VALIDATION OF TOXFATE
A CONTAMINANT FATE MODEL FOR A LARGE LAKE SYSTEM（LAKE ONTARIO）
by
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A TOXic contaminants FATE model, TOXFATE, has been used to predict the fate of four chlorobenzenes which have entered Lake ontario fram the Niagara River since 1999, they are: 1,2,4-trichlorobenzene (1,2,4-TCB), 1,2,3,4tetrachlorobenzene ( $1,2,3,4-\mathrm{TeCB}$ ), penta- ( $Q C B$ ) and hexachlorobenzene (HCB). TOXFATE predicts concentrations in the water colum and bottom sediments as well as volatilization, burial and transport rates. The mathematical model has been validated by comparing computed compartment masses of contaminants and process losses to those estimated from field data. Computed sedimentation, volatilization and transport rates to the St. Lawrence River agree with field estimates within a few percentage points. Simulations show that on average about $67 \%$ of HCB loadings and 73 to $89 \%$ of the other three chlorobenzenes are lost from the lake through volatilization; the model predicts that only $3 \%$ of $1,2,4-\mathrm{TCB}$ loadings remain in the bottom sediments while 13 to $29 \%$ of the other congeners do. Despite total chemical inputs to the lake of 15000 to 310060 kg , the water fraction in 1983 contains anly a minute proportion, namely 60 to $8 \emptyset \emptyset \mathrm{~kg}$ for each chemical over the whole lake, resulting in very low water concentrations on the order of $\varnothing .1 \mathrm{ng} / \mathrm{L}$ or 0.1 ppt. The amount in fish is much lower, 6 kg for HCB and as little as 0.14 kg for $1,2,4-\mathrm{TCB}$ in the whole lake.

On s'est servi du modèle TOXFATE, qui permet de déterminer la destination ultime des contaminants toxiques, pour prévoir celle de quatre chlorobenzènes que la rivière Niagarạ déverse dans le lac Ontario depuis 1909. Ce sont : le $1,2,4$-trichlorobenzène (1,2,4-TCB), le $1,2,3,4$-tétrachlorobenzène ( $1,2,3,4-\mathrm{TeCB}$ ), le pentachlorobenzène ( QBC ) et 1 'hexachlorobenzène (HCB). À l'aide du modèle TOXFATE, on peut prévoir les concentrations d'un contaminant dans la colonne d'eau et dans les sédiments ainsi que les taux de volatilisation, de transport et d'enfouissement. Le modèle mathématique a êté validé en comparant les masses de contaminants aux différents paliers du système aquatique et les pertes dues à la transformation avec les valeurs estimées à partir des données réelles. Les taux modélisés de sédimentation, de volatilisation et de dēversement dans le fleuve Saint-Laurent concordent avec les valeurs estimatives rẻelles à quelques points de pourcentage près. D'après les ṣimulations, il semble qu'en moyenne 67 p. 100 des charges de HCB et entre 73 p. 100 et 89 p. 100 des trois autres chlorobenzènesase dispersent par volatiliṣation. De plus, le modèle prévoit qu'il ne reste que 3 p. 100 de la charge de $1,2,4$-TCB dans les sédiments du fond tandis que la proportion des autres chlorobenzènes se situe entre 13 p. 100 et 29 p. 100. Malgré un apport total de contaminants toxiques de $1^{\prime}$ 'ordre de 15000 à 310000 kg , en 1983 , on n 'a trouvé qu'une faible proportion de chaque produit chimique dans I'ensemble du lac entre 60 et 800 kg . Par conséquent, les concentrations dans $l^{\prime}$ eau se sont révélées très faibles, soit de l'ordre de $0,1 \mathrm{ng} / \mathrm{L}$ ou de $0,1 \mathrm{ppt}$. Les quantités décelées dans le poisson ont été encore moindres, soit 6 kg de HCB et seulement $0,14 \mathrm{~kg}$ de $1,2,4-\mathrm{TCB}$ pour 1 'ensemble du lac.

