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Biofouling and Its Control in Membrane Separation
Bioreactors: A review

By:

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Biofouling and Its Control in Membrane Separation Bioreactors: A Review

B.Q. Liao, D.M. Bagley, H.E. Kraemer, G.G. Leppard and S. N. Liss

Abstract

Membrane separation technology is increasingly becoming an important innovation in biological wastewater treatment. Biofouling of the membrane is a major factor impacting the efficient and economic operation of membrane separation bioreactors (MBRs). This review summarizes the state-of-the-art progress in understanding the mechanisms and factors affecting membrane biofouling and the strategies for biofouling control. Biofouling mechanisms include the adsorption of soluble/suspended extracellular polymers on membrane surfaces and in membrane pores, the clogging of membrane pore structure by fine colloidal particles and cell debris, and the adhesion and deposition of sludge cake on membrane surfaces. Design and operating conditions of membrane modules, membrane materials, hydrodynamic conditions in MBR, process and environmental conditions of activated sludge systems and the physicochemical properties of the wastewater are the dominant factors determining membrane biofouling. Current strategies to control biofouling include periodic relaxation, backwashing, chemical cleaning and possible manipulation of hydrodynamic conditions and sludge properties. Achieving full integration of MBRs in wastewater treatment technology requires further research and development. Fundamental information on the bacteria/colloid/membrane interaction, developed through multi-method and multi-scale approaches, is particularly needed.

NWRI RESEARCH SUMMARY

Plain language title

What is the Nature of Biofouling in Water Treatment Apparatus and How Can We Minimize the Problem for the Case of Membrane Filters: A Review

What is the problem and what do scientists already know about it?

Biofouling of water treatment apparatus poses a serious economic problem. Progress in addressing the problem is impeded by a lack of fundamental information on the biofoulants, which are mainly bacteria and their colloidal secretions.

Why did NWRI do this study?

This sub-study evolved, as one of many sub-studies, from a joint NWRI/NSERC/environmental industry/university arrangement instituted in 1997. The university partners were McMaster, Toronto and Ryerson. The major funding initially was 326 K\$ coming from NSERC, supplemented by "in kind" contributions totalling 904 K\$ from the other partners. This sub-study was part of a major effort to understand (1)

what flocs and biofilms do in water treatment tanks, and (2) how to manipulate their activities for maximum cost-effectiveness.

What were the results?

1. We identified a specific scientific problem of considerable importance and addressed it with new technology developed by us. Fundamental aspects of the bacteria/colloid/membrane interaction, for the case of membrane separation bioreactors, are now amenable to analysis by multi-method multi-scale approaches placed in the literature by NWRI expertise. We prepared our review to show how our work fits into a general scheme of biofouling research.

How will these results be used?

1. These results will be used to improve ongoing research programs in many organizations.

Who were our main partners in the study?

1. Natural Sciences and Engineering Research Council of Canada, National Water Research Institute, McMaster University, Hamilton, Dept. of Biology, University of Toronto, Dept. of Chemical Engineering and Applied Chemistry, Ryerson University, Toronto, Dept. of Applied Chemical and Biological Sciences, Eight Canadian industries

L'encrassement biologique dans les bioréacteurs à séparation membranaire et les moyens de le limiter : une revue

B.Q. Liao, D.M. Bagley, H.E. Kraemer, G.G. Leppard et S. N. Liss

Résumé

La séparation membranaire est une innovation qui prend de plus en plus de place dans le traitement biologique des eaux usées. L'encrassement biologique des membranes est l'un des principaux facteurs qui se répercutent sur l'efficacité et les coûts d'exploitation des bioréacteurs à séparation membranaire (BSM). La présente revue résume les plus récents progrès dans la compréhension des mécanismes et des facteurs à l'origine de l'encrassement des membranes, ainsi que les stratégies permettant de le surveiller. Les mécanismes d'encrassement comprennent l'absorption de polymères extracellulaires en solution ou en suspension par les surfaces et les pores des membranes, l'obturation des pores par des particules colloïdales ou cellulaires fines, et l'adhésion et le dépôt de gâteaux de boue sur la surface des membranes. La construction et les conditions d'utilisation des membranes et de leurs matériaux constitutifs, les conditions hydrodynamiques dans les BSM, les conditions de traitement et les conditions ambiantes des systèmes de traitement par boues activées, et les propriétés physicochimiques des eaux usées sont les principaux facteurs d'encrassement des membranes. Les stratégies actuelles de lutte contre l'encrassement comprennent les relaxations périodiques, le lavage, le nettoyage chimique et, peut-être, des modifications des conditions hydrodynamiques et des propriétés des boues. Une intégration complète des BSM à la technologie de traitement des eaux usées exigerait plus de recherche et de développement. On a surtout besoin d'une information de base sur l'interaction bactéries-colloïdes-membrane, information qu'il faudrait obtenir par des approches multidisciplinaires et multi-échelles.

Sommaire des recherches de l'INRE

Titre en langage clair

La nature de l'encrassement dans les équipements de traitement des eaux usées et comment on peut atténuer le problème dans les filtres à membrane : une revue.

Quel est le problème et que savent les chercheurs à ce sujet?

L'encrassement des équipements de traitement des eaux usées pose un grave problème économique. Les progrès visant à résoudre ce problème sont ralentis par un manque d'information de base sur les agents d'encrassement, qui sont principalement les bactéries et leurs sécrétions colloïdales.

Pourquoi l'INRE a-t-il effectué cette étude?

À l'instar de nombreuses autres, la sous-étude actuelle est le résultat d'une entente entre l'INRE, le CRSNG, l'industrie de l'environnement et le monde universitaire, qui remonte

à 1997. Les partenaires universitaires étaient McMaster, Toronto et Ryerson. Initialement, le principal financement est venu du CRSNG (326 k\$), les autres partenaires ayant fait des contributions « en nature » totalisant 904 k\$. La sous-étude faisait partie d'un effort important pour comprendre 1) l'effet produit par les floes et les films biologiques dans les réservoirs de traitement des eaux usées et 2) comment moduler leurs activités pour maximiser la rentabilité.

Quels sont les résultats?

