Flash Point: No information. Vapour Density: 4.8 Specific Gravity: 2.2 to 2.3 (20°C).	Colour: Reddish-brown. Explosive Limits: No information. Melting Point: Sublimes at 398 to 416°C; melts at 595°C; ignites at 202 to 280°C.

Human Health	•	
Symptoms: inhalation: (dust) irritation of respirate Toxicology: Low toxicity by all routes. TLV [®] No information. Short-term inhalation Limits - No information.	ry tract. <u>Contact</u> : eyes - irritation. LC5g - No information. Delayed Toxicity - No information.	LD50 - No information. LD _{LO} - Oral: man = 4.41 g/kg
Fire		
Fire Extinguishing Agents: Use water to extinguish Behaviour in Fire: Heat may cause reversion to wirritating oxides of phosphorus (PO _x ). Ignition Temperature: 202 to 280°C.	<ul> <li>ite (yellow) phosphorus, which is toxic and sport Burning Rate: No information.</li> </ul>	ntaneously flammable upon contact with air. Burning yields
Reactivity		
With Water: No reaction. With Common Materials: Can react with reducing Stability: Stable.	materiais.	
Environment		
Water: Prevent entry into water intakes and water BOD: Not available. Land-Air: No information. Food Chain Concentration Potential: None.	ways. Fish toxicity = 0.105 mg/L/48 h/bluegill	/TLm/freshwater;

#### Special Hazards

#### FI AMMABLE.

# Inmediate Response

Keep non-involved people away from split site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Stop or reduce discharge, if this can be done without risk. Dike to prevent runoff from rainwater or water application. Notify manufacturer and environmental authorities.

### Protective Clothing and Equipment

In fire, Respiratory protection - self-contained breathing apparatus protective clothing. Otherwise, Goggles - (mone), tight fitting. Gloves - rubber.

# Fire and Explosion

Use water to extinguish. Heat may cause reversion to white (yellow) phosphorus which is toxic and spontaneously flammable upon contact with air. Burning yields irritating oxides of phosphorus (POx).

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin and eyes - remove contaminated clothing and wash eyes and skin thoroughly with plenty of water. Ingestion: give milk or water to conscious victim to drink. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### ENVIRONMENTAL PROTECTION MEASURES

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminated 4. Remove material by manual or mechanical means. bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# PHOSPHORUS, WHITE OR YELLOW PA

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UN No. 1381

Common Synonyms	Observable Characteristics :	Manufacturers
WHITE PHOSPHORUS YELLOW PHOSPHORUS WHITE DRY PHOSPHORUS	White to yellow, waxy solid. Odourless, but POx fumes from combustion have a garlic-like odour.	Erco Industries Limited, Varennes, Que., Long Harbour, Nfid.
Transportation and Storage Information		
Shipping State: Solid. Classification: Spontaneously combustible. Inert Atmosphere: Water cover. Venting: Pressure-vacuum.	Label(s): White and red label - SPONTANEOUSLY COMBUSTIBLE; Class 4.2 Storage Temperature: Ambient.	Grades or Purity: Technical, 99.9%. Containers and Materials: Cans, drums, tank trucks, tank cars (ship under water).
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.0003 g/100 mL (15°C). Molecular Weight: 123.9 (P4). Vapour Pressure: 0.028 mm Hg (21°C). i mm Hg (76.6°C). Boiling Point: 280°C.	Floatability (Water): Sinks. Odour: Odourless. Flash Point: Ignites spontaneously in air. Vapour Density: 4.4 Specific Gravity: 1.82 (20°C).	Colour: White to yellow. Explosive Limits: No information. Melting Point: 44°C.

#### HAZARD DATA

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Symptoms: Inhalation: irritation of respiratory tract, cough, weakness, nausea and vomiting, laboured breathing, anemia. Contact: eyes and skin - may cause severe burns. Ingestion: causes nausea, vomiting, jaundice, delirhum, coma and death. Symptoms after ingestion may be delayed from a few hours to 3 days.

Toxicology: Highly toxic by all routes. TLV*- (inhalation) 0.1 mg/m³.

Short-term Inhalation Limits - 0.3 mg/m³

(15 min).

LCsa - No information.

Delayed Toxicity - Severe attack of liver and bones. LDLo - Orali human = 0.0014 g/kg

Symptoms may persist from 5 to 15 days.

LDsn - intraperitoneal: rat = 0.1 g/kg

#### Fire

Fire Extinguishing Agents: Flood with water. After fire is extinguished cover with wet sand or dirt. Use extreme caution during cleanup, since reignition may

Behaviour in Fire: intense white smoke is formed and highly irritating and toxic POx fumes are released. Burning Rate: No information. ignition Temperature: 30°C.

#### Reactivity

With Water: No reaction.

With Common Materials: Ignites when exposed to air. Reacts vigorously with oxidizing materials. Reacts violently with alkaline hydroxides, halogens, chlorates, iodates, chlorosulfonic acid, nitrates, ammonium nitrate, bromates, bromine, calcium chlorate, ceslum, chlorine, chromic acid, chromic anhydride, copper, fluorine, iodine, iron, lead dioxide, lithium, manganese, mercuric oxide, nitric acid, oxygen, performic acid, platinum, potassium bromate, potassium chlorate, potassium hydroxide, potassium iodate, potassium permanganate, silver nitrate, sodium, sodium bromate, sodium chlorate, sodium chlorite, sodium hydroxide, sodium lodate, sodium peroxide, sulfur, sulfuric acid, zinc bromate and zinc chlorate.

# Stability: Ignites spontaneously on contact with air and is very reactive.

#### Environment

Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating: <1 ppm/96 h/TLm/freshwater; 0.045 ppm/96 h/bluegitl/LC50/freshwater; 0.025 ppm/96 h/Cod/LC50/saltwater.

Land-Air: Oral: duck = 3 mg/kg (6 to 33 hours) LD100-

Food Chain Concentration Potential: None.

# Special Hazards

SPONTANEOUSLY COMBUSTIBLE. REACTIVE. Ignites spontaneously when exposed to air.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warnings: "FLAMMABLE; POISON". Call Fire Department. Call manufacturer. Stop or reduce discharge, if this can be done without risk. Dike area of spill and cover with water. Notify environmental authorities.

# Protective Clothing and Equipment

In fires or confined spaces, Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, Goggles - (mono), tight fitting. If face shield is used, it must not replace goggles. Gloves - rubber. Boots - high, rubber (pants worn outside boots). Fire-resistant clothing (coveralls, suit, apron, etc.).

# Fire and Explosion

Flood discharge area with water (after diking). When fire is extinguished, cover with wet sand or earth. Cover fires with water spray, dry chemical, sand or earth. Use water to cool fire-exposed containers. Highly irritating and toxic PO_X fumes are released.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid immediately. Inhalation: If breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: immediately flush skin and eyes with plenty of water while removing contaminated clothing. Keep skin area wet until medical attention is obtained. Ingestion: if victim is conscious, give plenty of milk or water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or
- water application.
  4. Remove material by manual or mechanical means.
- Note: material requires special handling because of its spontaneous ignition in air. Soilled material should be covered at all times
- with wet sand or dirt.
- 5. Recover undamaged containers.
- 6. Motify environmental authorities to discuss discussing and cleanup of contaminated materials.

- I. Contact manfacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# PHOSPHORUS PENTASULFIDE (PoSs)2

UN No. 1340

Соттол Ѕунопутв	Observable Characteristics	Manufacturers
TETRAPHOSPHORUS DECASULPHIDE PHOSPHORIC SULPHIDE PHOSPHORUS PERSULFIDE THIOPHOSPHORIC ANHYDRIDE PHOSPHORUS SULPHIDE	Yellow to green solid. Hydrogen suifide- like (rotten eggs) odour.	No Canadian manufacturer. Selected U.S. manufacturer: Hooker Industrial Chemicals Division, Niagara Falls, N.Y., USA.
Transportation and Storage Information		
Shipping State: Solid. Classification: Fiammable. Inert Atmosphere: No requirement. Venting: Closed.	Label(s): Red and white striped label - FLAMMABLE SOLID; Class 4.1, Group II. Storage Temperature: Ambient.	Grades or Purity: Regular and reactive, 99+%. Containers and Materials: Cans, fibre or steel drums and tote bins.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): Reacts evolving H ₂ S and forming soluble phosphorus compounds. Molecular Weight: (P ₄ S ₁₀ ) 444.5 Vapour Pressure: 1 mm Hg (300°C). Boiling Point: 513 to 515°C.	Floatability (Water): Sinks and reacts, toxic H ₂ S gas evolved. Odour: Rotten-eggs (0.00021 to 0.00047 ppm as H ₂ S, odour threshold). Flash Point: Data not available but autoignition temperatures as dust; 260 to 290°C, and liquid; 275°C (approx.). Vapour Density: No information. Specific Gravity: 2.03 (17°C).	Colour: Yellow to green. Explosive Limits: Dust can be explosive in certain concentrations. Can be ignited by friction. Melting Point: 275 to 290°C.

# HAZARD DATA

	Symptoms: Contact of phosphorus pentasulfide with water or moisture produces H2S which is highly	ŧ
۱	headache, fatigue, difficulty breathing, kritation of respiratory tract. Contact: skin - irritation and burni	ng

toxic (see hydrogen sulfide). Inhalation: dizziness, ng: eyes - irritation and burning. Ingestion: produces symptoms similar to inhalation; however ingestion is extremely serious, as H2S is formed.

Toxicology: Highly toxic by all routes.

TLV - (inhalation) | mg/m3.

Short-term Inhalation Limits - 3 mg/m3

LCso - inhalation: rat = 713 ppm/l h (as H2S).

LD50 - Oral: rat = 0.39 g/kg

Delayed Toxicity - No information.

(15 min).

Human Health

Fire Extinguishing Agents: Carbon dioxide, dry chemical, sand.

Behaviour in Fire: When heated to decomposition, emits highly toxic SO_X and PO_X fumes. Reacts with water or moisture to produce toxic H₂S and P₂O₅ vapours. In contact with moisture, may ignite organic materials.

ignition Temperature: Dust, 260 to 290°C; liquid, 275°C.

Burning Rate: No information.

#### Reactivity

With Water: Reacts with water exothermically producing H2S gas and soluble phosphorus compounds (primarily P2O5). Can ignite spontaneously from heat of reaction with limited amount of moisture.

With Common Materials: Reacts with acids in a similar manner to water. Can react vigorously with oxidizing materials.

Stability: Stable (within the limits of the foregoing).

Water: Prevent entry into water intakes and waterways. Fish toxicity: 1.38 ppm/48 h/lathead minnow/TLm/freshwater (as H25).

Land-Air: No information.

Food Chain Concentration Potential: No information.

#### Special Hazards

FLAMMABLE. CONTACT WITH WATER RELEASES TOXIC AND POISONOUS H2S AND P2O5 GASES. MAY IGNITE IN PRESENCE OF MOISTURE.

#### Immediate Response

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE, POISONOUS AND REACTIVE WITH WATER". Call fire department (describe unusual properties - NO WATER). Evacuate area. Contact manufacturer for guidance and assistance. Notify environmental authorities.

#### Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Gloves - rubber, neoprene or vinyl. Boots - high, rubber (pants worn outside boots).

# Fire and Explosion

Use carbon dioxide, dry chemical or sand to extinguish. When heated to decomposition, emits highly toxic lumes of SO_X and PO_X. Reacts with water or moisture to produce toxic H₂S and P₂O₅ vapours. In contact with moisture, may ignite organic materials.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes and skin - flush affected areas with plenty of water. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Ingestion: give conscious victim water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Wate:

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove materials by manual or mechanical means. Caution must be observed when handling this material.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# PHOSPHORUS TRICHLORIDE PC13

IDENT	IFIC	۸	1к	ď	ú

UN No. 1809

Common Synonyms	Observable Characteristics	Manufacturers
PHOSPHORUS CHLORIDE CHLORIDE OF PHOSPHORUS	Colourless to slightly yellow liquid. Fumes on contact with air. Pungent, irritating, HCI-like odour.	No Canadian manufacturer. Selected U.S. manufacturer: Hooker Industrial Chemicals Division, Niagara Falls, NY, USA
Transportation and Storage Information		
Shipping State: Liquid. Classification: Corrosive. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Gear, centrifugal. Note: PC13 is corrosive to steel.	Label(s): White and black label - CORROSIVE; Class 8, Group II. Storage Temperature: Ambient. Hose Type: Nickel-lined stainless steel.	Grades or Purity: Technical, 98.5% PCl3 min. Containers and Materials: Drums, tank cars, tank trucks; nickel-lined stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Reacts to produce HCI and PH ₃ . Molecular Weight: 137.3 Vapour Pressure: 100 mm Hg (21°C). Boiling Point: 74.2 to 76.1°C.	Floatability (Water): Sinks and reacts violently with water to produce HCl and PH3 acids. Odour: Pungent (like HCl). Flash Point: Not flammable. Vapour Density: 4.8 Specific Gravity: 1.57 (20°C).	Colour: Colouriess to slightly yellow. Explosive Limits: Not fiammable. Melting Point: -112°C.

	HAZARD DATA	
Human Health		
Symptoms: Inhalation: sore throat, coughing, labor irritation, nausea, vomiting, burning. Textcology: Highly toxic by all routes.	-	skin - burning and inflammation; eyes - burning. <u>Ingestion</u> :
TLV - (irhalation) 0.2 ppm; 1.5 mg/m ³ . Short-term inhalation Limits - 0.5 ppm; 3 mg/m ³ (15 min).	I.C50 - Inhalation: rat = 104 ppm/4 h LC50 - Inhalation: guinea pig = 50 ppm/4 h Delayed Toxicity - No information.	LD50 - Oral: rat = 0.55 g/kg
Fire		
Fire Extinguishing Agents: Use carbon dioxide or d Behaviour in Fire: When heated to decomposition, ignition Temperature: No information. Reactivity		cts with water or moisture to produce HCl and PH3.
With Water: Reacts violently and exothermically with Common Materials: Reacts with oxidizing m dioxide, nitric acid, nitrous acid, organic matter, p Stability: Stable (within the limits of the foregoing	naterials. Reacts violently with acetic acid, a potassium and sodium.	luminum, dimethyl sulfoxide, fluorine, hydroxylamine, lead
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: None.	ways. Aquatic toxicity rating = 10 to 100 ppm/9	6 h/TLm/freshwater; BOD: Not available.

#### Special Hazards

CORROSIVE. REACTIVE WITH WATER.

#### Immediate Remonse

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department (describe ususual properties - NO WATER). Contact manufacturer for guidance, and assistance if possible. Avoid contact and inhalation. Contain spill area by diking. Stop or reduce discharge if this can be done without risk. Notify environmental authorities.

## Protective Clothing and Equipment

Resolvatory protection - self-contained breathing apparatus and totally encapsulated suit. Gloves - rubber. Boots - rubber, high (pants worn outside boots).

#### Fire and Explosion

Use carbon dioxide or dry chemical to extinguish. When heated to decomposition, emits highly toxic chloride and PO_X fumes. Reacts with water or moisture to produce HCl and PH₃.

#### First Ald

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: skin and eyes - immediately flush with large amounts of water and remove contaminated clothing. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Ingestion: dilute concentration of chemical by giving water or milk to conscious victim to drink. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means. Caution is required when handling materials.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal
- and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# PHTHALIC ANHYDRIDE C6H6(CO)2O

IDENTIFICATION

UN No. 2214

Manufacturers Observable Characteristics Common Synonyms BASF Canada, Cornwall, Ontario. White powder or crystals. Choking odour. **PHTHALANDIONE** PHTHALIC ACID ANHYDRIDE 1.3 - ISOBENZOFURANDIONE 1,2 - BENZENEDICARBOXYLIC ACID ANHYDRIDE Transportation and Storage Information Grades or Purity: Commercial, 99.8%. Labelish Black and white label -Shipping State: Solid. Containers and Materials: Multiwall paper CORROSIVE: Class 8, Group III. Cinssification: Corrosive. Storage Temperature: Ambient. bags, drums. Inert Atmosphere: No requirement. Venting: Open (flame arrester). Physical and Chemical Characteristics Floatability (Water): Sinks. Colour: White. Physical State (20°C, I atm): Solid. Solubility (Water): 1g/100mL (20°C): Odour: Choking (0.32 to 0.72 mg/m3, Explosive Limits: 1.7 to 10.5%. Melting Point: 130 to 132°C. odour threshold). Molecular Weight: 148.1 Flash Point: 152°C (c.c.). Vapour Pressure: 0.0002 mm Hg (20°C); Vapour Density: 5.1 0.001 mm Hg (30°C). Specific Gravity: 1.53 (20°C). Bolling Point: Sublimes 284.5°C.

# HAZARD DATA

	HAZARD DATA			
Symptoms: Inhalation: sore throat, irritation of nose, eyes and respiratory tract, coughing, difficulty breathing, cyanosis. Contact: skin - itching, irritation, inflammation, brown staining; eyes - stinging, watering, inflammation. Ingestion: burning sensation, stomach cramps, nausea and vomiting, general weakness, diarrhea, shock, convulsions, coma.  Toxicology: Moderately toxic by ingestion.  TLV 1 ppm, 6 mg/m ³ .  LC50 - No information.  LD50 - Oral: rat = 4.02 g/kg  Short-term inhalation Limits - 4 ppm; 24 mg/m ³ Delayed Toxicity - No information.				
Fire				
Fire Extinguishing Agents: Use carbon dioxide and/or dry chemical. Water or foam may cause frothing.  Behaviour in Fire: No information.  Ignition Temperature: 570°C.  Burning Rate: No information.				
Reactivity		•		
With Water: Slowly reacts to form benzoic acid. Stability: Reacts violently with nitric acid. With Common Materials: Stable.				
Environment				
Water: Prevent entry into potable water intakes. ppm/96 h/fathead minnow/TLm/freshwater; BOD: Land-Air: No information.	Harmful to aquatic life. Aquatic toxic 72 to 102%, 5 days.	ity rating = 10 to 100 ppm/96 h/TLm/freshwater; Fish toxicity: 756		

Food Chain Concentration Potential: None.

# Special Hazards

CORROSIVE.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department. Avoid contact and inhalation. Notify manufacturer or supplier. Dike to contain runoff. Notify environmental authorities.

### Protective Clothing and Equipment

In fires or confined spaces - Respiratory Protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, protective outer clothing as required.

### Fire and Explosion

Use carbon dioxide or dry chemical to extinguish. Water or foam may cause frothing.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with pienty of water; eyes - irrigate with pienty of water. Ingestion: give conscious victim pienty of water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor, or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming
- or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manfacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# POLYCHLORINATED BIPHENYLS CI2HIQ-xCIx

# IDENTIFICATION

UN No. 2315

	HATAIR CAIRM	0)1,100 2217
Common Trade Names or Synonyms	Observable Characteristics	Manufacturers
PCBs AROCHLOR ASKAREL PYRANOL INERTEEN	Clear to pale yellow mobile liquids and some solids. Weak, bitter odour.	Previous Canadian supplier: Monsanto Canada Limited, Mississauga, Ont. (PCBs no longer being manufactured)
Transportation and Storage Information	_	
Shipping State: Liquid or solid. Classification: Environmentally dangerous substance. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Most centrifugal types except those with rubber, neoprene, butyl rubber or PVC.	Label(s): Black and white label - POISON; Class 9.2. Storage Temperature: Ambient. Hose Type: Most types, except rubber, neoprene, butyl rubber or PVC. These should not be used after coming in contact with PCBs.	Grades or Purity: 11 grades (some liquid, some solid) which differ primarily in their chlorine content (20-68% by weight).  Containers and Materials: Drums. Electrical equipment.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid or solid. Solubility (Water): Low; 9.901 mg/100 ml. (freshwater); 9.0004 mg/100 ml. (saltwater). Molecular Weight: Variable; 327 (average). Vapour Pressure: 9.006 mm Hg (25°C) (Arochior 1016); 9.001 mm Hg (9°C); 29 mm Hg (200°C).	Floatability (Water): Sinks. Odour: Weak and bitter (odour threshold <0.5 mg/m²). Flash Point: Greater than 140°C. Vapour Density: No information. Specific Gravity: 1.3 to 1.5 (20°C). Boiling Point: 360 to 390°C.	Colour: Colouriess, to pale yellow. Explosive Limits: Not flammable. Melting Point: No information.

	HAZARD DATA	
Human Health		
remiting loss of weight, isundice, edem	pimples, dark pigmentation of exposed areas, followed a, abdominal pain, possible coma and death. Igher the chlorine content of the compound, the more to LC50 - No information.  TCL0 - Inhalation: human = 12 mg/m ³ Arochlor 1242 (42% chlorine)  Delayed Toxicity - Possible carcinogen; adverse reproductive effects and atrophy of the liver.	
Fire Extinguishing Agents: Water, foam, dry che Behaviour in Fire: Highly toxic, irritating gases ignition Temperature: "1000°C.	emical or carbon dioxide. Use water only if safe to do so (chlorides and chlorine) can be evolved. Burning Rate: No information.	
Reactivity With Water: No reaction.	With Common Materials: Reacts with liquid chloring	e. Stability: Stable.
Environment		
Water: Prevent entry into water intakes and water. Prevent entry into water intakes and water. ILm/freshwater; 0.005 ppm/336-1 080 h/pinfish 0.024 µg/L/24 h/conc. should not exceed 0.2 µg Land-Ale: LD50: 2 000 ppm (mailard duck). Food Chain Concentration Potential: Known to a		ions. Fish toxicity: 0.278 mg/L/96 h/bluegill/ onc. should not exceed 6.2 µg/L (freshwater);

#### Special Hazards

POISON.

#### bamedate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Call Fire Department. Avoid contact and inhalation. Contact supplier. Stop or reduce discharge if this can be done without risk. Contain any spillage. Notify environmental authorities.

#### Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Goggles, Boots - polyethylene, neoprene, Viton. Note: contaminated clothing should be disposed of in same manner as PCBs.

#### Fire and Explosion

Use water, foam, dry chemical or carbon dioxide to extinguish fires. Use water only if safe to do so. In fire, toxic gases (chlorides and chlorine) can be evolved.

#### First Aid

Move victim out of spill area to fresh air. Call for doctor, but start first aid at once. <u>Contact</u> - using appropriate gloves, remove contaminated clothing. Clothing should be disposed of in the same manner as PCBs. Wash eyes and affected skin thoroughly with plenty of water. <u>Ingestion</u> have conscious victim rinse mouth with water. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### ENVIRONMENTAL PROTECTION MEASURES

### Response

#### Water

- I. Notify environmental authorities immediately.
- 2. Stop or reduce discharge if safe to do so.
- 3. Contact manufacturer or supplier for advice.
- 4. If possible, contain discharge by damming or water diversion.
- 5. Bredge or vacuum pump to remove contaminants,
- liquids and contaminated bottom sediments.

  6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Notify environmental authorities immediately.
- 2. Stop or reduce discharge if safe to do so.
- 3. Contact manufacturer or supplier for advice.
- 4. Contain spill by diking with earth or other barrier.
- 5. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 6. Remove material by manual or mechanical means.
- 7. Absorb residual liquid on natural or synthetic sorbents.
- 3. Remove contaminated soil for disposal.
- 9. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 3

# POTASH (POTASSIUM CHLORIDE) KCI

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
POTASSIUM MURIATE MURIATE OF POTASH	Colourless to whitish-pink crystals. Odourless.	International Minerals and Chemicals, Esterhazy, Saskatchewan. Potash Corporation of Saskatchewan, Rocanville, Saskatchewan. Kaium Chemicals, Belle Plaine, Saskatchewan. Allan Potash Mines, Saskatoon, Saskatchewan.
Transportation and Storage Information		
Shipping State: Solid Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Industrial.  Containers and Materials: Multiwall paper bags.  Bulk by truck or train; steel.
Physical and Chemical Characteristics	•	
Physical State (20°C, 1 atm): Solid. Solubility (Water): 23.5 g/100 mL (20°C); 56.7 g/100 mL (100°C). Molecular Weight: 74.6 Vapour Pressure: No information. Boiling Point: 1420 to 1500°C (sublimes).	Pioatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not liammable. Vapour Density: No information. Specific Gravity: 1.99 (20°C).	Colour: Colourless to whitish-pink. Explosive Limits: Not flammable. Melting Point: 770°C.

	SHITUMEN CHILLS	
Human Health  Supertones Inhalation elight irritation of page	enserving Ingestions nauses and vomiting	. Contact: skin - irritation and inflammation; eyes - irritation,
inflammation and watering.	Strong. Mignitude innage out saiden.	
Toxicology: Low toxicity by all routes.		LD 50 - Oral: mouse = 0.383 g/kg
TLV No information.	LC ₅₀ - Ne information.  Delayed Toxicity - No information.	LD50 - Orak: guinea pig = 2.5 g/kg LD50 - Orak: Infant = 0.938 g/kg
Short-term Inhalation Limits - No information.	Delayed Toxicity - No anormation.	ED-20 - Orde: Birger - 41539 Et./2
Fire		
Fire Extinguishing Agents: Not combustible, Most Behaviour in Fire: Not combustible, Ignition Temperature: Not combustible,	fire extinguishing agents may be used on fi Burning Rate: Not combustible.	res involving potash.
Reactivity		
With Water: No reaction; soluble. Stability: Reacts violently with bromine trifluorid With Common Materials: Stable.	e and (potassium permanganate and sulfuric	acid).
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No informati		

## Special Hazards

#### Immediate Responses

Keep non-involved people away from spill site. Notify manufacturer or supplier. Dike to prevent runoff from rainwater or water application. Notify environmental authorities.

### Protective Clothing and Equipment

Outer protective clothing as required.

### Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving potash.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give conscious victim plenty of water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 400

# POTASSIUM CARBONATE K2CO3

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
SALT OF TARTAR PEARL ASH POTASH (laboratory usage)	White powder. Odourless.	
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Calcined 80 to 85%, 85 to 90%, 90 to 95%, 96 to 99%; hydrated 80 to 85%.  Containers and Materials: Multiwall paper bags, kegs, drums; bulk by train.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm); Solid. Solubility (Water); 112 g/100 mL (20°C); 156 g/100 mL (100°C). Molecular Weight: 138.2 Vapour Pressure: No information. Boiling Point: Decomposes, 891°C.	Floatability (Water): Sinks and mixes.  Odour: Odourless. Flash Point: Not flammable.  Yapour Density: No information.  Specific Gravity: 2.43 (20°C).	Colour: White. Explosive Limits: Not flammable. Meiting Point: 891°C.

Human Hesith		
Symptoms: Inhalation: irritation of nose, eyes ampain, watering. Ingestion: burning sensation, pain, Toxicology: Moderately toxic by ingestion. Low to TLV? No information.  Short-term Inhalation Limits - No information.	stomach cramps, nausea and vomiting.	ghing; Contact: skin - itching, burning and inflammation; eyes -  LD50 - Oral: rat = 1.87 g/kg
Fire		
Fire Extinguishing Agents: Not combustible. Most Behaviour in Fire: Not combustible. Ignition Temperature: Not combustible.	fire extinguishing agents may be used on fir Burning Rate: Not combustible.	es involving potassium carbonate.
Reactivity		
With Water: No reaction; soluble. With Common Materials: Reacts violently with ci Stability: Stable.	olorine trifluoride.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information.		

### Special Hezards

CORROSIVE.

#### Immediate Responses

Keep non-involved people away from spill site. Call Fire Department. Avoid contact and inhalation. Notify manufacturer or supplier. Dike to prevent runoff. Notify environmental authorities.

# Protective Clothing and Equipment

Protective outer clothing as required. Boots - high rubber.

#### Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving potassium carbonate.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion; give water to conscious victim to drink, do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# POTASSIUM HYDROXIDE KOH

#### IDENTIFICATION

UN No. 1813 solid 1814 liquid

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Common Synonyms	Observable Characte	eristics	Manufacturers		
CAUSTIC POTASH POTASSHIM HYDRATE LYE POTASSA		Clear, colourless liquid. H: White solid. Odour-	Canadian manufacturer C-I-L Cornwall, Ontari Canadian supplier: C-I-L Toronto, Ontario Canada Colours and Ch Toronto, Ontario. Hooker Chemical, Calg Canadian Occidental P Calgary, Alberta.	o. emicals Ltd., ary, Alta.	Originating from: International Minerals and Chemical, USA Hooker Chemical, USA
Transportation and Storage	Information				
Shipping State: Solid, liquid Classification: Corrosive. Inert Atmosphere: No requi Venting: Open. Pump Type: Gear, centrifu stainless steel (solutions).	irement.	Label(s): White and Black Class 8, Group II. Storage Temperature: A Hose Type: Natural rubb propylene, Chemiflex 95	mbient. er, polyethylene, poly-	solid, 88 to pellets, stic Containers drums, tank	urity: Solution, 45% KOH, 92% KOH (ground, Ilakes, ks, lumps). and Materials: Solutions – cars, tank trucks; steel. and drums (steel).
Physical and Chemical Cha	racteristics				
Physical State (20°C, 1 at: Solubility (Water): 107 g/10 178 g/100 ml. (100°C). Molecular Weight: 56.1 Vapour Pressure: 1 mm Hg Bolling Point: 133°C, (45% 1320°C (Solid KOH).	00 mL (15°C); (719°C).	Ficatability (Water): Mir Odour: Odourless. Flash Point: Not Hamma Vapour Density: No info Specific Gravity: 1.46, 4 2.04, solid (20°C).	ble.	Solid - white Explosive L Melting Poi	uid - colourless, clear. e. imits: Not flammable. nt: 45% solution KOH, -30°C; 80°C (depending on water content).