Varying amounts of toxic contaminants have entered Lake Ontario from the Niagara River since 1909 and these contaminants represent a potential hazard to drinking water and fish populations. Prediction of expected concentrations in the aquatic environment is important in order to provide much needed inputs to the decision making processes associated with the sound management of potable water supplies, of fish used for human consumption and of bottom sediments which can remain a major source of internal loadings after external inputs are reduced. In this paper, results from TOXFATE, a simulation model, are compared to field measurements for four chlorinated organic compounds for Lake Ontario. The agreement between model predictions and field measurements was good for all compounds. These results demonstrate the potential use of TOXFATE for predicting the fate of other toxic contaminants in Lake Ontario. Predictions of contaminant concentrations are essential for a large lake that may respond very slowly (average water residence time is about eight years) to reduce contaminant loadings.

Depuis 1909, la rivière Niagara déverse dans le lac Ontario dès quantités variables de contaminants toxiques. Ceux-ci risquent dẹ détériorer la qualité de l'eau potable et de mettre certaines populations de poissons en danger. Il est important de pouvoir prévoir les concentrations auxquelles on peut s'attendre dans le milieu aquatique pour être en mesure de prendre des décisions éclairées eṇ matière de gestion des réserves d'eau potable, des populations de poissons destinés à la consommation et des sédiments lacustres desquels peuvent resurgir des quantitês importantes de contaminants toxiques accumulés, même après une diminution des apports externes. Dans le présent rapport, on compare les résultats obtenus à l'aide du modèle mathématique TOXFATE à ceux des mesures expérimentales en milieu aquatique pour quatre composés organiques chlorés dans le lac Ontario. Les mesures sur le terrain et les prévisions du modèle concordent pour tous les composés à l'étude. Ces résultats laissent à penser que le modèle TOXFATE pourra être utile à l'avenir pour prévoir le cheminement $d^{\prime}$ autres contaminants toxiques que 1 'on trouve daṇs le lac Ontario. Il est essentiel de pouvoir prévoir les concentrations des contaminants dans une étendue d'eau de grande envergure puisqu'elle rêagit lentement (le dēlai moyen de renouvellement de l'eau du lac Ontario est de huit ans) aux fluctuations des charges de contaminants.

## INIRODUCTIGN

Lake ontario, one of the Great Lakes of North America, receives large amounts of toxic ontaminants from inflowing rivers and from the atmosphere (1). The problem of toxic contaminants is of particular concern since Lake ontario is a source of freshwater to many towns along its shores, Kingston, Toronto, Hamilton, Rochester and oswego, in addition to being a potentially rich source for commercial fishing. Unfortunately, the continuous inputs of various chemicals has lead to significant residue levels in fish (2).

Fate models have been proposed as tools to predict the fate of toxic contaminants. The spatial scales involved - the lake is 300 km long with a volume of $1 \rho^{12} \mathrm{~m}^{3}$ and wind driven currents transport a water mass ten times greater than the Niagara River flow ( $6000 \mathrm{~m}^{3} / \mathrm{s}$ ) (3) - make any modelling effort difficult; the physical transport plays an important role in determining the spatial distribution, fate and persistence. The contaminants may undergo a number of chemical transformations and changes of phases: they may be biodegraded, oxidized or photolyzed, they can move to the atmosphere by volatilization or be buried in the bottom sediments, they can leave the lake via the St. Lawrence River or bioaccumulate in the food chain. A mathematical model must take all these factors into consideration to produce an integrated and reliable prediction of a contaminant's fate.

TOXFATE, an organic TOXic contaminants FATE model, was developed (4) in 1983 for the specific purpose of modeling the fate of toxic contaminants in large lakes after a preliminary effort of using the EXAMS framework (5) was partially successful (6). In its present configuration (4,7) which includes the water circulation of Lake ontario, TOXFATE describes and predicts the fate of a toxic contaminant which enters Lake ontario fran the Niagara River; the river receives large discharges from industries and leachings from dumps located along its shores and has been a source of contaminants to the lake
since the beginning of the century (8). TOXFATE has been developed and verified (4) by simulating the fate of nine toxic contaminants, some of which have been detected in the lake, such as mirex, lindane and pentachlorophenol; at the time, however, validation was not possible because of the lack of long term loadings and concentration data. From the analysis of a radiodated sediment core from Lake ontario (8) and measurements of contaminant loadings to lake ontario from 1981 to 1983 (9) we have been able to estimate past loadings of several contaminants to Lake ontario from the Niagara River. In this paper the predictive ability of the model is tested and campared to field data for four chlorobenzenes. The model includes degradation processes but biodegradation and photolysis terms were set to zero since these processes are negligible for the studied chlorobenzenes.