Nous avons identifié un problème scientifique particulier d'une grande importance et l'avons résolu grâce à une nouvelle technologie que nous avons mise au point. Certains aspects fondamentaux de l'interaction bactéries-colloïdes-membrane dans les bioréacteurs à séparation membranaire peuvent maintenant être analysés au moyen d'approches multidisciplinaires et multi-échelles élaborées et publiées par l'INRE. Nous avons préparé notre revue dans le but de montrer comment nos travaux s'intègrent à la recherche sur l'encrassement.

Comment ces résultats seront-ils utilisés?

Ces résultats seront utilisés pour améliorer les programmes de recherche courants de nombreux organismes.

Quels étaient nos principaux partenaires dans cette étude?

Conseil de recherches en sciences naturelles et en génie (CRSNG); Institut national de recherche sur les eaux (INRE); université McMaster, Hamilton, Dept. Of Biology; université de Toronto, Dept. of Chemical Engineering and Applied Chemistry; université Ryerson, Toronto, Dept. of Applied Chemical and Biological Sciences; huit industries canadiennes.

1. Introduction

The integration of membrane separation technology into conventional activated sludge processes for biomass separation represents an important innovation in the evolution of biological wastewater treatment technology. Membrane separation bioreactors (MBRs) operate without secondary clarification. MBRs offer several advantages over the conventional activated sludge process, including a high biomass concentration, a compact footprint, better uncoupling of the hydraulic retention time (HRT) and sludge retention time (SRT), low sludge production, resistance to shock loading, and, most importantly, superior permeate quality (Brindle and Stephenson, 1996; Manem and Sanderson, 1996; Yamamoto *et al.*, 1997; Van Dijk and Roncken, 1997; Gander *et al.*, 2000; Stephenson *et al.*, 2001).

The primary role of the membrane is to retain solids. The membrane pore size, consequently, is in the range of ultrafiltration and microfiltration, typically 0.04-0.45 μ m. The performance of a membrane is determined by either its flux at a given trans-membrane pressure (TMP), where a greater flux is desirable, or by the TMP at a specific flux, where a lower TMP is better. MBRs can be generally classified into two categories: submerged MBRs and external crossflow MBRs (Heiner and Bonner, 1999; Gander *et al.*, 2000). In submerged MBRs the membrane modules are directly placed in the aeration tank and the permeate exits the aeration tank either by vacuum pump or by gravity (Stephenson *et al.*, 2001). For crossflow MBRs, a recirculating pump is used to transfer the activated sludge from the aeration tank to the membrane module where the biomass is separated. Water permeates from inside to outside of the module and the concentrated biomass returns to the aeration tank. At present, there are more than 500 MBR plants in the world treating municipal and industrial wastewaters (Stephenson *et al.*, 2001). Submerged MBRs are the most common systems employed for the treatment of municipal

1 wastewaters (Manem and Sanderson, 1996; Adham *et al.*, 1999; Gander *et al.*, 2000; Stephenson
2 *et al.*, 2001). Crossflow MBRs are mainly used for small-scale wastewater treatment applications
3 (e.g. commercial buildings). Owing to the high-energy consumption rate of these systems larger
4 scale applications are unlikely.

5 Membrane fouling, particularly biofouling, represents a significant limitation to the wide
6 spread application of MBRs. Biofouling results in reduced performance, severe flux decline,
7 high energy consumption, and frequent membrane cleaning or replacement (Manem and
8 Sanderson, 1996; Nagaoka *et al.*, 1996; Nagaoka *et al.*, 1998; Tardieu *et al.*, 1996, 1998 and
9 1999; Wisniewski and Grasmick, 1997 and 1998; Gander *et al.*, 2000; Stephenson *et al.*, 2001;
10 Wisniewski, 2001). The following review examines the current understanding of the
11 mechanisms and factors associated with membrane biofouling. Strategies for controlling
12 membrane fouling are explored, and the requirements to better control biofouling in MBRs are
13 proposed and discussed.

14

15 **Mechanisms of Biofouling in a Membrane Separation Bioreactor**

16 Fouling is a generic term that is closely associated with deterioration in membrane
17 performance. In practice, where flux is held constant, fouling results in an increase in TMP.
18 Fouling can be attributed to an inorganic foulant (e.g. iron) that relates to the constituents present
19 in the wastewater entering the MBR. Because the MBR membrane is designed to allow soluble
20 constituents to permeate, inorganic fouling due to precipitation and concentration gradients, as is
21 common in reverse osmosis membrane, may not be a dominant mechanism. Biofouling is
22 specifically related to the interaction of biosolids with the membrane. This could consist of
23 formation of biofilms or the accumulation of bioorganic material, including extracellular

1 polymeric substances (EPS), on the membrane surface. Due to the nature of wastewater and the
2 contents of mixed liquor suspended solids, composite fouling (i.e. a combination of biofouling
3 and inorganic fouling) (Sheikholeslami, 1999) often occurs in MBRs. Biofouling, though, is a
4 significant phenomenon due to the high biomass concentration in the bioreactor.

5 Biofouling is the result of interactions between the membrane surface and those
6 components of the biomass or sludge consisting of bacterial cells, or aggregates, and EPS.
7 During permeation, the mixed liquor and soluble components in water are transported to the
8 membrane surfaces where a concentration polarization occurs due to the dewatering process near
9 the boundary layer of the membrane surface. Based on a few studies (Choo and Lee, 1996;
10 Tardieu *et al.*, 1996 and 1998; Roorda *et al.*, 2000; Wisniewski, 2001), biofouling can be
11 characterized on the basis of three fouling patterns (Figure 1): adsorption of EPS to the
12 membrane surface; pore clogging by cells; and sludge cake, or film, formation arising from
13 deposition of cells or aggregates. All three mechanisms are likely to be involved. However, the
14 relative importance of each mechanism is site specific and may depend on a number of factors.