# **HAZARD DATA**

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ſ	Human Health				
	Symptoms: Contact with solutions can result in severe burns. KOH is a strong irritant. Contact: skin - burns; eyes - burns, severe damage. Inhalation: (of dust or mist) may damage upper respiratory tract and even lung tissue. Ingestion: can result in severe damage to mucous membranes, nausea, vomiting and diarrhea. Toxicology: Highly toxic by ingestion and inhalation.				
ı	TLY (inhalation) 2 mg/m ³ .	LV= (inhalation) 2 mg/m ³ . LC ₅₀ - No information. LD ₅₀ - Oral: rat = 0.365 g/kg			
l	Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.			
Γ	Fire				
	Behaviour in Firet Not combustible. Solid KOH in contact with organic chemicals.	fire extinguishing agents may be used on a contact with water may generate sufficient burning Rate: Not combustible.	fires involving KOH. Use water only in flooding amounts.  In heat to ignite combustible material. May cause ignition on		
L	Ignition Temperature: Not combustible.	Basining Rate: Not Compostible.			
l	Reactivity				
	With Water: Dissolves exothermically. With Common Materials: When wet may attack alu 1,2-dichloroethylene, maleic anhydride, nitroethan tetrahydrofuran and trichloroethylene. Stability: Stable.	minum, tin, lead and zinc to generate H ₂ e, nitromethane, nitropropane, nitropheno	gas. Reacts violently with acetic acid, acrolein, acrylonitrile, il, N-nitrosomethylurea, phosphorus, tetrachloroethane,		

# Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 80 ppm/24 h/mosquito fish/TLm/freshwater; Aquatic toxicity rating = 10 to 100 ppm/96 h/TLm/freshwater; BOD: None.

Land-Air: No information.

Food Chain Concentration Potential: None.

#### Special Hazards

CORROSIVE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department. Contain spill by diking with earth or other available material. Contact manufacturer for guidance. Avoid contact and inhalation. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

#### Protective Clothing and Equipment

Respiratory protection - filter or dust type respirators are normally sufficient protection against dusts and mists. Goggles - (mono), tight fitting. If face shield used, it must not replace goggles. Gloves - rubber. Boots - high, rubber (pants worn outside boots). Outerwear - as required, coveralls, aprons.

# Fire and Explosion

Most fire extinguishing agents may be used. Water should be used only in flooding amounts.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: If breathing has stopped, give artificial respiration; If laboured, give oxygen. Contact: eyes - irrigate with plenty of water; flush with plenty of water while removing contaminated clothing. Ingestion: give large quantities of milk or water to conscious victim to drink. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### **Valer**

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. If possible, dredge or vacuum pump to remove contaminants, liquid and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undammaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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# POTASSIUM SULFATE K2SO4

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
	White powder or crystals; odourless.	Potash Corp. of Saskatchewan, Cory, Sask. Potcal, Parry Sound, Ontario. Shamrock Chemicais, Port Stanley, Ontario.
Transportation and Storage Information		
Shipping State: Soild. Classification: None. Inert Atmosphere: No requirements. Venting: Open.	Label(sk: Not regulated. Storage Temperature: Ambient.	Grades or Purity: Commercial; technical.  Containers and Materials: Bags, drums. Bulk by truck or rail.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid.  Solubility (Water): 8.5 g/100 mL H ₂ O (10°C) 10.0 g/100 mL (20°C); 14.2 g/100 mL (50°C).  Molecular Weight: 174.3  Vapour Pressure: No information.  Bolling Point: 1 689°C.	Floatability (Water): Sinks and slowly mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 2.7	Colour: White. Explosive Limits: Not flammable. Melting Point: I 069 to 1 072°C, undergoes transition at 588°C.

Human Health			
Symptoms: Inhalation: irritation of nose and eyes Texicology: Low toxicity by all routes.	oms: <u>inhalation:</u> irritation of nose and eyes. <u>Ingestion:</u> stomach pains; <u>Contact</u> : skin - irritation, eyes - irritation.		
TLV [®] No information. Short-term Inhalation Limits - No information.	LC5g - No information. Defayed Toxicity - No information.	LD _{LO} Orai: human = 0.80 g/kg LD _{LO} Scutaneous: guinea pig = 3.0 g/kg	
Fire	**************************************		
Fire Extinguishing Agents: Not combustible; most Behaviour in Fire: Not combustible.	• • •	involving potassium sulfate.	
Ignition Temperature: Not combustible.	Burning Rate: Not combustible.		
Reactivity			
With Water: No reaction; slightly soluble. With Common Materials: May react with finely-d Stability: Stable.	ivided aluminum under certain circumstances.		
Environment			
Water: Prevent entry into water intakes or water Land-Air: No information. Food Chain Concentration Potential: None.	ways.		

# Special Hazards

#### Immediate Responses

Keep non-involved people away from spill site. Stop or reduce discharge if this can be done without risk. Notify supplier and environmental authorities.

# Protective Clothing and Equipment

Protective outerwear as required.

#### Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving potassium sulfate.

### Piret Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Contact: skin and eyes - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: give conscious victim plenty of water to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### ENVIRONMENTAL PROTECTION MEASURES

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, contaminated liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other material.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Material may be buried in a sanitary landfill (environmental authorities' approval required).

# PROPANE CH3CH2CH3

	IDENTIFICATION	UN No. 1978
Common Synonyms	Observable Characteristics	Manufacturers
LPG (see also Butane) DIMETHYLMETHANE	Colouriess gas. Odouriess when pure.	Superior Propane Ltd., Toronto, Ont. Consumers Co-op Refineries Ltd., Regina, Sask. Dome Petroleum Ltd., Calgary, Alta. Goliad Oli Canada Ltd., Calgary, Alta. Mobil Oil Canada Ltd., Calgary, Alta. Pacific Petroleums Ltd., Calgary, Alta. Home Oil Ltd., Calgary, Alta.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Flammable gas. Inert Atmosphere: No requirement. Venting: Safety relief. Pump Type: Rotary LPG.	Label(s): Red label - FLAMMABLE GAS.  Storage Temperature: Ambient.  Hose Type: LPG type; reinforced high pressure.	Grades or Purity: Commercial, technical 97.5%. Containers and Materials: Cylinders tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): Slight, 0.012 g/100 mL (17.8°C). Molecular Weight: 44.1 Vapour Pressure: 400 mm Hg (-56°C); 6 400 mm Hg (21°C). Boiling Point: -42.1°C.	Floatability (Water): Liquefied propane floats and boils.  Odour: Odourless when pure, added mercaptans give 500 to 20 000 ppm, odour threshold, natural gas odour. Flash Point: -104°C (c.c.).  Vapour Density: 1.5 (20°C).  Specific Gravity: (liquid) 0.58 (-44°C).	Colour: Colourless. Explosive Limits 2.1 to 9.5%. Melting Point: -187.7° to 189.9°C.

#### HAZARD DATA

# Harman Health Symptoms: Inhalation: asphyxiant; rapid irregular breathing, headache, fatigue, nausea and vomiting, convulsions, loss of consciousness; Contact: skin and eyes - with propane liquified causes frostbite, burning sensation. Texicology: An asphyxiant. Low texicity. TLV - (inhalation) Asphyxiant. LC50 - Human: no effect 10 000 ppm brief LD50 - No information. Short-term Inhalation Limits - No information. exposures; slight dizziness in a few minutes at 100 000 ppm. Delayed Toxicity - No information. Fire Fire Extinguishing Agents: Stop or reduce discharge if safe to do so. Do not attempt to extinguish fire until leak has been shut off. Use water spray to cool tanks exposed to fire. Behaviour in Fire: When exposed to heat and flame, containers may rupture. Flash back may occur along vapour trail. ignition Temperature: 432°C. Burning Rate: 8.2 mm/min. Reactivity With Water: No reaction. With Common Materials: Reacts vigorously with strong oxidizing agents. Reacts violently with chlorine dioxide. Stability: Stable. Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating > 1 000 ppm/96 h/TLm/freshwater. Land-Air: No information. Food Chain Concentration Potential: None.

#### Special Hazards

#### FI AMMABLE.

# immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all ignition sources. Contact supplier or manufacturer. Flash back may occur along vapour trail. Do not attempt to extinguish fire until leak has been shut off. Contact environmental authorities.

# Protective Clothing and Equipment

In fire or confined spaces - Respiratory protection - self-contained breathing apparatus. Goggles - (mono), tight fitting should be worn, to protect from liquid propane, which could cause eye injury from frostbite burn.

## Fire and Explosion

Stop or reduce discharge if safe to do so. Do not attempt to extinguish fire until leak has been controlled. Let fire burn. Use water spray to cool fire-exposed containers.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give exygen. Contact: eyes and skin-if exposed to liquid propane, immediately remove contaminated clothing irrigate eyes and flush skin with water. Treat as for frostbite. Do not rub affected areas. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Recover undamaged containers.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Burn or flare at spill site (under knowledgeable supervision).

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# PROPYLENE CH3CH:CH2

UN No. 1077

Common Synonyms	Observable Characteristics	Manufacturers
PROPENE METHYLETHYLENE METHYLETHENE	Colourless gas. Mild, aromatic odour.	Petrosar Limited, Coruma, Ont. Esso Chemical Canada, Sarnia, Ont. Petromont, Varennes, Que.
Transportation and Storage Information		
Shipping State: Liquid (compressed gas). Classification: Flammable gas. Inert Atmosphere: No requirement. Venting: Safety relief. Pump Type: Rotary LPG.	Label(s): Red label - FLAMMABLE GAS. Storage Temperature: Ambient. Hose Type: LPG types reinforced high pressure.	Grades or Purity: Technical min. 95%; Pure min. 99%. Containers and Materials: Cylinders, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): 0.02 g/100 mL (20°C). Molecular Weight: 42.1 Vapour Pressure: 7600 mm Hg (20°C); 400 mm Hg (-61°C); 3800 mm Hg (-5°C). Boiling Point: -47.7°C.	Fioatability (Water): Liquid floats and boils.  Odour: Mild, aromatic (22.5 to 67.6 ppm, odour threshold).  Flash Point: -108°C (c.c.).  Vapour Density: 1.5 (20°C).  Specific Gravity: (liquid) 0.51 (-48°C).	Colour: Colouriess. Explosive Limits: 2.0 to 11.1%. Melting Point: -185.2°C.

Human Health		
Symptoms: Inhalation: asphyxiant, headache, fatigu liquified, may cause frostbite. Texicology: Low toxicity. Asphyxiant. TLV® Asphyxiant. Short-term Inhalation Limits - No information.	e, mental confusion, nausea and vomit  LC50 - No information.  Delayed Toxicity - No information.	ing, convulsions, loss of consciousness. <u>Contact</u> : skin and eyes - with  LD 50 - No information.
Fire		
Fire Extinguishing Agents: Stop or reduce discharg exposed containers. Behaviour in Fire: Exposed containers may rupture. Ignition Temperature: 455°C.		extinguish fire until leak has been shut off. Use water to cool fire-
Reactivity		•
With Water: No reaction. With Common Materials: Can react vigorously with oxidizing materials. Reacts violently with nitrogen dioxide, nitrogen tetroxide, nitrous exide. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating >1 000 ppm/96h/TLm/treshwater;  Land-Air: Sweet pea: declination in seedlings, 1000 ppm, 3 days; tomato: 50 ppm, 2 days growth deformation of petiole.  Food Chain Concentration Potential: None.		

#### Special Hazards

FLAMMABLE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Contact supplier or manufacturer for assistance. Flash back may occur along vapour trail. Control of discharge should be attempted only by experienced persons. Do not attempt to extinguish fire until leak has been shut off. Notify environmental authorities.

#### Protective Ciothing and Equipment

In fires or confined areas; Respiratory protection - self-contained breathing apparatus. Otherwise, Goggles - (mono), tight fitting, to protect against liquid (and frost burn). Gloves - rubber. Outerwear - as required.

#### Fire and Explosion

Stop or reduce discharge if safe to do so. Use water spray to cool fire-exposed containers. Stop flow of gas before attempting to extinguish fire.

#### First Ald

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalation</u>: if breathing has stopped, give artificial respiration; if laboured, give oxygen. <u>Contact</u>: skin and eyes (liquid) remove contaminated clothing, irrigate eyes and flush skin with water; treat as for frostbite. Do not rub affected areas. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Recover undammaged containers.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Nesseal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Burn or flare at spill site (under knowledgeable supervision).

# PROPYLENE GLYCOL CH3CH(OH)CH2OH

# IDENTIFICATION

Observable Characteristics	Manufacturers.
1,3 PROPYLENE GLYCOL  - 2-DIOXYGLYCOL  - TRIMETHYLENE GLYCOL  - B-PROPYLENE GLYCOL  - 1,3-HYDROXYPROPANE	Dow Chemical Canada Inc., Sarnia, Ontario. Suppliers: Originating from: Hall Chemical, Montreal, Oxirane, USA Quebec.
Label(s): Not regulated. Storage Temperature: Ambient. Hose Type: Seamless stainless steel.	Grades or Purity: Technical, 95+%; pure, 99%. Containers and Materials: Drums, tank cars, tank trucks; steel.
Odour: Odourless (practically). Flash Point: (1,2) 99°C (c.c.).	Colour: Colouriess. Explosive Limits: (1,2) 2.6 to 12.5%. Melting Point: (1,2) -85°C.
ł,	1,3 PROPYLENE GLYCOL  - 2-DIOXYGLYCOL  - TRIMETHYLENE GLYCOL  - β-PROPYLENE GLYCOL  - 1,3-HYDROXYPROPANE  Label(s): Not regulated.  Storage Temperature: Ambient.  Hose Type: Seamless stainless steel.  Floatability (Water): Sinks in freshwater; may float in saltwater.  Odour: Odourless (practically).  Flash Point: (1,2) 99°C (c.c.).  Vapour Density: (1,3) (1,2) 2,5

HAZARD DATA			
Human Health			
Symptoms: Inhalation: intoxication, nose and ey gastrointestinal irritation and symptoms similar to	e irritation, staggering, headache, mental cor o inhalation. <u>Contact</u> : skin - may be absorbed;	nfusion, nausea and vomiting, drowsiness, stupor. eyes - redness, pain.	ingestion
Texicology: Low texicity by all routes. TLY*- No information. Short-term inhalation Limits - No information.	LC50 - No information. Delayed Toxicity - No information.	LD59 - Oral: rat = 20 g/kg (1,2) LD59 - Oral: mouse = 4.77 g/kg (1,3) TDL0 - Oral: child = 70 g/kg (1,2)	
Fire			
Fire Extinguishing Agents: Use alcohol foam, dry Behaviour in Firer No information.  Ignition Temperature: (1,2) 371°C.	chemical or CO ₂ . Water may be ineffective, bu Burning Rate: 1.5 mm/min	it may be used to coof fire-exposed containers.	
Reactivity			
With Water: No reaction; soluble. Stability: Can react with oxidizing materials. With Common Materials: Stable.			
Environment			
Water: Prevent entry into water intakes and water Land-Air: No information.	rways. Aquatic toxicity rating >1 000 ppm/96 h	/TLm/freshwater; BOD: 96%, 5 days.	

#### Special Hezards

#### COMBUSTIBLE.

#### komediate Responses

Keep non-involved people away from spill site. Call Fire Department. Call manufacturer for advice. Dike to prevent runoff. Notify environmental authorities.

#### Protective Clothing and Equipment

Outer protective clothing as required.

#### Fire and Explosion

Use alcohol foam. Water may be ineffective, but should be used to cool fire-exposed containers.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give plenty of water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- i. Stop or reduce discharge if sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- i. Contact manfacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

# PROPYLENE OXIDE CH3CHCH2O

IDENTIFICATION

UN No. 1280

Common Synonyms	Observable Characteristics	Manufacturers
METHYLOXIRANE 1,2-PROPYLENE OXIDE 1,2-EPOXYPROPANE PROPENEOXIDE METHYL ETHYLYLENE OXIDE	Colouriess liquids. Ethereal odour.	Dow Chemical Canada Inc., Sarnia, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: Inerted. Venting: Safety-relief. Pump Type: Rotary or centrifugal; steel, stainless steel.	Label(s): Red and white label - FLAMMABLE LIQUID; Class 3.1, Group I. Storage Temperature: Ambient. Hose Type: Seamless stainless steel.	Grades or Purity: Technical.  Containers and Materials: Drums, tank cars; steel.
Physical and Chemical Characteristics	-	•
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 65 g/100 mL (30°C); 40.5 g/100 mL (20°C). Molecular Weight: 58.1 Vapour Pressure: 400 mm Hg (18°C). Boiling Point: 34°C.	Figatability (Water): Floats and mixes. Odour: Ethereat (9.9 to 35.0 ppm, odour threshold) Flash Point: -37°C (c.c.). Vapour Density: 2.0 Specific Gravity: 0.83 (20°C).	Colour: Colourless. Explosive Limits: 2.8 to 37.0%. Melting Point: -10% to -112°C.

# HAZARD DATA

Firman Health  Symptoms: Inhalation: headache, coughing, nausea, Inflammation, burns; eyes – very irritating, waterin Toxicology: Moderately toxic by all routes. TLV*- 20 ppm, 50 mg/m³.  Short-term inhalation Limits – No information.	, vomiting, unconsciousness, mild depression of cenng, corneal burns, inflammation. Ingestion: abdom LC50 - inhalation: mouse = 1 740 ppm/4h. Delayed Toxicity - Suspected carcinogen.	entral nervous system and lung irritation. Contact: skin- ninal pain, nausea and vomiting, diarrhea, convulsions. LD50 - Oral: rat = 0.93 g/kg
Fire Fire Extinguishing Agents: Use dry chemical, alcoh		tive but should be used to cool fire-exposed containers and see polymerization which may result in container rupture.
Reactivity With Water: No reaction; soluble.	sterials. May polymerize violently with anhydro	ous chlorides of iron, tin, aluminum; peroxides of iron or acid, hydrochloric acid, hydrofluoric acid, nitric acid,

# With Common Materials: Stable, within the limits of the foregoing.

Water: Prevent entry into potable water intakes and waterways. Harmful to aquatic life. Aquatic toxicity rating >1 000 ppm/96 h/TLm/freshwater; BOD: 75%, 5 days.

aluminum and alkaline metal hydroxides. Reacts violently with ammonium hydroxide, chlorosulfonic acid, hydrochloric acid, hydrochloric acid, hydrochloric acid, hydrochloric acid, hydroxides.

Land-Air: No Information.

oleum and sulfuric acid.

Food Chain Concentration Potential: None.

# Special Hazards

#### FLAMMABLE.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning, "FLAMMABLE". Call Fire Department. Eliminate all ignition sources. Dike to prevent runoff. Notify manufacturer. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Clothing and boots: rubber.

#### Fire and Explosion

Use dry chemical, alcohol foam or carbon dioxide to extinguish. Water may be ineffective, but should be used to cool fire-exposed containers and disperse vapours. Flashback may occur along vapour trail. Contact with certain materials may cause polymerization which may result in container rupture.

#### First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected areas with plenty of water; eyes - irrigate with plenty of water for at least 15 minutes. Ingestion: give plenty of water to conscious victim to drink. Induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manfacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmental authorities required).

#### SODIUM Na

#### IDENTIFICATION

UN No. 1428 sodium metal 1429 sodium metal dispersions in organic liquids

Common Synonyms	Observable Characteristics	Manufacturers
SODIUM METAL METALLIC SODIUM NATRIUM	Silvery-white. Solid, turning greyish-white upon exposure to air. Odourless.	Canadian suppliers: Ethyl Corporation of Canada Ltd., Corunna, Ont., Toronto, Ont.
Transportation and Storage Information		
Shipping State: Solid. Classification: Flammable solid (dangerous when wet), Inert Atmosphere: Under naphthas or parrafins for shipping and storage. Venting: Pressure-vacuum.	Label(s): Blue label - FLAMMABLE; Class 4.3, Group II. Storage Temperature: Ambient.	Grades or Purity: Commercial, technical; 99.9%, Containers and Materials: Drums and smaller containers (solid form); under naphthas or parrafins. May also be shipped as a dispersion in organic liquids.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (water): Reacts violently forming H ₂ gas and NaOH. Molecular Weight: 23 Vapour Pressure: 0.014 mm Hg (300°C); 1.6 mm Hg (400°C). Boiling Point: 881 to 893°C.	Floatability (Water): Reacts violently with water, producing H ₂ gas and NaOH. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 0.97 (20°C); ilquid, 0.93.	Colour: Silvery-white; greyish-white upon exposure to air. Explosive Limits: Not flammable (Note melting point). Hydrogen: 4.1 to 74.2% (for hydrogen). Melting Point: 97.8°C.

#### HAZARD DATA

#### Human Health

Symptoms: Inhalation: (NaOH fumes) irritation of nose, eyes and throat, difficulty breathing, coughing. Ingestion: (Improbable) immediate burning sensation, vomiting, stomach cramps, rapid breathing, diarrhea, loss of consciousness. Contact: skin - painful ulcerations irreparable tissue damage, shock; eyes - very painful, intense watering, burns and ulcerations.

Texicology: Highly toxic upon contact. TLV- inhalation: 2 mg/m³ (as NaOH).

TLY - inhalation: 2 mg/m - (as NaOri).

Short-term inhalation Limits - No information.

LC50 - No Information.

Delayed Toxicity - No information.

LD50 - No information.

#### Fire

Fire Extinguishing Agents: Soda Ash, dry sodium chloride or dry graphite may be used (written in order of preference). DO NOT USE WATER.

Behaviour in Fire: Burns violently accompanied by explosions which cause splattering of the material. Vapour ignites spontaneously at room temperature. Ignition Temperature: >1.15°C (as hydrogen evolved Burning Rate: No information. from contact with water or moisture).

#### Reactivity

With Water: VIOLENT - generates hydrogen and is very exothermic. Reaction with water also produces sodium hydroxide (NaOH).

With Common Materials: Reacts violently with aluminum bromide, aluminum chloride, aluminum fluoride, ammonium nitrate, antimony halides, arsenic halides, bismuth halides, carbon dioxide, carbon tetrachloride, chlorinated hydrocarbons, chlorine, chloroform, chromium halides and oxides, copper halides and oxides, i.z-dichloroethylene, dichloromethane, hydrochloric acid, hydrofluoric acid, hydrogen peroxide, hydrogen sulfide, iodine, iron, iron halides, lead oxide, maleic anhydride, manganese halides, mercury halides, methyl chloride, monoammonium phosphate, nitric acid, nitrous oxide, phosgene, phosphorus, phosphorus halides and oxides, potassium oxides, silver halides, sulfur, sulfur halides and oxides, sulfuric acid, tin halides, tetrachloroethane, trichloroethylene, vanadium halides and zinc halides.

Stability: Reactive with moisture, air and many common compounds.

#### Environment

Water: Sodium with water produces NaOH which raises the pH of a water body; this can be harmful to aquatic life; BOD: None. Land-Air: No information.

Food Chain Concentration Potential: None.

### Special Hazards

FLAMMABLE, REACTS VIOLENTLY WITH WATER, LIBERATING HYDROGEN. VERY REACTIVE.

### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE AND REACTIVE WITH WATER". CALL FIRE DEPARTMENT (warn them of the unusual properties - no water). Contact manufacturer for assistance. Contain spill area by diking with earth or other available material. Avoid contact and inhalation of fumes. Notify environmental authorities.

# Protective Clothing and Equipment

In fires or confined spaces - Respiratory protection - self-contained breathing apparatus. Otherwise, Goggles - (mono), tight fitting. If face shield is used it must not replace goggles. Gloves - rubber. Boots - safety, rubber (pants worn outside boots). Outerwear - flame-proofed overalls or coveralls.

# Fire and Explosion

Use soda ash, dry sodium chloride or dry graphite to extinguish. DO NOT USE WATER. Burns violently, accompanied by explosions which cause splattering of the material. Vapour ignites spontaneously at room temperature. Vapour droplets ignite spontaneously at 121°C. Pools of molten sodium ignite at 204 to 427°C.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first ald at once. Contact: eyes - irrigate with plenty of water; skin: remove particles of sodium quickly but carefully and flush affected area with a copious amount of water for at least 30 minutes. Remove contaminated clothing at once. Inhalation: in case of exposure to fumes from fire, give oxygen if breathing is difficult, give artificial respiration if breathing has stopped. Ingestion: (improbable) give water to conscious victim to rinse mouth. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.

Avoid contact with water or moisture.

- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM ALUMINATE Na2A12O3 or NaA1O2

IDENTIFICATION

UN No. 2812 anhydrous 1819 solution

Common Synonyms	Observable Characteristics	Manufacturers
META-SODIUM ALUMINATE ALUMINUM SODIUM OXIDE	White powder. Odourless.	Handy Chemicals, Laprairie, Quebec.
Transportation and Storage Information		
Shipping State: Solid or liquid (aqueous solution).  Classification: Solution - Corrosive. Inert Atmosphere: No requirement. Venting: Open. Pump Type: No information.	Label(s): Solution - Black and white label - CORROSIVE. Storage Temperature: Ambient. Hose Type: No information.	Grades or Purity: Technical or 27°Bé solution. Consainers and Materials: Solid - multi-wall paper sacks, drums. Solution - drums and bulk; steel or stainless steel.
Physical and Chemical Characteristics		
Physical State Solid. Solubility (Water): Very soluble. Molecular Weight: 82 (Na Al O2). Vapour Pressure: No information. Boiling Point: >1 800°C.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: >1.5 (20°C); 27°Bé solution 1.23	Colour: White. Explosive Limits: Not flammable. Melting Point: 1650°C

Human Health		
Symptoms: Inhalation: (dust) irritation of respirate redness and pain.  Toxicology: Moderately toxic by all routes.  TLV® No information.  Short-term Inhalation Limits - No information.	bry tract. <u>Ingestion</u> : sore throat, abdominal p LC50 - No information. Delayed Toxicity - No information.	pain, diarrhea. <u>Contact</u> : skin - redness and pain; eyes - watering,  LD50 - No information.
Fire		
Fire Extinguishing Agents: Not combustible. Most Behaviour in Fire: Not combustible. Ignition Temperature: Not combustible.	fire extinguishing agents may be used in fire Burning Rate: Not combustible.	es involving sodium aluminate.
Reactivity		
With Water: No reaction; soluble. With Common Materials: No information. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No informati		

#### Special Hazards

CORROSIVE.

# Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call manufacturer or supplier for advice. Dike to prevent runoff from rainwater or water application. Notify environmental authorities.

# Protective Clothing and Equipment

Protective outer clothing as required.

### Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving sodium aluminate.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalations if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contacts skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestions give conscious victim plenty of water to drink. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and clearup of contaminated materials.

#### -

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM ARSENITE NaAsO2

IDENTIFICATION

UN No. 1686 Aqueous solutions 2027 Anhydrous

Common Synonyms	Observable Characteristics	Manufacturers
SODIUM META ARSENITE ARSENEOUS ACID MONOSODIUM SALT	White to grey-white powder. Odourless.	No Canadian Manufacturers.
Transportation and Storage Information		
Shipping State: Solid, aqueous solutions. Classification: Poison. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: No information.	Label(s): Black and white label - POISON; Group 6. Storage Temperature: Ambient. Hose Type: No information:	Grades or Purity: Pure; Technical (55 to 98%); Solution 75% arseneous oxide. Containers and Materials:
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Solid. Solubility (Water): Soluble. Molecular Weight: 129.9 Vapour Pressure: No information. Boiling Point: Decomposes.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not fiammable. Vapour Density: No information. Specific Gravity: 1.87 (20°C).	Colour: White to greyish-white. Explosive Limits: Not flammable. Melting Point: 615°C.

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Human Health			
Symptoms: <u>Inhalation and Ingestion</u> : irritation of stomach and intestines, abdominal pain, nausea and vomiting, mental confusion, weakness, diarrhea, coma Contact: skin - irritation and burning; eyes - irritation and burning.  Toxicology: Highly toxic by ingestion and contact.			
TLV (as As) 0.2 mg/m ³	LC50 - No information.	LD50 - Oral: rat = 0.041 g/kg	
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	$LD_{50}$ - Skinz rat = 0.15 g/kg	
Fire			
Fire Extinguishing Agents: Not combustible. Most material.  Behaviour in Fire: Not combustible.  Ignition Temperature: Not combustible.	fire extinguishing agents may be used. Water  Burning Rate: Not combustible.	should be used sparingly to avoid contamination of area with	
Reactivity			
With Water: No reaction; soluble. With Common Materials: No information. Stability: Stable.			
Environment			
Water: Prevent entry into water intakes and water freshwater.	ways. Fish toxicities: 14 to 39 mg/L/96 h/rainb	ow trout/LC50/freshwater; 21 to 42 mg/L/96 h/LC50/	
Land-Air: Waterfowi = 32 mg/kg, LD ₅₀ . Food Chain Concentration Potential: No informati	on.		

#### Special Hazards

POISON.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning "POISON". Avoid contact and inhalation. Notify manufacturer or supplier. Dike to prevent runoff. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally-encapsulated suit.

### Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving sodium arsenite. Water should be used sparingly to avoid contamination of area with material.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give conscious victim plenty of water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- I. Stop and reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Removel material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM BOROHYDRIDE NaBHis

IDE		

UN No. 1426

Common Synonyms	Observable Characteristics	Manufacturers
BOROHYDRIDE SODIUM TETRAHYDROBORATE	White to grey-white powder or lumps. Odourless.	No Canadian Manufacturers. Mailinckrodt ind. Chem., St. Louis, MO.
Transportation and Storage Information	**************************************	
Shipping State: Solid. Classification: Flammable solid. Inert Atmosphere: No requirement. Venting: Closed.	Label(s): Blue and white label - FLAMMABLE SOLID; Class 4.3, Group I. Storage Temperature: Ambient.	Grades or Purity: 95 to 98% minimum purity; dry powder pellets; 12% solution in 43% aqueous sodium hydroxide. Containers and Materials: Bottles, polyethylene - lined metal drums.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 55 g/100 mL (25°C); reacts with hot water producing H ₂ and Na ₂ B ₆ O ₇ . 10H ₂ O. Molecular Weight: 37.8 Vapour Pressure: No information. Boiling Point: Decomposes (490°C).	Floatability (Water): Sinks and mixes or reacts. Odour: Odourless. Flash Point: Not flammable, but the hydrogen released from decomposition is. Vapour Density: No information. Specific Gravity: 1.07 (20°C).	Colour: White to grey-white. Explosive Limits: Not flammable, but the hydrogen released from decomposition is. Melting Point: Decomposes (400°C).

Human Health		
Symptoms: Inhalation: sore throat, shortness of brodiarrhea, headache, dizziness.  Toxicology: Highly toxic by ingestion.  TLY*- (as sodium borate anhydrous) 1 mg/m³; as sodium borate decahydrate 5 mg/m³.  Short-term inhalation Limits - No information.	Eath, coughing, headache; <u>Contact</u> : eyes - redr LC50 - No Information. Delayed Toxicity - No Information.	ness, pain; skin-redness; <u>Ingestion</u> ; sore throat, abdominal pain,  LD50 - Intraperitoneal; rat = 0.018 g/kg
Fire	-	
Fire Extinguishing Agents: Graphite, limestone, so Behaviour in Fire: Decomposes and produces highly Ignition Temperature: Not combustible by itself - hydrogen is ignited at 500°C.	flammable hydrogen gas.	water, carbon dioxide, or halogenated extinguishing agents.
Reactivity		
With Water: Soluble; reacts with hot water to prod With Common Materials: Can react with oxidizing Stability: Stable.	uce hydrogen gas and sodium borate. materials.	
Environment		
Water: prevent entry into water intakes and water \$ 200 ppm/48 h/mosquito fish/TLm/freshwater. Land-Airs No information. Food Chain Concentration Potential: None.	ways. As sodium borater Fish toxicity: 3 000 to	3 300 ppm/tns/minnow/minimum lethal dose/freshwater;

# Special Hazards

FLAMMABLE - flammable gas released (H2) upon contact with water.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning "FLAMMABLE". CALL FIRE DEPARTMENT. Notify manufacturer or supplier. Dike to prevent runoff.

Notify environmental authorities.

# Protective Clothing and Equipment

In fires Respiratory protection - self-contained breathing apparatus. Otherwise, Goggles, Gloves rubber; boots - rubber (pants worn outside boots) and protective outer clothing as required.

# Fire and Explosion

Use graphite, limestone, soda ash and sodium chloride powders to extinguish. Do not use water, carbon dioxide, or halogenated extinguishing agents. Decomposes and produces highly flammable hydrogen gas.

#### Pirst Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured give oxygen. Contact: skin - remove contaminated clothing, and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give plenty of water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or, clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

# Water

- 1. Stop and reduce discharge if sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming
- or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

# Discosal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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# SODIUM CARBONATE Na₂CO₃ (anhydrous)

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
CALCINED SODA SODA ASH SODA MONOHYDRATE CRYSTAL CARBONATE CARBONIC ACID, DISODIUM SALT	White to grey crystalline solid or powder. Odourless.	Allied Chemical Limited, Amherstburg, Ontario.
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Dense (58%), light (58%), extra light, natural and refined.  Containers and Materials: Bags, barrels, drums and bulk by truck or train; steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Solid. Solubility (Water): 7.1 g/100 mL (0°C); 22 g/100 mL (20°C); 45.5 g/100 mL (100°C). Molecular Weight: 106.0 Vapour Pressure: No information. Boiling Point: Slowly begins to decompose at 400°C.	Pioatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 2.53 (20°C).	Colour: White to grey . Explosive Limits: Not flammable. Melting Point: 851 °C.

HAZARD DATA				
Human Health				
contact can cause "soda ulcers" and perforation vomiting, diarrhea and possible circulatory collapse	of nasal septum. <u>Ingestion</u> : of large amounts se.	Contact: with dust causes eye and skin irritation. Excessive is corrosive to the gastrointestinal tract, causing cramps,		
Toxicology: Moderately toxic by ingestion. Low to	exicity by inhalation and contact.	1 m		
TLV*- No information.  Short-term Inhalation Limits - No information.	LC 50 - No information. Delayed Toxicity - No information.	LD50 - intraperitoneal: mouse = 0.117 g/kg LD _{LO} - Oral: rat = 4 g/kg		
Fire Extinguishing Agents: Not combustible. Most Behaviour in Fire: Not combustible. Begins to decignition Temperature: Not combustible.  Reactivity	ompose at 400°C producing CO2 gas.	nvolving sodium carbonate.		
With Water: No reaction; moderately soluble. With Common Materials: Can react violently with Stability: Stable.	aluminum phosphorus pentoxide and fluoride su	ifuric acid.		
Environment				
Water: Prevent entry into water intakes and wate Land-Air: No information. Food Chain Concentration Potential: None.	rways. Fish toxicity: 265 mg/L/48 h/ <u>Daphnia</u> <u>ma</u>	agna/TLm/freshwater; BOD: No information.		

# Special Hazards

# mmediate Responses

Keep non-involved people away from spill site. Avoid contact and inhalation of dust. Stop or reduce discharge, if this can be done without risk. Notify supplier. Notify environmental authorities.

# Protective Clothing and Equipment

If dust is present wear dust respirator, industrial (tight fitting) goggies, gloves and coveralls.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving sodium carbonate.

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give warm water to conscious victim to drink. Do not induce vomiting. If medical attention is not immediately available, transport victim to hospital, clinic or doctor.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### **Vater**

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, 4. Remove material by manual or mechanical liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Contaminated materials may be buried in a secured landfill site (approval of environmental authorities required).

# SODIUM CHLORATE NaCIO3

**IDENTIFICATION** 

UN No. 1495 solid 2428 solution

Common Synonyms	Observable Characteristics	Manufacturers
CHLORATE OF SODA CHLORIC SALT OF SODIUM	Colouriess to white crystalline solid, colouriess in solution. Commercial material may be pale yellow.	Erco Industries Ltd., Buckingham, Que., Vancouver, B.C., Thunder Bay, Ontario. PPG Industries (CPI), Beauharnois, Que. Canadian Occidental Petroleum, Squamish, B.C., Nanaimo, B.C. B.C. Chemicals, Prince George, B.C.
Transportation and Storage Information		
Shipping State: Solid, liquid (aqueous solution). Classification: Oxidizer. Inert Atmosphere: No requirement. Venting: Open. Pump Type: For solutions, most types.	Label(s): Yeliow label - OXIDIZER; Class 5.1, Group II. Storage Temperature: Ambient. Hose Type: Polyethylene, polypropylene, natural rubber.	Grades or Purity: Commercial, 99% NaClO ₃ ; technical, 99.5% minimum. Containers and Materials: Anhydrous: drums; steel, plastic-lined steel. Solutions: drums, tank cars; stainless steel, plastic-lined steel.
Physical and Chemical Characteristics		**************************************
Physical State (20°C, 1 atm): Solid or liquid (solution) Solubility (Water): 72 g/100 mL (-15°C); 79 g/100 mL (0°C); 101 g/100 mL (20°C); 126 g/100 mL (40°C). Molecular Weight: 106.4 Vapour Pressure: Not pertinent. Boiling Point: Decomposes at 300°C releasing oxy	Floatability (Water): Mixes and sinks. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: Solid: 2.49 (15°C); Saturated Solution: 1.43 (20°C); i.38 (-15°C). vgen.	Colour: Colouriess to white (or pale yellow). Explosive Limits Not flammable. Melting Point: 248 to 261°C.

#### HAZARD DATA

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Symptoms: Inhalation: sneezing, small nasal ulcerations, irritation of respiratory tract. Contact: Prolonged exposure to dust may result in irritation to the skin, mucous membranes and eyes. Ingestion: vomiting, abdominal pain, cyanosis, diarrhea, unconsciousness and death. Toxicology: Moderately toxic by ingestion. Low toxicity by inhalation and contact.

TLY - No information.

Short-term Inhalation Limits - No information.

LC50 - No information.

Delayed Toxicity - No information.

LDso - Oral: rat = 1.2 g/kg

#### Fire

Fire Extinguishing Agents: Not combustible by itself but causes combustion. In fires involving sodium chlorate, use water in flooding amounts. Behaviour in Fire: Upon decomposition, releases toxic chloride fumes. Upon heating, can release oxygen, making extinguishment difficult. ignition Temperature: Not combustible, but causes combustion of other materials.

# Reactivity

With Water: No reaction; soluble.

With Common Materials: In contact with strong acids, can release toxic chlorine and chlorine dioxide gases. Reacts with aluminum, ammonium thiosulfate, arsenic, arsenic trioxide, carbon, charcoal, copper, manganese dioxide, metal sulfides, organic acids, organic matter, phosphorus, potassium cyanide, sulfur, sulfuric acid, thiocyanates and zinc.

Stability: Stable within the limits of the foregoing.

#### Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Aquatic toxicity rating = >1 000 ppm/96 h/Tim/freshwater; Fish toxicities: 3.8 ppm/Scenedesmus/threshold toxicity/freshwater; 4 200 ppm/24 h/rainbow trout/LCsg/freshwater; 3 157 ppm/24 h/channel catfish/LCsg/freshwater. Land-Air: Detrimental to vegetation - is used as a weed killer.

Food Chain Concentration Potential: None.

#### Special Hazards

OXIDIZER.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning "OXIDIZER". Call Fire Department. Notify manufacturer. Avoid contact and inhalation. Contain spill by diking to prevent runoff. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

# Protective Clothing and Equipment

In fires - Respiratory protection - self-contained breathing apparatus. Otherwise, if dust is present, respirator should be used. Goggles - (mono), tight fitting, or face shield. Gloves - rubber. Outer clothing - nonliammable coveralls, etc. Do not use organic material such as leather, etc. Boots - rubber; avoid contact of contaminated clothing with ignition sources.

# Fire and Explosion

Not combustible by itself, but can cause most organic material to ignite. Use water in flooding amounts. Clothing impregnated with sodium chlorate is highly flammable when dry.

### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: eyes - irrigate immediately with plenty of water; skin - remove contaminated clothing and flush affected areas with plenty of water. Ingestion: give milk or water to conscious victim to drink and induce vomiting. Repeat several times. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means (for solid spills).
- 5. Remove undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleamp of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM CHLORIDE NaCI

# IDENTIFICATION

Domtar, Goderich, Ontario. Canadian Salt, Ojibway, Ontario.
Dow Chemical Canada Fort Saskatchewan, Alberta.
Grades or Purity: Variable.  Containers and Materials: Multiwall paper bags, barrels, drums, bulk by train or truck; steel.
Colour: Colourless to white. Explosive Limits: Not flammable. Melting Point: 801°C.
-

•	HAZARD DATA	
Human Health		
- irritation, watering, inflammation.  Texicology: Low texicity by all routes.		nd vorniting; Contact: skin - irritation and inflammation; eyes
TLY No information.	LD50 - No information. Delayed Toxicity - No information.	LD50 - Oral: rat = 3.0 g/kg TLLO - Oral: human = 12.36 g/kg
Short-term Inhalation Limits - No information.	Delayed Toxicity - 140 milot mattorn	120 - 0101 10010 - 1112
Fire		
Fire Extinguishing Agents: Not combustible. Most Behaviour in Fire: Not combustible.		involving sodium chloride.
Ignition Temperature: Not combustible.	Burning Rate: Not combustible.	
Reactivity		
With Water: No reaction; soluble. With Common Materials: Can react violently with Stability: Stable.	bromine trifluoride and lithium.	•
Environment		
backs it Continghanters 2.7 att 16th h/Danbaia ma	ena/I-Cco/Ireshwater, BOO: None,	ish/TLm/freshwater; 2.9 to 11.7/192 to 216 h/stickle- 2.6 to 5.0 g/L restricts yields of many crops. Greater than 5.1

# Special Hazards

### **Immediate Responses**

Keep non-involved people away from spill site. Notify manufacturer or supplier. Dike to prevent runoff from rainwater or water application. Notify environmental authorities.

# Protective Clothing and Equipment

Outer protective clothing as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving sodium chloride.

# First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: move victim to fresh air. Contact: skin - flush affected areas with water; eyes - irrigate with water. Ingestion: give conscious victim plenty of water to drink and induce vomiting. Transport victim to hospital, doctor, or clinic if necessary.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Contaminated materials may be buried in a secured landfill site (approval of environmental authorities required).

# SODIUM CYANIDE NaCN

# IDENTIFICATION

UN No. 1689

Common Synonyms and Trade Names	Observable Characteristics	Manufacturers	
CYANIDE OF SODIUM CYANOGRAN (Du Pont) CYANOIDS (Kraft)	White, crystalline solid, powder or granules. Odourless when dry; when wet gives typical cyanide almond-like odour.	Canadian supplier: Canadian Industries Ltd., Montreal, Quebec.	Originating from: ICI, United Kingdom.
Transportation and Storage Information			
Shipping State: Solid. Classification: Poison. Inert Atmosphere: No requirement. Venting: Closed.	Label(s): White label - POISON; Class 6.1, Group I. Storage Temperature: Ambient.	Grades or Purity Containers and I	y: 97 to 99%. Materials: Drums; steel.
Physical and Chemical Characteristics			······································
Physical State (20°C, 1 atm); Solid. Solubility (Water); 48 g/100 mL (10°C); 82 g/100 mL (35°C). Molecular Weight; 49.0 Vapour Pressure; 1 mm Hg (817°C). Bolling Point: 1496°C.	Floatability: (Water): Sinks and mixes Odour: Odourless when dry; when wet cyanide almond-like odour. Flash Point: Not flammable. Vapour Density: 0.93 as HCN (25°C). Specific Gravity: 1.6 (25°C).		s: Not flammable. 60 to 564°C.

# HAZARD DATA

Human Health		
	rapid breathing, anguish, convulsions, foaming at moth symptoms similar to inhalation; eyes - irritation, we LC50 - inhalation: rat = 484 ppm/l h (as HCN). LCL0 - inhalation: human = 120 mg/m ³ /l h	
more could suppose them - the wine mention	(as HCN); = 200 mg/m ³ /10 min (as HCN). Delayed Toxicity - No information.	rofo - crass sousses - greats Plug
Fire		,
Behaviour in Fire: Not combustible. In fires may e ignition Temperature: Not combustible. Reactivity	volve toxic fumes.  Burning Rate: Not combustible.	
With Water: Contact with water or moist air may With Common Materials: Contact with acids or we oxidizing agents, and chiorates.  Stability: Stable when dry.	produce HCN; soluble. tak alkalis produces poisonous and flammable HCN ga	s. Reacts violently with nitrates, nitrites and other
Environment		·
Water: Prevent entry into water intakes and water BOD: 6%, 7 days (theoretical). Land-Air: No information.	rways. Fish toxicity: 0.15 ppm/96 h/bluegill/TLm/fres	hwater; 0.25 ppm/48 h/prawn/LC5g/saltwater;

# Special Hazards

POISON. Contact with acids or weak alkalis liberates HCN.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Avoid contact and inhalation. Contact supplier or manufacturer for guidance. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Gloves - rubber.

# Fire and Explosion

Not combustible. In fires involving sodium cyanide water may be used to extinguish; however, area should be diked to prevent runoff.

#### First Ald

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing, and flush affected areas with plenty of water; eyes -krigate with plenty of water. Ingestion: induce vomiting in conscious victim, and repeat until vomitus is clear. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- if possible, dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- Avoid contact with material.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- I. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

#### SODIUM DICHLOROISOCYANURATE NaNCIOINCICOINCICO

# **IDENTIFICATION**

Manufacturers Observable Characteristics Common Synonyms White powder. Bleach-like odour. No Canadian Manufacturers. NaDCC Monsanto, St. Louis, MO. **POOL CHLORINATOR** FMC. Philadelphia, PA. 1-SODIUM-3,5-DICHLORO-S-TRIAZINE-2,4,6-TRIONE ISOCYANURIC ACID, DICHLORO SODIUM SALT SODIUM DICHLOROISOCYANURATE (dihydrate). Transportation and Storage Information Grades or Purity: Technical, 39 to 60% Label(s): Not regulated. Shipping State: Solid. available chlorine. Storage Temperature: Cool, ambient. Classification: None. Containers and Materials: Plastic-lined fibre Inert Atmosphere: No requirement. drums, drums; steel, stainless steel. Venting: Pressure-vacuum. Physical and Chemical Characteristics Colour: White. Figatability (Water): Figats, mixes and reacts. Physical State (20°C, 1 atm): Solid. Explosive Limits: Not flammable. Odour: Bleach-like. Solubility (Water): Soluble in all proportions, Melting Point: Decomposes 230°C. Flash Point: Not flammable. reacts to release chlorine. Vapour Density: (as C12) 2.5 (0°C). Molecular Weight: 220.0 (anhydrous). Specific Gravity: 0.96 (20°C). Vapour Pressure: No information. Boiling Point: Decomposes 230°C.

#### HAZARD DATA

Human Health		
Symptoms: Inhalation: (dust) irritation of respirato irritation of gastrointestinal tract, difficulty breath Toxicology: Moderately toxic by all routes.  TLY 2 (as chlorine) 1 ppm; 3 mg/m ³ .  Short-term Inhalation Limits - (as chlorine) 3 ppm; 9 mg/m ³ (15 min).	ing. LC ₅₀ - No information.	(of chlorine)-burning, irritation, difficulty breathing. <u>Ingestion</u> :  LD50 - Oralt rat = 1.40 g/kg some
Fire		
be used in flooding amounts.		filres involving sodium dichloroisocyanurate. Water should only Contact with water produces chlorine gas. Contact with organic
Reactivity		
With Water: Reacts to release chlorine gas. With Common Materials: Reacts with many organiand hydrated salts.	c materials, to initiate spontaneous combus	tion. Can react violently with ammonia, ammonium salts, urea
Environment		
Water: Prevent entry into water intakes and waters 96 h/rainbow trout/TL50/freshwater. Land-Air: No information.	ways. Fish toxicity = 0.07 to 0.15 ppm/96 h/f	athead minnow/LC50/freshwater (as C1); 0.14 to 0.29 ppm/

Food Chain Concentration Potential: None.

# Special Hazards

OXIDIZER, CORROSIVE, Reacts with water to release chlorine.

## Immediate Responses

Keep non-involved people away from spill site. Call Fire Department. Avoid contact and inhalation. Notify manufacturer or supplier. Notify environmental authorities.

#### Protective Clothing and Equipment

In fires or confined spaces Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, outer protective clothing as required and self-contained breathing apparatus.

# Fire and Explosion

Not combustible. Use dry chemical or carbon dioxide to extinguish fires involving sodium dichloroisocyanurate. Water should only be used in flooding amounts. When heated to decomposition emits carbon monoxide and chloride lumes.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with water; eyes - irrigate with water. Ingestion: give conscious victim water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain spill by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater
- or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM DICHROMATE Na2Cr207-2H2O

# **IDENTIFICATION**

UN No. 1479

Common Synonyms	Observable Characteristics	Manufacturers		
SODIUM ACID CHROMATE SODIUM BICHROMATE SODIUM BICHROMATE DIHYDRATE	Red-orange crystalline solid. Odourless.	No Canadian manufacturers. Selected U.S. manufacturers Essex Chemical Company, Clifton, NJ, USA		Canadian supplier: Canadian industries Limited, Toronto, Ontario. Allied Chemical Ltd., Toronto, Ont., Montreal, Quebec.
Transportation and Storage Information				
Shipping State: Solid, liquid (aqueous solution).  Classification: Oxidizer. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Gear, centrifugal; steel, stainless steel (for solutions only).	Label(s): Yellow label - OXI Group II. Storage Temperature: Ambi Hose Type: Natural rubber, polypropylene.	ient.	(98.8-99%) Containers paper bags	Purity: Technical, crystalline  1. Technical, solution (69-70%),  2. and Materials: Dry - multiwall  2. fibre drums. Solution - tank  2. trucks; steel.
Physical and Chemical Characteristics		· · · · · · · · · · · · · · · · · · ·		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 238 g/100 mL (0°C); 508 g/100 mL (80°C). Molecular Weight: 298.1 Vapour Pressure: No information. Bolling Point: Decomposes (400°C).	Floatability (Water): Sinks a Odour: Odourless. Flash Point: Not flammable. Vapour Density: No informa Specific Gravity: 2.52 (13°C	tion.	Melting Po	ed-orange. Limits: Not flammable. int: 357°C (anhydrous). solution, -48.2°C.

### HAZARD DATA

		iel.
	-	la h

Symptoms: Inhalation: irritation of nose and eyes, tingling and burning sensation in respiratory tract, sneezing. Contact: extremely corrosive to skin and mucous membranes. May cause rash or external ulcers on skin. Ingestion: of large quantities may be fatal, burning sensation in mouth and throat, nausea and vomiting.

Toxicology: Highly toxic by contact and ingestion. TLV •- (inhalation) 0.05 mg/m³ (as Cr).

TLV - (inhalation) 0.05 mg/m³ (as Cr). Short-term inhalation Limits - No information. LC50 - No information.

Delayed Toxicity - Suspected carcinogen.

LD50 - Intraperitoneal: rodent = 0.16 g/kg

LDLo - Intraperitoneal: guinea pig = 0.335 g/kg

# Fire

Fire Extinguishing Agents: Not combustible. In fires involving sodium dichromate, flood with water.

Behaviour in Fire: Not combustible but may ignite combustible materials upon contact. Decomposes to produce oxygen when heated.

Ignition Temperature: Not combustible. Burning Rate: Not combustible.

# Reactivity

With Water: No reaction; soluble. Sodium dichromate is a strong oxidizer and may ignite organic materials.

With Common Materials: Reacts violently with hydrazine.

Stability: Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: 145 ppm/24 h/bluegili/TLm/freshwater; 10 ppm/48 h/Daphnia magna/TLm/freshwater; Aquatic Toxicity rating = 100 to 1 000 ppm/96 h/TLm; BOD: None.

Land-Air: No Information.

Food Chain Concentration Potential: None.

# Special Hazards

OXIDIZER.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "OXIDIZER". Avoid contact and inhalation. Stop or reduce discharge if this can be done without risk. Dike to prevent runoff. Notify manufacturer or supplier and environmental authorities.

# Protective Clothing and Equipment

Goggles - (mono), tight fitting. Gloves - rubber. Dust respirator. Protective work clothing - rubber apron, coveralis. Boots - high, rubber (pants outside boots).

# Fire and Explosion

Not combustible. In fires involving this material, flood discharge area with water.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured give oxygen. Contact: eyes - irrigate with plenty of water; skin - flush affected areas with plenty of water and remove contaminated clothing. Ingestion: do not induce vomiting; however, vomiting may occur spontaneously. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM DITHIONITE Na2S2O4*xH2O (x = 6 to 2)

# IDENTIFICATION

UN No. 1384

Observable Cha	eracteristics	Manufacturers		
		Canadian supplier:		Originating from: Virginia Chem, USA Royce Chemical, USA (direct)
nation		**************************************		
•	Class 4.2, Group II. Re label - SPONTANEOUS	d, black and white COMBUSTIBLE.	Containers	urity: Technical. and Materials: Barrels, kegs, boxes; al lined. Drums; steel.
istics				
20°C).	Odour: Faint sulfurous Flash Point: Not flamm Vapour Density: No inf	odour. nable by Itself. ormation.	Explosive Li with water.	te to yellow. imits: Not flammable but can react nt: Decomposes, 52 to 55°C.
	White to yellow	Label(s): Red and whit spontaneous Class 4.2, Group II. Re iabei - SPONTANEOUS nt. Storage Temperature:  istics  lid. Ploatability (Water): Si Odour: Faint sulfurous Flash Point: Not flamm Vapour Density: No inf	White to yellow powder or crystals.  Sulfurous odour.  No Canadian manufact Canadian supplier: Vir Chem of Canada, it  nation  Label(s): Red and white label - FLAMMABLE;  spontaneous  Class 4.2, Group II. Red, black and white label - SPONTANEOUS COMBUSTIBLE.  nt.  Storage Temperature: Ambient.  istics  lid.  Ploatability (Water): Sinks and mixes. Odour: Faint sulfurous odour. Flash Point: Not flammable by Itself. Vapour Density: No information.	White to yellow powder or crystals.  Sulfurous odour.  Label(s): Red and white label - FLAMMABLE; Grades or P.  Spontaneous Class 4.2, Group II. Red, black and white Containers: label - SPONTANEOUS COMBUSTIBLE.  Storage Temperature: Ambient.  Storage Temperature: Ambient.  Storage Temperature: Sinks and mixes.  Odour: Faint sulfurous odour.  Flash Point: Not flammable by Itself.  Vapour Density: No information.  Melting Point

# HAZARD DATA

INCARD DATA	
ness of breath. <u>Contact</u> : eyes and skin - redustion and inhalation. LC50 - No information. Delayed Toxicity - No information.	ness and pain. <u>Ingestion</u> : causes abdominal pain and nausea.  LD50 - No information.
an cause fire.	e used in flooding amounts. The addition of small amounts of
	tion. In large amounts of water; soluble.
*	
ways. Harmful to aquatic life. BOD: No Info on.	ormation.
	ness of breath. Contact: eyes and skin - redrestion and inhalation.  LC 50 - No information.  Delayed Toxicity - No information.  ioxide, or dry chemical; water should only be an cause fire.  id.  Burning Rate: No information.  th a highly humid atmosphere can cause igning sulfur oxides.

#### Special Hazards

FLAMMABLE (spontaneously combustible).

#### immediate Responses

Keep non-involved people away from spill area. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Warn about water and moisture hazard. Contact manufacturer or supplier for guidance. Contain spill area by diking with earth or other available material. Dike to contain runoff. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - in fire, use self-contained breathing apparatus. Goggles - (mono), tight fitting. Gloves - rubber. Boots - high, rubber (pants worn outside boots). Protective clothing - as required - coveralls, acid suit, etc.

#### Fire and Explosion

Extinguishing agents: dry sand, dry chemical, carbon dioxide. Use water only in flooding amounts. In fires, SO2 is released.

# First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Contact: eyes - irrigate with plenty of water; skin - flush with plenty of water. Ingestion: give large amounts of water or milk to conscious victim. Keep warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# ENVIRONMENTAL PROTECTION MEASURES

#### Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss discuss and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM HYDROSULFIDE NaHS*xH>O (solution) (x = 0 to 3)

#### IDENTIFICATION

UN No. 2923 solid >25% water of crystallization 2318 solid <25% water of crystallization NA2922 solution

Common Synonyms	Observable Characteristics	Manufacturers
SODIUM SULPHYDRATE SODIUM BISULFIDE SODIUM HYDROGEN SULFIDE SODIUM MERCAPTAN	Colourless to yellow crystals. Clear to light straw-coloured (yellow to black at higher temperatures). Offensive, rotten-eggs odour (H ₂ S).	Cornwall Chemical Ltd., Montreal, Quebec. Canadian Industries Ltd., (CIL), Montreal, Quebec.
Transportation and Storage Information		
Shipping State: Solid or liquid (aqueous solution).  Classification: Spontaneous combustible; corrosive.  Inert Atmosphere: No requirement.  Yenting: Pressure-vacuum.  Pump Type: Gear, centrifugal; steel, stainless steel (solutions only).	Label(s): Black and white label - CORROSIVE (>25% water of crystallization). Red, black and white label - SPONTANEOUS COMBUSTIBLE (<25% water of crystallization).  Storage Temperature: Ambient. Hose Type: Natural rubber, Hypalon, polyethylene, polypropylene.	Grades or Purity: Anhydrous 70 to 72% NaHS. Solutions 40 to 44% NaHS. Containers and Materials: Drums, tank cars, tank trucks; steel, stainless steel (aluminum, zinc, copper, brass, bronze are not resistant).
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Liquid (typically shipped as solution). Solubility (Water): Soluble in all proportions. Molecular Weight: 56.1 (solute). Vapour Pressure: No information. Boiling Point: 100 to 122°C (solution).	Floatability (Water): Sinks and mixes. Odour: Rotten-egg odour (0.047 ppm). Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 1.31 (15.5°C).	Colour: Clear, light straw colour (yellow to black at higher temperatures).  Explosive Limits: Not fiammable (4.3 to 46% (H2S)).  Melting Point: 17°C; (solution) loses water;  55°C (anhydrous).

#### HAZARD DATA

# Human Health

Symptoms: Corrosive alkaline liquid. Contact: eyes and skin - causes severe burns. Ingestion: causes severe burning and corrosion of the gastrointestinal tract, pain in throat and abdomen, nausea and vomiting, diarrhea. In severe cases, collapse, unconsciousness and paralysis of respiration may occur. inhalation: of NaHS mist will cause irritation of the respiratory tract and possibly systemic poisoning. Vapours are extremely dangerous as the sense of smell is rapidly lost on exposure to high concentrations or prolonged exposure to dilute concentrations of HoS in air. Texicology: Highly toxic by all routes.

TLV[®] (inhalation) 10 ppm; 14 mg/m³ (as H₂S). Short-term inhalation Limits - 15 ppm;

LC 50 - Inhalation: mouse = 673 ppm/1 h (H2S). Delayed Toxicity - No information.

LDso - No information.