PAST LOADINGS ESTIMATES
The approximate loadings of chlorinated contaminants to Lake ontario from the Niagara River for the period 1981 to 1983 can be estimated using the weekly sampling data of Oliver and Niool (9). During the course of that study, a sediment core was also collected, analyzed and dated using radiochemical methods (8). The major chlorinated campounds foum in both of these studies were chlorobenzenes, or CB's, $s 0$ we chose the four chemicals, hexachlorobenzene (HCB), pentachlorobenzene ( $Q C B$ ), 1,2,3,4-tetrachlorobenzene ( $1,2,3,4-\mathrm{TeCB})$, and $1,2,4$-trichlorobenzene $(1,2,4$-TCB) to test and validate the model. In addition to their presence in all samples, these campounds span a fairly wide range of physical chemical properties.

Present loadings (1981-1983) and historical loadings (1909-1981) were estimated fram the sediment core analysis (8) using the assumption that a constant percentage of incoming contaminants were sequestered to setting particulates, which became bottom sediments. For example, the 1,2,4-TCB
concentration in the 7 to 8 om interval of the sediment core $(1966-67)$ was approximately 3 times higher than in the surficial sediment ( 0 to 1 cm ). Thus, loadings for this compound in $1966 / 67$ were estimated to be 7406 kg per year, three times the measured 1981 to 1983 loadings of approximately 2400 kg per year. These year to year loadings estimates were used in the modeling exercise (Table 1) and these data compare well with the overall loadings estimates made by Oliver (10) and reported in Table 2. The laboratory studies of Karickhoff (11) have shown that at low concentrations, such as those observed in the river ( $n g / L$ ), the sorption of hydrophobic pollutants by sediments is linear, reversible and characterized by a partition coefficient. Therefore, our assumption should be correct, but, since no historical records or concentrations of these contaminants in the river exist, we can not test our hypothesis.

## LAKE CNTARIO DATA

Table 2 shows the model's camputed masses of CB's in the various lake campartments as well as the camputed percentage losses of chemicals due to sedimentation, volatilization and transportation (St. Lawrence River). The measured $C B$ masses in the lake campartments fram the field data by Oliver (10) are also listed in Table 2. In that study, the masses of CB's lost via the St. Lawrence River were estimated by analyzing two large volume water samples near the outflow of the lake. The CB's in suspended sediments were estimated by analyzing centrifuged material from three stations at two depths an ane occasion. The $C B ' s$ in lake biota were take fram the literature (12,13). Botton sediments were sampled more extensively - one core from each of the major sedimentation basins (Niagara, Mississauga, Rochester) was analyzed in detail together with 15 surficial sediment samples fram each of these basins
as well as three samples fram the Kingston Basin to obtain the CB masses in lake bottan sediments (10).

Also included in Table 2 are estimates of the CB's associated with dewnfluxing material. Oliver and Charlton (14) measured the proportion of Niagara River contaminant input that was sedimenting in lake ontario using sediment traps over a six month period in 1982. This short tem study yielded somewhat lower amounts of CB's sedimenting than Oliver's other study (10). The average suspended sediment concentration in the Niagara River over the trap deployment period was $5 \mathrm{mg} / \mathrm{L}$, somewhat lower than values reported in other studies $-8.4 \mathrm{mg} / \mathrm{L}$ (15) and $22 \mathrm{mg} / \mathrm{L}(16)$. Higher suspended sediment concentrations in the river would tend to increase the percentage of CB's sedimenting to the lake bottom. In any case, considering the limited number of measurements in both field studies, the agreement between them can be considered fairly good.