15 The presence of soluble/suspended EPS leads to the accumulation of this material on
16 membrane surfaces and within the pore structure (Nagaoka *et al.*, 1996; Nagaoka *et al.*, 1998 and
17 2000; Chang and Lee, 1998; Mukai *et al.*, 1999). This may change the friction factor in the flow
18 channels and a decrease in the flow area, which leads to greater TMP. Both physical and
19 chemical adsorption of EPS may occur during the permeation process. Physical adsorption
20 involves weak interactions, which may be just a simple deposition, between EPS and membrane
21 surfaces, and thus EPS can be removed by the shear stress of fluid mechanics, like air scouring
22 and liquid flow on membrane surfaces. On the other hand, chemical adsorption, involving greater
23 adhesion strength and energy between EPS and membrane surfaces, is more resistant to shear

1 stress. Consequently, chemical agents, like acid, base and oxidants, must be used to remove
2 chemically adsorbed EPS. Irreversible biofouling by EPS may also occur depending on the nature
3 of interactions.

4 Pore clogging may occur due to cell debris and colloidal particles (Karr and Keinath,
5 1978; Urbain *et al.*, 1993) found in mixed liquor, whose sizes are similar to the membrane pore.
6 During permeation these fine particles can become entrapped in the pores and thus reduce the
7 surface area for filtration. The pore clogging decreases the flow area of the permeate resulting in
8 a greater TMP to achieve the same flux. A number of studies have found that pore clogging may
9 be the dominant biofouling mechanism. Backwashing and chemical cleaning are primarily the
10 ways to reduce membrane pore clogging (Bouhabila *et al.*, 1998 and 2001; Tardieu *et al.*, 1996,
11 1998 and 1999; Wisniewski and Grasmick, 1997 and 1998; Gan, 1999).

12 Adhesion or deposition of sludge cake on membrane surfaces may also contribute to
13 membrane biofouling (Kiat *et al.*, 1992; Ozaki and Yamamoto, 1996 and 1997; Pound *et al.*,
14 1997; Bouhabila *et al.*, 1998 and 2001; Tardieu *et al.*, 1996, 1998 and 1999; Wisniewski and
15 Grasmick, 1997 and 1998; Wisniewski, 2001). In the region close to the membrane surface, a
16 concentration gradient is expected as a result of dewatering effects. This leads to a higher MLSS
17 at the surface of the membrane than in the bulk phase. If the mass flow of mixed liquor to the
18 membrane surface is larger than the back transport of mixed liquor from the surface, sludge cake
19 formation or deposition is anticipated. The transport mechanisms describing the formation of
20 sludge cake on membrane surfaces includes shear-induced diffusion, inertial lift and surface
21 transport (Tardieu *et al.*, 1998). Backwashing plays a major role in controlling the fouling
22 process (Tardieu *et al.*, 1998). Air scouring in submerged MBRs is usually used to generate an
23 adequate shear stress for removing sludge cakes from membrane surfaces, while the shear stress

1 caused by the high flow velocity of mixed liquor in crossflow MBRs reduces sludge deposition at
2 surfaces.

4 **Factors Which Affect Membrane Biofouling**

5 A number of factors may affect membrane biofouling in MBRs, as shown in Figure 2.

6 These factors include the hydrodynamic conditions in MBRs, membrane module design (which
7 in turn affects the hydrodynamic conditions in MBRs), the operating conditions of the biological
8 process and membranes, membrane materials, and the physicochemical properties of the sludge.

9 The physicochemical properties of the sludge will be directly influenced by the operating
10 conditions of the treatment process (e.g. hydrodynamic conditions and solids retention time)
11 resulting in subsequent impacts on membrane fouling (Liao et al., 2001).

12 Hydrodynamic Conditions

13 Hydrodynamic conditions can impact on the performance of MBRs through shear stress
14 and air scouring. The flow velocity of mixed liquor in crossflow MBRs and the aeration
15 intensity and time in submerged MBRs are critical parameters that can influence these
16 phenomena and the performance of MBRs (Heiner and Bonner, 1999).

17 Ozaki and Yamamoto (1996 and 1997) investigated the impact of hydraulic conditions on
18 sludge accumulation in submerged MBRs and identified that the shear rate is the main hydraulic
19 factor that affects sludge accumulation. Shear stress caused by the cross flow velocity and the
20 recirculating pump is crucial in controlling biofouling in the crossflow MBRs, as the shear stress
21 induced by the recirculating pump affects not only the formation of sludge cake but also floc
22 structure (Tardieu *et al.*, 1996, 1998 and 1999; Wisniewski and Grasmick, 1997 and 1998). At a
23 low flow velocity of 0.5-2.0 m/sec, sludge cake layer can quickly form on membrane surfaces. A

1 high flow velocity greater than 4 m/sec appears to prevent sludge deposition on membrane
2 surfaces (Tardieu *et al.*, 1998). However, shear stress from the recirculating pump breaks up the
3 flocs, which negatively impacts membrane flux (Wisniewski and Grasmick, 1998). This is
4 further supported by a recent study from Korea (Kim *et al.*, 2001) that reported shear stress
5 caused by the recirculating pumps affects membrane biofouling. The vane-type rotary pump
6 imposed a much stronger shear than the turbine-type centrifugal pump and thus resulted in a
7 much smaller floc size. The net result of shearing sludge flocs is an increase in fouling potential.
8 Consequently, a low shear pump is recommended for the crossflow MBR process.

9 There have been extensive studies in Japan, Europe and North America on the impact of
10 aeration intensity on the performance of submerged MBRs. It is generally believed that shear
11 stress caused by the up flow of an air-water mixture on the membrane surfaces can be optimized
12 to reduce membrane biofouling in submerged MBRs. Pound *et al.* (1997) found that the
13 turbulence induced by the air bubbles from the aerators causes the membranes to collide and
14 produce a self-cleaning effect. Furthermore, Ueda *et al.* (1997) reported that the air flow rate
15 determines the sludge cake thickness in submerged MBRs. There appeared to be an optimum
16 value above which no further decrease in sludge cake thickness was observed. When the air flow
17 rate decreased from 700 to 350 L/min, the TMP increased from 0.2 to 0.5 bar in 20 hrs and an
18 increased sludge cake accumulation on membrane surfaces was observed. Cake removal
19 efficiency was improved either by increasing the air flow rate or by augmenting aeration intensity
20 by concentrating membrane modules over a smaller floor area. More recently, in France,
21 Bauhabila *et al.* (1998 and 2001) investigated the impact of air flow rate on the performance of
22 hollow fiber membranes. An increase in the air flow rate from 1.2 to 3.6 m³/m²/hr resulted in a
23 decrease in total filtration resistance and thus increased the membrane flux by a ratio of 3. These

1 results indicate that air scouring on the membrane surface is important in preventing sludge
2 accumulation. A recent survey on the application of MBRs for municipal wastewater treatment
3 (Adham *et al.*, 1999) indicates that the specific design of aerators, the location of the aerators and
4 the air flow patterns are crucial for reducing membrane biofouling due to sludge cake formation,
5 as these factors affect the hydrodynamic conditions in MBRs.