21 mg/m³ for 15 min (as H₂S).

Fire Extinguishing Agents: Not combustible. Most fire extinguishing agents may be used. Water should be used in flooding amounts. However, flammable and toxic H₂S is liberated when heated.

Behaviour in Fire: Not combustible, but releases flammable H2S when heated.

ignition Temperature: Not combustible. Burning Rate: Not combustible.

# Reactivity

With Water: No reaction: soluble.

With Common Materials: NaHS is stable while alkaline. Reacts with acids to release hydrogen sulfide gas. Reacts violently with diazonium salts. Stability: Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 206 mg/L/96 h/mosquito fish/TLm/freshwater. Land-Air: No information.

Food Chain Concentration Potential: No information.

# Special Hazards

CORROSIVE. Upon heating releases H2S (poisonous and flammable). <25% water of crystallization, spontaneously combustible.

# Immediate Responses

Keep non-involved people away from spill site. Issue warnings: "CORROSIVE, SPONTANEOUSLY COMBUSTIBLE". Call Fire Department. Avoid contact or inhalation. Contact manufacturer for advice. If H₂S fumes are present, evacuate people downwind. Contain spill by diking with earth or other available material. Attempt to control discharge, if this can be done without risk. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - if H₂S present, use self-contained breathing apparatus. Goggles - (mono), tight fitting. If face shield used, it must not replace goggles. Gloves - rubber. Boots - high, rubber (pants worn outside boots). Outerwear - as required, acid suit, aprons, etc., rubber or vinyl.

# Fire and Explosion

Not combustible by itself. H₂S is released upon heating; when containing <25% water of crystallization contact with water or moisture may cause spontaneous combustion. Most fire extinguishing agents may be used. Water should be used in flooding amounts.

#### First Ak

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Ingestion: if victim is conscious give large amounts of milk or water. Induce vomiting. Contact: eyes - immediately irrigate with plenty of water; skin - immediately flush affected areas with water and remove contaminated clothing. If medical assistance is not immediately—available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### **Valer**

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- t. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain soill by diking with earth
- or other barrier.
- 4. If fiquid, remove material with pumps or vacuum equipment and place in appropriate containers.
- 4a. If solid, remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice or disposal.

#### SODIUM HYDROXIDE NAOH

# IDENTIFICATION

UN No. 1823 solid 1824 solution

ommon Synonyms Observable Characteristic			
ommon Synonyms Observable Characteristic		Manufacturers	
AUSTIC SODA AUSTIC YE ODA LYE ODIUM HYDRATE HITE CAUSTIC  Clear to slightly turbid lit to slightly coloured (white or slightly coloured pellets or flakes)	e to yellow) appearance. id: White to slightly	Canadian Industries Limited, Becancour, Que. Dow Chemical Canada Inc., Sarnia, Ont., Fort Saskatchewa	Canadlan Occidental Petroleum Ltd., Vancouver, B.C. FMC, Squamish, B.C. an, Alta.
ansportation and Storage Information			
ipping State: Solid and liquid (aqueous dution). lassification: Corrosive. ert Atmosphere: No requirement. enting: Open. amp Type: Gear, centrifugal; steel, ainless steel (for solutions).	Class 8, Group II. Storage Temperature Hose Type: Natural re	ack label - CORROSIVE;	Grades or Purity: Solution, 59%, 73% NaOH. Dry - (anhydrous, 99+% NaOH), flake, powder solid. Containers and Materials: Liquid - tank cars, tank trucks. Dry - cans, drums, hopper cars.
hysical and Chemical Characteristics			
hysical State (20°C, I atm): Solid. Subility (Water): Soluble, 42 g/100 mL °C); I i i g/100 mL (25°C); 347 g/100 mL 00°C). Secular Weight: 40.0 (solute). Sapour Pressure: I mm Hg (739°C); 42 mm Hg 000°C). Solling Point: 50% solution: I42 to I48°C; 3% solution: I88 to 195°C; I 390°C (anhydrous)	73% sofn, 2.00 (15.5°	nmable. information. 3 (anhydrous); 50% sot'n, 1.53;	Colour: Liquid - clear to slightly coloured. Solid (dry) - white. Explosive Limits: Not flammable. Melting Point: Anhydrous 318°C; 50% solution, 12 to 15°C; 73% solution, 63°C.
		7APD DATA	

#### HAZARD DATA

# Human Health

Symptoms: Contact: skin - severe burns, often resulting in deep ulceration and ultimate scarring, can result from contact with solid or liquid forms. Eyes solid or liquid; very rapidly causes severe damage. Inhalation: of dust or mist may cause damage to upper respiratory tract and even to lung tissue proper-ingestion: (liquid or solid) severe damage to mucous membranes or deeper tissues. If perforation occurs, subsequent severe scar formation may occur. Penetration into vital areas may be fatal.

Toxicology: Highly toxic upon contact and ingestion.

TLY*- (inhalation) 2 mg/m³ (particulate).

Short-term inhalation Limits - No Information.

LC₅₀ - No information.

Delayed Toxicity - None known.

LD_{Lo} - Oral: rabbit = 0.5 g/kg (10% solution).

#### Fire

Pire Extinguishing Agents: Not combustible. Most fire extinguishing agents may be used in fires involving sodium hydroxide. Use water in flooding amounts. Anhydrous form in contact with water may generate sufficient heat to ignite combustible materials. May cause ignition on contact with organic chemicals. Behaviour in Fire: Not combustible.

ignition Temperature: Not combustible.

Burning Rate: Not combustible.

# Reactivity

With Water: Anhydrous form dissolves with great heat evolution. Boils and splatters.

With Common Materials: Reacts violently with acetaldehyde, acetic acid, acetic anhydride, acrolein, acrylonitrile, allyl alcohol, allyl chloride, aluminum, chlorohydrin, chloronitrotoluenes, chlorosulfonic acid, 1;2-dichloroethylene, ethylene cyanohydrin, glyoxyl, hydrochloric acid, hydrofluoric acid, hydroquinone, maleic anhydride, nitric acid, nitroparaffins, oleum, phosphorus, phosphorus pentoxide, sulfuric acid, tetrahydrofuran and trichloroethylene.

Stability: Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life in high concentrations. Fish toxicity: 25 ppm/24 h/brook trout/LC100/ freshwater; 99 ppm/48 h/bluegitl/TLm/freshwater; 10 to 33 ppm/48 h/shrimp/LC50/saltwater; BOD: None. Land-Air: No information.

Food Chain Concentration Potential: None.

# Special Hazards

CORROSIVE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department. Contain spill by diking with earth or other available material. Contact manufacturer for advice. Avoid contact and inhalation. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - suitable respirator. Goggles - (mono), tight fitting. If face shield used, it must not replace goggles. Gloves - rubber. Boots - high, rubber (pants should be worn outside boots). Outerwear - as required; coveralls, aprons, suits - rubber, vinyi.

# Pice and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving sodium hydroxide. Use water in flooding amounts. Anhydrous form in contact with water may generate sufficient heat to ignite combustible materials. May cause ignition on contact with organic materials.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - irrigate with plenty of water; skin -flush with plenty of water, while removing contaminated clothing. Continue washing (eyes and skin) for an additional 30 minutes if considered necessary. Ingestion: give large quantities of milk or water to conscious victim only, to dilute chemical. Yomiting may occur spontaneously but should not be induced. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss
- disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

#### SODRIM HYPOCHLORITE NaOCI

UN No. 1791

Common Synonyms/Trade Names	Observable Characteristics	Manufacturers
CHLOROX LIQUID BLEACH	Green to yellow watery liquid. Chlorine-type odour (like bleach). Pure sodium hypochlorite is a powder which is unstable in air, and rarely shipped.	Bristol-Myers Canada, Toronto, Ont. Canadian Industries Ltd., Becancour, Que. MacMillan Bioedel, Nanaimo, B.C. Prince Albert Pulp, Prince Albert, Sask.
Transportation and Storage Information		
Shipping State: Liquid (aqueous solution). Classification: Corrosive liquid. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Steel, stainless steel.	Label(s): Black and white label - CORROSIVE LIQUID; Class 8, Group II.  Storage Temperature: Ambient. Hose Type: Natural rubber, Hypaion, polyethylene, polypropylene, flexible stainless steel.	Grades or Purity: Technical >7% available chlorine; solutions are typically 7 to 15% chlorine. Containers and Materials: Carboys, drums and tanks cars.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): Solutions are soluble in all proportions (26.0 g/100 mL (0°C) (anhydrous). Molecular Weight: 74.4 (solute). Vapour Pressure: No information. Boiling Point: Decomposes at temperatures >100°C.	Floatability (Water): Sinks and mixes. Odour: Like chlorine bleach. Flash Point: Not flammable (solutions). Vapour Density: No information. Specific Gravity: 1.00 to 1.025 at 20°C (solution; depending on concentration).	Colour: Green to yellow.  Explosive Limits: Anhydrous explosive in air; solutions not flammable.  Melting Point: 5% solution, -6°C.

# **HAZARD DATA**

Limbsu	LICEL	<b>4</b> 1	

(15 min) (as chlorine).

Symptoms: Contact: eyes and skin - irritation, burning and inflammation. Inhalation: of fumes causes severe pulmonary irritation with coughing and choking followed by pulmonary edema. Ingestion: irritation and corrosion of mucous membranes with pain and vomiting, stomach cramps, and diarrhea. Toxicology: Moderately toxic by inhalation and ingestion.

TLV - 1 ppm; 3 mg/mL (as chlorine).

Short-term Inhalation Limits - 3 ppm; 9 mg/mL

LC50 - No information. Delayed Toxicity - None known. LDso - No information.

Fire Extinguishing Agents: Solutions are not combustible; however, in fires involving sodium hypochlorite solution, most fire extinguishing materials may be used. In fires involving anhydrous form, water should only be used in flooding amounts.

Behaviour in Fire: Decomposes in fire, generating toxic and corrosive chlorine gas. NaOCI+5H2O highly unstable, anhydrous form explosive in air. ignition Temperature: Solutions not combustible. Burning Rate: Solutions not combustible.

# Reactivity

With Water: No reaction. Soluble. Anhydrous form reacts violently with water.

With Common Materials: Concentrated forms may cause combustible materials to ignite. Reacts violently with amines, ammonium acetate, ammonium carbonate, ammonium nitrate, ammonium phosphate, cellulose and ethyleneimine.

Stability: Solutions are stable. (Anhydrous and pentahydrate are not stable.)

# Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: 5.9 ppm/96 h/fathead minnow/LC50/freshwater (4 to 6% aqueous solution); 52 ppm/96 h/grass shrimps/LC50/freshwater (4 to 6% aqueous solution); BOD: No information.

Land-Air: No information.

Food Chain Concentration Potential: None.

# Special Hazards

CORROSIVE. Concentrated and other forms can be strong oxidizers.

# Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department. Stop discharge if safe to do so. Contain spill by diking. Contact manufacturer. Contact environmental authorities.

# Protective Clothing and Equipment

In lires, Respiratory protection - self-containing breathing apparatus; otherwise rubber gloves and goggles, protective clothing - with pants worn outside boots.

# Fire and Explosion

Solutions not combustible. Most fire extinguishing agents may be used in fire involving NaOC1. Cool fire-exposed containers with water. Pentahydrate and anhydrous forms are not stable. Decomposes in fires generating toxic and corrosive chlorine gas.

# First AL

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - flush with water. Ingestion: if victim is conscious, give milk or water to drink. Keep victim warm. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- t. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 44

# SODIUM NITRATE NaNO3

IDENTIFICATION	
CONTRACTOR SOUTH SOUTH	

UN No. 1498

Common Synonyms	Observable Characteristics	Manufacturers
CHILE SALTPETER SODA NITER NITRATINE	Colourless to white crystals or powder. Odourless.	No Canadian manufacturer. Selected U.S. manufacturer: Olin Corporation, Lake Charles, Louisiana, USA Canadian supplier: Canadian Industries Limited, Toronto, Ont. (and branches across Canada).
Transportation and Storage Information		
Shipping State: Solid. Classification: Oxidizing material. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Yellow label - OXIDIZER; Class 5.1, Group III. Storage Temperature: Ambient.	Grades or Purity: Technical, synthetic - 99.4% NaNO3 min. Containers and Materials: Polyethylene-lined paper bags. Carloads or bulk.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 73 g/100 mL (0°C); 92.1 g/100 mL (25°C); 180 g/100 mL (100°C). Molecular Weight: 85.01 Vapour Pressure: No information. Boiling Point: Decomposes at 380°C.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 2.26 (25°C).	Colour: Colourless to white. Explosive Limits: Explodes at 537°C. Melting Point: 307°C.

	HAZARD DATA	,
Human Health		
Symptoms: Contact: reddening of skin and eyes. <u>In</u> yorniting, convulsions, cyanosis, loss of conscious: Toxicology: Highly toxic by ingestion.		hortness of breath. <u>Ingestion</u> : dizziness, abdominal cramps,
TLV No information.	LCsn - No information.	LDsa - No information.
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	LDLo - Oral: rat = 0.20 g/kg
Fire	······································	
Fire Extinguishing Agents: Not directly combusti Behaviour in Fire: May explode at 537°C. Decon Ignition Temperature: 537°C.	ible; explodes. In fires involving sodium nitrate, nposes with heat to produce toxic NO _x fumes.  Burning Rate: No information.	use flooding amounts of water.
Reactivity		
With Water: No reaction; soluble. With Common Materials: Strong oxidizer, can ign charcoal. Stability: May explode when detonated or heated		ith antimony, cyanides, sodium hypophosphite, sulfur and
	<u></u>	
Environment		

Food Chain Concentration Potential: No information.

# Special Hazards

OXIDIZER.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "OXIDIZER". Call Fire Department. Stop discharge if safe to do so. Remove unbroken bags or drums from spill area. Dike to contain spill. Contact manufacturer. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - in fires, self-contained breathing apparatus. Otherwise; Gloves - rubber. Safety glasses and work clothing, coveralls, apron, etc.

## Fire and Explosion

Not directly combustible; explodes. Emits toxic No. tumes upon heating. Use flooding amounts of water to extinguish.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: skin - remove clothing and flush with large amounts of water. Inhalation: If in fire, fumes of nitrous oxide may be present. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. Ingestion: give milk or water to conscious victim. Induce vomiting. If medical assistance is not quickly available, transport victim to hospital, doctor or clinic.

# ENVIRONMENTAL PROTECTION MEASURES

#### Response

#### **Valer**

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM PENTACHLOROPHENATE NaC6C15O

IDE			

UN No. 2567

Common Synonyms	Observable Characteristics	Manufacturers
SODIUM PENTACHLOROPHENOLATE DOWICIDE-G SODIUM PENTACHLOROPHENOL SODIUM PENTACHLOROPHENOXIDE PENTACHLOROPHENOL, SODIUM SALT	White to light-brown powder. Slight, characteristic odour (phenolic or carbolic odour).	No Canadian manufacturers. Canadian suppliers: P. Leiner & Sons, Toronto, Ont. Monsanto Canada, Mississauga, Ont. Originating from: W.R. Grace & Co. USA, Monsanto, USA
Transportation and Storage Information		
Shipping State: Solid. Classification: Poison. Inert Atmosphere: No requirement. Venting: Open.	Label(s): White label - POISON; Class 6.1, Group II. Storage Temperature: Ambient.	Grades or Purity: Technical.  Containers and Materials: Bags, drums, carlots.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 33 g/100 mL (25°C.) Molecular Weight: 288.4 Vapour Pressure: 0.0002 mm Hg (25°C). Boiling Point: No information.	Floatability (Water): Sinks and mixes. Odour: Slight, characteristic (phenolic or carbolic). Flash Point: Not flammable. Vapour Density: 9.2 Specific Gravity: 2.0 (25°C).	Colour: White to light brown. Explosive Limits: No information. Melting Point: No information.

# HAZARD DATA

•••••		
Symptoms: Inhalation: nose irritation, nausea and consciousness, death due to heart failure, risk of s inflammation and blisters.  Toxicology: Highly toxic by all routes.	vomiting, coughing, high fever, profuse sweating severe pulmonary edema if victim survives. <u>Inges</u>	, rapid difficult breathing; cyanosis, convulsions, loss of stion: same as Inhalation. Contact: skin and eyes - irritation,
TLV (skin) 0.5 mg/m ³ (as pentachiorophenoi).	LCso - No information.	LD50 - Oral: rat = 0.21 g/kg LD50 - Subcutaneous: rat = 0.072 g/kg
Short-term Inhalation Limits - (skin) - 1.5 mg/m ³ (15 min).	Delayed Toxicity - No information.	FD 20 - 200 Chraneonz: Lat = 0.015 Blv8
Fire		
Fire Extinguishing Agents: Not combustible. Mos Behaviour in Fire: When heated to decomposition,	t fire extinguishing agents may be used in fires in text to take fumes of chlorides are evolved.	nvolving sodium pentachlorophenate.
Ignition Temperature: No information.	Burning Rate: No information.	
Ignition Temperature: No information. Reactivity	Burning Rate: No information.	
Ignition Temperature: No information.	Burning Rate: No information.	
Ignition Temperature: No information.  Reactivity  With Water: No reaction; soluble.  With Common Materials: No information.  Stability: Stable.  Environment	Burning Rate: No information.	growth stopped immediately/freshwater; 0.2-0.6 ppm/

# Special Hazards

#### POISON.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Avoid contact and inhalation of dust. Stay upwind. If water is used, dike area to prevent runoff. Notify supplier. Notify environmental authorities.

# Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus. Otherwise use dust respirator. Acid suit (jacket and pants) or coveralls. Goggies - (mono), tight fitting. Gloves - neoprene. Boots - high, rubber (pants worn over boots).

# Fire and Explosion

Not flammable; in fires involving sodium pentachlorophenate; use water, dry chemical, foam or carbon dioxide to extinguish. Cool fire-exposed containers with water. In fire, wear self-contained breathing apparatus to avoid toxic decomposition products (chlorides).

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid immediately. Inhalation: give artificial respiration if breathing has stopped (not mouth-to-mouth method); give oxygen if breathing is laboured. Contact: skin and eyes - remove dust-contaminated clothing and wash eyes and affected skin with plenty of warm water for at least 15 min. Ingestion: give warm water to conscious victim and induce vomiting. Repeat several times. Keep warm and quiet, if medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

#### Tate

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runolf from rainwater or water application.
- 4. Remove material by manual or mechanical
- 5. Recover undamaged containers.
- 6. Absorb residual fiquid on natural or synthetic sorbents.
- 7. Remove contaminated soil for disposal.
- 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM PHOSPHATE (dibasic) NazHPOA

**IDENTIFICATION** UN No. NA9147 Common Synonyms Observable Characteristics Manufacturers Colouriess, translucent powders DSP Commonly shipped in various forms: Erco industries. PHOSPHATE OF SODA anydrous NapHPOu and crystals. Odourless. Port Maitland, Ontario. dihydrate NazHPO4º2H2O Buckingham, Que. DISODIUM HYDROGEN PHOSPHATE DISODIUM PHOSPHATE heptahydrate Na2HPO4-7H2O DISODHUM MONOHYDROGEN PHOSPHATE dodecahydrate NapHPOu*12HpO SODIUM ORTHOPHOSPHATE, SECONDARY SECONDARY SODIUM PHOSPHATE Transportation and Storage Information Shipping State: Solid. Label(s): Not regulated. Grades or Purity: Commercial. Classifications Class 9.2 Storage Temperature: Ambient. Containers and Materials: Paper bags, drums, Inert Atmosphere: No requirement. barrels. Venting: Open. Physical and Chemical Characteristics Floatability (Water): Sinks and mixes. Colour: Colourless to translucent. Physical State (20°C, 1 atm): Solid. Odour: Odourless. Explosive Limits: Not flammable. Solubility (Water): dihydrate 100 g/100 ml. (50°C), 117 g/100 mL (80°C); anhydrous Flash Points Not flammable. Melting Point: Anhydrous 240°C converted to 1.63 g/100 mL (0°C); heptahydrate 104 g/100 mL Vapour Density: No information. sodium pyrophosphate; dihydrate 95°C loses (40°C); dodecahydrate 4.2 g/100 mL (0°C). 2H2O: heptahydrate 48°C loses 5H2O: Specific Gravity: dihydrate 2.07 (15°C); hepta, 1.68 (25°C); dodeca, 1,52 (25°C). 87 g/100 mL (40°C). dodecahydrate 35.1 °C loses 5HoO. Molecular Weight: anhyd. 142; dihydrate 178; hepta. 268; dodeca. 358. Vapour Pressure: No information.

#### HAZARD DATA

Boiling Point: Aphydrous at 240°C is converted to sodium pyrophosphate; dihydrate at 92.5°C loses 2H2O; hepta hydrate at 48°C loses 5H2O; dodecahydrate at 100°C loses 12H2O.

Land-Air: No information.

Food Chain Concentration Potential: None.

# Human Health Symptoms: Inhalation: of dust may irritate nose and throat. Ingestion: injures mouth, throat and gastrointestinal tract resulting in vomiting, cramps and diarrhea: pain and burning in mouth may occur. Contact: eyes - local irritation to chronic damage: skin - dermatitis. Texicology: Low texicity by ingestion. LCsa - No information. LD₅₀ - Oral: rat = 12.9 g/kg (hepta). TLY*- No information. Delayed Toxicity - No Information. LDia - intraperitoneal: rat = 1.0 g/kg (anhydrous). Short-term Inhalation Limits - No information. Fire Extinguishing Agents: Not combustible. Most fire extinguishing agents may be used in fires involving sodium phosphate (dibasic). Behaviour in Fire: Anhydrous forms toxic PO, upon decomposition in fires. ignition Temperature: Not combustible. Burning Rate: Not combustible. Reactivity With Water: No reaction; soluble. With Common Materials: Reacts violently with magnesium. Stability: Stable. Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: 126 ppm/92 h/Daphnia magna/TLm/freshwater.

#### Special Hazards

#### **Inmediate Responses**

Keep non-involved people away from site. Avoid contact and inhalation of dust. Dike to prevent runoff. Notify manufacturer or supplier for advice. Notify environmental authorities.

# Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; otherwise, safety goggles or face shield, dust mask, rubber gloves, rubber boots, and protective clothing as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving sodium phosphate (dibasic). Toxic PO_X fumes are formed upon decomposition in fires.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Dust Contact: eyes - flush with water. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Solid Contact: skin - remove contaminated clothing and flush affected areas with water. Ingestion: give conscious victim plenty of water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Neposal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODRUM PHOSPHATE (monobasic) NaH2PO4

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manulacturers
MSP MONOSODIUM DIHYDROGEN PHOSPHATE MONOSODIUM PHOSPHATE SODIUM BIPHOSPHATE SODIUM ACID PHOSPHATE SODIUM ORTHOPHOSPHATE, primary (also shipped as a hydrate NaH2PO4+H2O)	White, crystalline powder. Odourless. Monohydrate - large translucent crystals.	Erco Industries, Port Maitland, Ont. Buckingham, Que.
Transportation and Storage Information		
Shipping State: Solid. Classification: Not regulated. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Technical; commercial.  Containers and Materials: Paper bags, drums and barrels.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Solid. Solubility (Water): 59.9 g/100 mL (0°C); 427 g/100 mL (100°C) (monohydrate); anhydrous and dihydrate very soluble. Molecular Weight: 120 (anhydrous); 138 (monohydrate). Vapour Pressure: No information. Boiling Point: monohydrate ~ loses water at 100°C; decomposes at 138°C; anhydrous form forms sodium acid pyrophosphate at 225 to 250°C; and sodium metaphosphate at 350 to 400°C.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 2.04 (monohydrate).	Colour: White. Explosive Limits: Not flammable. Melting Point: Decomposes at 204°C; monohydrate loses water at 100°C.

	HAZARD DATA	
Human Health		
diarrhea; pain and burning in mouth may occur. Co	d throat. Ingestion: injures mouth, throat and g intact: eyes - local irritation to chronic damage	pastrointestinal tract resulting in vomiting, cramps and e, skin - dermatitis.
Toxicology: Low by all routes. TLV® No information.	LC ₅₀ - No information.	LD ₅₀ - Intramuscular: rat = 0.25 g/kg
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	
Fixe		
Fire Extinguishing Agents: Not combustible. Mos Behaviour in Fire: At high temperatures, forms high Ignition Temperature: Not combustible.	t fire extinguishing agents may be used in fires ghly toxic PO _X fumes.  Burning Rate: Not combustible.	involving sodium phosphate (monobasic).
Reactivity		
With Water: None; soluble (forms a weak acidic so With Common Materials: Reacts violently with mu Stability: Stable.	lution with water). agnesium.	
Environment		
Water: Prevent entry into water intakes and wate Land-Air: No information. Food Chain Concentration Potential: None.	rways. Harmful to aquatic life. Fish toxicity:	126 ppm/72 h/Daphnia magna/TLm/freshwater.

# Special Hazards

#### Immediate Responses

Keep non-involved people away from site. Avoid contact and inhalation of dust. Dike to prevent runoff. Notify manufacturer. Notify environmental authorities.

# Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; otherwise, safety goggles or face shield, dust mask, rubber gloves, rubber boots, and protective clothing as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving sodium phosphate (monobasic). At high temperatures POx fumes are emitted.

#### Piest Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Dust: Contact: eyes - flush with water. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Solid: Contact: skin - remove contaminated clothing and flush affected areas with water. Ingestion: induce vomiting in conscious victim; give water to drink and keep warm. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Al

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM PHOSPHATE (tribasic) Na₃PO₄-12H₂O

IDENTIFICATION

UN No. NA9148

Common Synonyms		Observable Characte	ristics	Manufacturers
TSP SODIUM ORTHOPHOSPHATE SODIUM PHOSPHATE TRIBASIC SODIUM PHOSPHATE TRISODIUM ORTHOPHOSPHATE	Forms: dodecahydrate Na ₃ PO ₄ ·12H ₂ O Note: information on this sheet is for the dodecahydrate, the most commonly shipped form.	Crystals. Odourless. white.	Clear to	Erco Industries. Port Maitland, Ontario, Buckingham, Quebec.
Transportation and Storage inform	ation			
Shipping State: Solid. Classification: Class 9.2. Inert Atmosphere: No requirement Venting: Open.	Label(s): Not required. Storage Temperature: An .	bient.		or Purity: Commercial, CP. ers and Materials: Barrels and bags.
Physical and Chemical Characteris	stics			
Physical State (20°C, I atm): Solidbility (Water): 1.5 g/100 mL (0 28.3 g/100 mL (15°C); 157 g/100 m Molecular Weight: 380.12 Vapour Pressure: No information. Bolling Point: 73 to 77°C decomposes 12H ₂ O at 190°C.	°C); Odour: Odourless.  il. (70°C). Flash Point: Not flammal Vapour Density: No infor Specific Gravity: 1.62 (20	ole. mation.	Explosiv	Clear to white. re Limits: Not flammable. Point: Decomposes at 73 to 77°C.

# HAZARD DATA

Human Health		
Symptoms: inhalation: nose, eyes and throat, irritistomach cramps. Contact: skin - itching, burning Texicology: Moderately toxic by ingestion and inhalation.	sensation, inflammation; eyes - irritation a alation.	
TLV*- No information.	LC59 - No information.	LD50 - Oral: rat = 7.4 g/kg (hepta.)
Short-term Inhalation Limits - No Information.	Delayed Toxicity - No information.	
Fire		•
Fire Extinguishing Agents: Not combustible. Mor Behaviour in Fire: Not combustible, but at high te ignition Temperature: Not combustible. Reactivity	mperatures degrades and gives off toxic PC Burning Rate: Not combustible.	x fumes.
With Water: No reaction; soluble. Solutions are co With Common Materials: Reacts violently with me Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: None.	erways. Harmful to aquatic life; Fish toxic	ity: 126 ppm/72 h/ <u>Daphnia magna</u> /TLm/freshwater.

# Special Hazards

# immediate Responses

Keep non-involved people away from site. Avoid contact and inhalation of dust. Dike to prevent runoff. Notify manufacturer or supplier. Notify environmental authorities.

# Protoctive Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; otherwise, safety goggles or face shield, dust mask, rubber gloves, rubber boots, and protective clothing as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving sodium phosphate (tribasic). Emits toxic POX fumes upon heating.

#### First Aid

Move victim out of spiil site to fresh air. Call for medical assistance, but start first aid at once. Dust Contact: eyes - flush with water. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Solid Contact: skin - remove contaminated clothing and flush affected areas with water. Ingestion: give plenty of water to conscious victim and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# ENVIRONMENTAL PROTECTION MEASURES

# Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# SODIUM SILICATE Na2 O-nSiO2-xH2O (n = 1 to 5)

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
SILICATE OF SODA WATER GLASS SODIUM METASILICATE (n=1).	Colourless to greenish-white powder. Odourless. Solutions are also colourless to green-white and viscous in appearance.	National Silicates Ltd., Toronto, Ont. Valleyfield, Que.
Transportation and Storage Information		
Shipping State: Solid or liquid (aqueous solution). Classification: Not regulated. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Gear, centrifugal steel, stainless steel (for solutions).	Label(s): Not regulated.  Storage Temperature: Ambient.  Hose Type: Hypalon, polyethylene, polypropylene, natural rubber.	Grades or Purity: 40, 47, 52° Bé. Containers and Materials: Drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): Soluble to very soluble depending on form. Molecular Weight: Varies (122 to 288). Vapour Pressure: No information. Boiling Point: Solutions decompose at about 100°C.	Pioatability (Water): Sinks and mixes.  Odour: Odourless.  Flash Point: Not flammable.  Vapour Density: No information.  Specific Gravity: (solution) 1.1 to 1.8 (20°C); (solid) 2.4 (metasilicate).	Colour: Colourless to greenish-white. Explosive Limits: Not Ilammable. Melting Point: Variable, solutions decompose at about 100°C. Solids: disilicate 874°C; metasilicate I 088°C; orthosilicate I 018°C.