The estimation of volatilization losses from the field data can be obtained by difference in two ways. The sum of the short term (May to November, 1982) downfluxing material (14) plus the amount of material lost via the St. Lawrence River (10) can be subtracted from the total loading to give volatilization: 1,2,4-TCB (96\%), 1,2,3,4-TeCB (96\%), QCB (93\%) and HCB (80\%). Volatilization losses calculated from long term bottom sediment masses (10) and the same losses via the St. Lawrence River (10) are somewhat different: $1,2,4-\mathrm{TCB}$ (93\%), 1, 2,3,4-ТеСB (93\%), $\alpha$ CB (79\%) and HCB (38\%).

## MODEL SIMUAATIONS

The four chemicals represent a class of volatile contaminants that may remain in the lake temporarily or possibly be persistent: their final sink may be the atmosphere rather than the bottom sediments or the lake waters. The prediction of their fate would therefore be helpful in detemining permissible
loadings and to find the maximum chemical load permissible to Lake ontario before exceeding the limits set by the International Joint Commission. TOXFATE was used to predict their fate and to campute the percentage of the loadings that would be lost to the atmosphere or remain in the lake; a good agreement between predicted and observed processes and between predicted and observed mass balances would be considered a validation of the model, given the fact that the model was only allowed to see the estimated inputs and was not further tampered with. Resuspension rates fram the bottan sediments were set
$-2 \quad-1$ at 5 gm day (17); the chemical properties used in the model for $1,2,4 \mathrm{TCB}$, 1,2,3,4-TeCB, $Q C B$ and HCB were: molecular weight [amu] 181.5, 215.9, 250.3, 284.9; octanol-water partition ooefficient, $\log K, 4.0,4.5,4.9,5.5 ;$ water
 pressure at 20 C [tori] $0.42,0.01,1.7 \times 10,1.1 \times 10$, respectively. The simulations were run farm the period of 1999 to 1983 and took seven hours each on a CDC Cyber 171 ormputer.

## Masses of contaminants in Lake Ontario

Table 2 shows the masses of the four chlorobenzenes in four lake compartments, water, suspended sediments, biota and bottan sediments. The agreement between oliver's estimates (10) and the model's are quite good; masses are within a factor of two for bottan sediments and water and within a factor of three for suspended sediments and biota. Given the very low percentage of loadings that remains in the lake $-0.28 \%$ and $0.40 \%$ in the water compartment for $1,2,4-\mathrm{TCB}$ and for HCB, respectively, and in suspended sediments 0.003 of and $0.03 \%$, respectively - the agreement is quite good, indeed any agreenent within one order of magnitude would have been considered excellent, since the model was run blindly beginning at 1909 to obtain 1983 data. Similar good comparisons were obtained for the other contaminants (Table 2). Notwithstanding the very large inputs to the lake (150币Ø to $31 \varnothing \sigma \varnothing$
kg over the 74 year period for the examined compounds), the water fraction in 1983 contains only a minute proportion, 60 to 800 kg for each of the chemicals over the whole lake, resulting in very low water concentrations in the order of $\varnothing .1 \mathrm{ng} / \mathrm{L}$ or 0.1 ppt. The amount in fish is much lower, 0.1 to 6 kg in the whole lake for each contaminant, but concentrations are higher than in water because of the low fish biomass.

## Oncentrations of contaminants in Lake ontario

A mass balance of contaminants in the lake is important to assess the fate of the chemicals once they enter the lake. From a practical or management point of view the prediction of concentrations, especially in water used as potable, water supplies, in fish used for human consumption and in bottom sediments which can remain as a major source of internal loadings after external inputs are reduced, is of fundamental importance in a large lake that may respond very slowly (average water residence time is 7.8 years) to reduced contaminant loadings. The problem of predicting concentrations is very difficult because by definition a concentration is a ratio between the mass of a contaminant and the mass of the compartment of interest, for example, fish, benthos, suspended sediments etc. For scme compartments this mass can be estimated fairly readily, even if within reasonable limits; for example, suspended sediments are on average $0.4 \mathrm{mg} / \mathrm{L}$, in Lake ontario and this corresponds to 400,000 metric tons of suspended sediments in the whole lake assuming that the lake is hamogeneous. . Conversely, the estimation of bianass is very difficult and estimates can range within a factor of five; for example Oliver (10) estimates biota at $10 \mathrm{~g} / \mathrm{m}^{2}$ in the whole lake or 190,000 metric tons dry weight, with an estimate as low as 85,000 metric tons or as high as 450,060 metric tons. When either number is used as an estimate of bicmass, the predicted concentrations in fish can vary over the same range.