6 Module Configuration

7 Module design plays a major role in determining the performance of MBRs by affecting
8 the hydrodynamic conditions in the MBR, which in turn influences the fouling rate. The packing
9 density of hollow fibres and flat sheets, the choice of aerator and the specific location of aerator
10 under the membranes are critical design parameters for MBRs (Adham *et al.*, 1999). Kiat *et al.*
11 (1992) found that solids did not accumulate in the space between fibres when fibre packing
12 density was lower than a critical value. Maximum permeate flux and productivity were obtained
13 below and at the critical packing density, respectively. Significant solids accumulation and flux
14 decline were observed above the critical packing density. More recently, Ozaki and Yamamoto
15 (1996 and 1997) found that the module length and distance between membranes affect sludge
16 accumulation on membrane surfaces. Shorter distances between membranes resulted in sludge
17 accumulating in a shorter time. This was explained by the increase in the frequencies of collision
18 of flocs over a shorter distance between membranes.

19 Innovations in the modification and optimization of module design include the reduction
20 in leaf length, spacing adjustments, packing density, the development of new aerators, and even
21 the location for the permeate output and aerator. An important improvement in module design is
22 the utilization of gravity force to replace the vacuum pump to suck permeate (Kubota, Japan;
23 Millenniumpore Ltd, UK). This design significantly reduces energy consumption.

Membrane Flux and Operating Cycle Length

Membrane permeating flux determines the mass flow of mixed liquor to membrane surfaces and thus affects sludge cake formation. It has been generally assumed that fouling should be greatly reduced when the membranes are operated at a permeate flux below a critical level determined by the magnitude of the sum of back-transport mechanisms (Field *et al.*, 1995; Howell, 1995; Tardieu *et al.*, 1998; Fan *et al.*, 2000; Fane *et al.*, 2000; Gander *et al.*, 2000). However, the critical value of flux is very much system specific. The typical membrane operating flux in MBR plants for municipal wastewater treatment is about 15 gfd (Adham *et al.*, 1999).

Madaeni *et al.* (1999) found that the critical flux depends on sludge concentration and crossflow velocity. A higher critical flux is related to a higher crossflow velocity and lower sludge concentration. This is not surprising, as both a higher flow velocity and a lower sludge concentration reduce membrane biofouling. A higher critical flux is also related to a more hydrophilic membrane surface. More recently, Fan *et al.* (2000) found that membrane fouling is a function of the permeate flux and a higher permeate flux resulted in a shorter membrane cleaning interval.

The operating cycle length is important in determining the rate of membrane biofouling. Defrance and Jaffrin (1999) tested a hydrodynamic method for fouling control using intermittent filtration and backwashing. The cyclic operation consisted of alternating short periods of filtration (1-4 s) and short periods of backwashing (1-2 s) at low TMP and high velocity. The quick switch between filtration and backwashing yielded a 20% increase in permeate flux and a 10% energy reduction, compared to the conventional operating manner of MBRs with continuous filtration.

Membrane Materials and Properties

1 Membrane materials and properties have long been considered important factors that can
2 affect membrane fouling. The properties include the hydrophobicity, surface charge, pore size
3 and roughness. Presently, both tubular ceramic membrane and polymeric membranes have been
4 used for MBRs. However, there is no direct comparative study between those two membrane
5 materials. There is a particular lack of information regarding the role of polymeric membrane
6 materials and properties on membrane fouling in MBRs.

7 Chang and Lee (1999) reported for batch filtration tests a lower fouling rate that related to
8 a more hydrophobic membrane. It is possible to increase the hydrophilicity of a membrane via
9 surface modification. Sainbayar *et al.* (2001) were able to increase the hydrophilicity of a
10 membrane modified by UV grafting and this resulted in a decrease in the biofouling potential.
11 Gan (1999) reported that there was a significant difference in the start-up transient flux with
12 respect to pore size. A larger flux decline rate was related to a smaller pore size. However, the
13 difference in flux diminished progressively with permeation time and smaller pore size
14 membranes reached a stable condition earlier. This was probably due to the fact that more severe
15 in-pore fouling might occur in larger pores. Madaeni *et al.* (1999) found that the critical flux of
16 activated sludge is similar for different membrane pore sizes but is dependent on surface
17 hydrophobicity.

18 There is only limited information on the importance of membrane surface charge with
19 respect to biofouling. Shimizu *et al.* (1989) investigated the impact of aluminum membrane
20 surface charge on the filtration characteristics of methanogenic waste using three types of
21 charged aluminum membranes. It was found that the negatively charged membrane exhibited a
22 higher filtration flux than that of the non- and positively charged membranes. The impact of
23 membrane charge was mainly reflected by its importance in controlling electrostatic adsorption

1 and pore clogging. No impact of membrane surface charge on sludge cake formation was
2 observed.

3 Mixed Liquor Suspended Solids

4 The impact of mixed liquor suspended solids (MLSS) on sludge dewatering is well
5 known for conventional sludge dewatering processes. Analogous to this is the dewatering of
6 sludge that occurs on the surface of a membrane in MBRs. The influence of MLSS
7 concentrations on the performance or fouling of MBRs has been studied in both laboratory and
8 full-scale plants (Magara and Itoh, 1991; Sato and Ishii, 1991; Nagaoka *et al.*, 1996; Adham *et*
9 *al.*, 1999). A higher MLSS correlates to a higher TMP or a lower permeate flux due to severe
10 sludge cake formation on the membrane surfaces. This leads to the reduction of membrane
11 permeability. Typical MLSS concentrations in MBRs fall in the range of 10-30g/L with a median
12 value of 10g/L for submerged MBRs (Adham *et al.*, 1999).