#### HAZARD DATA

	HAZAKO DATA	
Human Health		
Symptoms: Contact: skin - burning and inflamma difficulty in breathing, coughing fits and chemical Toxicology: Low toxicity by all routes.	tion; eyes - pain, watering. <u>Ingestion</u> : burning bronchitis.	, nausea, vomiting and stomach cramps. <u>Inhalation</u> : sneezing,
TLV [®] No information.	LCso - No information.	LD50 - Oral: mouse = 1.10 g/kg
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	LD50 - Oral: rat = 1.30 g/kg
Fire		
Fire Extinguishing Agents: Not combustible. Mos Behaviour in Fire: Not combustible. Ignition Temperature: Not combustible.	t fire extinguishing agents may be used on fire Burning Rate: Not combustible.	s involving sodium silicate.
Reactivity		
With Water: No reaction, soluble. With Common Materials: Reacts violently with fit Stability: Stable.	iorine.	
Environment		
Water: Prevent entry into water intakes and water freshwater; 216 ppm/96 h/Daphnia magna/TLm/la Land-Akr: No information Potential: None.	rways. Harmful to aquatic life in high concentr ke water; BOD: None.	ations. Fish toxicity: 2 320 ppm/96 h/mosquite fish/TLm/

# Special Hazards

#### Immediate Responses

Keep non-involved people away from spill site. Contain spill by diking if there is water runoff from fire or rain. Avoid contact and inhalation. Notify manufacturer. Notify environmental authorities.

# Protective Clothing and Equipment

Safety gongles - tight fitting. Rubber gloves. Protective work clothing, as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving sodium silicate.

#### First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: remove contaminated clothing and wash eyes and skin thoroughly with plenty of water. Ingestion: give water or milk to conscious victim. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Water division.

  4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Contaminated materials may be buried in a secured landfill site (with environmental authorities' approval).

# SODIUM SULFATE Na₂SO_{4*xH₂O (x = 0, 7, 10)}

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
DISODIUM SULFATE GLAUBER'S SALT (Na ₂ SO ₄ ·10H ₂ O)(decahydrate) SALT CAKE Also occurs in two basic mineral forms: thenardite and metathenardite; heptahydrate and decahydrate forms.	White crystals or powder. Odourless.	Saskatchewan Minerals, Chaplin, Sask. Ormiston Mining and Smelting, Ormiston, Fox Valley, Sask., Alberta Sulfate, Metiskow, Alta.
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Technical and CP. Containers and Materials: Bags, drums or bulk.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid.  Solubility (Water): thenardite 4.7 g/100 mL (0°C);  42.7 g/100 mL (100°C); metathenardite 48.8 g/100 mL (40°C); heptahydrate 19.5 g/100 mL (0°C); 48 g/100°mL (20°C); decahydrate  11 g/100 mL (0°C); 36 g/100 mL (15°C);  92.7 g/100 mL (30°C).  Molecular Weight: 142 anhydrous; 268 hepta.;  322 decahydrate.  Yapour Pressure: No information.  Boiling Point: Loses 10H ₂ O at 100°C (decahydrate)	Plash Point: Not flammable.  Vapour Density: No information.  Specific Gravity: thenardite 2.66, metathenardenite 2.70, decahydrate 1.46 (25°C).	Colour: White. Explosive Limits: Not flammable. Melting Point: Hepta turns to anhydrous at 24.4°C; anhydrous 884°C; decahydrate 32.4°C.

# HAZARD DATA

,	HAZAKO DATA	
Human Health		
Symptoms: Ingestion: low toxicity. Contact: ski mechanical irritation and watering. Toxicology: Low toxicity by all routes.	n - not irritating, but prolonged contact	with concentrated solutions or powders should be avoided; eyes -
TLV*- No information.	LC59 - No information.	LDsg - Oral: mouse = 5.99 g/kg (anhydrous)
Short-term inhalation Limits - No information.	Delayed Toxicity - No information.	
Fire		
Fire Extinguishing Agents: Not combustible. Mos Behaviour in Fire: Not combustible. Ignition Temperature: Not combustible.	t fire extinguishing agents may be used in Burning Rate: Not combustible.	fires involving sodium sulfate.
Reactivity		
With Water: No reaction; soluble. With Common Materials: Reacts violently with alu Stability: Stable.	minum under certain circumstances.	
Environment		
Water: Prevent entry into water intakes and wate LC100/freshwater; 16,500 ppm/96 h/mosquito fish, Land-Air: 7 500 mg/L/15 days/poultry/LC33. Food Chain Concentration Potential: No informati	/TLm/treshwater.	kiegill/LC50/freshwater; 5 200 ppm/48 h/ <u>Daphnia magna</u> /

# Special Hazards

#### Immediate Responses

Keep non-involved people away from spill site. Avoid contact and inhalation. Dike to prevent runoff from water application or rainwater. Notify manufacturer. Notify environmental authorities.

# Protective Clothing and Equipment

Safety goggles - (tight fitting). Rubber gloves. Protective clothing, as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving sodium silicate.

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: remove contaminated clothing and wash eyes and skin thoroughly with plenty of water. Ingestion: give water or milk to conscious victim. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

# Response

# Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Contaminated materials may be buried in a secured landfill site (with environmental authorities' approval).

# 400

# SODIUM SULFITE Na2SO3

# IDENTIFICATION

	Observable Characteristics White crystals or powder. Odourless.	Manufacturers International Paper Consolidated. Matane, Que. Abitibi-Price, Kenogami, Que. Bathurst, Bathurst, NB. Lake Utopia Paper, St-George, N.B.
Transportation and Storage Information		·
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Furity: Technical anhydrous; 90 $\pm$ 1%. Containers and Materials: Bags and drums.
Physical and Chemical Characteristics		
	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 2.63 at 15°C (anhydrous); 1.54 at 15°C (heptahydrate).	Colour: White. Explosive Limits: Not flammable. Melting Point: Decomposes.

# HAZARD DATA

•	MAZAKO DATA	
Human Health		
Symptoms: <u>inhalation</u> : (dust) sore throat, coughin and chronic diarrhea, circulatory disturbances, ce Toxicology: Moderately toxic by ingestion.	g, shortness of breath. <u>Contact</u> : skin and eyes intral nervous system depression and death.	- causes redness, pain, inflammation. <u>Ingestion</u> : violent
TLV - No information.	LC50 - No information.	LD50 - intravenous: rat = 0.115 g/kg (anhydrous).
Short-term Inhalation Limits - No information.	Delayed Toxicity - No information.	LDLo - Oral: rabbit = 2.83 g/kg (anhydrous). LD50 - intraperitoneal: mouse = 0.277 g/kg (hepta).
Fire		
Fire Extinguishing Agents: Not combustible. In fit Behaviour in Fire: Not combustible. Ignition Temperature: Not combustible.	res involving sodium sulfite, most fire extinguish  Burning Rate: Not combustible.	ing agents may be used.
Reactivity		
With Water: No reaction; soluble. With Common Materials: No information. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water	rways. Hazardous to aquatic life in high concen	strations. Fish toxicity: 2 600 ppm/24, 48 and 96 h/

#### Special Hazards

## immediate Responses

Keep non-involved people away from spill site. Avoid contact. Dike area to prevent runoff from rainwater. Contact manufacturer. Contact environmental authorities.

## Protective Clothing and Equipment

Salety goggles - tight fitting. Rubber gloves. Protective clothing as required.

### Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving sodium sulfite.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: remove contaminated clothing and wash with plenty of water. Ingestion: give water or milk to conscious victim. If medical help is not immediately available; transport victim to clinic, doctor or hospital.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or
- or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoif from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Contaminated materials may be buried in a secured landfill site (with approval of environmental authorities).

## STEARIC ACID CH3(CH2)16CO2H

## IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
HYDROFOL ACID OCTADECANOIC ACID n-OCTADECYLIC ACID STEAROPHANIC ACID I-HEPTADECACARBOXYLIC ACID	White to pale yellow solid with a slight fatty odour.	Proctor and Gamble Canada, Hamilton, Ont. Emery Industries, Toronto, Ontario. Canada Packers, Toronto, Ontario. Armak Chemicals, Saskatoon, Saskatchewan.
Transportation and Storage Information		
Shipping State: Solid. Classification: Not regulated. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: USP; Commercial; triple pressed; double pressed. Most commercial "stearic acid" is 45% palmitic acid, 50% stearic acid, and 5% oleic acid.  Containers and Materials: Cans, barrels, bags, bulk by truck and train.
Physical and Chemical Characteristics	•	
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.034 g/100 mL (25°C); 0.1 g/100 mL (27°C). Molecular Weight: 284.5 Vapour Pressure: 1.0 mm Hg (173.7°C). Boiling Point: 361 to 383°C (may also decompose at these temperatures).	Floatability (Water): Floats. Odour: Tallow-like; 20 ppm odour threshold. Flash Point: 196°C (c.c.). Vapour Density: 9.8 Specific Gravity: 0.94 at 20°C.	Colour: White to pale yellow. Explosive Limits: No information. Melting Point: 69 to 72°C.

	HAZARU UATA	
Human Health		
Symptoms: Generally considered nontoxic. Inhalat Toxicology: Low toxicity by all routes.	ion of dust irritates nose and throat. Dust cause	es mild eye irritation.
TLV [®] No information. Short-term inhalation Limits - No information.	LC 50 - No information. Delayed Toxicity - No information.	LD50 - Intravenous: rat = 0.022 g/kg LD50 - Intravenous: mouse = 0.036 g/kg
Fire		
Behaviour in Fire: No severe toxic products are prignition Temperature: 395°C.  Reactivity	oduced by combustion.  Burning Rate: No information.	
With Water: No reaction. With Common Materials: No information. Stability: Stable.		
Environment		· · · · · · · · · · · · · · · · · · ·
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: None.	rways. Harmful to aquatic life. BOD: 144%, 5 da	ays.

#### Special Hazards

COMBUSTIBLE.

## immediate Responses

Keep non-involved people away from spill site. Call Fire Department. Avoid contact with solid and dust. Stop discharge if safe to do so. Notify manufacturer. Notify environmental authorities.

## Protective Clothing and Equipment

Safety gaggles - tight fitting. Rubber gloves. Protective clothing as required.

## Fire and Explosion

Dry chemical or carbon dioxide. Water or foam may cause frothing.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Dust Contact: eyes - flush with water. Inhalation: if breathing has stopped, give artificial respiration; if laboured, give oxygen. Solid Contact: remove contaminated clothing and flush affected areas with water; eyes - flush with water. Ingestion: if victim is conscious, give milk or water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other

## barrier.

- Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Contaminated materials may be buried in a secured landfill site (with environmental authorities' approval).

## STYRENE MONOMER (inhibited) C4H5CHtCH2

<b>IDENTIF</b>	10.1	THE	~
MATERIAL SECTION SECTI		11 P 3	V 3

UN No. 2055

Common Synonyms	Observable Characteristics	Manufacturers
VINYLBENZENE PHENYLETHYLENE CINNAMENE ETHENYLBENZENE STYROL	Colourless to yellowish liquid. Sweet odour.	Polysar, Sarnia, Ontario. Dow Chemical Canada Inc., Sarnia, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Open (flame arrester). Pump Type: Centrifugal, gear; steel, most materials not containing copper.	Label(s): Red label - FLAMMABLE; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Fluorelastomer, stainless steel, Teflon.	Grades or Purity: Commercial, 99.5% min; technical, 99.2%; polymer, 99.6%. Containers and Materials: Tank cars, tank trucks, drums, cans; steel, galvanized iron, black iron.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.028 g/100 mL (15°C); 0.03 g/100 mL (20°C); 0.04 g/100 mL (40°C). Molecular Weight: 104.2 Vapour Pressure: 4.3 mm Hg (15°C); 9.5 mm Hg (30°C). Bolling Point: 145.2°C.	Floatability (Water): Floats. Odour: Sweet odour (0.15 to 25 ppm, odour threshold). Flash Point: 37°C (o.c.); 32°C (c.c.). Vapour Density: 3.6 Specific Gravity: 0.91 (25°C).	Colour: Colourless to yellow. Explosive Limits: 1.1 to 6.1%. Melting Point: -31°C.

#### HAZARD DATA

### Human Health Symptoms: Contact: eyes - irritation, possibly slight transient corneal injury; skin - moderate irritation upon prolonged or repeated contact. Not likely to be absorbed through skin in toxic amounts. Ingestion: Irritation of lips, mouth and throat; pain in swallowing, abdominal pain, nauses and vomiting, diarrhea, shock, convulsions may occur. Inhalation: Irritation of mucous membranes, difficulty in breathing, coughing, bluish face and lips, dizziness, fatigue, pneumonia. Texicology: Moderate toxicity by inhalation or ingestion. TLV*- (inhalation) 50 ppm; 215 mg/m³. LC50 - No Information. LDso - Oral: rat = 5.0 g/kg TCLo - Inhalation: human = 376 to 600 ppm Delayed Toxicity - Possible kidney damage from Short-term inhalation Limits - 100 ppm; 425 mg/m³ (15 min).

ingestion. Suspected carcinogen.

Fire

Fire Extinguishing Agents: Use dry chemical, foam or carbon dioxide; water may be ineffective but may be used to keep fire-exposed containers cool and knock down vapours.

Behaviour in Fire: At elevated temperatures, polymerization may occur. Fire-exposed containers may rupture. Flashback may occur along vapour trail. ignition Temperature: 490°C. Burning Rate: 5.2 mm/min.

Reactivity

With Water: No reaction.

With Common Materials: Reacts vigorously with oxidizers and metallic halides. Reacts violently with chlorosulfonic acid, oleum and sulfuric acid. Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Fish toxicity: 22 ppm/96 h/bluegill/TLm/freshwater: 26 mg/L/24 h/gold(ish/LDso/freshwater: 68 mg/L/24 h/brine shrimp/TLm/saltwater; Aquatic toxicity rating = 10 to 100 ppm/96 h/TLm/treshwater; BOD: 55 to 195%, 5 days. Land-Air: No information.

Food Chain Concentration Potential: None.

## Special Hazards

FLAMMABLE. May polymerize violently upon heating.

## Immediate Responses

Keep non-involved people away from spill area. Issue warning: "FLAMMABLE". Call Fire Department. Eliminate all ignition sources. Avoid contact and inhalation. Call manufacturer or supplier for guidance. Contain spill by diking. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

## Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; otherwise, Coggles - (mono), tight fitting; Cloves - rubber. Boots - high, rubber (pants worn outside boots). Protective clothing - coveralls. In major spill or fire - totally encapsulated suit.

## Fire and Explosion

Dry chemical, foam or CO2 to extinguish. At elevated temperatures, such as in fire, polymerization may occur. Fire-exposed container, may rupture. Fire may flash back along vapour trail.

### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - Irrigate with pienty of water immediately for at least 15 minutes; skin - flush with pienty of water for at least 15 minutes; at same time, remove contaminated clothing. Inhalation: if breathing stops, give artificial respiration; if laboured, give oxygen. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge il sale to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier. Pump water into area to prevent soil penetration.
- Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Remove material by manual or mechanical
- 6. Recover undamaged containers.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## SULFUR Se

#### **IDENTIFICATION**

UN No. 1350 2448 molten

Common Synonyms	Observable Characteristics	Manufacturers
SULPHUR BRIMSTONE FLOWERS OF SULFUR SULFUR FLOUR	Yellowish powder or lumps. Pure is odourless; with impurities, faint rotteneggs type odour.	Aquitaine of Canada, Ram River, Alta. Chevron Standard, Kayob, Alta. Sheli Canada, Waterton, Alta. Amoco/Texasguif, Windfall, Alta.
Transportation and Storage Information		
Shipping State: Solid, liquid (molten). Classification: Not regulated. Inert Atmosphere: No requirement. Venting: No requirement. Pump Type: Centrifugal; steel, stainless steet. Relevant for molten material only.	Label(s): Voluntary red and white striped label - FLAMMABLE; Class 4.1, Group III. Storage Temperature: Ambient (molten, 142°C). Hose Type: Flexible steel or stainless steel. Molten material only.	Grades or Purity: Technical, crude, refined, high purity. Containers and Materials: Bags, barrels, bulk, carlots or truck loads.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): Insoluble. Molecular Weight: 256.5 (atomic weight: 32.1) Vapour Pressure: <0.0001 mm Hg (20°C); 0.11 mm Hg (140°C); 1 mm Hg (184°C); 10 mm Hg (244°C). Boiling Point: 444.6°C.	Floatability (Water): Sinks (finely divided powder may trap air and remain floating). Odour: Faint rotten-egg odour (impure); pure form odourless.  Flash Point: Pure sulfur, 188 to 207°C; impure sulfur as low as 168°C.  Vapour Density: 8.9  Specific Gravity: 1.92 to 2.07 (20°C); 1.80 (molten) (120°C).	Colour: Yellow.  Explosive Limits: 35 to 1 400 g/m ³ dust.  Melting Point: 113 to 122°C.

## HAZARD DATA

#### Human Health

Symptoms: Inhalation: sore throat, coughing. Contact: eyes and skin - (dust) irritation (burns with molten). Ingestion: irritation of mouth, sore throat, diarrhea.

Texicology: Low toxicity by all routes.

TLY .- No information.

Short-term Inhalation Limits - No information.

LC₅₀ - No information.

Delayed Toxicity - No information.

LD50 - No information.

## Fire

Fire Extinguishing Agents: Use dry chemical, carbon dioxide or water spray. Avoid straight streams of water which will scatter molten sulfur and dust. Behaviour in Fire: Produces toxic SO_x fumes. Dust may explode if there is an ignition source. Ignition Temperature: 232 to 266°C (190°C; dust).

Burning Rate: No information.

## Reactivity

With Water: No reaction.

With Common Materials: May react with oxidizing materials under some conditions. Reacts violently with halogens, carbides, zinc, tin, sodium, nickel, phosphorus, potassium, calcium, aluminum, ammonia, ammonium nitrate, potassium permanganate, ammonium perchlorate, barium halites, bromates, calcium chlorate, calcium hypochlorite, calcium iodate, charcoal, chlorates, chlorine oxides, chromic anhydride, fluorine, hydrocarbons, iodates, lead halides, lead dioxide. Lithium, magnesium halides, mercury oxides, perchlorates, potassium halides, potassium perchlorate, silver halides, sodium halides, sodium nitrate and zinc halides.

Stability: Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life in high concentrations. Fish toxicity: 10 000 ppm/96 h/mosquito fish/TLm/ freshwater; 16 000 ppm/5 h/goldfish/LC100/freshwater (as colloidal suspension); 2 100 ppm/1 h/goldfish/LC100 freshwater; BOD: No information. Land-Air: No information.

Food Chain Concentration Potential: None.

## Special Hazards

COMBUSTIBLE.

## immediate Responses

Keep non-involved persons away from spill site. Issue warning: "COMBUSTIBLE". Call Fire Department. Call supplier for advice. Avoid contact and inhalation. Stop discharge if this can be done without risk. Notify environmental authorities.

## Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; otherwise; Goggles - (mono), tight fitting. Gloves - boots - coveralls.

## Fire and Explosion

Use dry chemical, carbon dioxide or water spray. Avoid straight streams of water. Toxic SO_X fumes produced in fires.

### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - irrigate with plenty of warm water for at least 15 minutes. Inhalation: remove to fresh air. If breathing has stopped, give artificial respiration; if laboured, give oxygen. Ingestion: give water to conscious victim to drink. Induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## SULFUR DIOXIDE SO2

IDENTIFI	

UN No. 1079

Common Synonyms	Observable Characteristics	Manufacturers
SULFUROUS ACID, ANHYDRIDE SULFUROUS OXIDE	Colouriess liquid or gas. Strong, pungent odour.	Canadian Industries Ltd., Copper Cliff, Ont, Cominco Ltd., Trail, B.C.
Transportation and Storage Information		
Shipping State: Gas or liquid (compressed gas). Classification: Poison. Inert Atmosphere: No requirement. Venting: Safety-relief valve. Pump Type: No information.	Label(s): White label - POISON; Class 2.3. Storage Temperature: Ambient. Hose Type: Flexible steel or stainless steel.	Grades or Purity: Commercial, 99.9% SO ₂ ; refrigeration, 99.98% SO ₂ . Containers and Materials: Cylinders, ton containers, tank cars; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Gas. Solubility (Water): 23 g/100 mL (0°C); 0.58 g/100 mL (90°C). Molecular Weight: 64.1 Vapour Pressure: 2 538 mm Hg (21°C). Boiling Point: -10°C.	Floatability (Water): Mixes (liquid SO ₂ sinks).  Odour: Strong, pungent (3 ppm, odour threshold).  Flash Point: Not flammable.  Vapour Density: 2.3 (0°C).  Specific Gravity: (liquid) 1.45 (-10°C).	Colour: Colourless. Explosive Limits: Not flammable. Melting Point: -75.8°C.

HAZARD DATA			
Human Health			
Symptoms: Inhalation: sore throat, coughing redness, pain; skin - redness, pain, burns. Co Toxicology: Highly toxic by all routes.	<ul> <li>shortness of breath, laboured breathing, bronchitis, respintant with liquid produces frostbite.</li> </ul>	iratory paralysis. <u>Contact:</u> eyes - blurred vision,	
TLY (inhalation) 2 ppm; 5 mg/m ³ .	LC5g - No information.	LD50 - No information.	
Short-term Inhalation Limits - 5 ppm;	LCLo - Human: Inhalation = 400 ppm (1 min).		
10 mg/m ³ (15 min).	TC _{Lo} - Human: Inhalation = 4 ppm (I min). Delayed Toxicity - No information.		
Fire			
Fire Extinguishing Agents: Not combustible. involving sulfur dioxide. In fires involving SC Behaviour in Fire: Not combustible.	Stop flow of gas before attempting to extinguish fire. M 2, cool fire-exposed containers with water.	ost fire extinguishing agents may be used on fires	
ignition Temperature: Not combustible.	Burning Rate: Not combustible.		

## Reactivity

With Water: Soluble; reacts to form sulfurous acid.

With Common Materials: Reacts violently with acrolein, aluminum, chlorates, chromium, ferrous oxide, fluorine, manganese, potassium chlorate, sodium carbide and stannous oxide.

Stability: Stable.

## Environment

Water: Prevent entry into water Intakes and waterways. Harmful to aquatic life. Fish toxicity: 5 ppm/1 h/trout/LC100/freshwater; 16 ppm/1 h/sunfish/ LC100/freshwater; BOB: No information.
Land-Air: Concentrations >1 ppm injurious to plant foliage.
Food Chain Concentration Potential: None.

## Special Hazards

POISON.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Call Fire Department. Contact manufacturer for advice and assistance. Shut off leak; if safe to do so. In cold weather, if liquid SO2 is discharging, it may be possible to safely dike the spill, wearing full protective equipment. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - use self-contained breathing apparatus. Gloves - rubber. Outerwear - coveralls or gas tight suit in high gas concentrations.

## Fire and Explosion

Not combustible. Stop flow of gas before attempting to extinguish fire. Most fire extinguishing agents may be used on fires involving sulfur dioxide, cool fire-exposed containers with water.

## First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - Irrigate with plenty of water for at least 15 minutes. If pain persists, irrigate for another 15 minutes; skin - flush skin immediately with plenty of soapy water for at least 15 minutes. At same time remove contaminated clothing. Treat as for frostbite. Inhalation: if breathing has stopped, start artificial respiration at once. If breathing is laboured, administer oxygen. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier, if possible.
- 4. Recover undamaged containers.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## SULFURIC ACID H2SO4

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UN No. 1832

Common Synonyms	Observable Characteristics	Manufacturers
HYDROGEN SULPHATE FERTILIZER ACID BATTERY ACID DIPPING ACID (See Oleum for concentrated solutions).	Colourless to brown liquid. Sharp, penetrating odour.	(CIL) Canadian Industries Ltd., Copper Cliff, Ont. ESSO Chemicals Canada, Redwater, Aita. Texasgulf Canada, Timmins, Ontario. Western Co-operative Fertilizers, Calgary, Alta.
Transportation and Storage Information		•
Shipping State: Liquid. Classification: Corrosive liquid. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Centrifugal; alloy 20 (for 70% and up).	Label(s): White and black label - CORROSIVE; Class 8, Group II. Storage Temperature: Ambient. Hose Type: Chemiflex 951 (polypropylene), flexible stainless steel - rigid pipe and swivel joints.	Grades or Purity: Commercial, 52° Bé 65.1% H2SO4; 58° Bé 74.4% H2SO4; 60° Bé 77.7% H2SO4; 66° Bé 93.2% H2SO4. Containers and Materials: Bottles, carboys, (lined) drums, tank trucks, tank cars; stainless steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Liquid. Solubility (Water): Soluble in all proportions. Molecular Weight: 98.1 (pure). Vapour Pressure: "I mm Fig at 38°C for 66° Bé. Boiling Point: (66° Bé), "281°C.	Floatability (Water): Sinks and mixes. May react vigorously. Odour: Sharp, penetrating (1 mg/m ³ odour threshold). Flash Point: Not flammable. Vapour Density: 2.8 (20°C) (SO ₃ ). Specific Gravity: 52° Bé 1.56; 58° Bé 1.67; 60° Bé 1.71; 66° Bé 1.84.	Colour: Clear to dark brown. Explosive Limits: Not flammable. Melting Point: (66° Bé), -32°C; 100% 10.4°C; (52° Bé), - 40°C; 58° Bé, -44°C; (60° Bé), -8°C.

## HAZARD DATA

### Human Health

Symptoms: Highly concentrated sulfuric acid is rapidly destructive to body tissues on contact. Contact: skin - dermatitis and burns; eyes - rapidly causes severe damage, and possible loss of sight. Inhalation: of concentrated vapour or mist will cause damage to the upper respiratory tract and lung tissue, sore throat, coughing, laboured breathing. Ingestion: sore throat, abdominal pain, nausea and vomiting.

Texicology:

TLV*- (inhalation) 1 mg/m3.

Short-term inhalation Limits - No information.

LC50 - Inhalation: guinea pig = 18 mg/m³.

Delayed Toxicity - None known.

LD50 - Orak rat = 2.14 g/kg

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Fire Extinguishing Agents: Not combustible. Use dry chemical to fight adjacent fires.

Behaviour in Fires Not combustible. In lires, toxic SO_X fumes may be released. May react with metals producing flammable H₂ gas.

ignition Temperature: Not combustible.

Burning Rate: Not combustible.

#### Reactivity

With Water: Sokible. Concentrated solutions may react violently producing toxic SO_Y fumes.

With Common Materials: Powerful oxidizer; concentrated solutions may ignite organic materials. Can react violently with acetic anhydride, acetonitrile, acrolein, acrylonitrile, allyl alcohol, allyl chloride, ammonium hydroxide, aniline, n-butyraldehyde, carbides, chlorates, chlorosulfonic acid, epichlorohydrin, ethylenediamine, ethylene glycol, hydrochloric acid, hydrochloric acid, iron, isoprene, metals (powdered), perchlorates, phosphorus, potassium-t-butoxide, potassium permanganate, propylene oxide, pyridine, sodium, sodium carbonate, sodium chlorate, sodium hydroxide, steel, styrene monomer, vinyl acetate and zinc chlorate.

Stability: Stable (within the limits of the foregoing).

#### Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: 10 to 24.5 mg/L/24 h/bluegill/iethal/freshwater; 45.2 ppm/48 h/prawn/LC50/saltwater; 138 ppm/4 h/goldfish/lethal/freshwater; 80 to 90 ppm/48 h/shrimp/LC50/saltwater; 80D: None. Land-Air: No information.

Food Chain Concentration Potential: None.

## Special Hazards

CORROSIVE. Reactive with water and other common materials.

### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Call Fire Department. Eliminate all ignition sources. Contact manufacturer for advice. Contain spill by diking with earth or other material. Avoid inhalation and contact. Stop or reduce discharge if this can be done without risk. Notify environmental authorities.

#### Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Gloves - gauntlet type, rubber, vinyl, Boots - high, rubber or neoprene (pants worn outside boots).

### Fire and Explosion

Not combustible. Avoid use of water. Toxic SO_x fumes are released in fires.

## First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: eyes - irrigate immediately with plenty of water; skin - flush with plenty of water, and remove contaminated ciothing. Ingestion: give plenty of water to conscious victim to drink to reduce acid concentration. Do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Remove material by manual or mechanical means.
- 6. Remove contaminated soil for disposal or neutralize with lime.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## SULFURYL CHLORIDE SO2C12

IDE	NTIFIC	ATION

UN No. 1834

Common Synonyms	Observable Characteristics	Manufacturers
SULFONYL CHLORIDE SULFURIC OXYCHLORIDE CHLOROSULFURIC ACID SULFURIC CHLORIDE	Colouriess to yellow. Acrid, choking odour.	
Transportation and Storage Information		•
Shipping State: Liquid, Classification: Corrosive. inert Atmosphere: No requirement. Venting: Pressure-vacuum.	Label(s): Black and white label - CORROSIYE. Storage Temperature: Ambient.	Grades or Purity: Technical 99%. Containers and Materials: Drums, carbuoys; plastic-lined or steel.
Physical and Chemical Characteristics		
Physical State: Liquid. Solubility (Water): Reacts to form hydrochloric and sulfuric acids. Molecular Weight: 135.0 Vapour Pressure: 99.8 mm Hg (18°C). Boiling Point: 69.1°C.	Floatability (Water): Sinks and reacts to form hydrochloric and sulfuric acids. Odour: Acrid, choking. Flash Point: Not flammable. Vapour Density: 4.6 Specific Gravity: 1.67 (20°C).	Colour: Colouriess to yellow. Explosive Limits: Not flammable. Melting Point: -54.1°C.