Therefore, if we were satisfied with the prediction of the mass of contaminants within a factor of two, we should be equally pleased if we can estimate concentrations in biota within one order of magnitude.

Given the previous assumptions and limitations, Table 3 shows predicted and literature value concentrations for several lake compartments. Water concentrations agree well with measured values since the water mass is well known and Table 2 showed that contaminant masses were in good agreement, even if only a very small percentage of the input of these volatile CB's remained in the lake. Suspended sediment concentrations vary widely in the lake with standard deviations almost as large as the mean value; in this respect the estimated concentrations are close to the mean lake values. Concentrations in bottom sediments are predicted within a factor of two, similar to the agreement for masses.

In a large lake the most difficult prediction to make is the bioconcentration in fish, benthos and plankton because of the above mentioned problem of not knowing the biamass. For some contaminants, such as QcB the agreement is within a factor of two and for the others the agreement is within a factor of three and certainly within an order of magnitude. The anly discrepancy seems to be the low concentration predicted for HCB in fish. A similar discrepancy between enviromental residue levels and concentrations calculated fram bicooncentration factors has been reported for HCB (12). Some type of food chain transfer coefficient must be incorporated into the model for chemicals such as HCB which have a long half life in fish and, therefore, may never reach chemical equilibrium with the surrounding water. Improved estimates of biomass in the lake are also required before more precise concentrations can be predicted.

Ontaminants' fate and the lake's self-cleaning ability
The fact that only a low percentage of the four contaminants is found in the water confinms oliver's (1б) estimate that most of these contaminants leave the lake system by volatilization fram the surface. Volatilization rates vary according to the chemical, more for $1,2,4-\mathrm{TCB}$ and less for HCB. For example the average yearly loadings since 1977 have been about 120 kg for HCB and 2300 kg for $1,2,4-\mathrm{TCB}$ (Table 1) but concentrations in the bottom sediments top layer are very similar for the two compounds, about $0.1 \mathrm{ug} / \mathrm{g}$, indicating that $1,2,4-\mathrm{TCB}$ is much less persistent than HCB. As for fluxes, TOXFATE predicts that during the period 1977 to 1983 about $67 \%$ of HCB loadings were lost from the lake by volatilization processes, while oliver (10) estimated $80 \%$ during the period May-November 1982. Similarly for the other contaminants, the model produces lower volatilization estimates than Oliver's. The model predicts that the half life of the volatilization process is only 0.3 years for $1,2,4-$ TCB and 1.5 years for HCB, while the half life of the physical transport out of the St. Lawrence river is 3 years for $1,2,4-\mathrm{TCB}$, i.e. similar to the lake water half life, and 12 years for HCB. Lake ontario is a transition zone for the other volatile chemicals analyzed in this study but not for others, such as Mirex, which does not volatilize and is very persistent (1).
one last question that was investigated with the model was the time frame that it would take for some contaminants to be permanently buried in the bottom sediments if loadings were completely stopped. Two chemicals HCB and 1,2,4-TCB were chosen as representative of relatively persistent and more volatile compounds. As mentioned above, $1,2,4-\mathrm{TCB}$ is very volatile and therefore anly about 1 to 38 of the loadings are in the bottan sediments; if loadings were completely stopped it would take about 19 to 38 years for 1,2,4TCB to be buried under 8 am of clean sediments. This estimate is fairly
uncertain given the very small amounts which remain in the sediments. In comparison it will take about 150 years to completely eliminate fram active circulation mirex (1), and about 26 years for HCB.