13 Magara and Itoh (1991) found that a higher MLSS concentration is related to a lower
14 permeate flux. A 10% variation in MLSS resulted in a 5% variation in permeate flux. Sato and
15 Ishii (1991) developed an empirical model to correlate the membrane filtration resistance to the
16 MLSS, soluble COD, viscosity and TMP. MLSS was found to have the second largest effect on
17 total filtration resistance after the soluble COD concentration in the supernatant. More recently,
18 using a pilot-scale submerged MBR, Nagaoka *et al.* (1996) found that a decrease in MLSS was
19 related to a decrease in the filtration resistance (a decrease in TMP). This is because the increase
20 in MLSS caused an increase in viscosity.

21 It has been reported that MLSS ranging from 3.5 to 7.0 g/L had little impact on
22 membrane fouling (Fan *et al.*, 2000), implying that a higher MLSS does not cause more serious
23 membrane fouling. It is more likely, however, that this range was below the critical MLSS levels

1 that may lead to sludge cake formation. At these MLSS levels back transport mass of mixed
2 liquor is always larger than the mass flow of mixed liquor to the membrane surfaces. Above the
3 critical concentrations the net accumulation of sludge cake on membrane surfaces is proportional
4 to the MLSS concentration.

5 Process and Environmental Conditions of Activated sludge (SRT, F/M, Nutrient Conditions)

6 Process and environmental conditions of the activated sludge system are known to play
7 an important role in controlling sludge properties (Chao and Keinath, 1979; Andreadakis, 1993;
8 Liao *et al.*, 2001). These conditions include solids retention time (SRT), hydraulic retention time
9 (HRT), F(food)/M(microorganism) ratio and nutrient conditions. Consequently, an influence of
10 process and environmental conditions on the performance of MBRs is anticipated, as they affect
11 the nature of the foulants.

12 Recently, a number of studies have been conducted to elucidate the role of SRT and F/M
13 in controlling membrane biofouling (Chang and Lee, 1998; Fan *et al.*, 2000; Nagaoka *et al.*,
14 2000; Bouhabila *et al.*, 2001). Using membranes with different hydrophobicity and pore sizes,
15 Chang and Lee (1998) found that an increase in SRT from 3 and 8 days to 33 days resulted in a
16 significant increase in the sustainable membrane flux. This could be attributed to a lower level of
17 EPS content at higher SRTs. A similar conclusion was reached in the study of Fan *et al.* (2000)
18 at three different SRTs (20, 10 and 5 days). It was reported that there was no need for membrane
19 cleaning at an SRT of 20 days for more than 70 days of operation; on the other hand, a cleaning
20 interval of 3-5 days was required for an SRT of 5 days (Fan *et al.*, 2000). Nagaoka *et al.* (2000)
21 found that the membrane fouling rate at a high F/M rate (1.5 g/L/day) was much higher than that
22 at a low F/M rate (0.3g/L/day), due to the higher extracellular polymeric substances (EPS)
23 content at the higher F/M loading.

1 There are also some studies concerning the impact of environmental and biological
2 conditions on membrane fouling. Chang and Lee (1998) found that a foaming sludge had a much
3 higher filtration resistance than a non-foaming sludge, mainly due to the cake resistance caused
4 by foaming sludge. Nitrogen-deficient sludge produced less EPS and thus had a sustainable
5 membrane flux 30-40% higher than the control sludge produced under non-nutrient-limiting
6 conditions.

7 Because the foulant or biofouling originates from diverse components of activated sludge
8 broth, it is not surprising to see the importance of sludge properties in controlling biofouling. The
9 content and rate of membrane biofouling depend on the type and strength of interactions between
10 membrane surfaces and each component of activated sludge broth. Until now, studies have been
11 conducted on the floc size, EPS and viscosity, but there is little information about the role of
12 sludge surface charge and hydrophobicity in controlling biofouling.

13 Particles and Colloids

14 Floc size has long been considered as a dominant factor in controlling sludge dewatering
15 (Karr and Keinath, 1978). The same is true for MBRs. It is generally believed that floc size in
16 MBRs is smaller than that in the conventional activated sludge (Zhang *et al.*, 1996; Heiner and
17 Bonner, 1999). Recently, the influence of different fractions of the sludge on the membrane
18 fouling has been studied (Chang and Lee, 1998; Tardieu *et al.*, 1998 and 1999; Wisniewski and
19 Grasmick, 1998). All studies concluded that the soluble COD fraction of activated sludge broth
20 plays a significant role in determining membrane fouling, followed by the influence of colloidal
21 components. The larger particles were less important in determining the filtration resistance of
22 the membrane. This is probably not surprising, as the back-transport velocity of smaller
23 components, like colloids and solutes, is lower than that for the larger particles. More recently,

1 Bouhabila *et al.* (2001) investigated the impact of the three fractions (suspended solids, colloids
2 and solutes) of sludge on membrane fouling at different SRTs (10, 20 and 30 days). At all SRTs,
3 it was clear that colloids and solutes were the dominant fractions controlling filtration resistance.

4 Extracellular Polymeric Substances

5 The importance of EPS in controlling membrane fouling has been extensively studied. A
6 larger EPS content is related to a higher fouling rate. Sato and Ishii (1991) found that
7 soluble/suspended EPS had the largest effect on the filtration resistance and was even more
8 important than the MLSS. The presence of a large quantity of extracellular matrix on bacterial
9 surfaces increased the cake resistance of bacteria (Hodgson *et al.*, 1993). These results are
10 supported by two more recent studies (Nagaoka *et al.*, 1996; Chang and Lee, 1998; Mukai *et al.*,
11 1999). The accumulation of EPS in the aeration tank and on membrane surfaces resulted in an
12 increase in viscosity of the mixed liquor and an increase in the filtration resistance of the
13 membrane (Nagaoka *et al.*, 1996). In studying the impact of physiological and nutritional
14 conditions on membrane biofouling, Chang and Lee (1998) found a strong correlation between
15 filtration resistance and EPS content. Furthermore, Mukai *et al.* (1999) found that the retained
16 ratio of proteins to sugar polymers during filtration may be one of the factors affecting the
17 magnitude of time-dependent relative permeate flux decline. This suggests that, for permeate flux
18 decline, there are different degrees of interactive effects of proteins and sugar polymers, whose
19 quantity and quality in turn depend on microorganism species and their growth phases.