Human Health		
Symptoms: Inhalation: sore throat, shortness eyes-burns, blurred vision, inflammation.  Toxicology: Highly toxic by all routes.  TLV 2 (as SO ₂ ) 2 ppm, 5 mg/m ³ ; (as C1) 1 ppm, 3 mg/m ³ .  Short-term inhalation Limits - No information	LC 50 - No information. Delayed Toxicity - No information.	rns, abdominal pain. <u>Contact</u> : skin - burns, inflammation;  LD50 - No information.
Fire		
with production of hydrochloric and sulfuric	acids. Water may be used to cool fire-exposed conti ition emits highly toxic SO _X and chloride lumes. Burning Rate: Not combustible.	ept water. Contact with water may produce a violent reaction ainers.
Reactivity		
With Water: Reacts to produce hydrochloric With Common Materials: Reacts violently w Stability: Stable.	and sulfuric acids. Reaction may be violent. Ith lead dioxide.	·
Environment		
Water: Prevent entry into water intakes and Land-Air: No information. Food Chain Concentration Potential: None.	waterways. Aquatic toxicity rating = 10 to 100 ppm	n/96 h/TLm/íreshwater.

#### Special Hazards

CORROSIVE.

## Immediate Responses

Keep non-involved people away from spill site. Issue warning "CORROSIVE". Avoid contact and inhalation. Call manufacturer or supplier. Dike to contain material. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

## Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires involving sulfuryl chloride, except water. Contact with water may produce a violent reaction with production of hydrochloric and sulfuric acids. Water may be used to cool fire-exposed containers. When heated to decomposition, emits highly toxic SO_x and chloride lumes.

### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes -irrigate with water. Ingestion: give water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming
- or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- I. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## TALL OIL

## IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
TALLOL LIQUID ROSIN  (A mixture of rosin and fatty acids produced	Yellow to brown liquids. Acrid odour, from wood pulp.)	Hercules Canada, Burlington, Ontario.  B.C. Chemical, Prince George, B.C. Great Lakes Forest Products, Thunder Bay, Ontario. Boise Cascade, Fort Frances, Ontario. Prince Albert Pulp, Prince Albert, Saskatchewan.
Transportation and Storage Information		
Shipping State: Liquid. Classification: None. Inert Atmosphere: No requirement. Venting: Open (flame arrester). Pump Type: Most types.	Label(s): Not regulated. Storage Temperature: Ambient. Hose Type: Most types.	Grades or Purity: Crude, refined. Containers and Materials: Drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Liquid. Solubility (Water): Insoluble. Molecular Weight: Variable. Vapour Pressure: No information. Boiling Point: High-	Floatability (Water): Floats. Odour: Acrid. Flash Point: 182°C (c.c.) Vapour Density: No information. Specific Gravity: 0.95 to 0.99 (20°C).	Colour: Yellow to brown. Explosive Limits: No information. Melting Point: Variable.

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Human Health		
		tale disember shallow remieration unconscious
ness. Contact: skin - irritation; eyes - irritation.	athing, respiratory tract irritation. ingestion	n: nausea, vomiting, diarrhea, shallow respiration, unconscious-
Toxicology: Low toxicity, by all routes.	1.C No information	LDsg - No information.
TLV No information.	LC50 - No information.  Delayed Toxicity - No information.	
Short-term Inhalation Limits - No information.	Delayed Toxicity - 140 antermetion	
Fire		
Fire Extinguishing Agents: Foam, dry chemical or Behaviour in Fire; No information.  Ignition Temperature: No information.	carbon dioxide, water may be ineffective bu Burning Rate: No information.	it may be used to cool fire-exposed containers.
Reactivity		
With Water: No reaction. With Common Materials: Can react with oxidizing Stability: Stable.	agents.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: None.	-ways.	

## Special Hazards

COMBUSTIBLE.

## Immediate Responses

Keep non-involved people away from spill site. Notify manufacturer or supplier. Dike to contain material. Stop or reduce discharge if safe to do so. Notify environmental authorities.

## Protective Clothing and Equipment

Outer protective clothing as required.

## Fire and Explosion

Use foam, dry chemical or carbon dioxide to extinguish. Water may be ineffective, but may be used to cool fire-exposed containers.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Ingestion: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water: eyes - irrigate with plenty of water. Ingestions give plenty of water to conscious victim to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Remonte

#### **Valer**

- i. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
  4. Remove material with pumps or vacuum equipment and and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal
- and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Incinerate (approval of environmetal authorities required).

# 412

# TEREPHTHALIC ACID C6H4(COOH)2

## IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
1,4-BENZENEDICARBOXYL ACID p-BENZENEDICARBOXYLIC ACID TPA p-PHTHALIC ACID	White crystals or powder. Slight acidic odour.	No Canadian manufacturers Canadian supplier: Millhaven Fibres, Millhaven, Ontario. Originating from: Amoco, USA
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Commercial, fibre.  Containers and Materials: Bags, barrels, bulk, truck, carlots.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 9.002 g/100 ml. (25°C). Molecular Weight: 166.1 Vapour Pressure: 0.5 mm Hg (120°C). Boiling Point: Sublimes >300°C.	Ploatability (Water): Sinks. Odour: Slight acidic. Flash Point: 260°C (o.c.). Vapour Density: 5.74 (21°C). Specific Gravity: 1.5 (20°C).	Colour: White. Explosive Limits: (dust) 0.05 g/L (LEL). Melting Point: Sublimes >300°C.

Human Health		
Symptoms: Contact: skin - slight irritation, rednes Toxicology: Low toxicity by all routes. TLV® No information. Short-term inhalation Limits - No information.	s, pain. <u>Inhalation</u> : of dust or vapours irritating, LC50 - No information. Delayed Toxicity - No information.	, sore throat, coughing.  LD 50 - Oral: rat = 18.8 g/kg  LD 50 - Intraperitoneal: mouse = 1.43 g/kg
Fire \		
Fire Extinguishing Agents: Use dry chemical or ca	rbon dioxide. Water or foam may cause frothing	g.
Behaviour in Fire: No information. Ignition Temperature: 496°C.	Burning Rate: No information.	
Reactivity		
With Water: No reaction. With Common Materials: May react with oxidizing Stability: Stable.	g agents.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: Decomposition by soil microorganisms i Food Chain Concentration Potential: No informati	n 2 days.	

## Special Hazards

COMBUSTIBLE.

#### immediate Responses

Keep non-involved people away from spill site. If burning, call Fire Department. Avoid contact and inhalation. Contact manufacturer. Contact environmental authorities.

## Protective Clothing and Equipment

Safety goggles - tight fitting. Rubber gloves. Protective clothing as required.

## Fire and Explosion

Extinguish fire with dry chemical or carbon dioxide. Water or foam may cause frothing.

## First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - flush with water; skin - remove contaminated clothing and flush affected areas with water. Inhalation: move to fresh air. Ingestion: give water to conscious victim to drink. Transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 4/4

# TERPHENYLS (C6H5)2C6H4(o,m,p)

## IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
DIPHENYLBENZENE BENZENE-DIPHENYL	Colouriess or light yellow solids. Odouriess.	No Canadian manufacturers.
Transportation and Storage Information	· · · · · · · · · · · · · · · · · · ·	***************************************
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Technical. Containers and Materials: Bags, drums.
Physical and Chemical Characteristics	· · · · · · · · · · · · · · · · · · ·	
Physical State (20°C, 1 atm): Solid. Solubility (Water): Insoluble. Molecular Weight: 230.3 Vapour Pressure: No information. Bolling Point: 0, 332°C; m, 365°C; p, 405°C.	Floatability (Water): Sinks. Odour: Odourless. Flash Point: 0, 163°C; m, 191°C; p, 207°C (o.c.) Vapour Density: 7.95 Specific Gravity: 0, 1.14; m, 1.16; p, 1.24 (20°C).	Colour: Colourless to light yellow. Explosive Limits: No information. Melting Point: 0,58°C; m, 86 to 89°C; p, 213°C.

Numen Health		
Symptoms: Contact: eyes and skin - irritating. In Texticology: Moderately toxic by ingestion or inhal	gestion: burning sensation in throat. <u>Inhalation:</u> burni ation.	ng sensation, lung damage.
TLV - 0.05 ppm; 5 mg/m ³ . Short-term inhalation Limits - No information.	LC50 - No infórmation. Delayed Toxicity - Possible injury to liver and kidneys.	LD ₅₀ - No information. LD _{LO} - Oral: rat = 0.5 g/kg
Fire		
Fire Extinguishing Agents: Use dry chemical or ca Behaviour in Fire: No information. Ignition Temperature? Not available.	rbon dioxide; water or foam may cause frothing.  Burning Rate: No information.	
Reactivity		
With Water: No reaction. With Common Materials: Reacts with strong oxidi Stability: Stable.	zers.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No informat		

## Special Hazards

POISONOUS, COMBUSTIBLE.

## immediate Responses

Keep non-involved people away from spill site. If burning, call Fire Department. Avoid contact and inhalation. Contact manufacturer. Contact environmental authorities.

## Protective Clothing and Equipment

In fires or enclosed spaces; Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Safety goggles - tight fitting. Rubber gloves. Protective clothing as required.

## Fire and Explosion

Use dry chemical or carbon dioxide to extinguish; water or foam may cause frothing.

#### First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - flush with water immediately; skin - flush with water, remove contaminated clothing. Inhalation: give artificial respiration if necessary (not mouth-to-mouth method). Ingestion: do not induce vomiting. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or
- or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce dischagrge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## TETRAETHYL LEAD (C2H5)4Pb

## IDENTIFICATION

UN No. 1649

Common Synonyms	Observable Characteristics	Manufacturers
MOTOR FUEL ANTIKNOCK COMPOUND TEL TETRAETHYL PLUMBANE TML, TETRAMETHYL LEAD	Liquid, usually containing a coloured dye (red, blue) (colourless in pure state). Sweet, pleasant odour.	Dupont of Canada Ltd., Maitland, Ont. Ethyl Corporation of Canada Ltd., Corunna, Ont.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Poison. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Most types.	Label(s): White label - POISON; Class 6.1, Group I. Storage Temperature: Ambient. Hose Type: Most types. Containers and Materials: Drums, trailers, tank cars, tank trucks; steel.	Grades or Purity: Technical (98%). "Ethyl" TML motor mix; tetramethyl lead 90.3%, ethylene dibromide 17.9%, ethylene dichloride 18.8%, toluene 12.2%. "Ethyl" TEL motor mix; tetraethyl lead 61.5%, ethylene dibromide 17.9%, ethylene dichloride 18.8%, kerosene 1.2%. "Ethyl" TEL aviation mix; tetraethyl lead 61.4%, ethylene dibromide 35.7%, kerosene 2.5%.
Physical and Chemical Characteristics	*	
Physical State (20°C, 1 atm): Liquid. Solubility (Water): Insoluble. Molecular Weight: 323.4 Vapour Pressure: 0.15 mm Hg (20°C); 1.0 mm Hg (38.4°C) (technical); 7.5 mm Hg (20°C) TEL aviation mix; 35.9 mm Hg (20°C) TEL motor mix; 39.6 mm Hg (20°C) TML motor mix. Boiling Point: Technical decomposes 198 to 202°C; TEL aviation mix 138°C; TEL motor mix 105°C; TML motor mix 97°C.	Floatability (Water): Sinks.  Odour: Sweet, pleasant.  Flash Point: 85°C (o.c.); 93°C (c.c.) (technical); 32°C (o.c.) (TML); 118°C (o.c.) (TEL motor mix); 121°C (o.c.) (TEL aviation).  Vapour Density: 7 to 11.2 (technical); 4.9 (TML); 3.7 (TEL motor mix); 6.6 (TEL aviation).  Specific Gravity: Technical 1.7; TEL (motor), TML, 1.59; TEL aviation 1.74 (all at 20°C).	Colour: Red, blue or orange depending on dye added. Explosive Limits: No information. Melting Point: -137°C (technical).

	HAZARD DATA	
Human Health		- dumities incomin and nightmans. Contacts
skin - (absorbed) itching, inflammation, blisters	ss, headache, trembling hands, pale complexion, nausea eyes – irritation, watering inflammation. Extreme ex	posure could be fatal.
Toxicology: Highly toxic by all routes.  TLV ² (skin) 0.1 mg/m ³ tetraethyl (as Pb);  0.15 mg/m ³ (tetramethyl).  Short-term inhalation Limits - 0.3 mg/m ³ tetraethyl 15 min (skin)(as Pb); 0.5 mg/m ³	LC 50 - Inhalation: rat = 850 mg/m ³ (60 min) Delayed Toxicity - Lead poisoning. methyl (as Pb).	LD 50 - Via skim: rat = 0.015 g/kg LD _{LO} - Orah: rat = 0.017 g/kg
Pire		•
	chemical, foam or carbon dioxide to extinguish. Use w	vater to cool fire-exposed containers and disperse
Vapours.  Rehaviour in Fires Heating to decomposition or	oduces toxic lead vapours. Burning (TEL mixes) in air i	s only sustained in presence of burning combustibles.
TML will undergo self-sustained burning.		
ignition Temperature: Closed containers may e	xplode above 80°C.	Burning Rate: No information.
Reactivity		
With Water: No reaction. With Common Materials: Can react vigorously to Stability: Commercial grades are stable. Technology.	vith oxidizing materials. ical grade is only stable below 80 °C.	<u>-</u>

Water: Prevent entry into water intakes and waterways. Fish toxicity: 0.2 mg/L/96 h/bluegill/TLm/treshwater; Aquatic toxicity rating = <1 ppm/96 h/TLm/treshwater; 1.4 mg/L/48 h/bluegill/TLm/treshwater; 2 mg/L/24 h/bluegill/TI m/treshwater; POD: No. 10.000

Environment

## Special Hazards

POISON.

## Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Call Fire Department. Evacuate area and keep all persons upwind. Avoid contact and inhalation. Notify manufacturer and request assistance. Dike spill area to prevent liquid from entering water intakes, watercourses or sewers. Stop or reduce discharge, if this can be done without risk. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated protective clothing. Gloves - neoprene, rubber Boots - high, rubber foants worn outside boots).

## Fire and Explosion

Use water mist, dry chemical, foam or carbon dioxide to extinguish. Use water to cool fire-exposed containers and disperse vapours. Heating to decomposition produces toxic vapours. Burning (TEL mixes) in air is only sustained in presence of burning combustibles. TML will undergo self-sustained burning.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: while removing contamined clothing, immediately flush skin and eyes with plenty of warm water for at least 15 minutes. Speed is of utmost importance. Ingestion: in case of swallowing, vomiting should be induced for conscious victim only, give milk or water to drink. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### ENVIRONMENTAL PROTECTION MEASURES

#### Response

#### l'ate:

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- i. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Remove contaminated soil for disposal.
- 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 4/2

# TITANIUM DIOXIDE TIO2

## IDENTIFICATION

Сопинов Synonyms	Observable Characteristics	Manufacturers
TITANIC OXIDE TITANIC ANHYDRIDE TITANIUM WHITE	White powder. Odourless.	NL Chemical Canada, Varennes, Quebec. Tioxide of Canada, Tracy, Quebec.
Transportation and Storage Information		
Shipping State: Solid. Classification: Not regulated. Inert Atmosphere: No requirement. Venting: No requirement.	Labelis): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Technical and pure. Containers and Materials: Fibre drums, paper bags; bulk by truck or train.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): Insoluble. Molecular Weight: 79.9 Vapour Pressure: No information. Builing Point: 2 500 to 3 000°C.	Fioatability (Water): Sinks (finely divided powder may trap air and remain floating). Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 4.26 (20°C).	Colour: White Explosive Limits: Not flammable. Melting Point: I 830 to I 850°C.

Human Health				
Symptoms: Inhalation: coughing (nuisance partic Toxicology: Relatively nontoxic.	le) - eyes, redness. <u>Ingestion</u> : generally n	•		
TLV [®] (dust) 10 mg/m ³ .	LC5g - No information.	LD50 - No information.		
Short-term inhalation Limits - 20 mg/m ³ for 15 min (dust).	Delayed Toxicity - None.	LD _{LO} - intramuscular: rat = 0.3 g/kg		
Fire				
Fire Extinguishing Agents: Not combustible. Me Behaviour in Fire: Not combustible. Titanium di Ignition Temperature: Not combustible.  Reactivity	ost fire extinguishing agents may be used it oxide is very inert even in a fire. Burning Rate: Not combustible.	n fires involving titanium dioxide.		
кевситиу				
With Water: No reaction.  With Common Materials: Reacts violently with lithium.  Stability: Stable.				
Environment				
Water: Prevent entry into water intakes and wat Land-Air: No information. Food Chain Concentration Potential: No informa		m/96 h/TLm/freshwater.		

Special Hazards

Immediate Responses

Notify manufacturer. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - dust respirators should be worn if required; otherwise, goggles and protective clothing as required.

Fire and Explosion

Not combustible. Most fire extinguishing agents may be used on fires containing titanium dioxide.

First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: remove victim to fresh air, allow to rest. Contact: eyes or skin - rinse with plenty of water; remove contaminated clothing. Ingestion: give water to conscious victim to drink. If necessary, transport victim to doctor, clinic or hospital.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Remove material by manual or mechanical
- 4. Recover undamaged containers.
- 5. Notify environmental authorities to discuss

disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Can be placed in a landfill site (approval of environmental authorities required).

## TOLUENE C6H5CH3

IDENTIFICATION

UN No. 1294

Common Synonyms	Observable Characteristics	Manufacturers
TOLUOL METHYLBENZENE METHYLBENZOL PHENYLMETHANE	Clear, colourless liquid. Aromatic odour.	Finachem Canada, Montreal, Quebec. Sunchem, Sarnia, Ontario. Esso Chemical Canada, Sarnia, Ontario. Shell Canada, Corunna, Ontario. Dow Chemical Canada Inc., Sarnia, Ont.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable. Inert Atmosphere: No requirement. Venting: Open (flame arrester) or pressure-vacuum. Pump Type: Gear, centrifugal; steel, all - iron, stainless steel.	Label(s): Red label - FLAMMABLE LIQUID; Class 3.2, Group II.  Storage Temperature: Ambient. Hose Type: Polyethylene, Viton, butyl, flexible stainless steel.	Grades or Purity: Industrial, 94+%, nitration, 99.8+%.  Containers and Materials: Drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.047 g/100 mL (16°C); 0.087 g/100 mL (20°C). Molecular Weight: 92.2 Vapour Pressure: 10 mm Hg (6.4°C); 22 mm Hg (20°C); 36.7 mm Hg (30°C); 40 mm Hg (31.8°C). Boiling Point: 110.6°C.	Floatability (Water): Floats. Odour: Aromatic (0.17 to 40 ppm. odour threshold). Flash Point: 12.8°C (o.c.); 4°C (c.c.). Vapour Density: 3.1 Specific Gravity: 9.87 (20°C).	Colour: Colouriess.  Explosive Limits: 1.2 to 7.1%.  Melting Point: -95°C.

Human Health		
Symptoms: Inhalation: Headache, nausea, lossystem depression, depression of bone marrow Texicology: Moderately toxic by ingestion and TLV*- (skin) 100 ppm; 375 mg/m ³ . Short-term Inhalation Limits - (skin) 150 ppm; 560 mg/m ³ .	s of appetite, rapidly developing pulmonary edema. Inget, depressed respiration. Contact: eyes - irritation; skin linhalation. Moderately toxic by contact.  LC50 - Inhalation: mouse = 5 320 ppm/8 h LCL0 - Inhalation: rat = 4 000 ppm/4 h Delayed Toxicity - No information.	estion: vomiting, nausea, irritability, central nervous  1 - may be absorbed.  LD50 - Oral: rat = 5 g/kg  LD50 - skin: rabbit = 14 g/kg
Fire		
Fire Extinguishing Agents: Use carbon dioxide Behaviour in Fire: Flashback may occur along ignition Temperature: 480°C.  Reactivity	e, dry chemical or foam. Water may be ineffective but in vapour trail.  Burning Rate: 5.7 mm/min.	may be used to cool fire-exposed containers.
With Water No reaction	y with oxidizing materials. Reacts violently with (nitric	and sulfuric acids), nitrogen tetroxide and silver
Environment		
Water: Prevent entry into water intakes and sunfish/TLm/freshwater; 24 mg/L/24 h/blueg Land-Air: No information. Food Chain Concentration Potential: No information	waterways. Aquatic toxicity rating: 10 to 100 ppm/96 iii/TLm/freshwater; I 340 mg/L/24 h/mosquito fish/TLm mation.	h/TLm/freshwater; Fish toxicity: 1 180 mg/L/96 h/ n/freshwater; BOD: 86%, 5 days.

## Special Hazards

## FLAMMABLE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all ignition sources. Call manufacturer or supplier for advice. Avoid contact and inhalation. Contain spill by diking. Prevent entry into water intakes and sewers. Stop or reduce discharge if this can be done without risk. Notify environmental authorities.

Protective Clothing and Equipment

Respiratory protection - In fires or enclosed spaces, self-contained breathing apparatus; otherwise, chemical goggles - (mono), tight fitting. Face shield must not replace goggles. Gloves - rubber. Boots - rubber, high (pants worn outside boots). Outer protective clothing as required. Totally encapsulated protective clothing in fires or enclosed spaces.

## Fire and Explosion

Use dry chemical, carbon dioxide or alcohol foam. Water may be ineffective, but may be used to cool fire-exposed containers. Flash back may occur along vapour trail.

#### Pirst Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: skin and eyes - wash affected parts with soap and water. Flush eyes with copious quantities of water for at least 15 minutes. Ingestion: do not induce vomiting. Give water or milk to conscious victim. Inhalation: if breathing has stopped, give artificial respiration. If breathing is laboured, give oxygen. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Resconse

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or
- synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Can be taken to chemical incinerator (approval of environmental authorities required).

# TOLUENE DIISOCYANATE 1-CH3C6H3(NCO)2-2,4

# IDENTIFICATION

UN No. 2078

Common Synonyms	Observable Characteristics	Manufacturers	
2,4-TOLULENE DIISOCYANATE DI-ISO-CYANATOLUENE 2,4-TOLYLENE DIISOCYANATE META-TOLYLENE DIISOCYANATE TOLUENE 2,4-DIISOCYANATE ISOCYANIC ACID, METHYLPHENYLINE ESTER 2,4-TOI VORANATE T-80"	Water white to pale yellow liquid with sharp, pungent odour.	No Canadian Manufacturers. Canadian Suppliers: BASF Canada, Montreal, Que. Dow Chemical Canada Inc. Mobay Canada, Toronto Olin Chemical Canada	Originating from: BASF Wyandotte, USA Dow Chemical, USA, Freeport, TX. Mobay USA, Pittsburgh. Olin, USA, Lake Charles LA.
Transportation and Storage information			
Shipping State: Liquid. Classification: Poisonous liquid. Inert Atmosphere: Inerted. Venting: Pressure-vacuum. Pump Type: Centrifugal or positive displacement; steel or stainless steel.	Label(s): White label - POISONOUS LIQUID; Class 6.1; Groups I or II. Storage Temperature: Ambient. Hose Type: Seamless stainless steel, Teston, Viton.		2,6-TDI; 2,4-TDI 65% and
Physical and Chemical Characteristics			
Physical State Liquid. Solubility (Water): Reacts producing C02 and other products. Molecular Weight: 174.2 Vapour Pressure: 0.01 mm Hg (20°C); 1 mm Hg (80°C). Bolling Point: 251°C.	Floatability (Water): Sinks and reacts. Odour: Sharp pungent (0.4 to 2.4 ppm, odour threshold). Flash Point: 132°C (o.c). Vapour Density: 6.0 Specific Gravity: 1.22 at 25°C.	Colour: Water white to Explosive Limits: 0.9 t Melting Point: 20 to 2 80/20 14°C (freezing p	o 9.5%. 2°C.

HAZARD DATA
Human Health
Symptoms: Inhalation: Potent sensitizer and lung irritant if inhaled. May produce bronchospasm (asthma), pneumonitis, bronchitis and pulmonary edema.  Nocturnal cough and shortness of breath. Contact: skin - irritation; eyes - severe irritation. Ingestion: corrosive to digestive tract causing severe irritation, nausea, vomiting and diarrhea.  Toxicology: Highly toxic by inhalation, moderately toxic by ingestion.  TLV*- 0.005 ppm, 0.04 mg/m³  LC50 - Inhalation: mouse = 10 ppm/4 h  LD50 - Oral: rat = 5.8 g/kg  Short-term Inhalation Limits - 0.02 ppm,  0.15 mg/m³ for 15 min.
Pire-
Fire Extinguishing Agents: Use water spray, dry chemical or carbon dioxide. Water or foam may cause frothing. Cool exposed tanks with water.  Behaviour in Fire: When heated to decomposition, emits highly toxic fumes.  Ignition Temperature: No information.  Burning Rate: No information.
Reactivity
With Water: Reacts violently to form carbon dioxide gas and other products. With Common Materials: Reacts with acids, amines, bases and alcohols. May react with metals and compounds of copper, tin and mercury. Stability: Stable.
Environment
Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating = 1 to 10 ppm/96 h/TLm/freshwater.  Land-Air: No information.  Food Chain Concentration Potential: None.

## Special Hazards

POISON.

#### **Immediate Responses**

Keep non-involved people away from spill site. Issue warning: "POISON". Call Fire Department. Keep upwind. Avoid contact and inhalation. Evacuate from downwind. Dike to prevent runoff. Call manufacturer or supplier. Call environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated suit.

## Fire and Explosion

Use water spray, dry chemical or carbon dioxide to extinguish. Water or foam may cause frothing. When heated to decomposition, emits highly toxic fumes.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water for at least 15 minutes; eyes - irrigate with water; lagestion: give water to conscious victim to drink. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- Shorts and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers (caution: avoid contact and inhalation).
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# å

# TRIALLATE C10H16C13NOS

IDENTIFICATION

UN No. 2761

Danger Group According to Percentage of Active Substancer Group III, Liquid 30 to 100%

Common Synonyms	Observable Characteristics	Manufacturers
S-2,3,3-TRICHLOROALLYL DISOPROPYLTHIOCARBAMATE	Light brown liquid or grey-brown solid.	Monsanto Company, Winnipeg, Manitoba
Common Trade Names AVADEX	•	
(A herbicide used for pre- or post-emergent control	of wild oats).	
Transportation and Storage Information		
Shipping State: Solid or liquid (formulation). Classification: None. Inert Atmosphere: No requirement. Venting: Open. Pump Type: No information.	Label(s): Not regulated. Storage Temperature: Ambient. tiose Type: No information.	Grades or Purity: Various, as shown below. Containers and Materials: Glass bottles; cans, drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid (technical). Solubility (Water): 0.0004 g/100 mL (25°C); EC is dispersible in water. Molecular Weight: 304.7 Vapour Pressure: 0.00012 mm Hg (25°C) (technical). Bolling Point: 136°C (at 1 mm Hg) (technical) (decomposes >200°C).	Floatability (Water): Sinks; EC mixes. Odour: No information. Flash Point: 95°C (o.c.). Vapour Density: No information. Specific Gravity: 1.27 (25°C) technical.	Colour: Liquid-light brown; Solid-grey brown. Explosive Limits: Not flammable. Melting Point: 29 to 30°C (technical).

Human Health		
Symptoms: Ingestion: CNS depression, headache a Toxicology: Low toxicity by all routes. TLV 2 No information. Short-term inhalation Limits - No information.	nd rash. LC ₅₀ - No information. Delayed Toxicity - No information.	LD ₅₀ - Orai: rat = 1.47 g/kg LD ₅₀ - Skin: rabbit = 2.23 g/kg
Fire		,
Fire Extinguishing Agents: Foam, carbon dioxide	or dry chemical.	
Behaviour in Fire: Releases toxic fumes. Ignition Temperature: No information.	Burning Rate: No information.	
Reactivity		
With Water: No reaction. With Common Materials: No information. Stability: Stable.	•	
Environment		
Water: Prevent entry into water intakes and water Land-Air: LD50 - Oral: Duck = >5000 ppm. LC50 Food Chain Concentration Potential: No informat	- Oral: Quail =>5000 ppm.	out/TLm/freshwater; 4.9 ppm/96 h/bluegill/TLm/freshwater.

Special Hazards

POISON.

#### komediate Responses

Keep non-involved people away from spill site. Stop-or reduce discharge if safe to do so. Notify manufacturer or supplier. Dike to contain material or water runoff. Notify environmental authorities.

## Projective Clothing and Equipment

in fires or confined spaces - Respiratory protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, approved pesticide respirator and impervious outer clothing.

## Fire and Explosion

Use carbon dioxide, foam or dry chemical to extinguish. Releases toxic fumes in fires.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contacts skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim to drink and induce vomiting; in the case of petroleum distillates, do not induce vomiting for fear of aspiration and chemical pneumonia. If medical assistance is not immediately available, transport victim to hospital, doctor, or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

1. Stop or reduce discharge if safe to do so.

2. Contact manufacturer or supplier for advice.

3. If possible contain discharge by booming.

4. If floating, skim and remove.

Sinks or mixes

3. If possible contain discharge by damming or water diversions.

4. Dredge or vacuum pump to remove contaminants, liquids and con-

taminated bottom sediments.