## DISCUSSION

A simulation model, TOXFATE, has been employed to quantify the behaviour of four chlorobenzenes, $1,2,4-\mathrm{TCB}, 1,2,3,4-\mathrm{Te} C B, Q C B$ and HCB , in Lake ontario. The purpose of this study was to validate and to verify the prediction ability of the model using data, not used in model development. The predicted and observed data in the different components of the ecosystem agree within a factor of two; a very satisfactory conclusion considering the four orders of magnitude range between fish and bottom sediments, 2 to 3 kg of contaminants and 10,000 to $11,080 \mathrm{~kg}$, respectively, the uncertainty of the data set and the uncertainty in same model parameters, e.g. resuspension. computed sedimentation, volatilization and transport rates from the St. Lawrence River agree within a few percentage points with field estimates (Table 2). Given the good agreement with the data, TOXFATE, originally developed and verified for nine other contaminants, can be considered validated and a useful tool to predict the fate of persistent and/or volatile campounds in large lake systems where physical effects, such as wind driven water circulation, play an important role. From a management perspective, TOXFATE could now be used to predict the fate of other contaminants entering lake ontario from the Niagara River and to assess the enviromental effects of different loadings reduction alternatives.

## CONCLUSIONS

As pointed out by Falfon (1) large lake toxic contaminant models are sensitive to one parameter, mainly the resuspension rate from the sediments; fideling or calibrating the model with that parameter allows an almost perfect fit to the data, within a few percentages for the mass of the four contaminants. Therefore, the question is whether the model should be calibrated using the resuspension parameter or whether we should only use measured parameters, such as those available from charlton (17). In this exercise the measured value was used even if a calibrated model would have given a better fit. Comparison with other contaminant data in the future will permit the resolution of this problem. The inescapable conclusion, however, is that resuspension rates should be measured accurately wherever possible since they have fundamental role in controlling contaminants fate in Lake ontario (see also Halfon (6)).

The main problem of the prediction of toxic contaminants fate, ance a model has been satisfactorily validated for a few compounds, is the lack of loading data; this problem is particularly evident in a large lake system. presently, the Niagara River is regularly monitored for compounds in the water and in suspended sediments (15), however, such monitoring has taken place only since the late 197ø's and earlier loadings data can only be inferred from bottom sediments data. For new or recently released compounds, or for compounds not yet identified in the lake, these data are not available and may not be available for a foreseeable future; for contaminants which are already entering the lake, the bottan sediments are a good source of past loadings . data provided that a comprehensive data base with concentrations and dating is available, as shown by Oliver (10).