20 Nagaoka *et al.* (1998 and 2000) have developed a mathematical model to simulate
21 temporal changes in suction pressure, flux and filtration resistance, based on the mechanisms of
22 accumulation, detachment and consolidation of EPS on membrane surfaces. The measured
23 patterns of flux, TMP and filtration resistance were well predicted by the model. The impact of

1 process conditions, like organic loading, flux and shear stress, on the performance can be
2 evaluated through the model.

3 Viscosity

4 Viscosity of the mixed liquor has a significant impact on the performance of MBRs
5 (Magara and Itoh, 1991; Sato and Ishii, 1991; Nagaoka *et al.*, 1998). Both Magara and Itoh
6 (1991) and Sato and Ishii (1991) found that a higher viscosity was related to a lower permeate
7 flux. Viscosity is the third most important factor, after soluble COD and MLSS, in determining
8 the permeate flux (Sato *et al.*, 1991). More recently, Nagaoka *et al.* (1996) found that there was a
9 strong correlation between the filtration resistance of sludge and the viscosity. A higher TMP is
10 related to a more viscous sludge. This is probably not surprising, as the more viscous sludge has
11 a larger potential to cause membrane fouling.

13 **Characterization of Foulants in MBRs**

14 Knowledge of foulants is essential in selecting the most economic and effective cleaning
15 method. The type and concentration of cleaning agents depend on the type of membrane and the
16 nature of the foulants. The most effective approach to foulant identification is by extensive
17 analysis of the foulants through a destructive autopsy (Fane *et al.*, 2000). At present, an
18 increasing number of techniques is available for autopsy to characterize the nature and location
19 of microbial structures (Liss, 2002). These techniques include multi-scale microscopic
20 observation, by conventional light/optical microscopy (COM), scanning confocal laser
21 microscopy (SCLM), two photon laser scanning microscopy (2P-LSM), atomic force microscopy
22 (AFM) and Raman confocal microspectroscopy, conventional scanning electron microscopy
23 (SEM), environmental scanning electron microscopy (ESEM) and transmission electron

1 microscopy (TEM). They also include element analyses by a microprobe (energy-dispersive
2 spectroscopy, or EDS) coupled to a scanning/transmission electron microscope (STEM-EDS),
3 ion chromatographic (IC) analyses and GC/MS/Pyrolysis of foulants.

4 Most of the studies were conducted by the membrane-manufacturing industry to develop
5 better cleaning methods for specific MBR plants. Only limited information is available in the
6 published literature.

7 Correlative microscopy (CM) is a strategy of using multiple microscopic techniques
8 (Leppard, 1992) such as COM, SCLM, TEM, and ESEM (Liss et al., 1996; Liss, 2002) which
9 facilitates the detection, assessment and minimization of artifacts that might arise from using one
10 technique only. Collectively, these techniques enable the microscopist to potentially visualize
11 microbial floc or films and identify physicochemical parameters, three-dimensional arrangements
12 of constituents, topography, chemical composition and forces governing interfacial phenomena.
13 The use of only one microscopic technique can bias or limit the information acquired because of
14 the artifacts that arise in specific sample preparations and the resolution constraint associated
15 with a particular technique.

16 CM has been successfully used by Liss *et al.* (1996) with a minimal perturbation
17 approach in studying natural and engineered flocs. A recent minimal perturbation approach
18 (Droppo *et al.*, 1996a, b) involves the use of sample stabilization in low melting point agarose
19 and a four fold multi-preparatory technique. This was found to maintain the structural integrity
20 of the samples through the stabilization, staining and washing procedures. Nanoplast resin is
21 particularly effective as a stabilization medium because it is a hydrophilic embedding resin that
22 holds the fibrillar EPS (Liss et al., 1996).

23 SCLM is one of the most recent microscopic techniques used to study activated sludge

1 flocs (Wagner *et al.*, 1994), and has been shown to be a useful technique in bridging the
2 resolution gap between COM and TEM (Liss *et al.*, 1996). 2P-LSM permits examination of floc
3 and films approaching 1 mm in thickness while minimizing photobleaching and phototoxicity
4 (Cowan and Holloway, 1996; Holloway and Cowen, 1997; Decho and Kawaguchi, 1999).

5 A binocular microscope has been used to observe the nature of deposits on ceramic
6 surfaces (Tardieu *et al.*, 1998). The thickness of a deposit was from 0-100 μm , which coincides
7 with the observation of Shimizu *et al.* (1996). GC/MS/Pyrolysis analyses showed that the
8 composition of the deposited foulant varied according to the biological conditions and was
9 similar to that of the supernatant of the biological suspension after centrifugation. The deposited
10 foulants contained a substantial quantity of mineral elements (Tardieu, 1995). In another study,
11 Yoon *et al.* (1999) used a SEM and IC to characterize the nature of a crystalline precipitate on
12 ceramic membranes for anaerobic treatment. A significant amount of inorganic ions, including
13 Mg^{2+} , Fe^{3+} and PO_4^{3-} , were detected in the fouling layer, implying inorganic fouling was
14 important under anaerobic conditions. More recently, Bouhabila *et al.* (2001) observed the
15 existence of a deposit layer of foulant using microscopes. The deposit was a thin layer of biofilm
16 equivalent to one or two layers of filamentous bacteria. These results suggest that both organic
17 fouling and inorganic fouling could be important, depending on the wastewater characteristics
18 and the nature of activated sludge.

19 TEM, applied to ultrathin sections of embedded samples, has been used rarely but
20 effectively since the 1980s to examine nanoscale associations between microbes, organic
21 colloidal materials, colloidal minerals and membrane filters (Buffle *et al.*, 1992). Recently, there
22 has been an attempt to correlate results from multi-scale multi-method observations employing
23 COM, ESEM, TEM and STEM-EDS (Leppard *et al.*, 1999) applied to fouled membranes. In this

1 latter study, the ESEM (facilitated by large volume/low resolution observations from COM) was
2 employed to bridge the scalar and resolution gaps between moderate resolution techniques (near
3 micrometer range) and the high resolution of TEM (near nanometer range). The ESEM made two
4 major contributions to the analysis: (1) it permitted structural descriptions of hydrated fouled
5 membranes and their foulants in an essentially unperturbed state; (2) it oriented spatially the
6 subsequent TEM and STEM-EDS examinations well enough to overcome the cost-prohibitive
7 barrier of having to use TEM-based techniques alone, which would mean examining serial
8 ultrathin sections (0.07 μm thick) taken across millimeter sized samples.