5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### i.and-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. If liquid, remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. If solid, remove material by manual or mechanical means.
- 6. Recover undamaged containers.
- 7. Absorb residual liquid on natural or synthetic sorbents.
- 8. Remove contaminated soil for disposal.
- 9. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

## Disposal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## **Available Formulations**

Technical Grade: Purity: 90+%.

Properties: Combustible, insoluble in water.

Formulations

GR - granular

Type: EC - emulsifiable concentrate

Parity:

- typically 35%

- typically 10% in inerts (clay)

Properties:

- dispersible in water

## TRICHLORFON (CH3O)2P(O)CHIOH)CCI3

**IDENTIFICATION** 

UN No. 2783

Danger Group According to Percentage of Active

Substance: Group III, Solid 2 to 20% Liquid 0.5 to 20%

Common Synonyms

Observable Characteristics

Manufacturers

2.2.2-TRICHLORO-I-HYDROXYETHYL-

White or brownish powder or liquid.

interprovincial Co-ops Ltd...

PHOSPHONATE TRICHLORPHON

Saskatoon, Sask.

Common Trade Names DYLOX, DUTOX

Chemagro Ltd., Mississauga, Ontario. Chipman Inc., Stoney Creek, Ontario.

(An insecticide used for the control of a variety of pests.)

Les produits Marquette, Longueuil, Québec.

Transportation and Storage information

Shipping State: Solid or liquid (formulation). Classification: None.

Label(s): Not regulated. Storage Temperature: Ambient. Grades or Purity: Various, as described

Inert Atmosphere: No requirement.

Hose Type: No information.

below. Containers and Materials: Glass bottles:

Venting: Open.

cans, drums; steel.

Pump Type: No information.

Physical and Chemical Characteristics

Physical State (20°C, I atm): Solid (technical). Solubility (Water): 15.4 g/100 mL (25°C) technical; EC is dispersible in water. SN and SP are soluble.

Molecular Weight: 257.4 Vapour Pressure: 0.0000078 mm Hg (20°C)

technical.

Boiling Point: 100°C (0.1 mm Hg) technical.

Floatability (Water): Sink and mix.

Odeur: No information. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 1.73 (20°C) technical. Colour: White or brownish. Explosive Limits: Not flammable. Melting Point: \$3 to \$4°C (technical).

#### HAZARD DATA

#### Human Health

Symptoms: inhalation, Skin Absorption or Ingestion: nausea, salivation, tearing, abdominal cramps, vomiting, sweating, muscular tremors, cyanosis-

Texicology: Highly toxic by ingestion. Moderately toxic by inhalation and skin contact.

TLY No information. Short-term Inhalation Limits - No information. LC50 - inhalation: rat = 1.3 mg/m³ Delayed Toxicity - Suspected mutagen.  $LD_{50}$  - Oral: rat = 0.45 g/kg LD50 - Stutancoes rat = 0.40 g/kg

Fire Extinguishing Agents: Use foam, carbon dioxide or dry chemical.

Behaviour in Pire: Releases toxic fumes.

lenition Temperature: No information.

Burning Rate: No information.

## Reactivity

With Water: May hydrolyse to dichloroves in alkaline conditions.

With Common Materials: Decomposes to dichlorvos in alkaline solutions.

Stability: Stable.

#### Environment

Water: Prevent entry into water intakes and waterways. Fish toxicities: 0.0081 mg/L/48 h/Daphnia magna/LC50/freshwater; 1.2 to 2.4 mg/L/96 h/rainbow trout/LCsq/freshwater; 6.7 to 9.2 mg/L/96 h/fathead minnow/LCsp/freshwater; 2.6 to 3.7 mg/L/96 h/bluegill/LCsq/freshwater.

Land-Air: LDso - Oral: Chicken = 0.125 g/kg; Oral: Wild bird = 0.04 g/kg

Food Chain Concentration Potential: Suspected to convert to more toxic dichloryos.

## Special Hazards

POISON.

### immediate Responses

Keep non-involved people away from spill site. Stop or reduce discharge if safe to do so. Notify manufacturer or supplier. Dike to contain material or water runoff. Notify environmental authorities.

## Protective Clothing and Equipment

In fires or confined spaces - Respiratory Protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, approved pesticide respirator and impervious outer clothing.

## Fire and Explosion

Use carbon dioxide, foam or dry chemical to extinguish. Releases toxic fumes in fires.

#### Pleas Ald

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim and induce vomiting; in the case of petroleum distillates, do not induce vemiting for fear of aspiration and chemical pneumonia. if medical assistance is not immediately available, transport victim to hospital, doctor, or clinic.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so. 2. Contact manufacturer or supplier for advice.

## 3. If possible contain

discharge by booming.

4. If floating, skim and remove.

Sinks or mixes

3. If possible contain discharge by damming or water diversions. 4. Dredge or vacuum pump to remove

contaminants, liquids and contaminated bottom sediments.

5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

## Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. If liquid, remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. If solid, remove material by manual or mechanical means.
- 6. Recover undamaged containers.
- 7. Absorb residual liquid on natural or synthetic sorbents.
- 8. Remove contaminated soil for disposal.
- 9. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

#### Disposal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## Available Formulations

Technical Grade: Purity: 90+%.

Properties: moderately soluble in water, combustible.

#### Permulations:

GR - granular

Properties: Purity: Type: - dispersible in water - typically 80% EC - emulsifiable concentrate - miscible in water - typically 85% SN - solution - miscible in water - typically 80% SP - soluble powder - typically 5%, remainder is inert

Other Possible Ingredients Found in Formulations: dichlorvos, oxydemeton methyl.

# 1,1,1-TRICHLOROETHANE CH3CCI3

IDEN	IF IC	ATE	ON

UN No. 2831

Common Synonyms	Observable Characteristics	Manufacturers
TRICHLORETHANE METHYLCHLOROFORM CHLOROTHENE AEROTHENE	Colourless, liquid with a sweetish odour.	Dow Chemical Canada, Inc., Sarnia, Ontario.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Poisonous liquid. inert Atmosphere: No requirement. Venting: Pressure-vacuum Pump Type: Steel, stainless steel (not aluminum); centrifugal or positive displacement.	Label(s): White label - POISONOUS LIQUID; Class 6.1, Group III. Storage Temperature: Ambient. Hose Type: Seamless stainless steel, Tellon (not natural rubber).	Grades or Purity: Commercial and technical.  Containers and Materials: Drums and tanks cars;  steel (not aluminum).
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): 0.44 g/100 mL (20°C). Molecular Weight: 133.4 Vapour Pressure: 100 mm Hg (20°C); 127 mm Hg (25°C); 155 mm Hg (30°C). Bolling Point: 74.1°C.	Floatability (Water): Sinks. Odour: Sweetish; 100 to 700 ppm, odour threshold. Flash Point: Not flammable under ordinary temperatures and pressures. Vapour Density: 4.6 Specific Gravity: 1.34 at 20°C.	Colour: Colourless. Explosive Limits 8 to 10.5% (at elevated temperatures). Melting Point: -30 to -33°C.

RAZAKU	mu.

Human Health		
Symptoms: Contact: skin - irritation and derma respiratory depression. Toxicology: Moderately loxic by ingestion.	utitis; eyes - pain and irritation. <u>Inhalation or ingestion</u>	dizziness, nausea, fainting, unconsciousness and
TLV ⁶ - 350 ppm; 1 900 mg/m ³ .	LC50 - No information.	LD50 - Oral: rat = 10.3 g/kg
Short-term inhalation Limits - 450 ppm; 2 450 mg/m ³ (15 min).	LC _{Lo} - Inhalation: man = 4 900 ppm/10 min. Delayed Toxicity - No information.	LD30 - Oral: mouse = 11.2 g/kg
Fire		
Behaviour in Fire: Toxic and irritating gases ar ignition Temperature: 537°C.  Reactivity	oam or carbon dioxide. Water may be used to cool fire- e generated in fires (Cl ₂ , HCl, phosgene, etc.). Burning Rate: 2.9 mm/min.	
With Water: No reaction. With Common Materials: Reacts violently with Stability: Stable.	acetone, oxygen, sodium metal and sodium hydroxide.	
Environment		
Water: Prevent entry into water intakes and w toxicity rating = 10 to 100 ppm/96 h/TLm/frest Land-Air: No information. Food Chain Concentration Potential: None.	aterways. Hazardous to aquatic life. Fish toxicity: 75 hwater.	to 150 ppm/tns/pinfish/TLm/saltwater; Aquatic

Special Hazards

POISON.

#### immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Call Fire Department. Avoid contact and inhalation. Dike to prevent runoff of material. Contact manufacturer. Contact environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus and totally encapsulated protective clothing.

## Fire and Emplosion

Use dry chemical, foam or carbon dioxide to extinguish fires. Use water to keep fire-exposed containers cool. Toxic and irritating gases (Ci 2, HCI, phosgene) are generated in fires.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: skin - remove contaminated clothing and wash affected area with soap and water; eyes - irrigate with water. Inhalation: give artificial respiration if necessary. Ingestion: give water to drink; do not induce vomiting. Transport to hospital; advise doctor not to give adrenalin or other stimulants.

## **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### **Tate**

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum equipment
- and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Adsorb residual liquid on natural or

synthetic sorbents.

- 7. Remove contaminated soil for disposal.
- 8. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# ŧ

# TRIFLURALIN F3C(NO2)2C6H2N(C3H7)2

## IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
a,a,a-TRIFLUORO-2,6-DINITRO-N, N-DIPROPYL-p-TOLUIDINE Common Trade Names TREFLAN  (A herbicide used for the pre-emergent control of	Orange or brown solid, brown liquid. Slight odour. grasses and broadleaf plants.)	Interprovincial Co-ops Ltd., Saskatoon, Sask. Eli Lilly and Co. (Canada Ltd.), Toronto, Ontario. Chevron Chemical (Canada Ltd.), Burlington, Ontario. Saskatchewan Wheat Pool, Regina, Saskatchewan.
Transportation and Storage Information		
Shipping State: Solid or liquid (formulation). Classification: None. Inert Atmosphere: No requirement. Venting: Open. Pump Type: No information.	Labef(s): Not regulated. Storage Temperature: Ambient. Hose Type: No information.	Grades or Purity: Various as described below.  Containers and Materials: Glass bottles; cans, drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid (technical). Solubility (Water): 0.0001 g/100 mL (technical); EC is dispersible.  Molecular Weight: 335 Vapour Pressure: 0.0002 (30°C) technical.  Boiling Point: 139-149°C (at mm Hg) technical.	Floatability (Water): Sinks, EC is dispersible. Odour: Slight. Flash Point: EC may be flammable. Vapour Density: No information. Specific Gravity: 1.29 (25°C) technical.	Colour: Orange to brown. Explosive Limits: EC may be flammable. Melting Point: 49°C (technical).

Human Health		
Symptoms: <u>Inhalation:</u> 'respiratory tract irritation, Toxicology: Moderate toxicity by all routes. TLV [®] No information. Short-term Inhalation Limits - No information.	convulsions, coma; <u>Contact</u> ; skin - irritation.  LC 50 - No information.  Delayed Toxicity - Suspected carcinogen.	Ingestion: irritation of gastrointestinal tract.  LD50 - Orai: mouse = 5.0 g/kg
Fire	`	
Fire Extinguishing Agents: Use foam, carbon dioxid	e or dry chemical.	
Behaviour in Fire: Releases toxic lumes. Ignition Temperature: No information.	Burning Rate: No information.	
Reactivity		
With Water: No reaction. With Common Materials: No information. Stability: Stable.		
Environment		
Water: Prevent entry into water intakes and water fathead minnow/LC50/freshwater; 0.047 to 0.070 n Land-Air: LD50 - Oral: Chicken = 72.0 g/kg; (no el Food Chain Concentration Potential: None.	ng/L/96 h/bluegill/LC50/freshwater; 0.320 to 1	rainbow trout/LC50/freshwater; 0.083 to 0.134mg/L/96 h/ .0 mg/L/96 h/ <u>Daphnia</u> magna/LC50/freshwater.

# 491

#### **EMERCENCY MEASURES**

## Special Hazards

POISON.

#### Immediate Responses

Keep non-involved people away from spill site. Stop or reduce discharge if safe to do so. Notify manufacturer or supplier. Dike to contain material or water runoff. Notify environmental authorities.

## Protective Clothing and Equipment

In fires or confined spaces - Respiratory Protection - self-contained breathing apparatus and totally encapsulated suit. Otherwise, approved pesticide respirator and impervious outer clothing.

## Fire and Explosion

Use carbon dioxide, foam or dry chemical to extinguish. Releases toxic lumes in fires.

#### First Add

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. Inhalations if breathing has stopped, give artificial respiration (not mouth-to-mouth method); if laboured, give oxygen. Contacts skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give water to conscious victim to drink and induce vomiting; in the case of petroleum distillates, do not induce vomiting for fear of aspiration and chemical pneumonia. If medical assistance is not immediately available, transport victim to hospital, doctor, or clinic.

### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.

#### ###---

- 3. If possible contain discharge by booming.
- 4. If floating, skim and remove.
- remove.

## Sinks or mixes

- 3. If possible contain discharge by damming or water diversions.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleams of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. If liquid, remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. If solid, remove material by manual or mechanical means.
- 6. Recover undamaged containers.
- 7. Absorb residual liquid on natural or synthetic sorbents.
- 8. Remove contaminated soil for disposal.
- 9. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

#### Disposal

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

## Available Formulations

Technical Grade: Purity: 95+%.

Properties: combustible, insoluble in water.

Formulations:

Type:

Purity

Properties:

- EC emuisifiable concentrate
- typically 35% remainder aromatic
- petroleum distillates
- dispersible in water,
- combustible and possible flammable.

GR - granular

- typically 5%

# TRINITROTOLUENE C6H2(CH3)(NO2)3

IDENTIFICATION

UN No. 0209 dry 1356 wetted (containing 10% water)

Common Synonyms	Observable Characteristics	Manufacturers
METHYLTRINITROBENZENE TNT	Pale yellow crystals or flakes. Wet is a yellow slurry or sludge. Odourless.	Canadian Industries Ltd., Montreal, Quebec.
Transportation and Storage Information		
Shipping State: Solid. Classification: Explosive. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Orange and black label - EXPLOSIVE; Class 1.1.D. Storage Temperature: Ambient.	Grades or Purity: Technical.  Containers and Materials: Wooden cases or kegs.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.02 g/100 ml. (15°C); 0.07 g/100 ml. (100°C). Molecular Weight: 227.1 Vapour Pressure: 0.043 mm Hg (81°C). Bolling Point: Explodes at 240°C.	Ficatability (Water): Sinks. Odour: Odourless. Flash Point: Explodes. Vapour Density: No information. Specific Gravity: 1.65 (20°C).	Colour: Pale yellow.  Explosive Limits: Explodes if vigorously shocked or heated to 240°C.  Melting Point: \$1°C.

Symptoms: Inhalation, Skin absorption, or ing Texticology: Moderately toxic by ingestion and TLY® (skin) 0.3 mg/m³.  Short-term inhalation Limits – (skin) 3 mg/m³ (15 min).	zestion: jaundice, dermatis, cyanosis pallour, loss of appeti id contact.  LC50 - No information.  Delayed Toxicity - Damage to liver, bones and kidneys.	le, nausea. LD50 - No information. LDL0 - Oral: rat = 0.7 g/kg
Fire		
Behaviour in Fire: Will explode. In explosion Ignition Temperature: Explodes 240°C.	releases toxic NO _X Iumes.  Detonation velocity: (6 900 m/s).	
<del></del>		<u> </u>
Reactivity With Water: No reaction. With Common Materials: Can react vigorous	ly with reducing materials. ure (240°C) will detonate TNT.	
Reactivity	ly with reducing materials. ure (240°C) will detonate TNT.	

# Special Hazards

EXPLOSIVE. Can detonate at 240°C or with shock.

# immediate Responses

Keep non-involved people away from spill site and evacuate people to a safe distance as soon as possible. Issue warning "EXPLOSIVE", CALL FIRE DEPARTMENT. Enforce strict evacuation of area. Contact manufacturer. Contact environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - after explosion has occured, self-contained breathing apparatus; otherwise, face shield rubber gloves, boots and protective clothing as required.

# Fire and Explosion

Explosive, light fires from explosive-resistant location. Use water spray to cool fire-exposed containers.

#### First AL

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - irrigate immediately; skin - wash with soapy water immediately. Ingestion: give water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to doctor, hospital or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

### Response

#### Water

- 1. Evacuate area and control spill from an an explosion resistant location.
- 2. Stop or reduce discharge if safe to do so.
- 3. Contact manufacturer or supplier for advice.
- Notify environmental authorities to discuss disposal and cleamin of contaminated materials.

#### Land-Air

- i. Evacuate area and control spiil from an explosion resistant location.
- 2. Stop or reduce discharge if safe to do so.
- 3. Contact manufacturer or supplier for advice.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### (Name of the

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# 434

# TURPENTINE C₁₀H₁₆ (primarily)

IDENTIFICATION

UN No. 1299

Record Chemical.

Napier ville, Ouebec.

Common Synonyms

SPIRITS OF TURPENTINE

TURPS GUM TURPENTINE

WOOD TURPENTINE
GUMTHUS

Mixture of several compounds, principally pinene, diterpene.

Transportation and Storage Information

Shipping State: Liquid.

Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Open (flame arrester).

Pump Type: Steel, stainless steel.

Label(s): Red label - FLAMMABLE LIQUID;

Class 3.2, Group III.

unpleasant odour.

Observable Characteristics

Colourless liquid, with penetrating,

Storage Temperature: Ambient. Hose Type: Most types (except rubber). Grades or Purity: Variety.

Containers and Materials: Bottles, cans, drums,

tank cars and trucks; steel.

Physical and Chemical Characteristics

Physical State (20°C, 1 atm): Liquid.

Solubility (Water): insoluble.
Molecular Weight: Variable, 136 (average).

Vapour Pressure: 5 mm Hg (25°C).
Boiling Point: 154 to 170°C.

Floatability (Water): Floats.

Odour: Aromatic, penetrating, unpleasant.

Flash Point: 35°C (c.c.). Vapour Density: 4.8

Specific Gravity: 0.85 to 0.88 (15°C).

Colour: Colourless.

Explosive Limits: 0.8% (LEL).

Melting Point: -55°C.

## HAZARD DATA

Human Health

Symptoms: Inhalation: vapour causes headache, confusion, respiratory distress. If liquid is taken into lungs, causes severe pneumonitis. Contact: skin-irritation. Ingestion: irritates entire digestive system and may injure kidneys.

Toxicology: Highly toxic by Inhalation.

TLV² (skin) 100 ppm; 560 mg/m³. Short-term inhalation Limits - 150 ppm; LC50 - No information.

TCLo - Inhalation: human = 175 ppm

LD 20 - Oral: rat = 5.8 g/kg

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LD20 - Inhalation: mouse = 0.06 g/kg

Fire

Fire Extinguishing Agents: Use foam, dry chemical or carbon dioxide. Water may be ineffective, but should be used to keep fire-exposed containers cool. Behaviour in Fire: Releases acrid fumes in fire.

ignition Temperature: 253°C.

840 mg/m³ (15 min).

Burning Rate: 2.4 mm/min.

Reactivity

With Water: No reaction.

With Common Materials: Can react with oxidizers. Reacts violently with calcium hypochlorite, chlorine, chromic anhydride, hexachioromelamine, stannic chloride and trichloromelamine.

Stability: Stable.

Environment

Water: Prevent entry into water intakes and waterways. Harmful to aquatic life. Fish toxicity: 100 ppm/tns/fish/toxic/freshwater; Aquatic toxicity rating = 10 to 100 ppm/96 h/TLm/freshwater.

Land-Air: No information.

Food Chain Concentration Potential: None.

### Special Hazards

#### FLAMMABLE.

### immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Avoid contact and inhalation. Dike to prevent runoff. Notify manufacturer. Notify environmental authorities.

# Protective Clothing and Equipment

In fires and confined spaces - Respiratory protection - self-contained breathing apparatus; otherwise, face shield, boots and protective clothing as required.

# Fire and Explosion

To extinguish use foam, dry chemical or carbon dioxide. Water may be ineffective, but may be used to keep fire-exposed containers cool.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: eyes - irrigate; skin - wash with soapy water and remove contaminated clothing. Ingestion: do not induce vomiting. Inhalation: If breathing has stopped, give artificial respiration; if laboured, give oxygen. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4: If Hoating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material with pumps or vacuum
- equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss cleanup and disposal of contaminated materials.

- I. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Can be burnt in an incinerator (approval of environmental authorities required).

# UREA NH2CONH2

# IDENTIFICATION

Common Synonyme	Observable Characteristics	Manufacturers
CARBAMIDE CARBONYLDIAMIDE	White crystals or powder. Odourless, or slight ammonia odour.	Canadian Fertilizers, Medicine Hat, Alta. Cominco, Carseland, Alta. Canadian Industries Ltd., Courtright, Ontario.
Transportation and Storage Information	<u> </u>	
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Labei(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Technical, CP, fertilizer (45 to 46% N), feed grade (72% N).  Containers and Materials: Bags, drums, bulk, trucks, train. Also shipped as solution in tank cars or tank trucks.
Physical and Chemical Characteristics		<u> </u>
Physical State (20°C, 1 atm): Solid. Solubility (Water): 78 g/100 mL (5°C); 119.3 g/100 mL (25°C). Molecular Weight: 60.1 Vapour Pressure: No information. Boiling Point: Decomposes above 133°C.	Floatability (Water): Sinks and mixes. Odour: Odourless, or slight ammonia odour. Flash Point: Not combustible or flammable. Vapour Density: No information. Specific Gravity: 1.34 at 20°C (solid).	Colour: White. Explosive Limits: Not flammable or combustible. Melting Point: 133 to 135°C.

HAZARD DATA			
Numan Health			
Symptoms: Inhalation: (dust) sore throat, coughing abdominat pain.	g, shortness of breath. Contact: skin or ey	es - Causes redness and irritation. <u>Ingestion</u> : sore throat,	
Toxicology: Moderately toxic by inhalation and co	ntact.		
TLV No information.	LC50 - No information.	LD50 - No information.	
Short-term Inhalation Limits - No Information.	Delayed Toxicity - None.	LDLo - Cutaneous: rabbit = 3.0 g/kg	
Fire		<u></u>	
ignition Temperature: Not combustible. Reactivity	Burning Rate: Not combustible.		
With Water: No reaction; soluble. With Common Materials: Reacts violently with hy Stability: Stable.	pochlorites.	•	
Environment			
Water: Prevent entry into water intakes and waterways. Fish toxicity: 16 000 to 30 000 mg/L/24 h/creek chub/critical range/freshwater; Aquatic toxicity rating >1 000 ppm/96 h/TLm/freshwater; BOD: 9 to 100%, 5 days (temperature dependent). Land-Air: No information. Food Chain Concentration Potential: None.			

### Special Hazards

# immediate Responses

Keep non-involved people away from spill site. Dike to prevent runoff from rainwater or water application. Contact manufacturer. Contact environmental authorities.

# Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; otherwise, face shield and protective clothing as required.

# Fire and Explosion

Not combustible; most fire extinguishing agents may be used in fires involving urea. Urea decomposes above 133°C producing toxic NO_x fumes and ammonia gas.

### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: skin - remove contaminated clothing and wash affected area with soap and water; eyes - irrigate with water. <a href="https://medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medical.google.com/medic

# ENVIRONMENTAL PROTECTION MEASURES

## Response

#### Fater

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. May be buried in landfill site (environmental authorities approval required).

# 4

# YANADIUM PENTOXIDE V205

	FIC	

UN No. 2862

Common Synonyms	Observable Characteristics	Manufacturers
VANADIC ACID ANHYDRIDE VANADIUM OXIDE	Yellow to red crystalline powder. Odourless.	No Canadian manulacturer. U.S. supplier Monsanto Co., St-Louis, MO, USA
Transportation and Storage Information		
Shipping State: Solid. Classification: Poison. Inert Atmosphere: No requirement. Venting: No requirement.	Label(s): White label - POISON; Class 6.1, Group II. Storage Temperature: Ambient.	Grades or Purity: Commercial, technical, CP. Containers and Materials: Drums, multiwall paper bags.
Physical and Chemical Characteristics		
Physical State (20°C; 1 atm); Solid. Solubility (Water): 0.8 g/100 mi. (20°C). Molecular Weight: 121.9 Vapour Pressure: No information. Boiling Point: Decomposes I 750°C.	Ploatability (Water): Sinks. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: 3.36 (18°C).	Colour: Yellow to red. Explosive Limits: Not flammable. Melting Point: 690°C.

Symptoms: Contact: skin - redness and itch; eyes - redness, pain, watering. <u>Ingestion</u> : metal taste, vomiting, pallour, tremors, gastrointestinal disturbances. <u>Inhalation</u> : coughing, chest pain, respiratory system irritation and greenish-black discolouration of tongue.				
Texicology: Highly toxic by all routes.  TLV = (dust or fume) 0.05 mg/m ³ .  Short-term inhalation Limits - No information.	LC50 - No information. LC _{LO} - inhalation: rat = 70 mg/m ³ (2 h) Delayed Toxocity - No information.	LD50 - Oral: rat = 0.01 g/kg		
Fixe \				
Fire Extinguishing Agents: Not combustible. Most fire extinguishing agents can be used in fires involving vanadium pentoxide.  Behaviour in Fire: Not combustible.  Ignition Temperature: Not combustible.  Burning Rate: Not combustible.				
Reactivity				
With Water: No reaction. With Common Materials: Can react violently with (Ca + S + H ₂ O), chlorine trifluoride and lithium. Stability: Stable.				
Environment				
Water: Prevent entry into water intakes and water Land-Air: No information.	ways. Fish toxicity: 13 ppm/96 h/fathead minno	w/TLm/softwater; BOD: None.		

### Special Hazards

POISON.

### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "POISON". Avoid contact or inhalation of fumes or dust. Stop or reduce discharge, if possible. Notify supplier. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus. Gloves - rubber or PVC. Outer protective clothing as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving vanadium pentoxide.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: wash eyes and affected skin thoroughly with plenty of water. Ingestion: give conscious victim plenty of water do not induce vomiting. Inhalation: give artificial respiration if required (not mouth-to-mouth method). If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, tiquids and contaminated bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# VINYL ACETATE CH3COOCH;CH2

IDENTIFICATION

UN No. 1301

Common Synonyms	Observable Characteristics	Manufacturers
ACETIC ACID VINYL ESTER ACETIC ACID ETHENYL YAC YAM (Vinyl Acetate Monomer)	Colourless liquid. Unpleasant in high concentrations; sweet pleasant in small quantities.	Celanese Canada Ltd., Edmonton, Alta.
Transportation and Storage Information		
Shipping State: Liquid. Classification: Flammable liquid. Inert Atmosphere: No requirement. Venting: Pressure-vacuum. Pump Type: Carbon or stainless steel (explosion-proof).	Label(s): Red label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Polyethylene, butyl, neoprene, flexible steel or stainless steel.	Grades or Purity: Technical Grade A, 99.8% (diphenylamine inhibited); grade H, 99.8% (hydroquinone inhibited).  Containers and Materials: Cans, drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, I atm): Liquid. Solubility (Water): 2.5 g/100 mL (20°C). Molecular Weight: 86.1 Vapour Pressure: 83 mm Hg (26°C); 100 mm Hg (23.3°C); 115 mm Hg (25°C). Boiling Point: 72-73°C.	Floatability (Water): Floats.  Odour: Sweet, unpleasant in high concentration (0.12 to 0.55 ppm, odour threshold).  Flash Point: -5°C (o.c.), -8°C (c.c.).  Vapour Density: 3.0  Specific Gravity: 0.93 (20°C).	Colour: Colouriess.  Explosive Limits: 2.6 to 13.4%.  Melting Point: -93 to -100°C.

# **HAZARD DATA**

Human Health				
Symptoms: Contact: skin and eyes - liquid and Toxicology: Moderately toxic by all routes.	vapours irritating. Inhalation: high concentrations pro-	oduce anaesthetic effect. <u>Ingestion</u> : nausea and vomiting.		
TLY*- (inhalation) 10 ppm; 30 mg/m ³ .				
Short-term Inhaiation Limits - 20 ppm;	LCLo - Inhalation: rat = 4 000 ppm (4 h)	• •		
60 mg/m ³ (15 min).	Delayed Toxicity - No information.			
Fire				
		ective, but may be used to cool fire-exposed containers.  ent rupture of tank. Flashback may occur along vapour		
Reactivity		<u> </u>		
With Water: No reaction; slightly soluble. With Common Materials: Reacts violently with hydrofluoric acid. Stability: Stable if at room temperature and in		ethylene dlamine, hydrochloric acid and peroxides, and		
Environment	,			
		life in very low concentrations. Fish toxicity: 18 ppm/		

96 h/bluegili/TLm/freshwater; Aquatic toxicity rating = 10 to 100 ppm/96 h/TLm/freshwater; 19 to 39 mg/L/96 h/TLm/fathead minnow/freshwater; BOD: 62%, 5 days (theoretical).

Land-Air: No information.

Food Chain Concentration Potential: None.

# Special Hazards

FLAMMABLE. Heat initiates polymerization which could proceed violently.

### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Avoid contact and inhalation. Stay upwind and use water spray to control vapour. Stop or reduce discharge, if this can be done without risk. Contact supplier for guidance. Dike runoff from rainwater or water application. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus. <u>Coggles</u> - (mono), tight fitting. If face shield is used, it should not replace goggles. <u>Suit</u> - (jacket and pants) or coveralls, rubber or plastic. <u>Boots</u> - high, rubber (pants worn outside boots). <u>Gloves</u> - rubber or plastic.