REFFERENCES
(1) R.J. Allan, A. Mudroch and M. Minawar, eds. 1983. The Niagara River - Lake ontario pollution problem. J. Great Lakes Res. (Special Issue on the Niagara River), 9:109-340.
(2) D.M. Whittle and J.D. Fitzsimmons, 1983 . The influence of the Niagara River and contaminant burdens of Lake Ontario biota. J. Great Lakes Res., 9:295-302.
(3) R.J. Murthy, D.C.L. Lam, T.J. Simons, J.A. Bull, W.M. Schertzer. 1984. Dynamics of the Niagara River plume in Iake ontario. Enviromment Canada, NWRI report \# 84-7.
(4) E. Halfon. in press. TOXFATE, a fate model of toxic substances in Lake ontario. J. Great Lakes Res.
(5) R.R. Lassiter, G.L. Baughman, L.A. Burns. 1979. Fate of toxic organic substances in the aquatic environment. In S.E. Jorgensen ed. State of the Art in Eoological Modelling, Intern. Society of Ecological Modelling, Copenhagen, Denmark, pp. 219-246.
(6) E. Halfon. 1984. Error analysis and simulation of mirex behavior in Lake Ontario, Eoological Modelling, 22:213-252.
(7) E. Halfon. 1984. Modelling the fate of toxic oontaminants in the Niagara River and Lake ontario. Part I and Part II, Enviroment Canada, NWR report NO. 84-39.
(8) R.W. Durham, and B.G. Oliver. 1983. History of Lake ontario contamination fram the Niagara River by sediment radiodating and chlorinated hydrocarbon analysis. J. Great Lakes Res.. 9:160-168.
(9) B.G. Oliver and K.D. Nicol. 1984. Chlorinated contaminants in the Niagara River, 1981-1983. Sci. Tot. Environ., 39: 57-70.
(10) B.G. Oliver. 1984. Distribution and pathways of some chlorinated benzenes in the Niagara River and Lake ontario. Water poll. Res. J. Canada, 19:47-58.
(11) S.W. Karickhoff. 1981. Semi-empirical estimation of sorption of hydrophobic pollutants on natural sediments and soils. Chemosphere, 10:833-846 .
(12) B.G. Oliver and A.J. Niimi. 1983. Bicooncentration of chlorobenzenes from water by rainbow trout: correlations with partition coefficients and environmental residues. Environ. Sci. Technol., 17:287-291.
(13) M.E. Fox, J.H. Carey and B.G. Oliver. 1983. Compartmental distribution of organochlorine contaminants in the Niagara River and the western basin of Lake ontario. J. Great Iakes, Res., 9: 287-294.
(14) B.G. Oliver and N.M. Charlton. 1984. Chlorinated organic contaminants on settling particulates in the Niagara River vicinity of Lake ontario. Environ. Sci. Technol., 18:903-9ø8.
(15) K.W. Kuntz and N.D. Warry. 1983. Chlorinated organic contaminants in water and suspended sediments of the lower Niagara River. J. Great Lakes Res., 9: 241-248.
(16) A.L.W. Kemp, N.S. Harper. 1976. Sedimentation rates and a sediment budget for Lake ontario. J. Great Lakes Res., 2:324-340.
(17) M.N. Charlton. 1983. Downflux of sediment, organic matter, and phosphorus in the Niagara River area of Lake ontario. J. Great Lakes Res., 9: 201211.

Table 1: Estimated loadings ( $\mathrm{kg} / \mathrm{year}$ ) to Iake ontario from the Niagara River 1909-1983.

| Year/Oampound | 1,2,4 - TCB | 1,2,3,4-ТеСВ | QCB | HCB |
| :---: | :---: | :---: | :---: | :---: |
| 1909-1920 | $15 ø \varnothing$ | 150 | 60 | 46 |
| 1921-1931 | 2400 | 290 | 140 | 230 |
| 1932-1939 | 3300 | 490 | 160 | 150 |
| 1940-1946 | 28.0 | 500 | 416 | 180 |
| 1947-1949 | $44 \varnothing \square$ | 880 | 450 | 180 |
| 1950-1953 | 3906 | 670 | 280 | 110 |
| 1954-1956 | 3900 | 1900 | 440 | 190 |
| 1957-1959 | 6100 | 1200 | 560 | 260 |
| 1960-1962 | $89 \varnothing \square$ | 2100 | $77 \square$ | 920 |
| 1963-1965 | $906 \square$ | 3806 | 870 | 710 |
| 1966-1967 | 7408 | 3200 | 710 | $39 \varnothing$ |
| 1968-1969 | 5700 | 1400 | 420 | 350 |
| 1976்-1972 | $39 \square 0$ | 1360 | 300 | 200 |
| 1973-1974 | $37 \varnothing \varnothing$ | 940 | 290 | 160 |
| 1975-1976 | $37 \varnothing \square$ | 880 | 200 | $10 \square$ |
| 1977-1978 | 2406 | 610 | 150 | 100 |
| 1979 | 2200 | 760 | 240 | 140 |
| 1980 | 2400 | 760 | 240 | 140 |
| 1981 | 2400 | 760 | 240 | 160 |
| 1982-1983 | 2400 | 760 | 240 | 120 |

Table 2: Camparibon of simulations and field estimates for the year 1983.


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[^0]:    - Fram Reference 14.
    ** From core data (8) using "one" contimetre slices and used as inputs to the model. Model inputs are differant fram oliver's (10 estimates since they are based on yearly estimates rather than on a global 74 year estimate.