9 Progress with this approach has been slow because engineers and materials scientists are
10 apprehensive about using biological preparatory techniques to prepare complex samples for
11 ultrathin section analysis. Such techniques are both time consuming and expensive, while
12 requiring consummate skill on the part of technicians and a fine tuning of the techniques to adapt
13 to the specific characteristics of a given sample. Despite the apprehension, however, a multi-
14 method TEM preparatory protocol has come into general use (Liss *et al.*, 1996); for fouled
15 membranes, it can provide structural information which relates well to the unperturbed state,
16 allows description of individual colloids and colloid associations with surfaces, and provides
17 element analysis on a "per colloid" basis. Additionally, the descriptions of colloids can often be
18 converted to valuable characterizations using literature standards based on shape, size, native
19 electron-opacity, electron-staining characteristics, internal morphology and microprobe element
20 analysis. A guide to the scattered literature on such characterizations can be found in Leppard
21 (1992), in a document endorsed by the International Union of Pure and Applied Chemistry.

22 Figure 3a shows a view by TEM (0.003 μm practical resolution) of an ultrathin section
23 (0.07 μm thickness) taken through a biofilm on the upper surface of a fouled membrane filter

1 (bottom of view). The bacterial cells and their fibrillar EPS are clearly identifiable, and several
2 kinds of colloidal debris are evident. Figure 3b shows a glancing section taken through the
3 uppermost layer of a fouled membrane filter. This view reveals the intimate association between
4 individual nanoscale fibrils (0.003-0.01 μm diameter) and inner membrane surfaces. Figure 4
5 presents a microprobe spectrum (STEM-EDS) taken from an individual selected colloid
6 positioned at the biofilm/filter interface.

7 SCLM is one of the most useful tools to investigate biofilms in situ (Lawrence and Neu,
8 1999). A particularly useful feature of SCLM and 2P-LSM is that these can be used in
9 combination with a variety of fluorescent molecular probes to study the spatial distribution of
10 extracellular polysaccharides, cell viability, pH gradient, proteins, RNA, lipids, and other
11 components of floc nondestructively. Measurements of the dimensions of colloidal matrix
12 material and their three-dimensional disposition are realistic. Figure 5 illustrates a SCLM image
13 showing the association of a microcolony with the surface of a membrane in a pilot scale
14 submerged system fed municipal wastewater. The image was derived from observations of thin
15 optical sections of a sample subjected to lectin-binding analysis (soybean agglutinin, wheat germ
16 agglutinin and concanavalin A). The combination of the three lectins reveals the distribution
17 glycoconjugates on the cell surface and within the film/aggregate matrix including the deposition
18 of EPS directly on the membrane surface.

19 While examination of fouled membranes by microscopes is highly informative, individual
20 autopsy techniques generally require a perturbing preparation of samples and invasion of
21 membranes. Therefore, simple and non-invasive characterization methods are desirable. For this
22 purpose, Fane and co-workers (Li *et al.*, 1998; Chang and Fane, 2000) developed two non-
23 invasive methods in recent years. One is a direct observation through the membrane (DOTM)

1 and the other involves the direct observation of the surface of the membrane (DOSM). Both
2 methods confirmed the existence of a critical flux and directly observed the deposition of
3 particles on membrane surfaces. These two methods provide an excellent way to monitor the
4 sludge cake formation on membrane surfaces but not for element analyses nor for important
5 nanoscale events.

6 7 **Strategies for Controlling Membrane Biofouling and Membrane Cleaning**

8 The strategies for biofouling control and cleaning involve both physical and chemical
9 methods. Based on the purpose of cleaning, these methods can be classified into two categories:
10 1) methods for preventing biofouling; and 2) methods for membrane cleaning once fouling
11 occurs.

12 **Optimal Hydrodynamic Conditions and Membrane Module Design**

13 Optimization of the hydrodynamic conditions is essential for preventing membrane
14 biofouling or at least reducing the rate at which biofouling occurs. In a MBR plant, optimization
15 of the hydrodynamic conditions can be achieved by controlling aeration intensity and time in
16 submerged MBRs and flow velocity of mixed liquor in crossflow MBRs, and also by appropriate
17 design of membrane modules. Pilot testing is required to find the optimal hydraulic conditions.

18 Because the design of membrane modules and spacers affects the hydrodynamics of flow
19 and contact time, the optimal packing density of fibres and flat sheets, the geometry and location
20 of aerators, and the orientation of fibres, have been active research topics in both industry and
21 academia. It is believed that an optimal performance (minimized rate of biofouling) can be
22 reached by design modification.

23 **Backwashing**

1 Once the membrane surface is fouled by pore clogging and sludge cake formation,
2 backwashing is an effective way to recover, at least partially, the membrane performance
3 (Srijaroonrat et al., 1999; Laitinen et al., 2001). In MBRs, backwashing with permeate will push
4 the colloidal particles and cell debris from the pore structure into the mixed liquor and partially
5 remove sludge cake from membrane surfaces. Laboratory studies indicate that backwashing
6 dramatically improves the performance of membrane modules (Visvanathan et al., 1997;
7 Defrance and Jaffrin, 1999; Gan, 1999; Chang and Fane, 2000; Bouhabila et al., 2001). The
8 frequency and flux of backwashing are related to the operating and environmental conditions in
9 the MBRs.

10 While both Gan (1999) and Bouhabila *et al.* (2001) found that periodic backwashing
11 decreased internal fouling (pore clogging), the total dissolved solids (TDS) in the permeate
12 increased with the application of a backwashing technique (Gan, 1999). In another study, Chang
13 and Fane (2000) found that injecting air into hollow fibres and tubular MBRs is effective in
14 controlling flux decline caused by concentration polarization and particle deposition. This was
15 attributed to the strong turbulence caused by the two-phase flow inside the fibres and MBRs.