#### Fire and Explosion

Use dry chemical, carbon dioxide or alcohol foam. Use water spray to knock down vapours and cool fire-exposed containers. Heat initiates polymerization which could proceed violently. Fire may cause violent rupture of tank. Flash back may occur along vapour trail.

#### First Aid

Move victim out of spill area to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalation</u>: give artificial respiration if breathing has stopped. Give oxygen if breathing is laboured. <u>Contact</u>: remove contaminated clothing and wash eyes and skin with plenty of warm water for at least 15 minutes. <u>Ingestion</u>: give conscious victim water to drink. If medical assistance is not immediately available, transport victim to hospital, clinic or doctor.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other material.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss discosal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. Spray into a chemical waste incinerator (approval of environmental authorities required).

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# VINYL CHLORIDE H2C:CHCI

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UN No. 1086

		014 1401 1000
Common Synonyms	Observable Characteristics	Manufacturers
/INYL CHLORIDE MONOMER /CM CHLOROETHYLENE CHLOROETHENE	Colourless, gas or liquid. Sweet odour.	Dow Chemical Canada Inc., Sarnia, Ont., Fort Saskatchewan, Alta.
ransportation and Storage Information		
hipping State: Liquid (compressed gas). Classification: Flammable. nert Atmosphere: No requirement. /enting: Safety-relief (under pressure); wessure-vacuum (atmospheric pressure). Pump Type: Steel or stainless steel; positive lisplacement or gas, explosion-proof.	Label(s): Red label - FLAMMABLE GAS; Class 2.1. Storage Temperature: Ambient. Hose Type: Tellon, Viton A; flexible stainless steel.	Grades or Purity: Commercial or technical, 99+% (inhibiter, phenol 40 to 100 ppm may be added).  Containers and Materials: Cylinders, tank cars (steel or stainless steel).
hysical and Chemical Characteristics		
hysical State (20°C, 1 atm): Gas. iolubility (Water): 0.006 g/100 mL (10°C); 1.003 g/100 mL (25°C). iolecular Weight: 62.5 /apour Pressure: 240 mm Hg (-40°C); 580 mm -lig (-20°C); 2 660 mm Hg (25°C). 30lling Point: -13.4 to -13.9°C.	Floatability (Water): Liquid, floats. Odour: Sweet odour (260 to 4 000 ppm, odour threshold). Flash Point: -78°C (o.c.). Vapour Density: 2.2 Specific Gravity: 9.97 (-20°C); 0.91 (15°C).	Colour: Colourless. Explosive Limits: 3.6 to 33%. Melting Point: -153 to -160°C.

### HAZARD DATA

	inflammation. Ingestion: nausea, vomiting, drowsing	irritation, headache, paralysis. Contact: skin - drying, ness, loss of consciousness, narcosis, shock.
TLV*- (inhalation) 5 ppm; 10 mg/m ³ .	LC50 - No information.	LDso - Oral: rat = 0.5 g/kg
Short-term inhalation Limits - No information.	LC _{Lo} - Inhalation: rat = 6 000 ppm/4 h TD _{Lo} - Inhalation: rat = 250 ppm/39 weeks Delayed Toxicity - Recognized carcinogen.	TDLo - Oral: rat = 34 g/kg
Fire		
	il leak has been shutoff. Water may be used to cool of tank. Burning releases hydrogen chlòride gas. Flo catalytic impurities, may violently polymerize.  Burning Rate: 4.3 mm/min.	

# Reactivity

Human Health

With Water: No reaction.

With Common Materials: Can react vigorously with oxidizing materials.

Stability: Stable.

# Environment

Water: Prevent entry into water intakes and waterways. Aquatic toxicity rating: >1 000 ppm/96 h/TLm/freshwater; BOD: No information. Land-Air: No information.

Food Chain Concentration Potential: None.

# Special Hazards

#### FLAMMABLE.

## Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all ignition sources. Contact manufacturer for guidance. Stay upwind and use water spray to control vapour. Dike runoff. Stop or reduce discharge if this can be done without risk. Notify environmental authorities.

#### Protective Clothing and Equipment

Respiratory protection - self-contained breathing apparatus. Boots - high, rubber (pants worn outside boots). Acid suit - (jacket and pants), "slicker suit" - neoprene, or coveralls. Gloves - rubber.

#### Fire and Explosion

Do not put out fire until leak has been shut off. Use water to cool fire-exposed containers. Fire may cause violent rupture of tank. Flash back may occur along vapour trail. Under high temperatures or in contact with certain catalytic impurities, may violently polymerize.

#### Piret Ale

Move victim from spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if breathing has stopped. Give oxygen if breathing is laboured. Contact: remove contaminated clothing and wash eyes and skin with pienty of warm water for at least 15 minutes. Ingestion: unlikely with vinyl chloride but should this happen, give warm water. Keep victim warm and quiet. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

### **ENVIRONMENTAL PROTECTION MEASURES**

### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# XYLENES (o-, m-, p-) C6H4(CH3)2

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UN No. 1307

Common Synonyms	Observable Characteristics	Manufacturers
1,2: 1,3: 1,4 - DIMETHYLBENZENE XYLOLS o-, m-, p-dimethylbenzene	Colouriess liquid. Sweet aromatic odour.	Finachem, Montreal, Quebec. Sunchem, Sarnia, Ontario. Shell Canada, Corunna, Ontario.
Xylenes are typically a mixture of the three isomers, ortho, meta and para, with the last two being the predominant portion.		
Transportation and Storage Information	· . · · · · · · · · · · · · · · · · · ·	
Shipping State: Liquid. Classification: Flammable. Inert Atmosphere: No requirement. Venting: Open or pressure-vacuum with flame arrester. Pump Type: Gear or centrifugal. Steel or stainless steel.	Label(s): Red label - FLAMMABLE LIQUID; Class 3.2, Group II. Storage Temperature: Ambient. Hose Type: Material should be resistant to aromatic chemical attack.	Grades or Purity: Technical, 99.2%; pure, 99.9%; nitration grade - xylene thinner.  Containers and Materials: Bottles, cans, drums, tank cars, tank trucks; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Liquid. Solubility (Water): (o-) 0.018 g/100 mL (20°C); (p-) 0.02 g/100 mL (25°C). Molecular Weight: 106.2 Vapour Pressure: (o-) 5 mm Hg (20°C); 9 mm Hg (30°C); (m-) 6 mm Hg (20°C); 11 mm Hg (30°C); (p-) 6.5 mm Hg (30°C); 12 mm Hg (30°C). Boiling Point: 144.4°C (o-); 139°C (m-); 138.4 (p-).	Fioatability (Water): Floats. Odour: Sweet aromatic odour (0.05 to 0.27 ppm, odour threshold). Flash Point: (m-) 27°C; (o-) 32°C; (p-) 27°C. Vapour Bensity: 3.7 (o-); 3.66 (m-); 3.65 (p-). Specific Gravity: 0.38 (o-); 0.36 (m-); 0.36 (p-); (20°C).	Colour: Colourless. Explosive Limits: (m-, p-) 1.1 to 7%; (o-) 1.0 to 6.0 %. Melting Point: -25°C (o-); -48 (m-); 13.0°C (p-).
	HAZARD DATA	

HAZARD DATA				
Human Health Symptoms: Inhalation: dizziness, coughing, nat	usea and vomiting, fatigue, drowsiness, narcosis; in	gestion: gastrointestinal irritation, dizziness, fatigue, loss		
Toxicology: Moderately toxic by inhalation.	ering, inflammation; skin - dryness, cracking, may			
TLV ² (skin) 100 ppm; 435 mg/m ³ .  Short-term inhalation Limits - (skin) 150 ppm; 655 mg/m ³ (15 min).  Delayed Toxicity - No information.	LC ₅₀ - No information. LC _{LO} - Inhalation: rat = 8 000 ppm/4 h (m-). LC _{LO} - Inhalation: rat = 4 912 ppm/24 h (p-)	LD30 - Orak rat = 50 g/kg (p-)		
Fire				
Fire Extinguishing Agents: Use dry chemical, co	arbon dioxide, foam. Water spray may be used to c	ool fire-exposed containers.		
Behavior in Fire: Containers may rupture in fire ignition Temperature: 463°C (0-); 527°C (m-);	es. Plash back may occur along vapour trail. 528°C (p-).	Burning Rate: 5.8 mm/min.		
Reactivity				
With Water: No reaction. With Common Materials: Can react with exidi: Stability: Stable.	zing materials.			
Environment		,		
Water: Prevent entry into water intakes and w meta 16 mg/L/24 h/goldfish/LD ₅₀ ; para 18 mg/ BOD: 64 to 235%, 5 days. Land-Air: 800 to 2 400 ppm threshold for comm Food Chain Concentration Potential: No inform	L/24 h/goldtish/LD ₅₀₁ 13.5 ppm/96 h/bluegiii/LC ₅ non crops.	L/96 h/TLm/freshwater; ortho 13 mg/L/24 h/goldfish/LD 3; g/freshwater; 8.2 ppm/96 h/rainbow trout/LC3/freshwater;		

### Special Hazards

#### FLAMMABLE.

#### Immediate Responses

Keep non-involved people away from spill site. Issue warning: "FLAMMABLE". CALL FIRE DEPARTMENT. Eliminate all sources of ignition. Avoid skin contact and inhalation. Stop or reduce discharge if this can be done without risk. Dike to contain spill and prevent runoff. Notify supplier and environmental authorities.

# Protective Ciothing and Equipment

Respiratory protection - in fires or confined areas, use self-contained breathing apparatus and totally encapsulated suit. Otherwise; Gloves - rubber. Eye protection - optional - goggies or face shield. Boots - rubber (pants worn outside boots).

## Fire and Explosion

Use dry chemical, carbon dioxide or foam to extinguish. Water spray may be used to cool fire-exposed containers. Flash back may occur along vapour trait.

#### First Ald

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Contact: remove contaminated clothing, wash affected parts with plenty of water (and soap if available). Flush eyes with plenty of water. Inhalation: give artificial respiration if breathing has stopped, give oxygen if breathing is laboured. Ingestions do not induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. If floating, skim and remove.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other material.
- 4. Remove material with pumps or vacuum equipment and place in appropriate containers.
- 5. Recover undamaged containers.
  6. Absorb residual liquid on natural or synthetic sorbents.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.
- 3. incinerate (approval of environmental authorities required).

# YELLOW CAKE UO2(OH)2-x(O-y+)x

x = 0,1.2 y+ = Na*, Mg* or NH4*

# IDENTIFICATION

UN No. 2912

Common Synonyms	Observable Characteristics	Manufacturers
URANIUM CONCENTRATE SYNTHETIC CARNOLITE	Yellow to yellow-brown powder. Odourless.	Eldorado Nuclear, Ottawa, Ont., Eldorado, Sask. Dennison Mines, Rio Algom Ltd., Elliot Lake, Ont. Madawaska Mines, Bancroft, Ont.
Transportation and Storage information		
Shipping State: Solid. Classification: Radioactive. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Yellow, white and black label - RADIOACTIVE SUBSTANCE; LSA 1. Storage Temperature: Ambient.	Grades or Purity: Various, by % Uranium. Containers and Materials: Drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): Insoluble. Molecular Weight: Variable. Vapour Pressure: No information. Boiling Point: No information.	Floatability (Water): Sinks. Odour: Odourless. Flash Point: Not flammable. Yapour Density: No Information. Specific Gravity: Variable >1.	Colour: Yellow to yellow-brown. Explosive Limits: Not flammable. Melting Point: No information.

Human Health		
Symptoms: Inhalation: (dust) severe respiratory tr Toxicology: Moderately toxic by all routes. TLY*- 0.2 mg/m³ (as uranium) Short-term Inhalation Limits - 0.6 mg/m³ (15 min). (as uranium)	act irritation. <u>Ingestion</u> : nausea, vomiting LC39 - No information. Delayed Toxicity - No information.	. <u>Contact</u> : skin - irritation, yellow colouration of skin.  LD ₅₀ - No information.
Fire		
Fire Extinguishing Agents: Not combustible. Mos Behaviour in Fire: Not combustible. Ignition Temperature: Not combustible.	t fire extinguishing agents may be used in Burning Rate: Not combustible.	fires involving yellow cake
Reactivity		
With Water: No reaction. With Common Materials: No information. Stability: Radioactivity relatively low level of ga	nma radiation.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No information.	•	

### Special Hazards

RADIOACTIVE

#### immediate Responses

Keep non-involved people away from spill site. Notify Atomic Energy Control Board. Notify manufacturer or supplier. Notify environmental authorities.

### Protective Clothing and Equipment

Outer protective clothing as required.

# Fire and Explosion

Not combustible. Most fire extinguishing agents may be used in fires involving yellow cake.

#### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: if breathing has stopped give artificial respiration; if laboured, give oxygen. Contact: skin - remove contaminated clothing and flush affected areas with plenty of water; eyes - irrigate with plenty of water. Ingestion: give conscious victim plenty of water to drink and induce vomiting. If medical assistance is not immediately available, transport victim to hospital, doctor or clinic.

#### ENVIRONMENTAL PROTECTION MEASURES

# Response

#### Water

- 1. Notify the Atomic Energy Control Board.
- 2. Stop or reduce discharge if safe to do so.
- 3. Contact manufacturer or supplier for advice.
- 4. If possible, contain discharge by damming or water diversion.
- 5. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- 6. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

# Land-Air

- I. Notify the Atomic Energy Control Board.
- 2. Stop or reduce discharge if safe to do so.
- 3. Contact manufacturer or supplier for advice.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Notify the Atomic Energy Control Board.
- 2. Contact manufacturer or supplier for advice on disposal.
- 3. Contact environmental authorities for advice on disposal.

# ZINC Zn

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Mamulacturers
GRANULAR POWDER ZINC ZINC DUST	Shiny white metal with bluish-grey lustre or blue-grey to black powder or dust. Odourless.	Cominco, Trail, B.C. Canadian Electrolytic Zinc, Valleylield, Que. Texasgulf Canada, Hoyle, Ontario.
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: No requirement.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Zinc dust, 99.0%; zinc powder, 99.0%; intermediate 99.5%; high grade 89.9%. Containers and Materials: Cans, drums, bulk by truck or rail.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.00007 g/100 mL (25°C). Molecular Weight: 65.4 Vapour Pressure: 0.13 mm Hg (487°C). Boiling Point: 907°C.	Floatability (Water): Sinks. Odour: Odourless. Flash Point: Not flammable. Yapour Density: No information. Specific Gravity: 7.1 (23°C).	Colour: Blue-grey to black. Explosive Limits: Dust forms explosive mixtures with air. Melting Point: 420°C.

Human Health	
Symptoms: inhaintion: (dust) coughing, fever as Textcology: Low textcity by inhaintion. TLY® (inhalation) 5 mg/m³ (as ZnO fume). Short-term inhalation Limits - 10 mg/m³ (as ZnO fume).	LC50 - No information.  LC50 - No information.  TCL0 - Inhalation: human = 124 mg/m ³ (50 min).  Delayed Toxicity - No information.
Fire \	
Fire Extinguishing Agents: In fires use carbon of Behaviour in Fire: When heated, releases highly ignition Temperature: 460°C (dust layer); 630°C (dust cloud).	toxic zinc fumes. Zinc dust may form explosive mixtures in air.
Reactivity	
With Water: No reaction. With Common Materials: Dust reacts with acide nitric acid, performic acid, potassium chromate acidic or alkali solutions can result in the evolu Stability: Stable.	s, chlorites, chlorine, fluorine, hydrazine, potassium nitrate, sulfur, cadmium, chlorates, chromic anhydride, potassium peroxide, selenium, sodium chlorate, sodium peroxide, and ammonium nitrate. In contact with tion of hydrogen.
Environment	
Water: Prevent entry into water intakes and wa 0.22 mg/L); Zn ² +/48 h/rainbow trout/TLm/frest Land-Air: No information. Food Chain Concentration Potential: No inform	

### Special Hazards

Dust concentrations may be explosive.

## Immediate Responses

Keep non-involved people away from spill site. If zinc dust is involved, caution should be taken in confined area because of explosive potential. Stop discharge if possible. Contact supplier or manufacturer. Notify environmental authorities.

## Protective Clothing and Equipment

Respiratory protection - in fires or confined areas use self-contained breathing apparatus; otherwise, protective clothing as necessary.

# Fire and Explosion

Use carbon dioxide and dry chemical.

#### First AM

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhatation: if breathing has stopped, give artificial respiration; if laboured give oxygen. Victim should be taken to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

## Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by booming.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recovery undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# ZINC CHLORIDE ZnC12

# IDENTIFICATION

UN No. 8159 solution 8160 anhydrous

Common Synonyms	Observable Characteristics	Manufacturers
BUTTER OF ZINC (granular) ZINC DICHLORIDE SOLUTION ZINC MURIATE SOLUTION	White, granular, crystal or powder or clear solution. Odourless.	Canadian Industries Ltd., Montreal, Quebec.
Transportation and Storage Information		
Shipping State: Solid or liquid (solution). Classification: Corrosive. Inert Atmosphere: No requirement. Venting: Open. Pump Type: Gear, centrifugal; rubber or plastic lined; stainless steel (solutions only).	Label(s): White and black label - CORROSIVE; Class 8, Group III. Storage Temperature: Ambient. Hose Type: Natural rubber, polyethylene, poly- propylene, Chemitiex 951 (solutions only).	Grades or Purity: Solid; CP or technical; Solutions, 50% or 62.5%. Containers and Materials: Drums, tank cars, tank trucks; lined stainless steel or fibreglass.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility: 432 g/100 ml. (25°C); 614 g/100 ml. (100°C). Molecular Weight: 136.3 Yapour Pressure: 1 mm Hg (428°C). Boiling Point: 732°C.	Floatability (Water): Sinks and mixes. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: Solution, 1.61 (15.5°C); Solid, 2.9 (25°C).	Colour: White, solid or clear solution. Explosive Limits: Not flammable. Melting Point: 283 to 290°C.

Human Health	
Symptoms: Contact: may cause severe skin irritation of nose and throat Toxicology: Highly toxic by inhalation. Moderately TLV - (inhalation) 1 mg/m ³ (as ZnCl ₂ fume). Short-term inhalation Limits - 2 mg/m ³ (15 min) (as ZnCl ₂ fume).	
Fire	
Fire Extinguishing Agents: Not combustible. Most in Behaviour in Fire: Not combustible, but when heate ignition Temperature: Not combustible.	lire extinguishing agents can be used on fires involving zinc chloride. d to decomposition, emits corrosive hydrogen chloride gas. Burning Rate: Not combustible.
Reactivity	
With Water: No reaction; soluble. With Common Materials: in contact with most acids Stability: Stable.	s, corrosive hydrogen chloride is produced. Reacts violently with potassium and is corrosive to most metals.
Environment	
	ways. Harmful to aquatic life in low concentrations. Fish toxicity: 7.2 ppm/96 h/bluegill/TLm/freshwater; h/Daphnia magna/LC50/continuous flow; BOD: None.

# Special Hazards

CORROSIVE.

# immediate Responses

Keep non-involved people away from spill site. Issue warning: "CORROSIVE". Stop or reduce discharge if this can be done without risk. Avoid contact. Contain spill by diking to prevent entry into watercourses and sewers. Notify manufacturer or supplier. Notify environmental authorities.

# Protective Clothing and Equipment

Respiratory protection - in fires or confined spaces use self-contained breathing apparatus; otherwise, Goggles - (mono), tight fitting. If face shield is used, it must not replace goggles. Gloves - rubber. Boots - high, rubber (pants worn outside boots). Outerwear as required, coveralls, apron, etc. For dusts (granular product), dust respirator may be needed.

# Fire and Explosion

Not combustible. In fires involving ZnCl2, most fire extinguishing agents may be used. When heated to decomposition, may emit corrosive hydrogen chloride

#### First Aid

Move victim out of split site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: remove victim to fresh air, get medical attention. Contact: eyes - irrigate with plenty of water for at least 15 minutes; skin - wash exposed skin with plenty of water; at same time, remove contaminated clothing. Ingestion: If victim is conscious, give large amounts of milk or water. If medical assistance is not quickly available, transport victim to hospital, doctor or clinic.

# **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminants bottom sediments.
- 5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Dike to prevent runoff from rainwater or water application.
- 4. Remove material by manual or mechanical means.
- Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# ZINC OXIDE ZAO

# IDENTIFICATION

Common Synonyms	Observable Characteristics	Manufacturers
ZINC WHITE CHINESE WHITE ZINCITE	White to yellow-white or grey powder. Odourless.	Zochem, Bramalea, Ontario. Pigment and Chemical, Milton, Ontario. Prodults Chemiques, GH, St. Hyacinthe, Quebec.
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated. Storage Temperature: Ambient.	Grades or Purity: Lead-free; leaded. Containers and Materials: Boxes, drums, multi-wall paper bags.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 0.00016 mg/100 mL (29°C). Molecular Weight: 81.4 Vapour Pressure: No information. Boiling Point: Sublimes >1 800°C.	Floatability (Water): Sinks. Odour: Odouriess. Flash Point: Not flammable. Vapour Density: 2.0 Specific Gravity: 5.6 (20°C).	Colour: White to yellow-white or grey. Explosive Limits: Not combustible. Not flammable. Melting Point: 1 975°C (sublimes at temperatures >1 800°C).

	HAZARD BATA	
Human Health		
Symptoms: Inhalation: (fumes) fever, chilis, naus Taxicology: Moderately toxic upon inhalation or TLV ² 5 mg/m ³ (ZnO iumes). Short-term inhalation Limits - 10 mg/m ³ (15 min) (ZnO iumes).	ea, vomiting, muscular aches, and weakness. In ingestion.  LC 50 - No information.  Delayed Toxicity - No information.  TCL0 - Inhalation = 600 mg/m ³ (30 min).	ngestion: metallic taste in mouth, nausea and vomiting.  LD50 - Oral: rat = 0.63 g/kg
Fire		
Fire Extinguishing Agents: Not combustible; more Behaviour in Fire: Not combustible; toxic ZnO filignition Temperature: Not combustible.  Reactivity	st fire extinguishing agents may be used in firesumes can be formed in fires.  Burning Rate: Not combustible.	s involving ZnO.
With Water: No reaction. With Common Materials: Reacts violently with a Stability: Stable.	nagnesium.	
Environment		
Water: Prevent entry into water intakes and water Land-Air: No information. Food Chain Concentration Potential: No information		

# Special Hazards

### Immediate Responses

Keep non-involved people away from spill site. Call manufacturer or supplier for advice. Stop or reduce discharge if safe to do so. Contact environmental authorities.

## Protective Clothing and Equipment

In fires, Respiratory protection - self-contained breathing apparatus; otherwise, suitable dust respirator and protective clothing as required.

### Fire and Expission

Not combustible; in fires involving zinc oxide; most fire extinguishing agents may be used.

#### First AL

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. Inhalation: give artificial respiration if necessary. Contact: skin - remove contaminated clothing and flush affected areas with water; eyes - irrigate with large amounts of water. Ingestion: give water to drink. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

#### **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 4. Dredge or vacuum pump to remove contaminants, liquids and contaminated bottom sediments.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

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# ZINC SULFATE $ZnSO_4 \cdot xH_2O(x = 0,1,7)$

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UN No. NA9161

Common Synonyms	Observable Characteristics	Manufacturers
ZINC SULFATE MONOHYDRATE ZINC SULFATE HEPTAHYDRATE - White Vitriol - White Copperas - Zinc Vitriol ZINC SULPHATE	Anhydrous form; colourless, crystals. Hydrous forms; White, crystalline powder. Odourless.	Cominco, Trail, B.C. Canadian Electrolytic Zinc, Valleyfield, Quebec. Texasgulf Canada, Timmins, Ontario.
Transportation and Storage Information		
Shipping State: Solid. Classification: None. Inert Atmosphere: No requirement. Venting: Open.	Label(s): Not regulated, Class 9.2, Group I. Storage Temperature: Ambient.	Grades or Purity: Commercial, Technical. Containers and Materials: Bags, fibre drums, drums; steel.
Physical and Chemical Characteristics		
Physical State (20°C, 1 atm): Solid. Solubility (Water): 96.5 g/100 mL (20°C); (hepta). Molecular Weight: 161.4 (anhydrous); 179.4 (mono); 287.5 (hepta). Vapour Pressure: No information. Bolling Point: Anhydrous 600°C (decomposes); heptahydrate loses H2O 280°C; monohydrate loses H2O >238°C.	Floatability (Water): Sinks. Odour: Odourless. Flash Point: Not flammable. Vapour Density: No information. Specific Gravity: Heptahydrate 1.96 (25°C); anhydrous 3.5 (23°C).	Colour: Colouriess to white. Explosive Limitss: Not flammable. Melting Point: (hepta) 100°C; loses water at 280°C; monohydrate loses water at 238°C; ail decompose above 500°C.

***************************************		
Human Health		
vomiting and diarrhea.		eyes and skin - redness and pain. <u>Ingestion</u> : abdominal spasms
Toxicology: Moderately toxic by inhalation and co TLV® No information.		I the a letter position calls are use a 0.020 allier
The term inhalation Limits - No information.	LC50 - No information.  Delayed Toxicity - No information.	LD50 - Intraperitoneal: mouse = 0.029 g/kg
ment-iciti hambition Philito - de Sifetiigibii	Desiyes toxicity - No intermation.	LDLo - Oral: rat = 2.2 g/kg (hepta)
Fire		
Fire Extinguishing Agents: Not combustible; most Behaviour in Fire: Not combustible.	fire extinguishing agents may be used in fires in	nvolving zinc sulfate.
ignition Temperature: Not combustible.	Burning Rate: Not combustible.	
Reactivity	**************************************	
With Water: No reaction.		
With Common Materials: No information.		
Stability: Stable.		
Environment		
	man a s. i.d. dozenić s.	the day of the beautiful and the said
	rways. Fish toxicity: 4.6 ppm/96 h/rainbow troui	t/LC5g/freshwater; 0.3 mg/L (Zn)/120 h/stickleback/lethal/
freshwater; BOD: Not available.		
Land-Air: No information.		
Food Chain Concentration Potential: No informat	ion	

### Special Hazards

#### Immediate Responses

Keep non-involved people away from spill site. Dike spill area, particularly if there is any danger of water runoff from tire fighting or rain. Lightly wet down dry spillage, if there is any danger of dust drift from wind. Notify environmental authorities. Notify supplier or manufacturer.

# Protective Clothing and Equipment

Wear dust mask if necessary. Other protective equipment; gloves, goggles, coveralls, if required.

### Fire and Explosion

Not combustible; in fires involving zinc sulfate, most fire extinguishing agents may be used.

### First Aid

Move victim out of spill site to fresh air. Call for medical assistance, but start first aid at once. <u>Inhalation</u>: give artificial respiration, if necessary. <u>Contact</u>: skin - remove contaminated clothing and flush affected areas with water; eyes - irrigate with large amounts of water. <u>Ingestion</u>: give water to drink. If medical assistance is not immediately available, transport victim to doctor, clinic or hospital.

## **ENVIRONMENTAL PROTECTION MEASURES**

#### Response

#### Water

- 1. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. If possible, contain discharge by damming or water diversion.
- 6. Dredge or vacuum pump to remove contaminants,
- liquids and contaminated bottom sediments.

  5. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

#### Land-Air

- I. Stop or reduce discharge if safe to do so.
- 2. Contact manufacturer or supplier for advice.
- 3. Contain spill by diking with earth
- or other barrier.
- 4. Remove material by manual or mechanical means.
- 5. Recover undamaged containers.
- 6. Remove contaminated soil for disposal.
- 7. Notify environmental authorities to discuss disposal and cleanup of contaminated materials.

- 1. Contact manufacturer or supplier for advice on disposal.
- 2. Contact environmental authorities for advice on disposal.

# Common Abbreviations

BOD	biological oxygen demand	MIC	maximum immission
b.p.	boiling point		concentration
c.c.	closed cup	min	minute or minimum
cm	centimetre	നന	millimetre
COD	chemical oxygen demand	m.p.	melting point
conc.	concentration	МŴ	molecular weight
g	gram	N	newton
ha	hectare	nm	nanometre
Hg	mercury	O	ortho
IDLH	immediately dangerous to	0.C.	open cup
	life and health	р	para
Imp. gal.	imperial gallon	PEL	permissible exposure level
in.	inch	рH	measure of acidity/
Ĵ	ioule	•	alkalinity
kg	kilogram	ppb	parts per billion
kĬ	kilojoule	ppm	parts per million
km	kilometre	$P_s$	standard pressure
kPa	kilopascal	psi	pounds per square inch
kt	kilotonne	S	second
L	litre	STEL	short-term exposure limit
lb.	pound	STIL	short-term inhalation limit
LC50	lethal concentration fifty	TCLo	toxic concentration low
LCLo	lethal concentration low	Td	decomposition temperature
LD50	lethal dose fifty	TDLo	toxic dose low
LDLo	lethal dose low	TLm	median tolerance limit
LEL	lower explosive limit	TLV	Threshold Limit Value
LFL	lower flammability limit	tns	time period not specified
m	metre	T _{S.}	standard temperature
m	meta	TWA	time weighted average
M	molar	UEL	upper explosive limit
MAC	maximum acceptable con-	UFL	upper flammability limit
	centration	v/v	volume per volume
max	maximum	w/s	weight per weight
mg	milligram		
-	· <del>-</del>		

μg	microgram		
μm	micrometre		

*Be degrees Baumé (density)