16 Chemical Cleaning

17 Chemical agents have been extensively used for both maintenance and recovery cleaning
18 of membranes in MBRs. Maintenance cleaning helps to maintain flux and reduce the frequency
19 of recovery cleaning. The inherent nature of membrane biofouling, however, makes recovery
20 cleaning to destroy the foulants that have accumulated on membrane surfaces and within the pore
21 structure unavoidable. There are no generic chemicals that are effective for recovering the
22 membrane flux in every situation, as the nature of foulants changes at different locations. The
23 chemical agents for membrane cleaning can generally be classified as acids, bases, oxidants and

1 surfactants. A chlorine solution has been widely used as an oxidant for membrane cleaning
2 (Gander et al., 2000). Immersion of the modules in an acid bath for membrane cleaning was also
3 reported (Pound et al., 1997).

4 A knowledge of the nature of foulants will assist in the selection of chemical agents and
5 the sequence for using more than one chemical agent. A destructive autopsy can be used to get
6 extensive information regarding the nature and locations of foulants (Fane *et al.*, 2000).

7 Optimal Design of Activated Sludge Process

8 Because the use of MBRs in the wastewater treatment industry is relatively recent, there
9 are opportunities to optimize performance with respect to biofouling through better process
10 design. Pretreatment, already widely used for conventional activated sludge systems, remains
11 important for MBRs. The presence of hair, rags and sharp materials entering the biological
12 process will significantly reduce membrane performance. A fine screen or a high degree of
13 primary treatment may be required for MBRs (Heiner and Bonner, 1999). Membrane fouling
14 from these materials is not biofouling, but must nevertheless be prevented.

15 Currently, polymers are widely used in conventional activated sludge facilities for sludge
16 thickening and dewatering. The MBR may remove the requirement for downstream thickening,
17 but solids dewatering will still be required. The polymers used for dewatering should be
18 examined not only for their solids capture efficiency in the dewatering process but also for
19 compatibility with the MBR membranes. The presence of dewatering polymers in the liquid
20 return from dewatering may accentuate biofouling in the MBR.

21 The key MBR design factor is the membrane flux. The operating permeate flux should be
22 lower than the critical flux, which is site specific and is a function of sludge concentration and
23 hydrodynamic conditions (Madaeni *et al.*, 1999). A number of biological process design factors,

1 such as MLSS and SRT, impact biofouling and thus flux. The sludge physico-chemical
2 properties that affect membrane biofouling (e.g. floc size, EPS, surface charge and
3 hydrophobicity) may be manipulated by process design (Higgins and Novak, 1997; Liao et al.,
4 2001). For example, a relatively higher SRT and HRT may reduce the soluble COD in mixed
5 liquor, thus reducing biofouling due to the presence of organic constituents (adsorption and pore
6 clogging). The addition of inorganic salts may enhance flocculation, thereby reducing
7 supernatant turbidity (Higgins and Novak, 1997) and thus reducing sludge cake formation. Other
8 opportunities to decrease the rate of biofouling by changing the biological design parameters
9 should also be examined.

11 **Summary and Recommendations**

12 Membrane biofouling is a major factor impacting the efficient and economic operation of
13 MBRs. Three mechanisms: adsorption of EPS and other bioorganics, pore clogging, and sludge
14 cake formation on membrane surfaces, are involved in membrane biofouling. Module design,
15 hydrodynamic conditions, operating conditions of membrane modules, membrane materials,
16 environmental and process conditions of a given activated sludge system and the nature of the
17 activated sludge mixed liquor are the dominant factors affecting membrane biofouling. Several
18 key physical issues for which optimal designs must be determined are:

- 19 1. The packing density of fibres and the space distance of the membrane modules.
- 20 2. The aeration intensity in submerged MBRs.
- 21 3. The mixed liquor recirculating flow velocity in crossflow MBRs, and
- 22 4. The shear imposed by the recirculating pump in crossflow MBRs.

1 Mixed liquor properties affect membrane biofouling. Four key properties are floc size,
2 EPS quantity and quality, and the viscosity of mixed liquor. A larger floc size, a lesser EPS
3 content and a lower viscosity are related to less membrane biofouling. Perhaps surprisingly, the
4 colloidal particles and solutes (including soluble/suspended EPS) in the mixed liquor are more
5 important than the suspended solids in controlling membrane biofouling.

6 Additional research is required to further understand and ultimately better control
7 membrane biofouling in MBRs. The specific phenomena causing membrane biofouling must be
8 further examined. An important starting place is to develop more knowledge about the chemical
9 and structural nature of the foulants. Characterization of foulants using destructive autopsy must
10 continue. Methodology used in understanding the nature of the foulants in membrane water
11 treatment (Wiesner and Chellam, 1999) may also be applicable to the MBR processes. The TEM
12 protocols of Liss *et al.* (1996) coupled to the TEM characterization approach of Leppard (1992)
13 should also contribute to a future understanding of the nature of foulants. Better characterization
14 of foulants will allow the development of better cleaning strategies (for recovering flux after
15 biofouling occurs), better membrane material selection (to prevent or slow biofouling) and better
16 opportunities to control the biological process to minimize foulant production (to prevent or slow
17 biofouling).

18 Further information about the role of surface charge and hydrophobicity of both the
19 mixed liquor and the membrane materials should be developed. Additionally, the relationships
20 between the mixed liquor properties known to impact biofouling, e.g. EPS content, and the
21 underlying operating and design variables such as SRT must be further examined. By combining
22 this information with increased knowledge of the foulants and the fouling mechanisms, improved
23 design and operating strategies can be implemented.

1 Mathematical models which predict and simulate membrane biofouling should be
2 developed. Although models have limitations, biofouling models will allow examination and
3 prediction of the relationships between key biofouling factors and design, operating and
4 membrane parameters. Model calibration and verification must be accomplished by conducting
5 comprehensive studies on the impact of specific factors on membrane biofouling in MBRs to
6 determine the relative importance of each factor for given conditions.

7 MBRs are an increasingly important technology for biological wastewater treatment.
8 Further understanding of the mechanisms of biofouling and further development of strategies to
9 prevent biofouling as well as recover from biofouling are required to allow full integration of this
10 technology.

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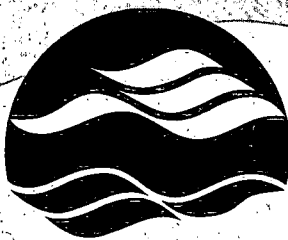
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